

## Electronic Supplementary Information

### Understanding the different reactivity of (*Z*)- and (*E*)- $\beta$ -nitrostyrenes in [3+2] cycloaddition reactions. An MEDT study

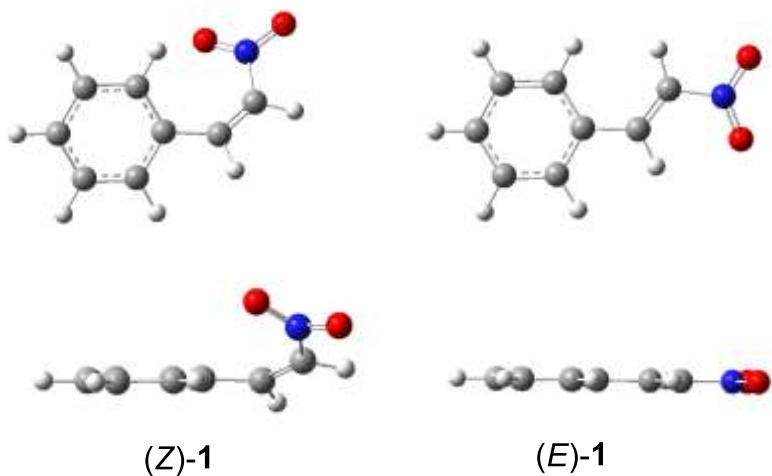
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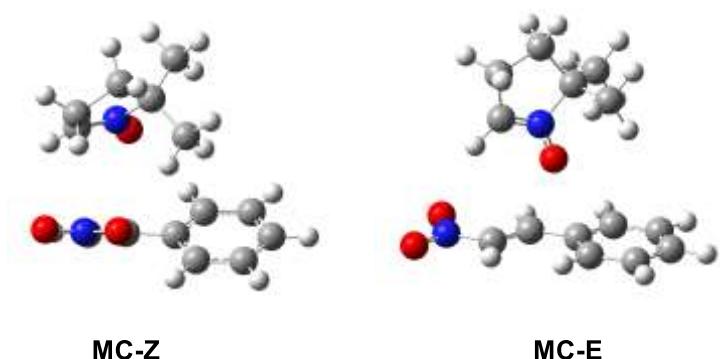
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#### Index

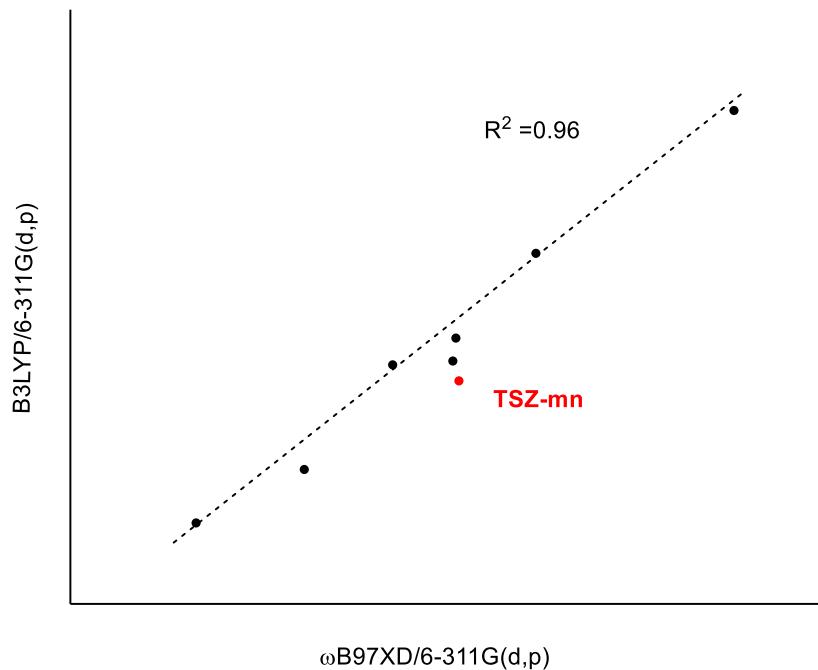
- S2** Figure S1.  $\omega$ B97X-D/6-311G(d,p) optimised geometries of (*Z*)- and (*E*)- $\beta$ -nitrostyrenes (*Z*)-**1** and (*E*)-**1**
- S3** Figure S2.  $\omega$ B97X-D/6-311G(d,p) optimised geometries of MCs **MC-Z** and **MC-E**
- S4** Figure S3. Plot of the B3LYP/6-311G(d,p) total energies vs the  $\omega$ B97X-D/6-311G(d,p) ones of the eight TSs involved in the 32CA reactions of (*Z*)- and (*E*)- $\beta$ -nitrostyrenes (*Z*)-**1** and (*E*)-**1** with nitrone **2**
- S5** Tables S1 and S2 with  $\omega$ B97X-D/6-311G(d,p) total energies, enthalpies, entropies and Gibbs free energies, computed at 25 °C and 1 atm. in benzene, of the stationary points involved in the 32CA reactione of (*Z*)- $\beta$ -nitrostyrene (*Z*)-**1** and (*E*)- $\beta$ -nitrostyrene (*E*)-**1** with nitrone **2**
- S6** Tables S3 and S4 with B3LYP/6-311G(d,p) total energies, enthalpies, entropies and Gibbs free energies, computed at 25 °C and 1 atm. in benzene, of the stationary points involved in the 32CA reactions of (*Z*)- $\beta$ -nitrostyrene (*Z*)-**1** and (*E*)- $\beta$ -nitrostyrene (*E*)-**1** with nitrone **2**
- S7** Table S5 and S6 with HF/6-311G(d,p) and M06-2X/6-311G(d,p) total and relative energies of **TS-Zmx** and **TS-Emn**, computed in benzene



**Figure S1.**  $\omega$ B97X-D/6-311G(d,p) optimised geometries of (*Z*)- and (*E*)- $\beta$ -nitrostyrenes. (*Z*)-**1** and (*E*)-**1**. The ArC–C–C–N dihedral angle at (*Z*)-**1** is 3.5 degrees.



**Figure S2.**  $\omega$ B97X-D/6-311G(d,p) optimised geometries of MCs **MC-Z** and **MC-E**.



**Figure S3.** Plot of the B3LYP/6-311G(d,p) total energies, in a.u., vs the  $\omega$ B97X-D/6-311G(d,p) ones, in a.u., in benzene, of the eight TSs involved in the 32CA reactions of (*Z*)- and (*E*)- $\beta$ -nitrostyrenes (*Z*)-**1** and (*E*)-**1** with nitrone **2** (see Tables S1-S4).

**Table S1.**  $\omega$ B97X-D/6-311G(d,p) total energies, E in a.u., enthalpies, H in a.u., entropies,  $\Delta S$  in cal·mol<sup>-1</sup>K<sup>-1</sup>, and Gibbs free energies, G in a.u., computed at 25 °C and 1 atm. in benzene, of the stationary points involved in the 32CA reaction of (*Z*)- $\beta$ -nitrostyrene (*Z*)-**1** with nitrone **2**.

	E	H	S	G
<b>(Z)-1</b>	-514.095089	-513.947175	97.7	-513.993607
<b>2</b>	-365.162654	-364.986862	85.0	-365.027262
<b>MC-Z</b>	-879.276365	-878.950385	140.8	-879.017290
<b>TS-Zon</b>	-879.235604	-878.909949	131.1	-878.972232
<b>TS-Zox</b>	-879.244029	-878.918790	132.9	-878.981933
<b>TS-Zmn</b>	-879.251824	-878.927000	129.3	-878.988422
<b>TS-Zmx</b>	-879.250870	-878.926078	132.9	-878.989233
<i>cis</i> - <b>9</b>	-879.302746	-878.974366	127.9	-879.035140
<i>cis</i> - <b>10</b>	-879.309221	-878.980853	129.6	-879.042428
<i>cis</i> - <b>8</b>	-879.302489	-878.974077	129.2	-879.035474
<i>cis</i> - <b>4</b>	-879.306082	-878.977762	129.3	-879.039187

**Table S2.**  $\omega$ B97X-D/6-311G(d,p) total energies, E in a.u., enthalpies, H in a.u., entropies,  $\Delta S$  in cal·mol<sup>-1</sup>K<sup>-1</sup>, and Gibbs free energies, G in a.u., computed at 25 °C and 1 atm. in benzene, of the stationary points involved in the 32CA reaction of (*E*)- $\beta$ -nitrostyrene (*E*)-**1** with nitrone **2**.

	E	H	S	G
<b>(E)-1</b>	-514.104821	-513.956924	96.1	-514.002608
<b>2</b>	-365.162654	-364.986862	85.0	-365.027262
<b>MC-E</b>	-879.283678	-878.957663	147.8	-879.027901
<b>TS-Eon</b>	-879.249251	-878.924225	132.8	-878.987317
<b>TS-Eox</b>	-879.250681	-878.925426	131.2	-878.987770
<b>TS-Emn</b>	-879.260751	-878.935842	131.5	-878.998323
<b>TS-Emx</b>	-879.257320	-878.932355	132.3	-878.995197
<i>trans</i> - <b>6</b>	-879.305497	-878.976913	128.2	-879.037837
<i>trans</i> - <b>7</b>	-879.306691	-878.978253	129.3	-879.039711
<i>trans</i> - <b>3</b>	-879.305044	-878.976738	130.9	-879.038919
<i>trans</i> - <b>5</b>	-879.299597	-878.971429	129.9	-879.033168

**Table S3.** B3LYP/6-311G(d,p) total energies, E in a.u., enthalpies, H in a.u., entropies,  $\Delta S$  in cal·mol<sup>-1</sup>K<sup>-1</sup>, and Gibbs free energies, G in a.u., computed at 25 °C and 1 atm. in benzene, of the stationary points involved in the 32CA reaction of (Z)- $\beta$ -nitrostyrene (*Z*)-**1** with nitrone **2**.

	E	H	S	G
<b>(Z)-1</b>	-514.280235	-514.133756	97.9	-514.180254
<b>2</b>	-365.280402	-365.106376	85.9	-365.147197
<b>MC-Z</b>	-879.568759	-879.246033	159.0	-879.321588
<b>TS-Zon</b>	-879.521630	-879.199731	133.6	-879.263203
<b>TS-Zox</b>	-879.530934	-879.209438	135.6	-879.273850
<b>TS-Zmn</b>	-879.534393	-879.213029	131.7	-879.275600
<b>TS-Zmx</b>	-879.537692	-879.216320	134.2	-879.280077
<i>cis</i> - <b>9</b>	-879.575272	-879.250846	130.9	-879.313045
<i>cis</i> - <b>10</b>	-879.581293	-879.256825	131.2	-879.319148
<i>cis</i> - <b>8</b>	-879.576419	-879.251870	131.2	-879.314186
<i>cis</i> - <b>4</b>	-879.579167	-879.254762	131.8	-879.317381

**Table S4.** B3LYP/6-311G(d,p) total energies, E in a.u., enthalpies, H in a.u., entropies,  $\Delta S$  in cal·mol<sup>-1</sup>K<sup>-1</sup>, and Gibbs free energies, G in a.u., computed at 25 °C and 1 atm. in benzene, of the stationary points involved in the 32CA reaction of (*E*)- $\beta$ -nitrostyrene (*E*)-**1** with nitrone **2**.

	E	H	S	G
<b>(E)-1</b>	-514.290326	-514.144048	96.0	-514.189649
<b>2</b>	-365.280402	-365.106376	85.9	-365.147197
<b>MC-E</b>	-879.580537	-879.258091	155.2	-879.331855
<b>TS-Eon</b>	-879.534733	-879.213061	133.2	-879.276354
<b>TS-Eox</b>	-879.534733	-879.213061	133.2	-879.276354
<b>TS-Emn</b>	-879.546866	-879.225429	133.5	-879.288846
<b>TS-Emx</b>	-879.541830	-879.220246	134.7	-879.284231
<i>trans</i> - <b>6</b>	-879.576122	-879.251621	130.9	-879.313822
<i>trans</i> - <b>7</b>	-879.582934	-879.258207	133.0	-879.321408
<i>trans</i> - <b>3</b>	-879.578279	-879.253805	132.2	-879.316597
<i>trans</i> - <b>5</b>	-879.583564	-879.258835	133.3	-879.322156

**Table S5.** HF/6-311G(d,p) total, E in a.u., and relative,  $\Delta E$  in kcal·mol<sup>-1</sup>, energies of **TS-Zmx** and **TS-Emn**, computed in benzene.

	E	$\Delta E$
(E)- <b>1</b>	-511.184701	
(Z)- <b>1</b>	-511.173066	
<b>1</b>	-362.937107	
<b>TS-Zmx</b>	-874.078897	19.6
<b>TS-Emn</b>	-874.084868	23.2

**Table S6.** M06-2X/6-311G(d,p) total, E in a.u., and relative,  $\Delta E$  in kcal·mol<sup>-1</sup>, energies of **TS-Zmn** and **TS-Zmx**, computed in benzene.

	E	$\Delta E$
<b>TS-Zmn</b>	-879.167674	
<b>TS-Zmx</b>	-879.164602	1.9