

## Supporting Information

### Oxygen Reduction Reaction at 68-Atoms Gold Cluster Supported on Carbon Nanotubes: Theoretical and Experimental Analysis

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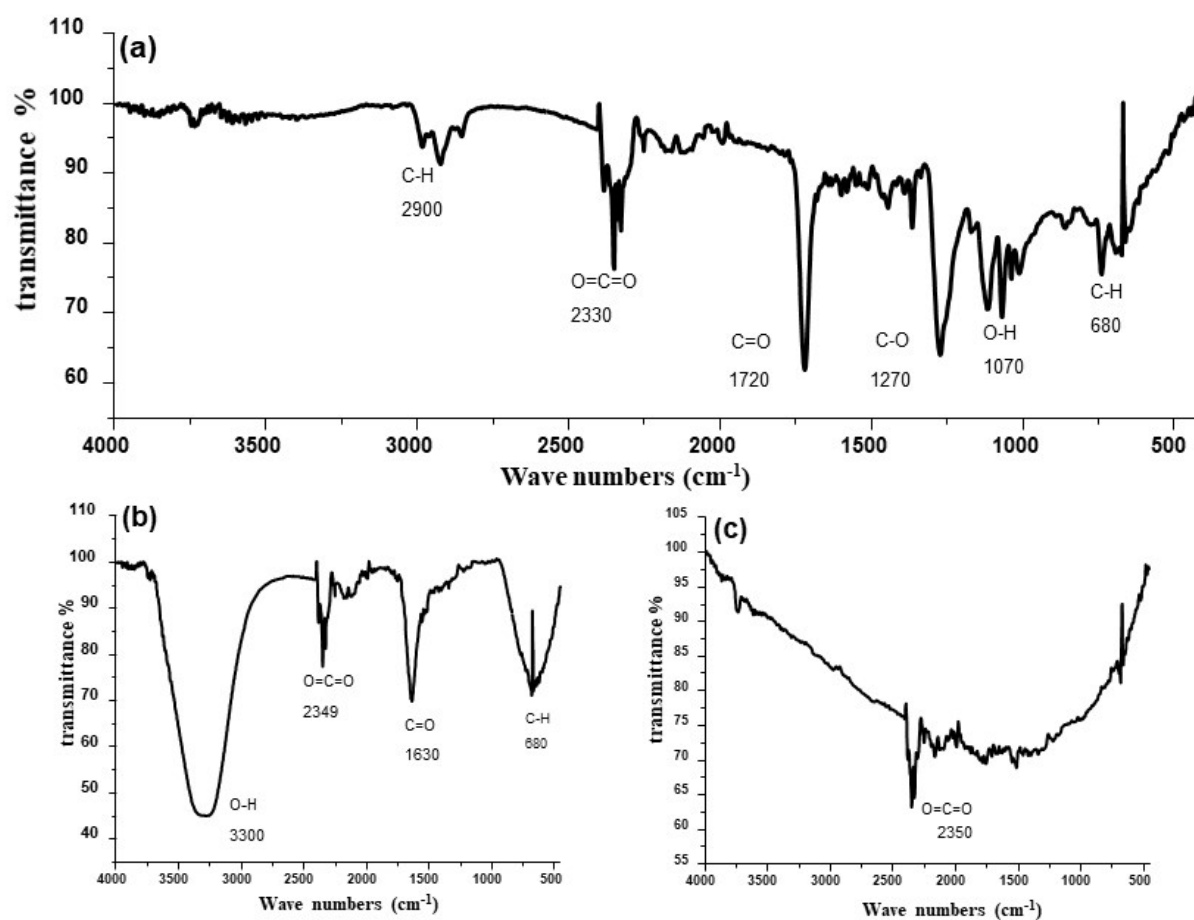
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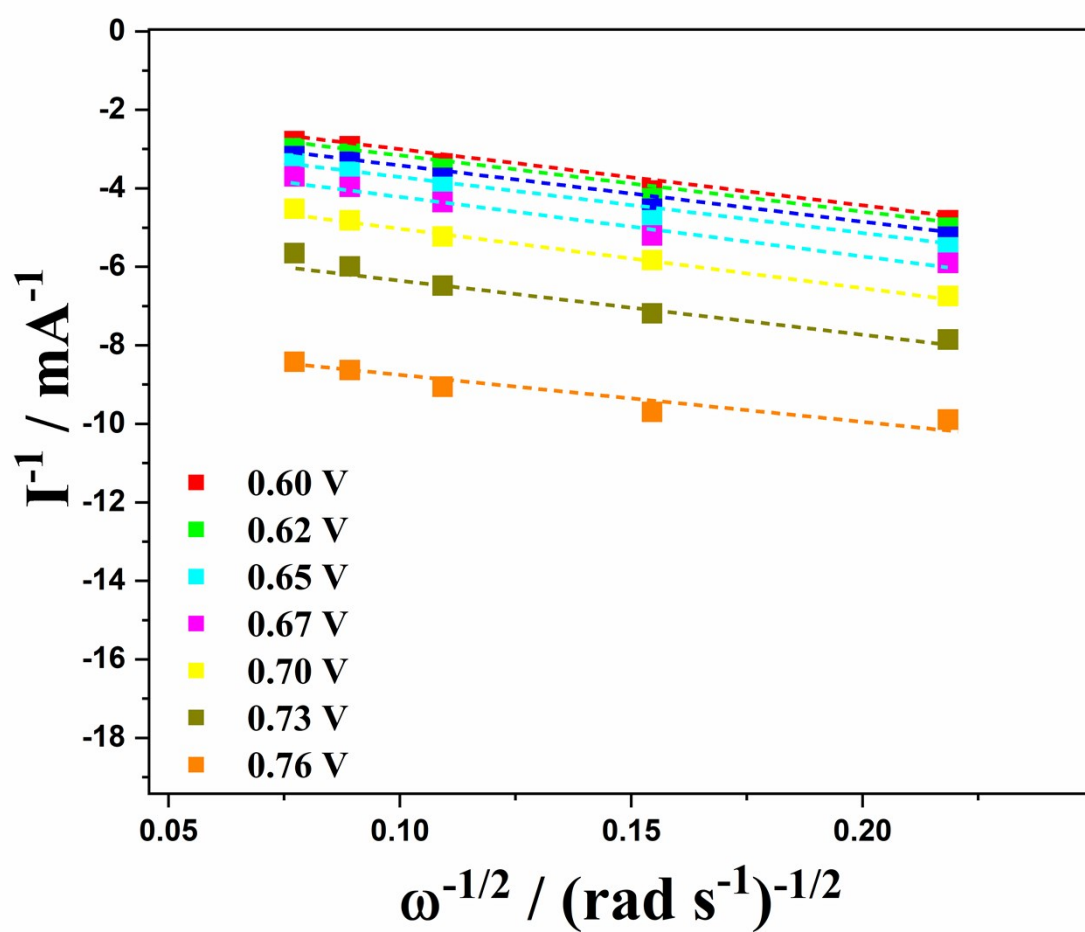
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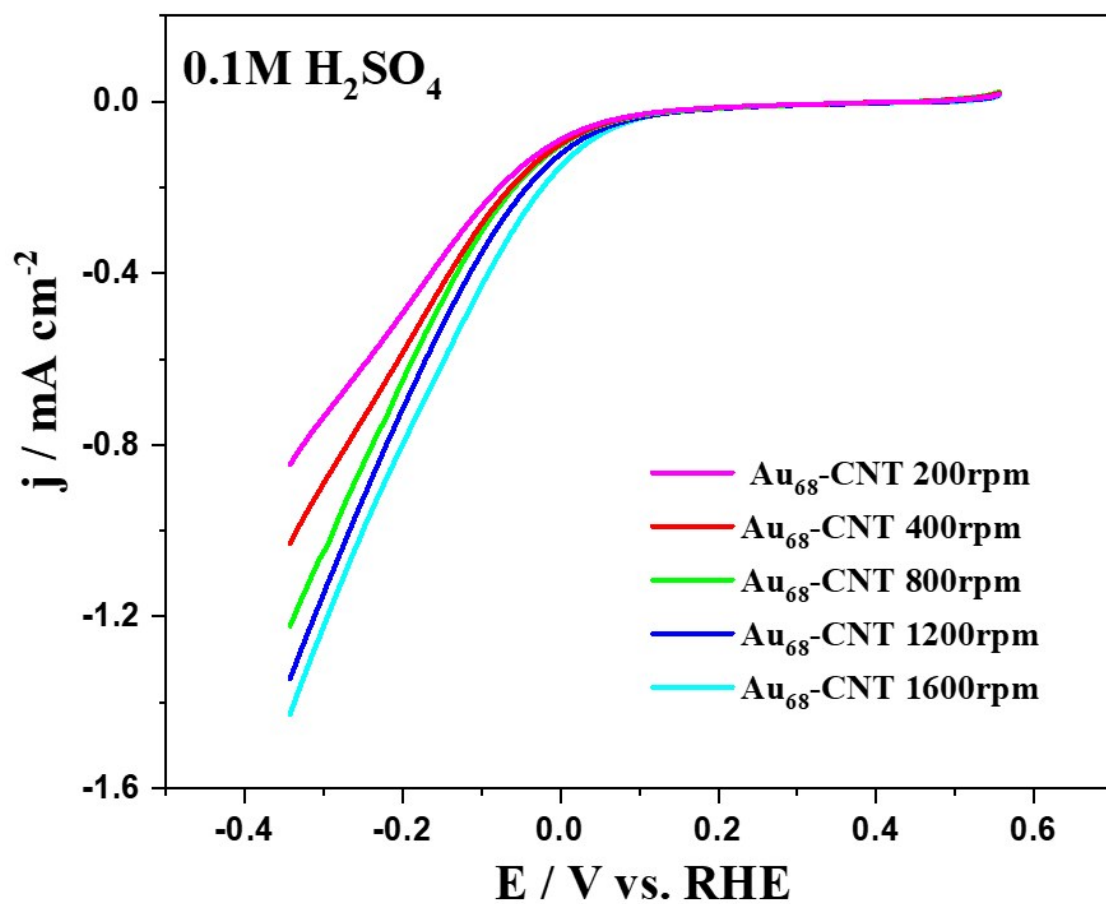
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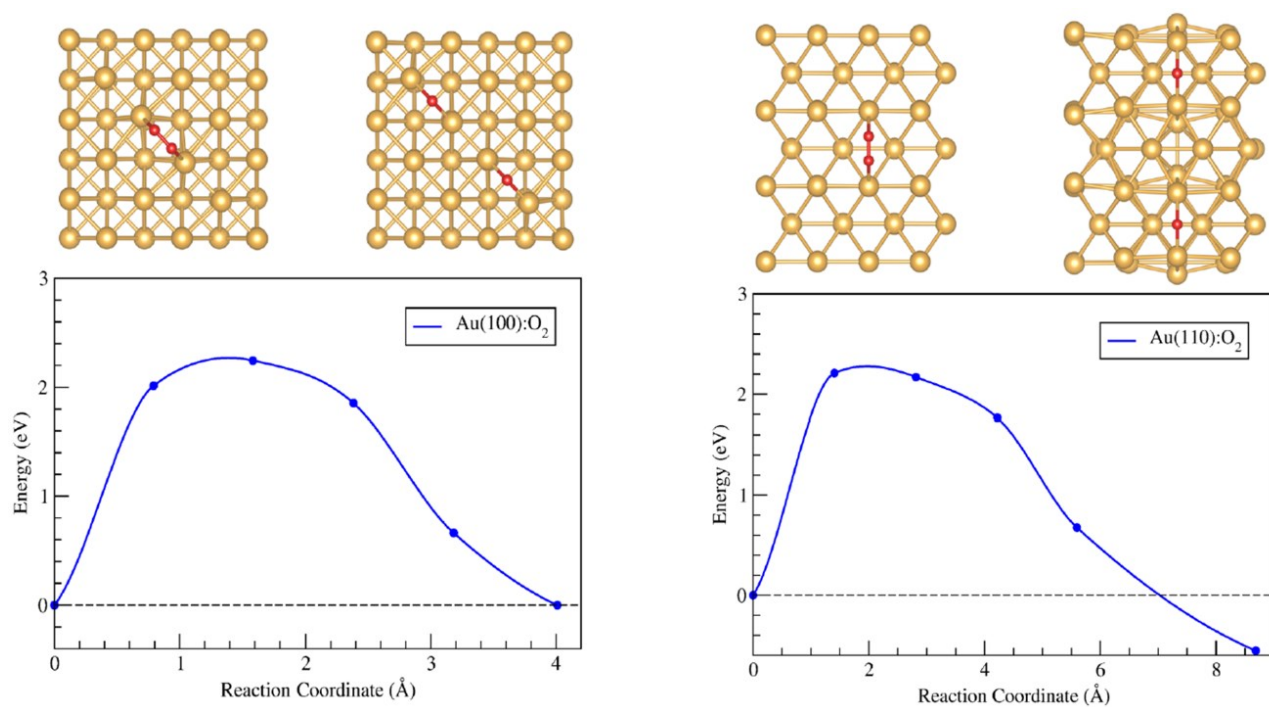
**Figure S1.** FTIR Spectra for (a) solid Au<sub>68</sub> nanoparticles with CNT, (b) Au<sub>68</sub> nanoparticles dispersed in liquid, and (c) carbon nanotubes.



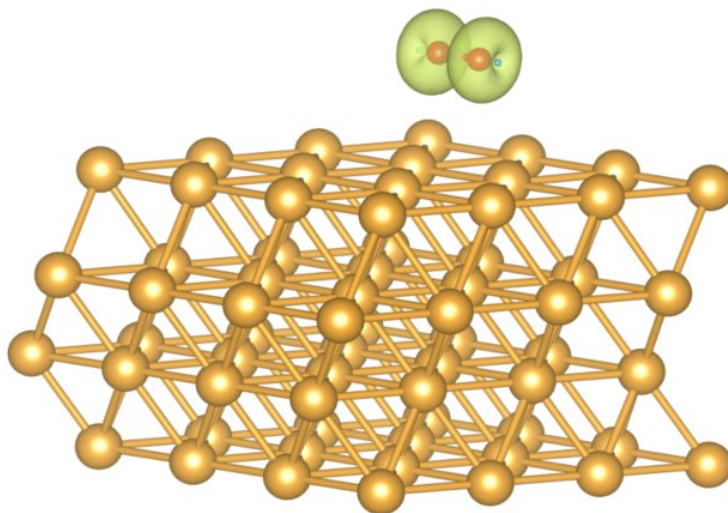
**Figure S2.** Koutecky-Levich plot for Au68-MWCNT at different electrode potentials.



**Figure S3.** Rotating disk electrode at different r.p.m. in 0.1M  $\text{H}_2\text{SO}_4$  with  $\text{O}_2$ -saturated atmosphere and using a scan rate of  $5\text{mV s}^{-1}$ .



**Figure S4.** Minimum energy path for the  $O_2$  dissociation on the Au(100) and Au(110) surfaces. The start and end of the reaction coordinate corresponds to the equilibrium positions shown in the figures.



**Figure S5.** Spin density of  $O_2$  adsorbed on Au(111) surface. Spin-up and spin-down densities are plotted in yellow and blue respectively for an isosurface of  $\rho = 1 \times 10^{-3} e/\text{\AA}^3$ .