

Electronic Supplementary Information

**AIE-based Donor-Acceptor-Donor Fluorenone
Compound as multi-functional luminescence
materials**

Xianchao Du^{a,b,d‡}, Huifang Su^{c‡}, Li Zhao^{b‡}, Xiaojing Xing^a, Bingnan Wang^d, Dongfang Qiu^{*a}, Jinyi Wang^{*b} and Mao-Sen Yuan^{*b}

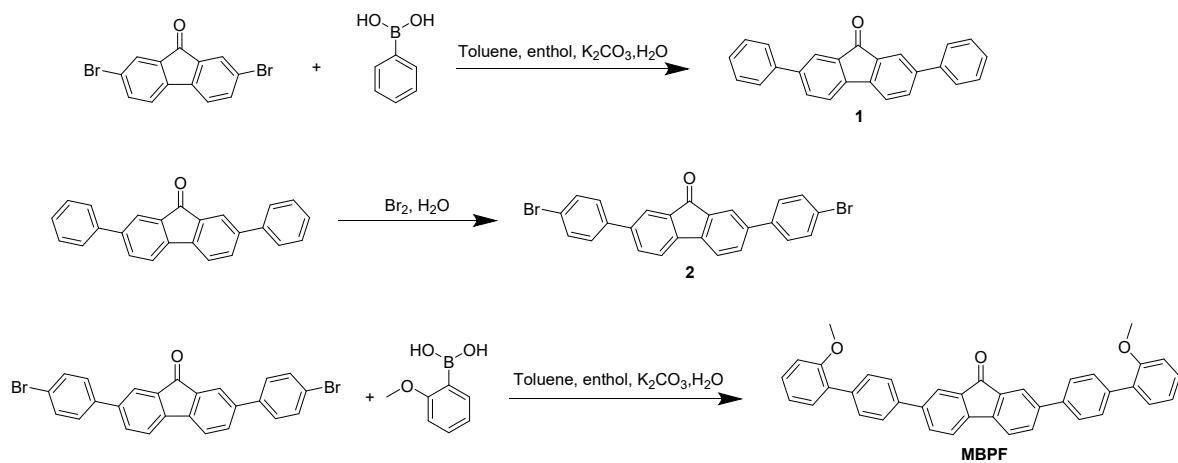
^a College of Chemistry and Pharmaceutical Engineering, Nanyang Normal University, Nanyang, Henan 473061, P. R. China

^b College of Chemistry & Pharmacy, Northwest A&F University, Yangling, Shanxi 712100, P. R. China

^c Department of Orthopaedic Surgery, The First Affiliated Hospital of Zhengzhou University, Zhengzhou, Henan 450052, P. R. China.

^d State Key Laboratory of Luminescent Materials and Devices, Guangdong Provincial Key Laboratory of Luminescence from Molecular Aggregates, Center for Aggregation-Induced Emission, South China University of Technology, Guangzhou, Guangdong 510640, P. R. China.

Synthesis of MBPF



Scheme S1. Synthetic routes to MBPF.

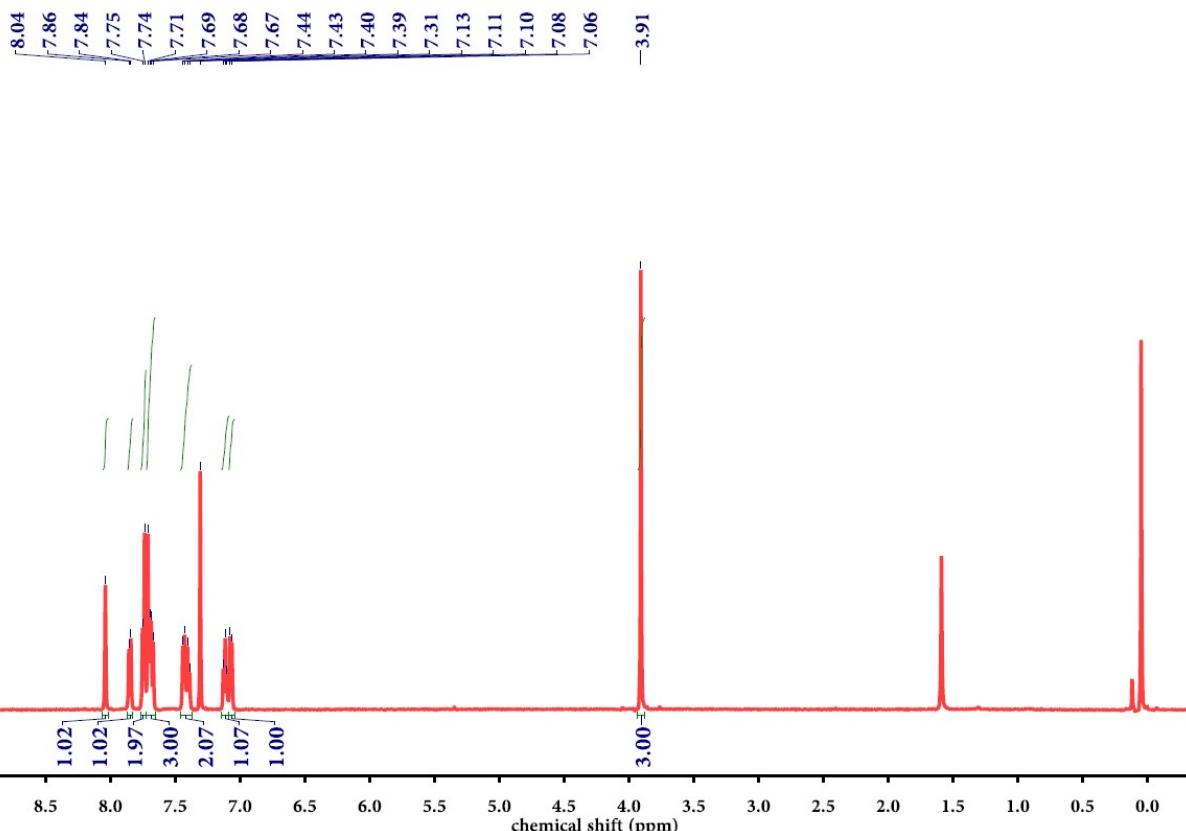


Fig. S1 1H NMR spectrum (500 MHz, $CDCl_3$) of compound MBPF.

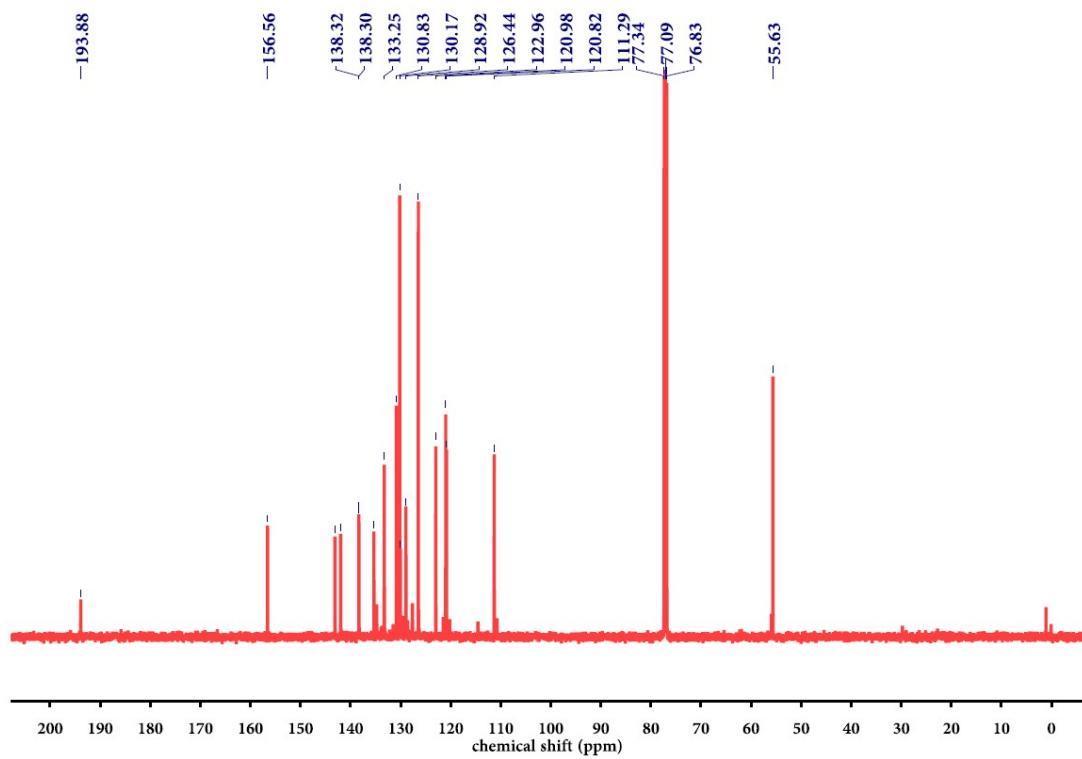


Fig. S2 ^{13}C NMR spectrum (126 MHz, CDCl_3) of compound **MBPF**.

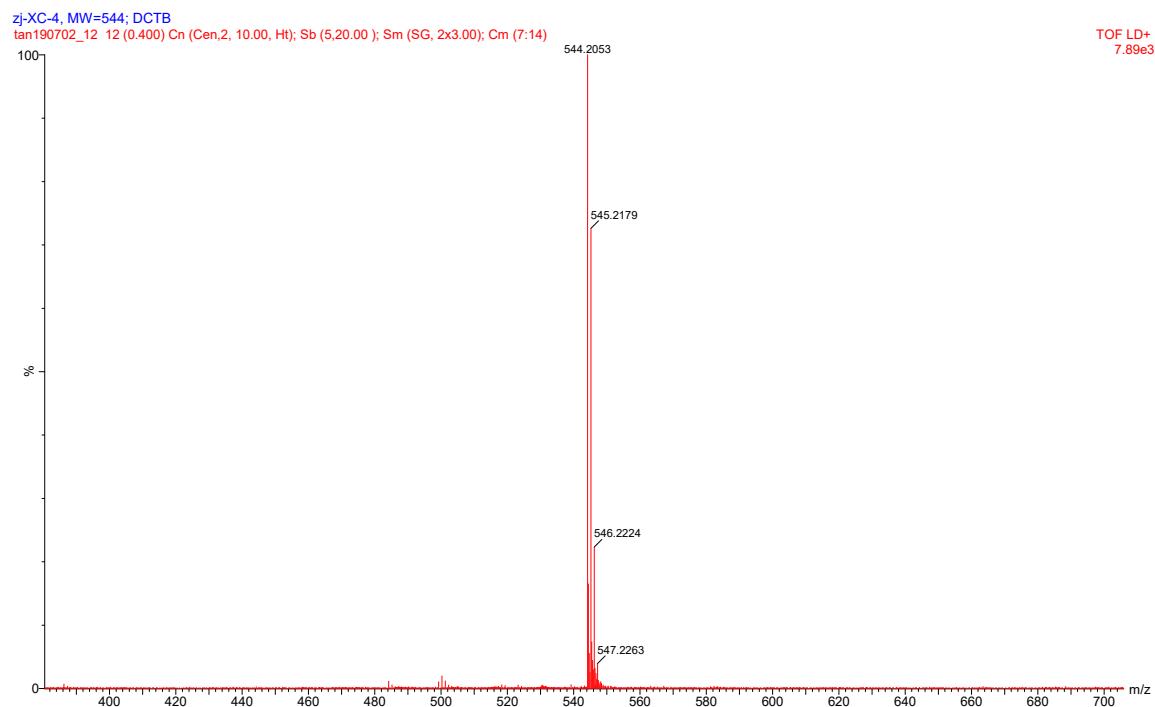


Fig. S3 HRMS spectrum of **MBPF**.

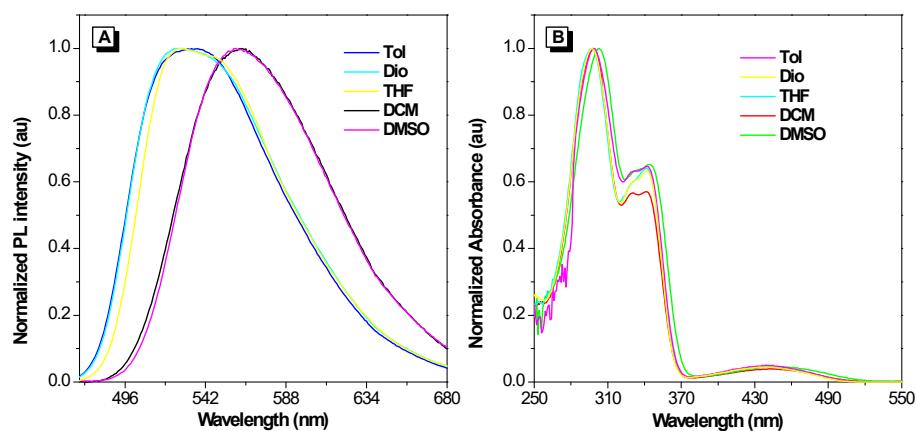


Fig. S4 Normalized PL spectra (A) and absorbance spectra (B) of **MBPF** in different solvents. Tol, toluene; Dio, dioxane; THF, tetrahydrofuran; DCM, dichloromethane; DMSO, dimethylsulfoxide.

Table S1. Photophysical properties of **MBPF** in various solvents

Solvent ^a	$\lambda_{\text{abs}}^{\text{b}}$	$\lambda_{\text{em}}^{\text{c}}$	$\Phi(\%)^{\text{d}}$	τ^{e}
Tol	297	530	0.25	6.19
Dio	297	527	0.15	9.26
THF	298	529	0.16	9.86
DCM	299	559	0.07	2.91
DMSO	303	563	0.07	3.01

^a Abbreviations: Tol, Toluene; Dio, Dioxane; THF, Tetrahydrofuran; DCM, Dichloromethane; DMSO, Dimethyl sulfoxide; ^b λ_{abs} , absorption maximum, ^c λ_{ex} , emission maximum, ^d Fluorescence quantum yield, ^e Fluorescence lifetime.

Table S2. Spectroscopic data for compound **MBPF**

	Solution in THF ^a				Crystalline powder		
	$\lambda_{\text{abs}}(\text{nm})$	$\lambda_{\text{em}}(\text{nm})$	Φ^c	$\tau^d(\text{ns})$	$\lambda_{\text{em}}(\text{nm})$	Φ	$\tau(\text{ns})$
Y-MBPF^b					532	0.43	11.41
O-MBPF	298, 340	530	0.16	9.86	558	0.35	9.05
R-MBPF					590	0.36	9.55

^a With $c = 1.0 \times 10^{-5} \text{ mol L}^{-1}$. ^b The amorphous, red crystal and orange crystal of compound **MBPF** are named **Y-MBPF**, **O-MBPF** and **R-MBPF**, respectively. ^c Fluorescence quantum yield. ^d Fluorescence lifetime.

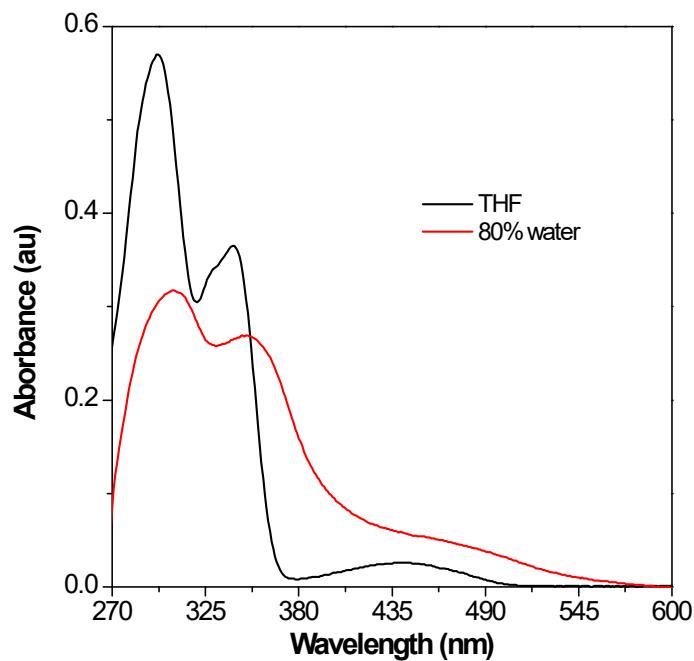


Fig. S5 The absorbance spectra of **MBPF** in pure THF and 80% water/THF.

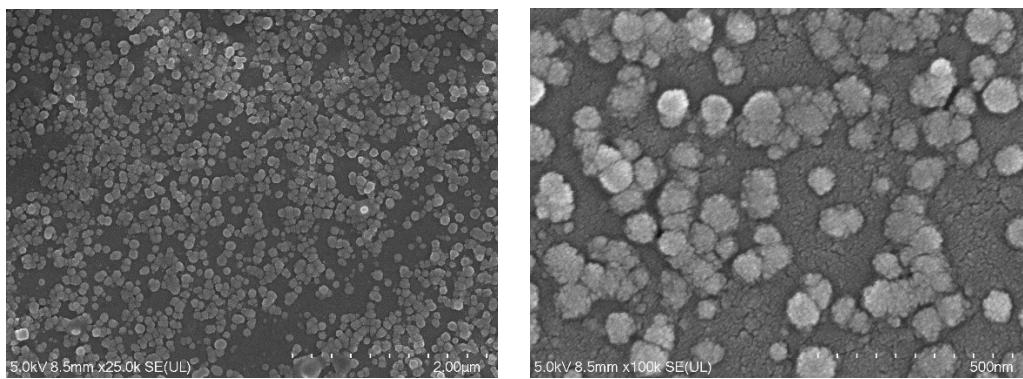


Fig. S6 The SEM of **MBPF** in water/THF (90 vol% water).

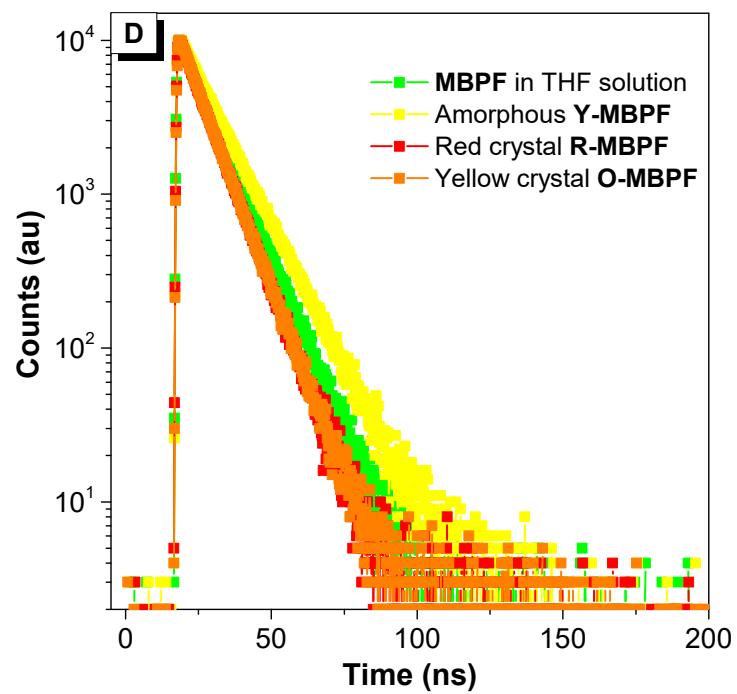


Fig. S7 The fluorescence decay profiles of **MBPF** in THF, Orange crystal **O-MBPF**, Red crystal **R-MBPF** and amorphous **Y-MBPF**.

Table S3. Selected crystallographic data for O-MBPF and R-MBPF.

crystals	O-MBPF	R-MBPF
formula	C ₃₉ H ₂₈ O ₃	C ₃₉ H ₂₈ O ₃
fw[g·mol ⁻¹]	544.61	544.61
crystal color	Orange	Red
crystal system	Monoclinic	Triclinic
space group	<i>P2₁/c</i>	<i>P-1</i>
<i>a</i> [Å]	10.028(11)	10.0977(9)
<i>b</i> [Å]	17.169(19)	13.5881(12)
<i>c</i> [Å]	32.96(4)	22.1586(19)
β [°]	94.169(19)	78.4920(10)
<i>V</i> [Å ³]	5660(11)	2853.9(4)
<i>Z</i>	8	4
ρ_{calcd} [g/cm ³]	1.278	1.268
μ [mm ⁻¹]	0.080	0.079
<i>T</i> [K]	298(2)	298(2)
θ_{\min} - θ_{\max} [°]	2.20-25.02	2.29 –25.02
<i>R/wR</i> [<i>I</i> > 2 σ ₍₁₎]	0.1058/0.1393	0.0507/0.0607

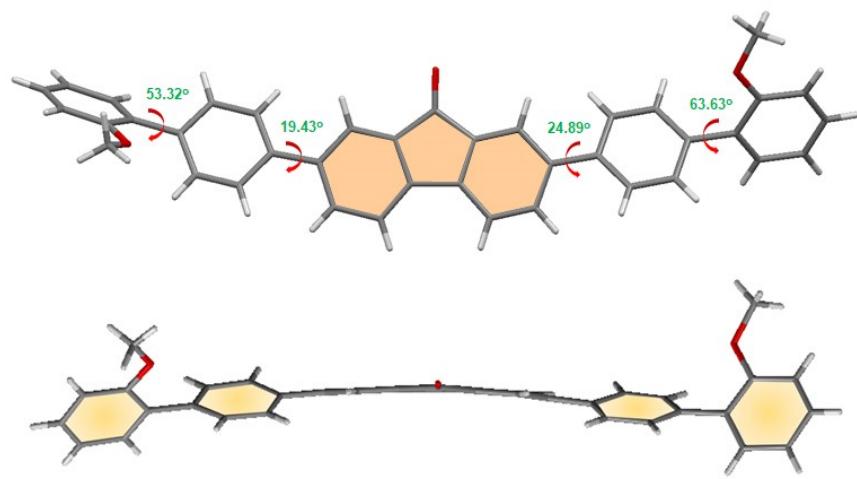


Fig. S8 The single crystal structure of **O-MBPF**.

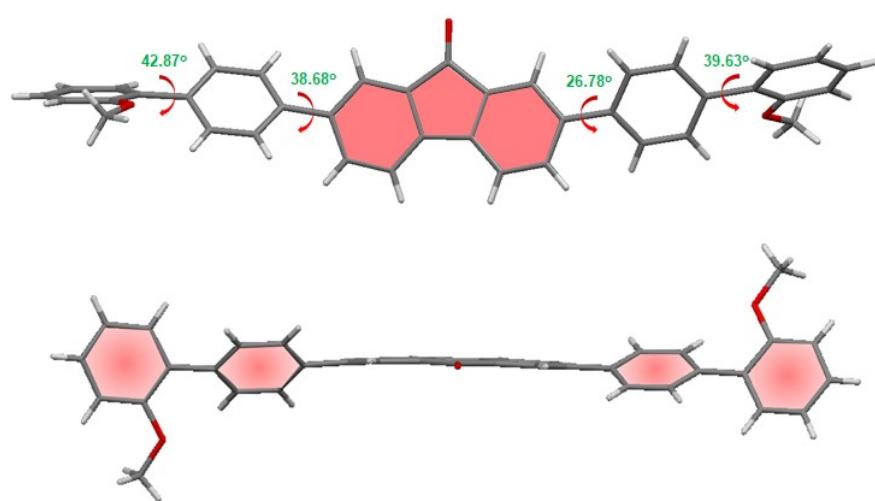


Fig. S9 The single crystal structure of **R-MBPF**.

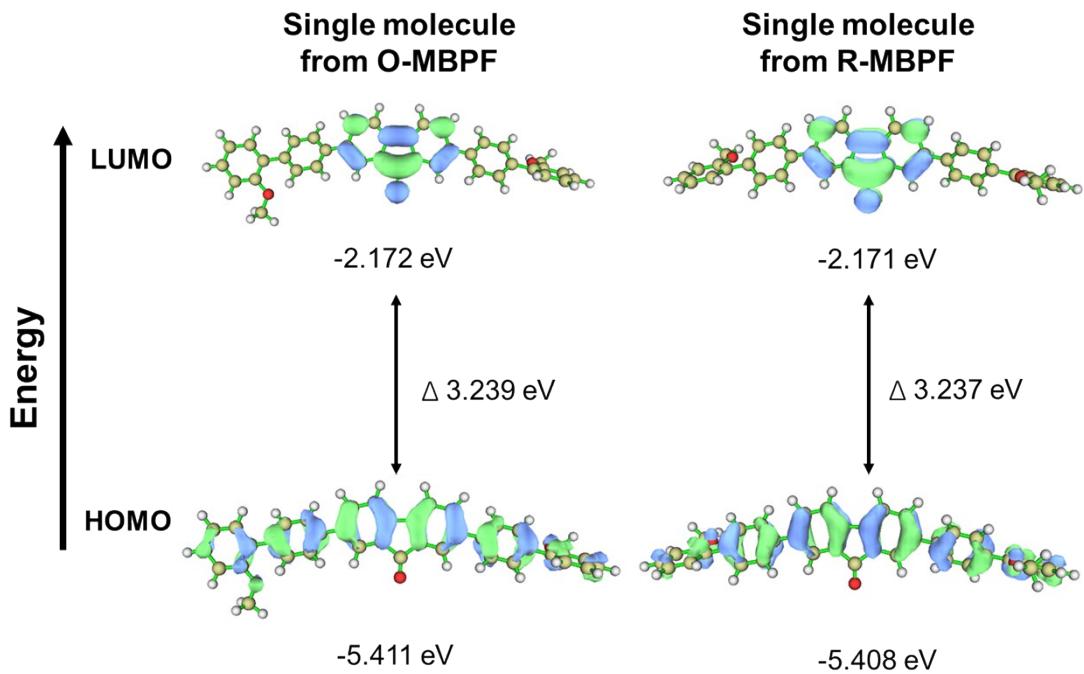


Fig. S10 Calculated HOMO and LUMO electron cloud for orange crystal **O-MBPF** and red crystal **O-MBPF** with their relative energy according to TD-DFT calculation.

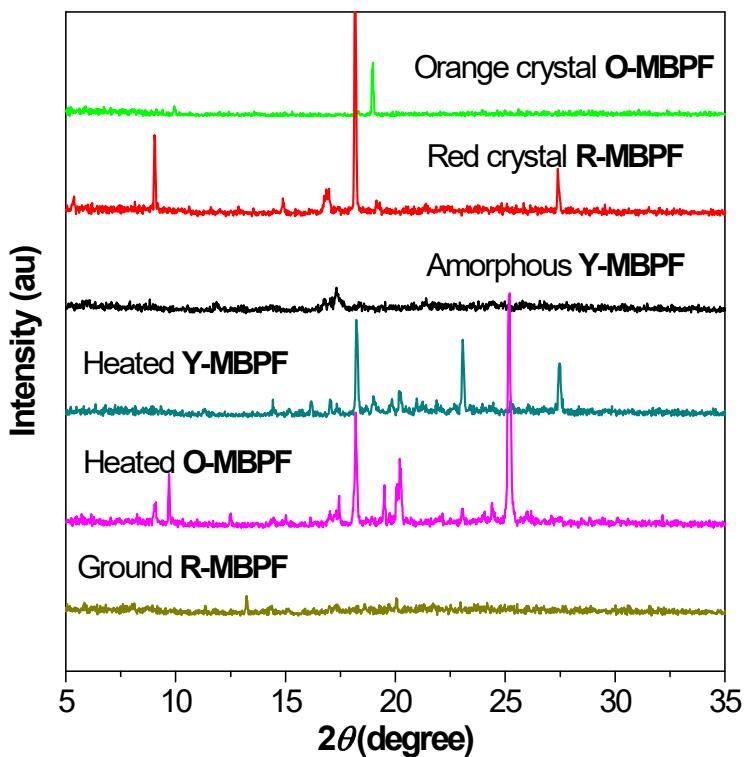


Fig. S11 Powder XRD profiles of the Orange crystal **O-MBPF**, Red crystal **R-MBPF** and amorphous **Y-MBPF**.

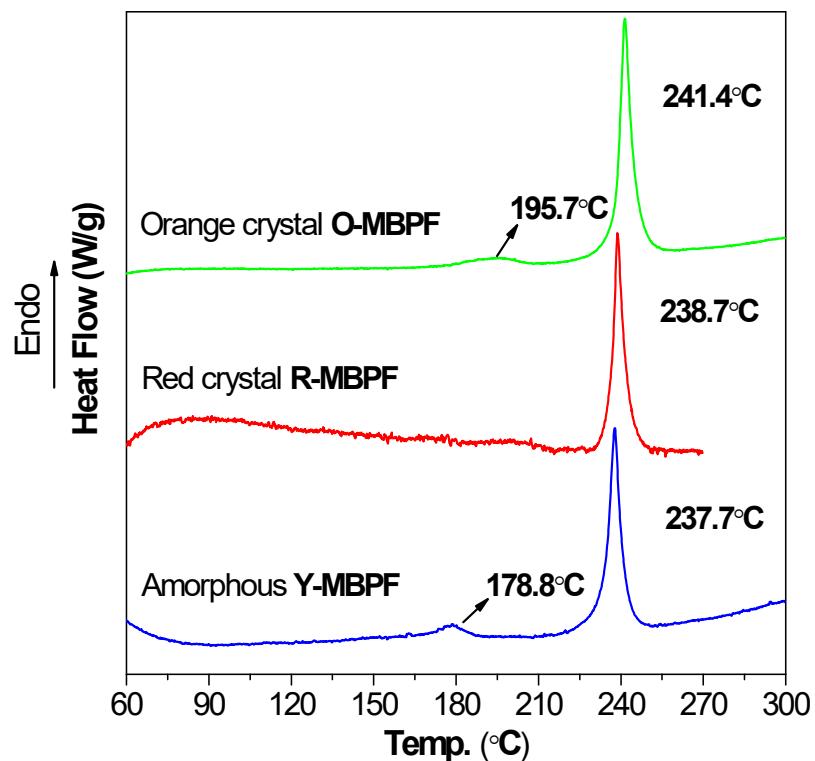


Fig. S12 DSC profiles of the orange crystal **O-MBPF**, red crystal **R-MBPF** and amorphous **Y-MBPF**.

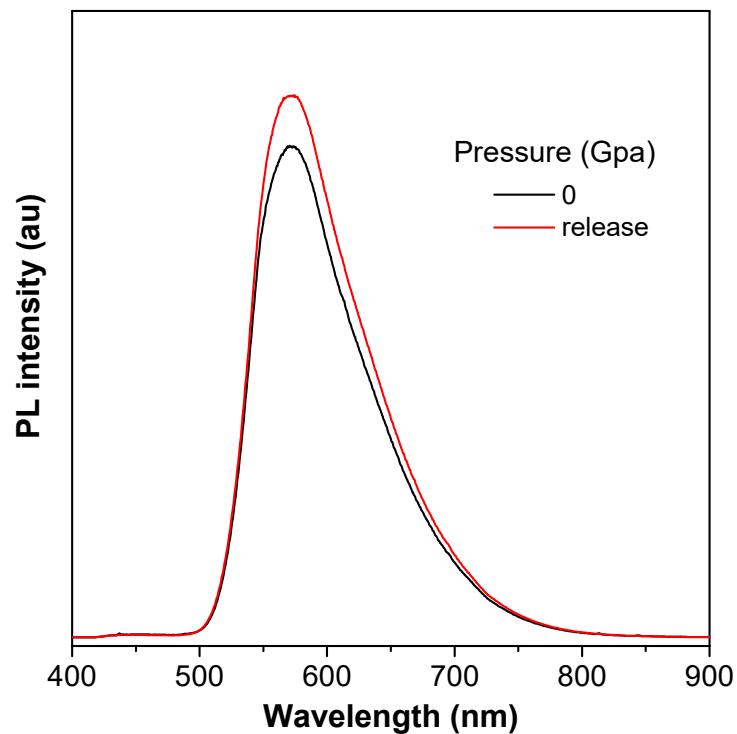


Fig. S13 The PL intensity of **MBPF** after release.

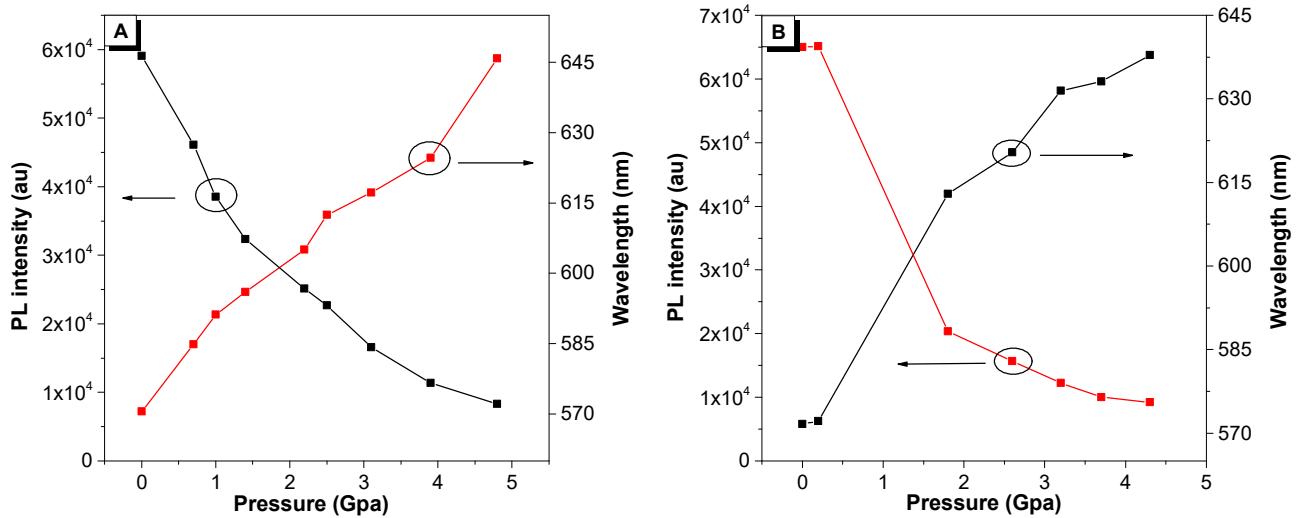


Fig. S14 Changes of PL intensity and wavelength of **R-MBPF** with increasing pressure (A) and decreasing pressure (B).

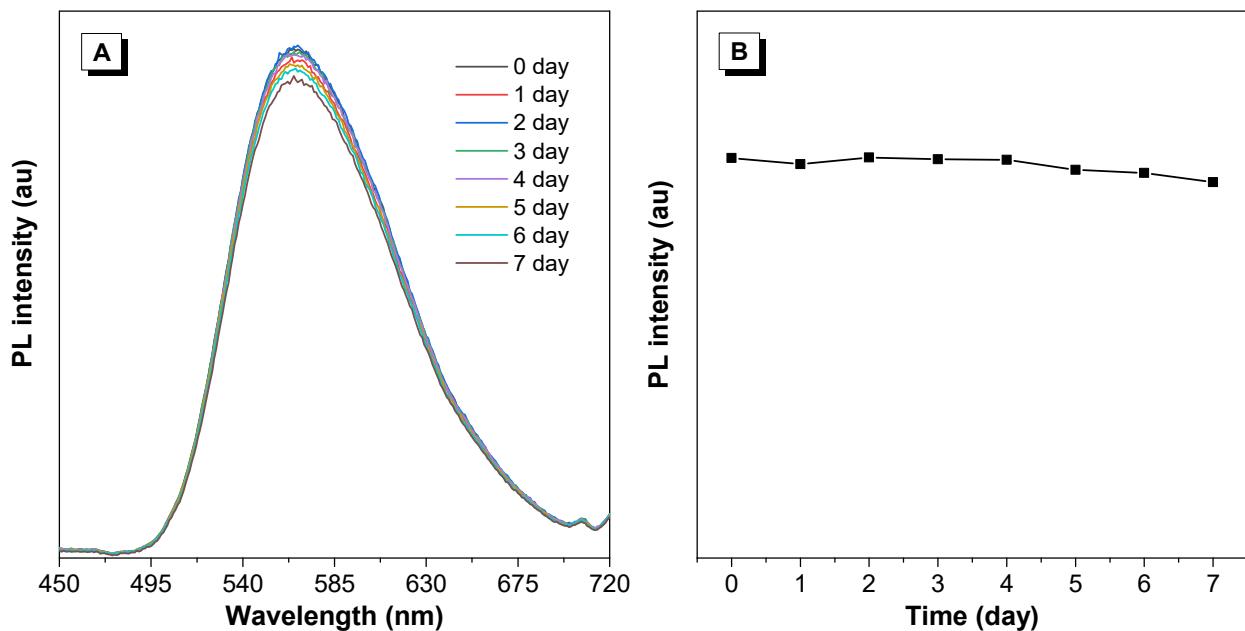
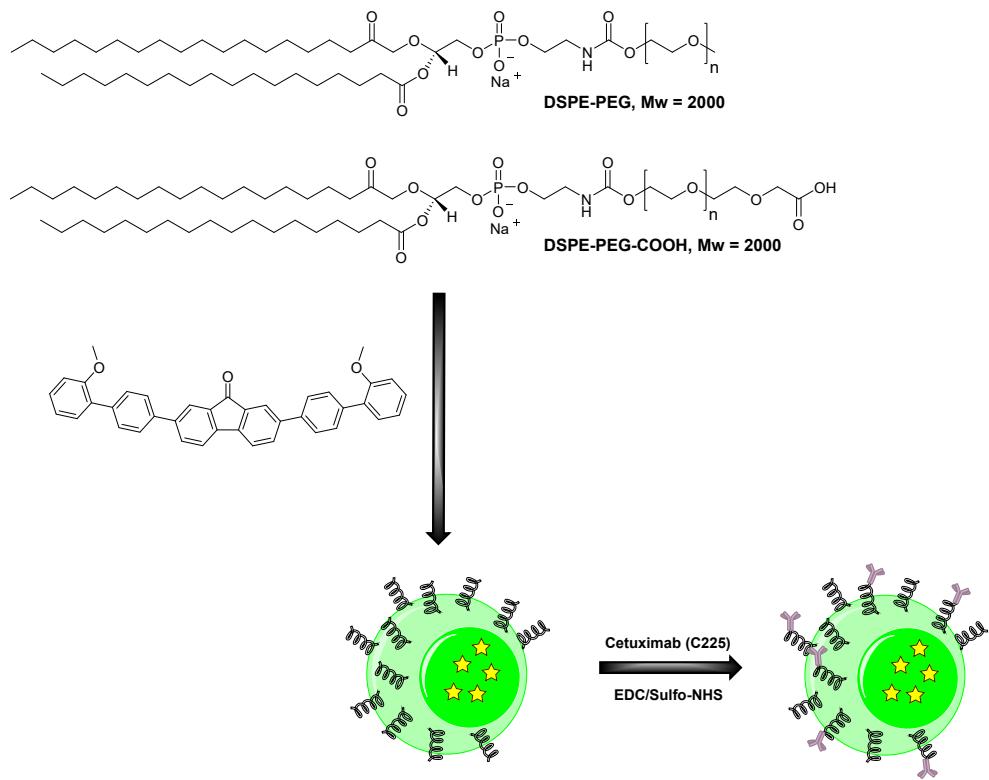


Fig. S15 Photostability of **MBPF** in 90% water/THF vs. storage time.



Scheme S2. Fabrication of **MBPF-C225** NPs.

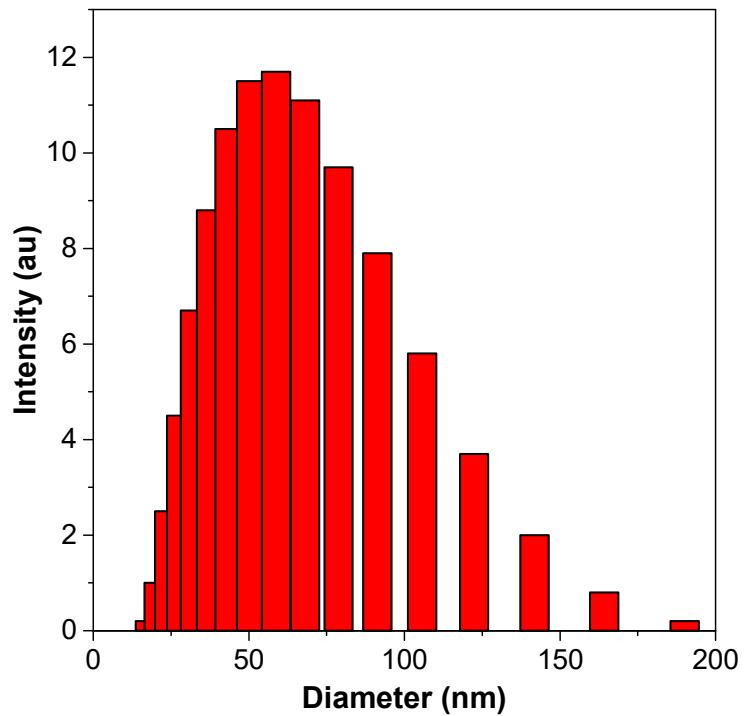


Fig. S16 Particle size distribution of **MBPF-C225** NPs studied by dynamic light scattering.

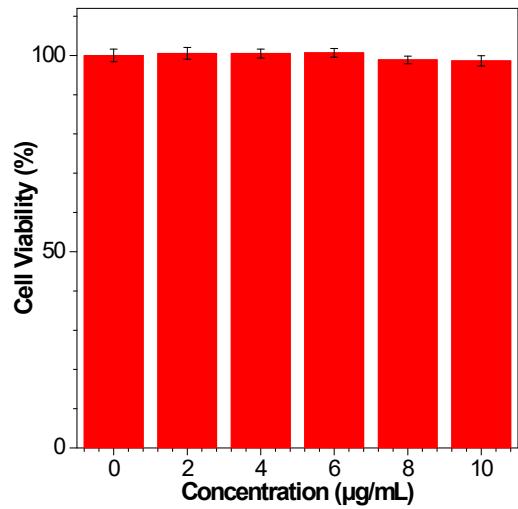


Fig. S17 Cell viability of HCC 827 cells treated with different concentrations of **MBPF-C225** NPs.

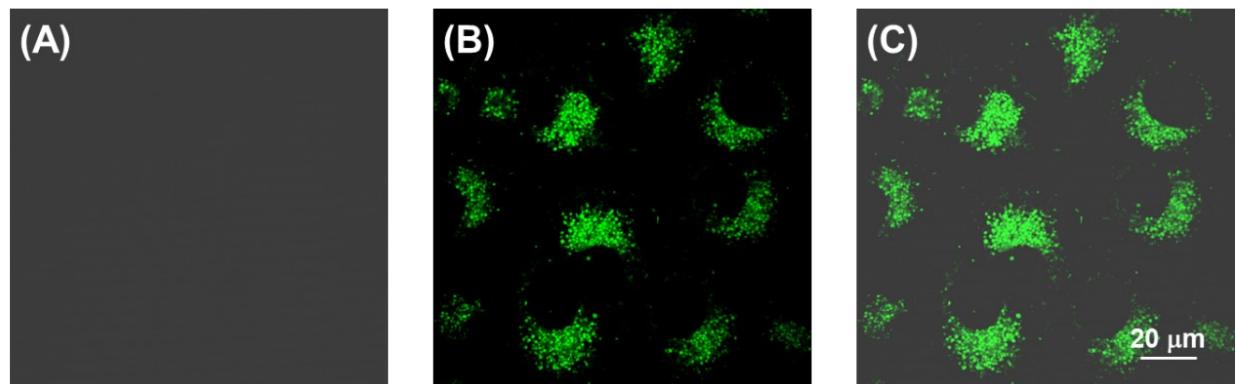


Fig. S18 The CLSM images of HCC827 cells after incubation with **MBPF-C225** NPs at 37 °C for 6 h. (A) Bright-field images; (B) fluorescence images; (C) the merged image of (A) and (B). Scale bar = 20 μm .

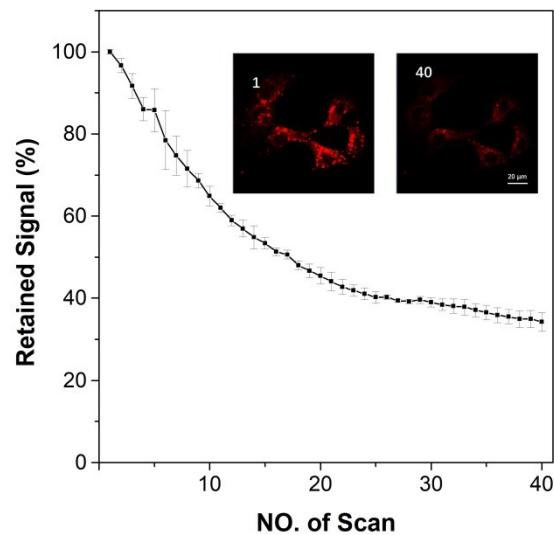


Fig. S19 Retained signal (%) of fluorescence of HCC827 cells stained with Lysotracker red with increasing number of scans. Inset: Fluorescence images of HCC827 cells with increasing number of scans (1 and 40 scans; the number of scans shown in upper left corner).