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ModifiableDoubledonorbasedonBis(N-Ethyl-N-hydroxyethyl)anilineforOrganicopticalnonlinearChromophores

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1. Structure of YLD124, JRD1 and HLD1-2





Figure S1. Structure of YLD124, JRD1 and HLD1-2



2. UV curves

Figure S2. UV-Vis absorption spectra of chromophores BLD1-BLD4 in solvents

3. Optimized structures of chromophores



Fig. S3 Optimized structures of chromophores BLD1-BLD4.

4. ¹H and ¹³C NMR spectra



Figure S4¹H NMR spectra of Chromophore BLD1



Figure S5 ¹³C NMR spectra of Chromophore BLD1



Figure S6¹H NMR spectra of Chromophore BLD2





Figure S8¹H NMR spectra of Chromophore BLD3





Figure S10¹H NMR spectra of Chromophore BLD4



Figure S11¹³C NMR spectra of Chromophore BLD4

5. DFT calculation and poling of the chromophore

Cmpd	BLD1		BLD2		BLD3		BLD4	
	номо	LUMO	номо	LUMO	номо	LUMO	номо	LUMO
donor	42.42%	14.48%	43.99%	14.40%	43.83%	14.41%	44.43%	15.41%
π bridge	39.67%	46.17%	40.14%	47.18%	37.86%	46.59%	38.47%	47.31%
acceptor	17.91%	39.35%	15.87%	38.42%	18.31%	38.99%	17.10%	37.28%

Table S1 The M.O Composition (%) in the Ground State for Chromophores A-D

The molecular orbital composition was calculated using Multiwfn program with Ros-Schuit (SCPA) partition.



Figure S12 Artist's concept of poling process of EO chromophores BLD1-BLD4.

6. Future Directions.

Bis(*N-Ethyl-N-hydroxyethyl*)aniline-based group modified with two OH groups were designed to enhance the electron-donating ability of chromophores and provide the linking sites for further modification of the donor for the first time. At present, there are few modification sites for donors with strong electron donor ability. As shown in Fig.S13, through esterification, groups that can be self-assembled, cross-linked or click chemical reaction can be introduced into the three OH groups on the donor and electron bridge of the chromophores. Cross-Linkable Molecular Glasses, Self-assembled dendrimers or click-type postfunctionalization of Nonlinear Optical Materials with large electro-optic coefficients are expected.



chromophores