Electronic Supplementary Information for:

A General Method for the Synthesis of Covalent and Ionic Amine Borane Complexes Containing Trinitromethyl Fragments

Jin Wang,^{a,c} Ming-Yue Ju,^a Xi-Meng Chen,^{*a} and Xuenian Chen^{*a,b}

^a School of Chemistry and Chemical Engineering, Henan Key Laboratory of Boron Chemistry and Advanced Energy Materials, Henan Normal University, Xinxiang, Henan 453007, China

^b Green Catalysis Center and College of Chemistry, Zhengzhou University, Zhengzhou, Henan 450001, China

^c College of Chemistry and Chemical Engineering, Mudanjiang Normal University, Mudanjiang, Heilongjiang 157011, China

E-mails: xnchen@htu.edu.cn; chenximeng@htu.edu.cn

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Entry	Comps.	А	¹ H NMR (ppm)	
1	3 a	NH ₃	2.61	
2	3b	CH ₃ NH ₂	2.53	
3	3c	$(CH_3)_2NH$	2.47	
4	3d	$(CH_3)_3N$	-	
5	3e	$(NH_2CH_2)_2$	2.02 ^b	
6	3f	$(CH_3)_3N$	-	
7	3g	(NH ₃) ₂	2.56	
^a "A" is a Lewis base; ^b molecular formula [NH ₂ CH ₂ BH ₂ C(NO ₂) ₃] ₂ .				

Table S1. The chemical shift of hydrides in BH₂ group in 3a-3g in ¹H NMR^a.



Fig. S1. IR spectrum of the residue from the TGA–MS experiment of NH₃BH₂C(NO₂)₃ (3a) (The spectrum is similar to that of B₂O₃).



Fig. S2. X-ray powder diffraction pattern of the residue from the TGA–MS experiment of $NH_3BH_2C(NO_2)_3$ (3a) (The peaks are related to B_2O_3).



Fig. S3. IR spectrum of the residue from the TGA–MS experiment of [H₂B(NH₃)₂][C(NO₂)₃] (3g) (The spectrum is similar to that of B₂O₃).



Fig. S4. X-ray powder diffraction pattern of the residue from the TGA–MS experiment of [H₂B(NH₃)₂][C(NO₂)₃] (3g) (The peaks are related to B₂O₃).



Fig. S5. ^{11}B (a) and $^{11}B{^{1}H}$ NMR (b) spectra of NH_3BH_2Cl (1a) in THF.



Fig. S6. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of CH₃NH₂BH₂Cl (1b) in THF.



Fig. S7. ^{11}B (a) and $^{11}B{^{1}H}$ NMR (b) spectra of $(CH_3)_2NHBH_2Cl$ (1c) in THF.





Fig. S9. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of ClBH₂NH₂CH₂CH₂NH₂BH₂Cl (1e) in THF.



Fig. S8. ^{11}B (a) and $^{11}B\{^{1}H\}$ NMR (b) spectra of $(CH_{3})_{3}NBH_{2}Cl$ (1d) in THF.





Fig. S12. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of NH₃BH₂C(NO₂)₃ (3a) in CD₃CN.



BH₂

Fig. S11. ^{11}B (a) and $^{11}B{^{1}H}$ NMR (b) spectra of $(NH_3)_2BH_2Cl$ (1g) in DMF.

(b)





Fig. S14. IR spectrum of NH₃BH₂C(NO₂)₃ (3a).



Fig. S15. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of CH₃NH₂BH₂C(NO₂)₃ (3b) in CD₃CN.



Fig. S16. ^{1}H (a) and $^{1}H{^{11}B}$ NMR (b) spectra of $CH_3NH_2BH_2C(NO_2)_3$ (3b) in CD_3CN .



-29.48

Fig. S17. ¹³C NMR spectrum of CH₃NH₂BH₂C(NO₂)₃ (3b) in CD₃CN.



Fig. S18. IR spectrum of CH₃NH₂BH₂C(NO₂)₃ (3b).



Fig. S19. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of (CH₃)₂NHBH₂C(NO₂)₃ (3c) in CD₃CN.



Fig. S20. ¹H (a) and ¹H{¹¹B} NMR (b) spectra of (CH₃)₂NHBH₂C(NO₂)₃ (3c) in CD₃CN.



Fig. S21. ¹³C NMR spectrum of (CH₃)₂NHBH₂C(NO₂)₃ (3c) in CD₃CN.



Fig. S22. IR spectrum of (CH₃)₂NHBH₂C(NO₂)₃ (3c).



Fig. S23. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of [CH₂NH₂BH₂C(NO₂)₃]₂ (3e) in CD₃CN.



Fig. S24. ^{1}H (a) and $^{1}H{^{11}B}$ NMR (b) spectra of $[CH_2NH_2BH_2C(NO_2)_3]_2$ (3e) in CD_3CN .



Fig. S25. ¹³C NMR spectrum of [CH₂NH₂BH₂C(NO₂)₃]₂ (3e) in CD₃CN.



Fig. S26. IR spectrum of [CH₂NH₂BH₂C(NO₂)₃]₂ (3e).



Fig. S27. ¹¹B (a) and ¹¹B{¹H} NMR (b) spectra of [H₂B(NH₃)₂][C(NO₂)₃] (3g) in CD₃CN.



Fig. S28. ¹H (a) and ¹H{¹¹B} NMR (b) spectra of [H₂B(NH₃)₂][C(NO₂)₃] (3g) in CD₃CN.



Fig. S29. IR spectrum of [H₂B(NH₃)₂][C(NO₂)₃] (3g).



Fig. S30. X-ray powder diffraction pattern of KCl.



Fig. S31. IR spectrum of K[C(NO₂)₃].