

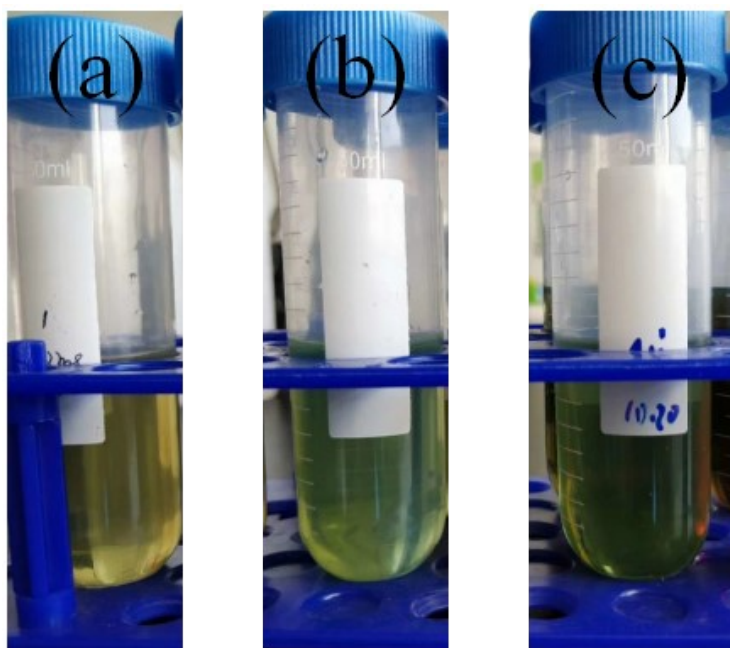
## Supporting Information

### Ethylene glycol assisted self-template conversion approach to synthesize hollow NiS microspheres for a high performance all-solid-state supercapacitor

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**Figure S1. Optical photographs of different solutions after solvothermal reaction using different amount of  $\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  as nickel resource. (a) low amount, (b) moderate amount, and (c) high amount.**

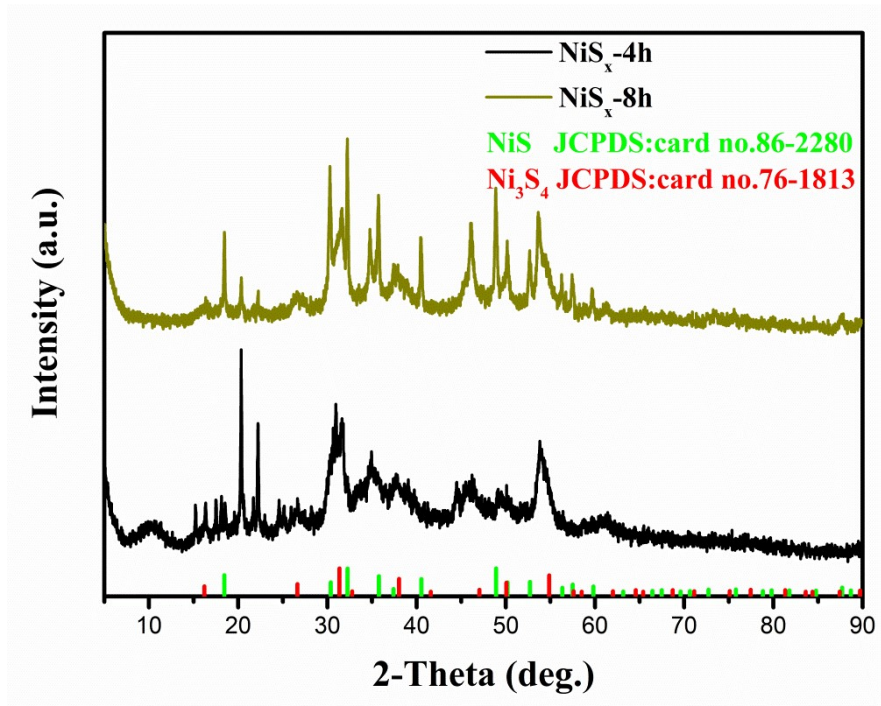


Figure S2. XRD patterns of different vulcanization time of NiS<sub>x</sub>

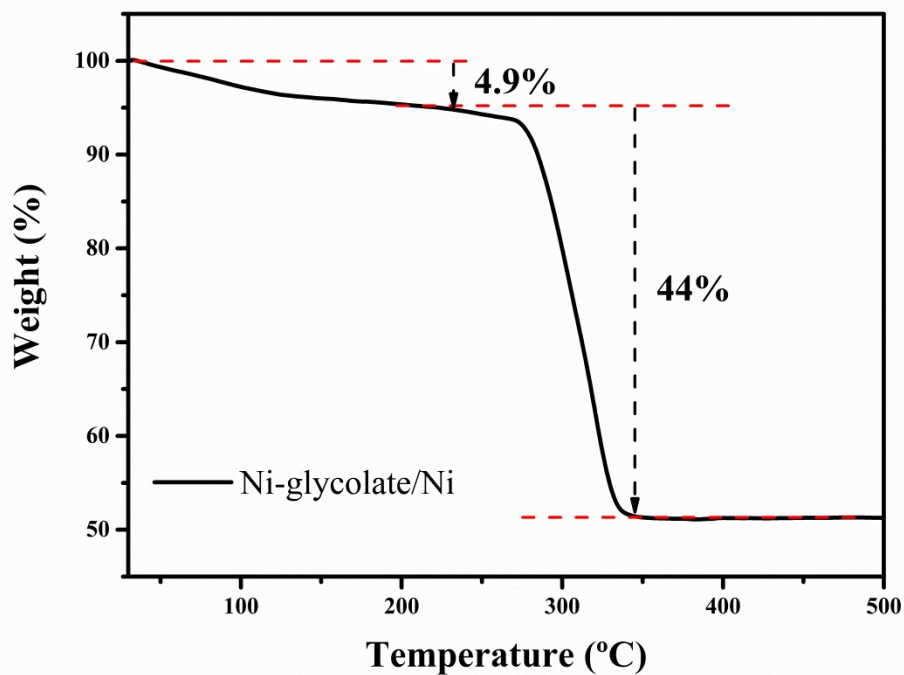


Figure S3. TGA curve of Ni-glycolate/Ni.

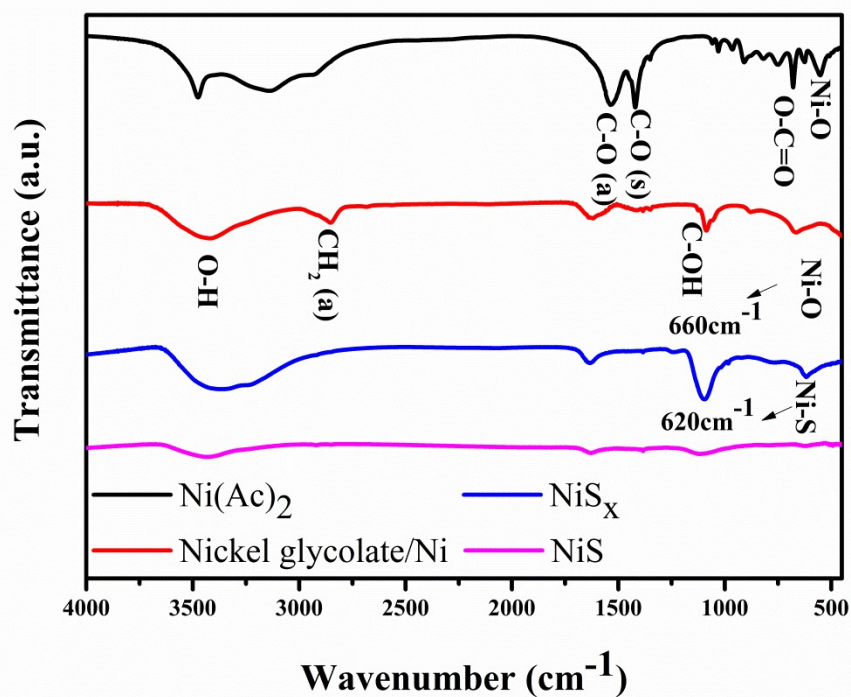


Figure S4. FT-IR spectra of Ni(Ac)<sub>2</sub>, Ni-glycolate/Ni, NiS<sub>x</sub> and NiS.

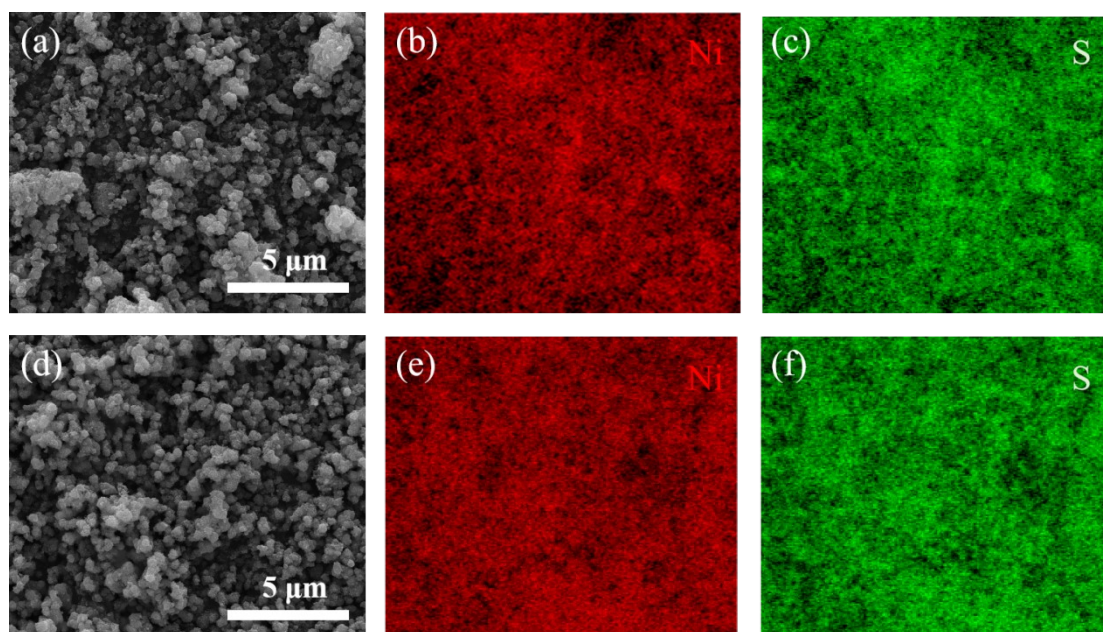
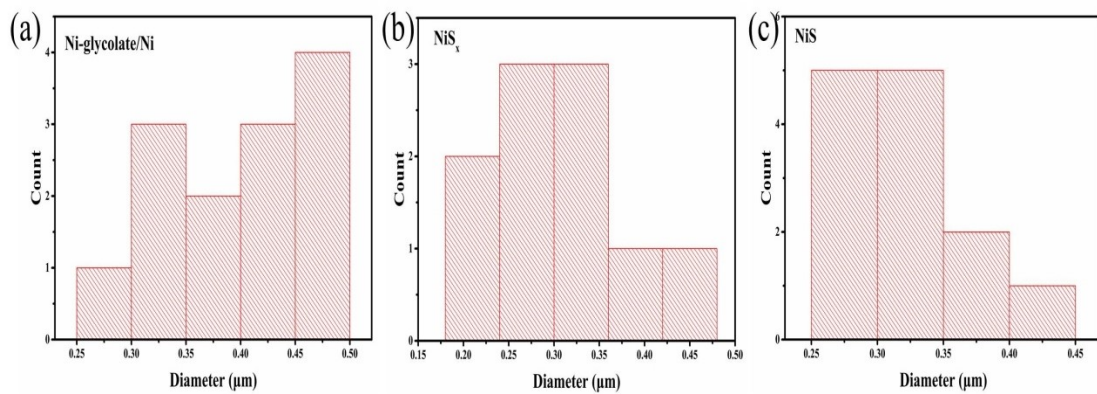
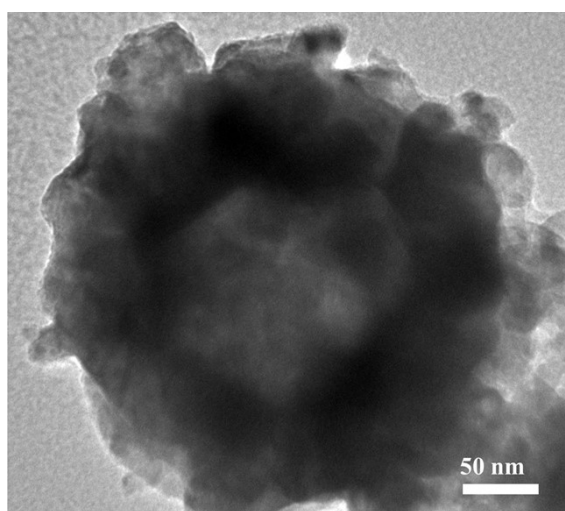


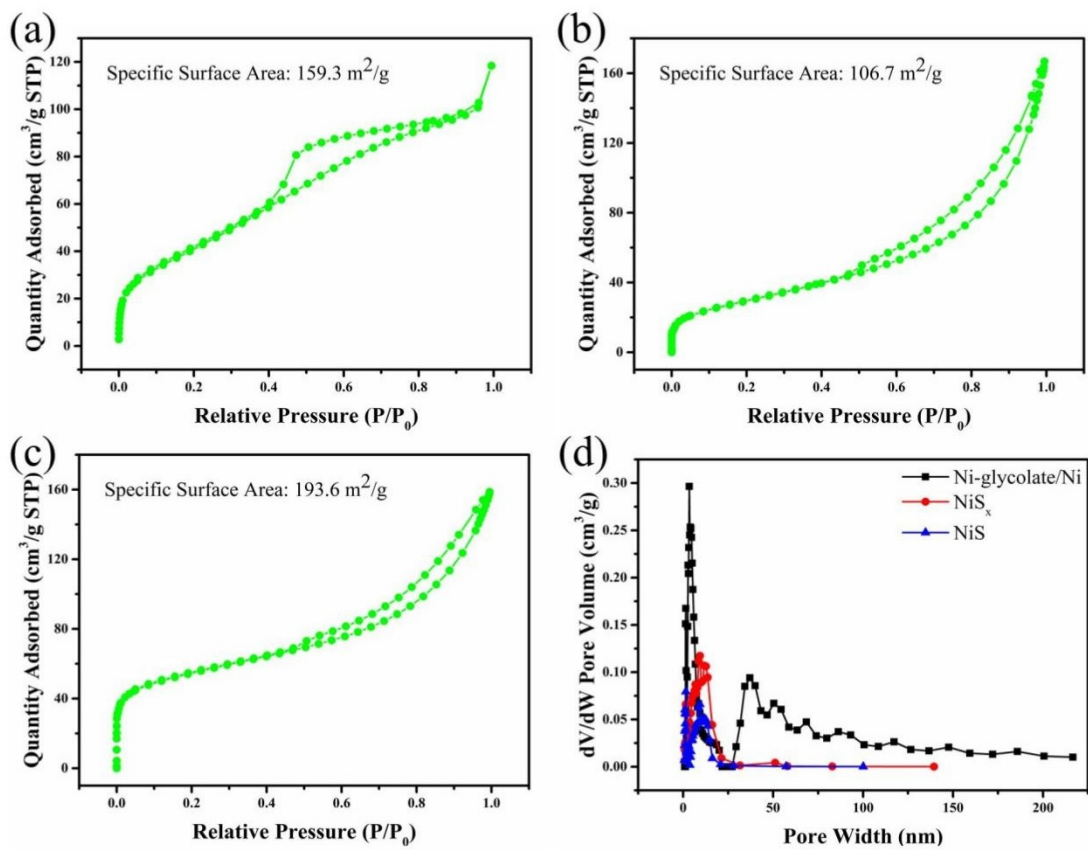
Figure S5. SEM images of (a) NiS, (b–c) EDS mapping of NiS<sub>x</sub>, (d) NiS, (e–f) EDS mapping of NiS.



**Figure S6. Diameter distribution of Ni-glycolate/Ni, (b) NiS<sub>x</sub>, (c) NiS.**



**Figure S7. TEM image of an individual NiS hollow microsphere.**



**Figure S8. Nitrogen adsorption-desorption isotherms of (a) Ni-glycolate/Ni, (b)  $\text{NiS}_x$ , (c) NiS and (d) their corresponding pore size distributions.**

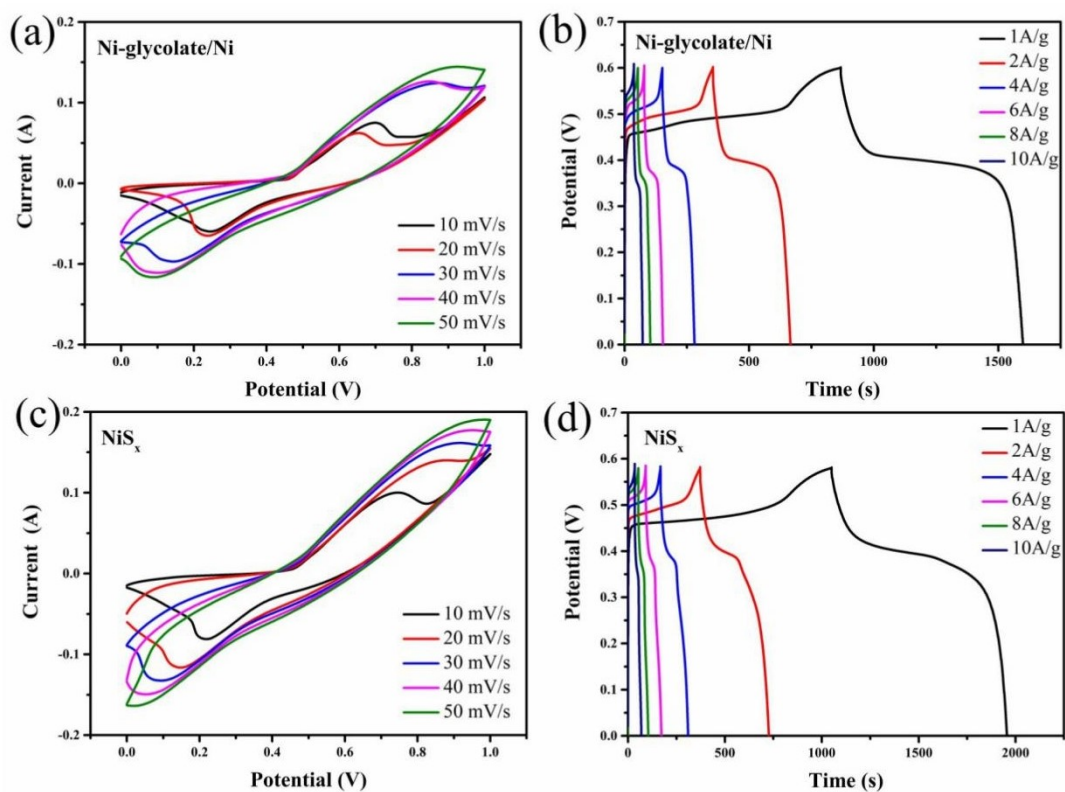


Figure S9. (a) CV curves of Ni-glycolate/Ni, (b) GCD curves of Ni-glycolate/Ni, (c) CV curves of NiS<sub>x</sub> and (d) GCD curves of NiS<sub>x</sub>.

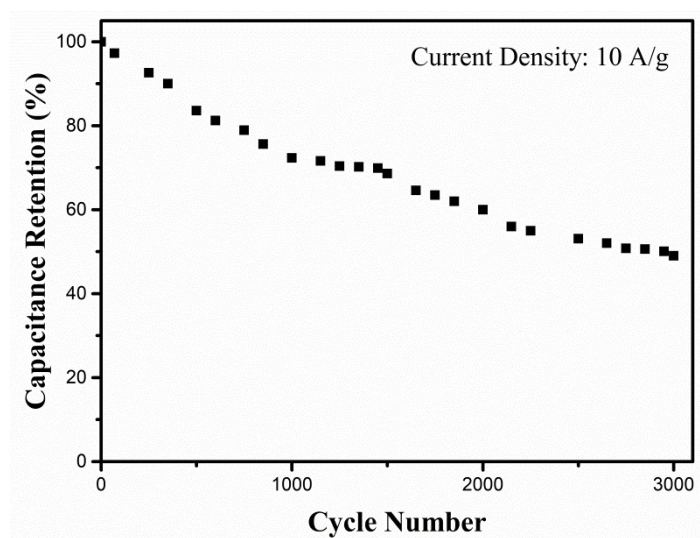


Figure S10. Cycling stability of NiS.

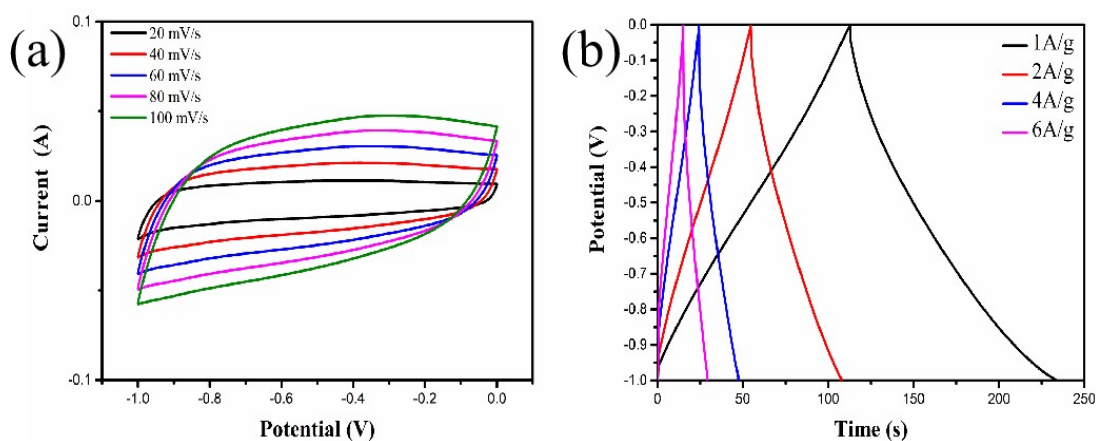


Figure S11. (a) CV curves of AC, (b) GCD curves of AC.

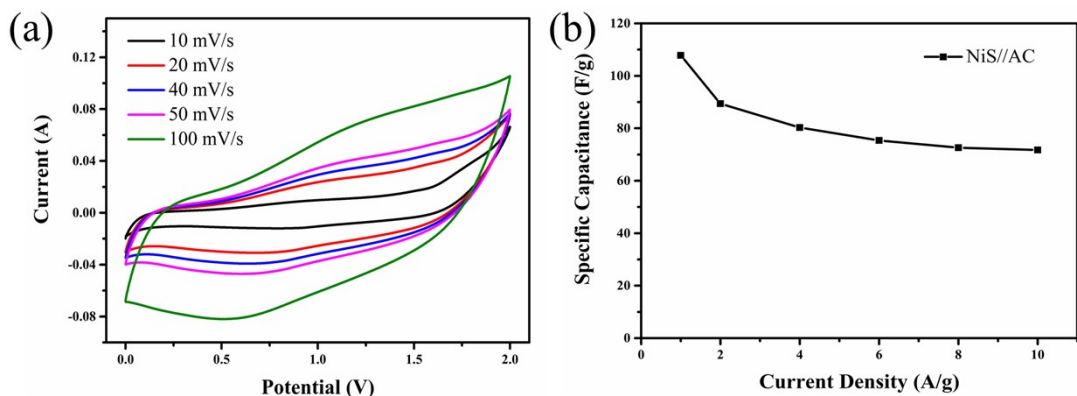


Figure S12. (a) CV profiles of NiS//AC at various sweep rates, (b) the calculated capacitances of the hybrid supercapacitor at 1–10 A/g.

Table S1. Components of the Equivalent Circuit Fitted for the Impedance Spectra

sample	$R_s$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )	$Z_w$ ( $\Omega$ )
Ni-glycolate/Ni	2.06	0.86	0.41
NiS <sub>x</sub>	1.88	0.85	0.51
NiS	1.63	0.13	0.43