

Supporting informations

Bimetallic Cobalt-Iron Diselenide Nanorods Modified Glassy Carbon Electrode: An Electrochemical Sensing Platform for the Selective Detection of Isoniazid

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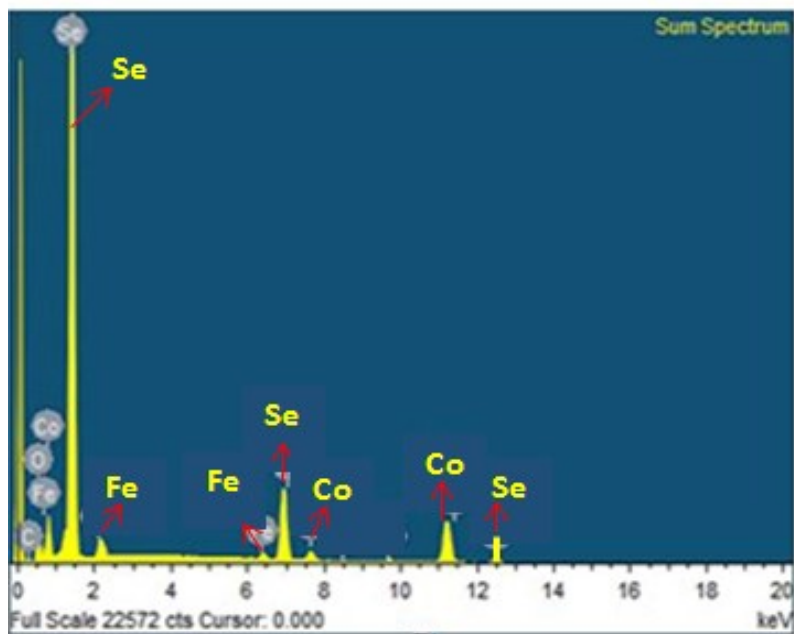


Figure S1. EDX profile of CoFeSe_2 nanocomposite.

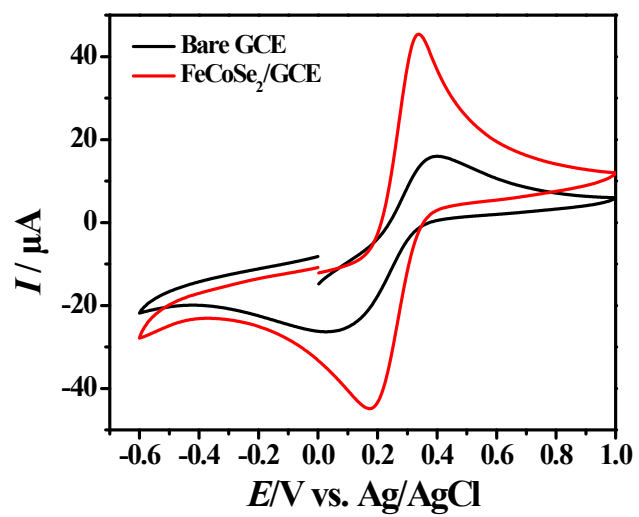


Fig. S2. Cyclic voltammograms for bare and FeCoSe₂/GCE in 5 mM solution of K₃[Fe(CN)₆] at the scan rate of 50 mV/s.

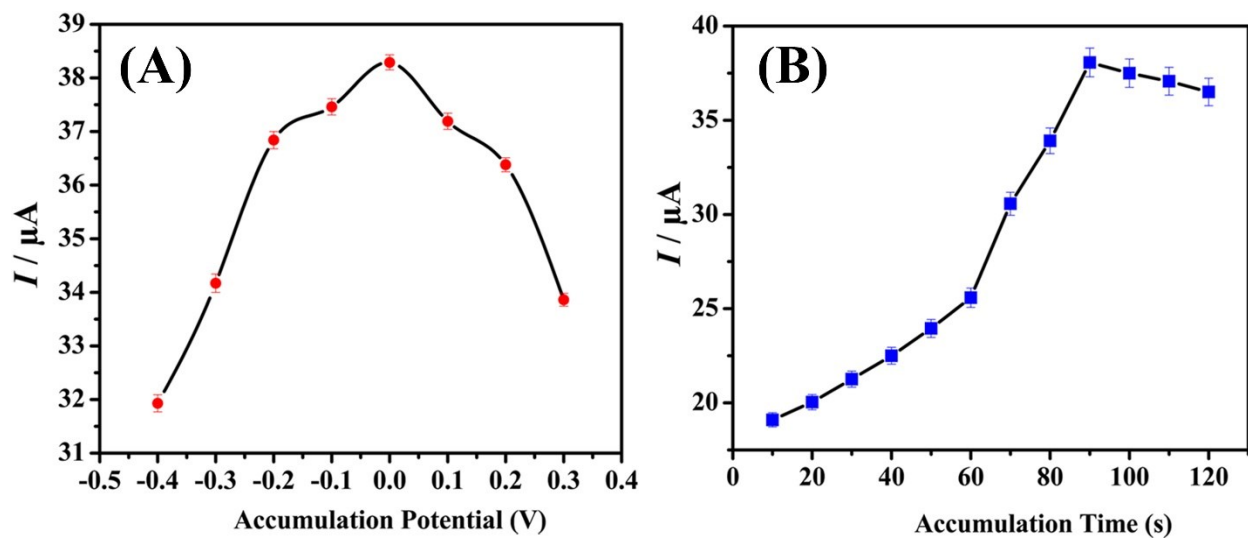


Fig. S3. (A) The effect of accumulation potential on oxidation peak current of INZ keeping accumulation time constant (90 s) and (B) the effect of accumulation time on oxidation peak current keeping accumulation potential constant (0.0 V) for FeCoSe₂/GCE in 18 μ M solution of INZ by applying SWASV technique.

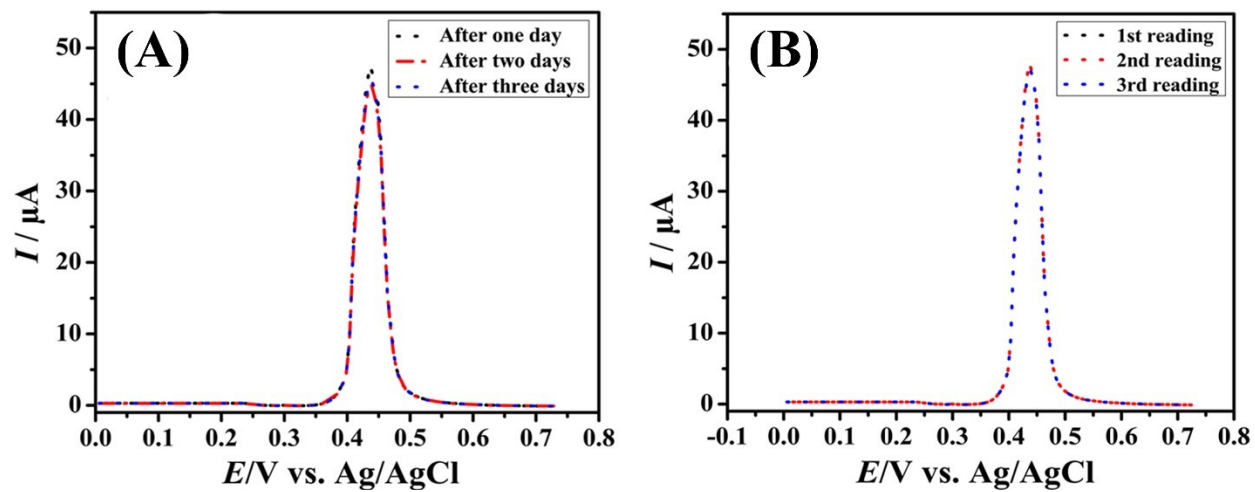


Fig. S4. Study of reproducibility and stability of designed sensor (A) Intra-day precision with RSD of 2.22%. (B) Inter-day precision with RSD of 0.56%.

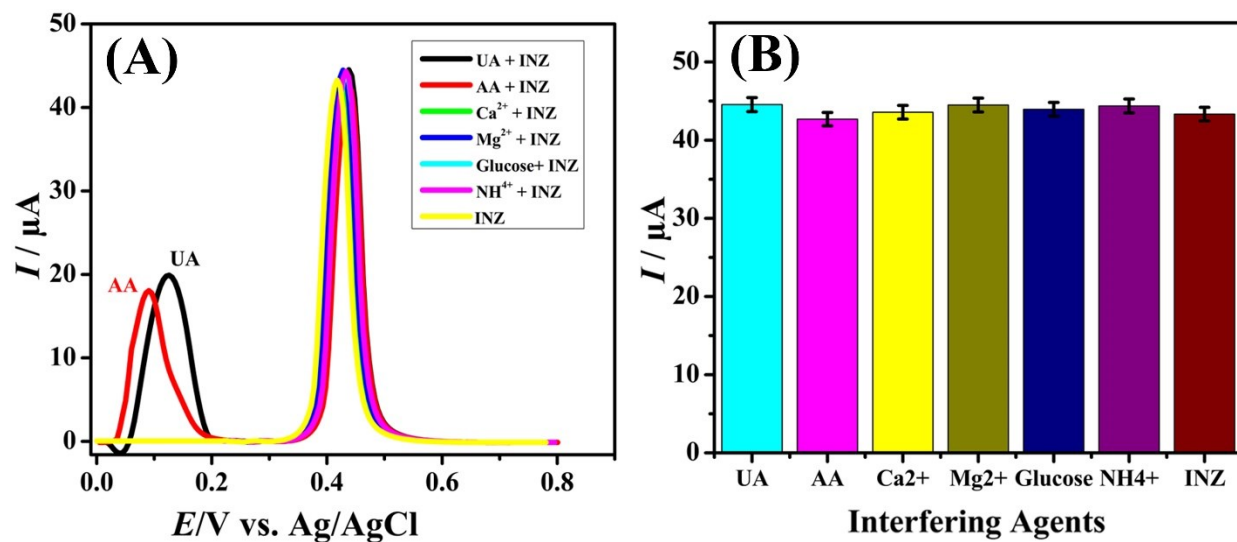


Fig. S5. (A) SWAS voltammograms of 25 μM solution of INZ for $\text{FeCoSe}_2/\text{GCE}$ in the presence of various co-existing interfering species using BRB (pH 7) as a supporting electrolyte and (B) Bar graph showing the effect of uric acid, ascorbic acid, Ca^{2+} , Mg^{2+} , glucose, NH_4^+ , and pure INZ solution.

Table S1. The regression Data and Various Parameters Obtained from Calibration Curves.

Parameters	Measured Values
Measured potential (mV)	433
Linearity range (μM)	0.03-1
Correlation coefficient (R^2)	0.998
LOD (M)	1.24×10^{-10}
LOQ (M)	4.14×10^{-10}