Supporting Information

Conjugated microporous polymer foams with excellent thermal insulation performance in humid environment

Experiment materials

1,1,2,2-tetrakis(4-bromophenyl) ethene (Kylpharm Co. Ltd), 1,3,6,8-tetrabromo-pyrene (Kylpharm Co. Ltd), 1,4-diethynylbenzene (Kylpharm Co. Ltd), 1,3,5-triethynyl-benzene (Kylpharm Co. Ltd), copper(I) iodide, tetrakis (triphenylphosphine) palladium, toluene, triethylamine.

Characterization

The structure of CMP foams was characterized by Solid-state ¹³C-CP/MAS NMR spectroscopy and Fourier Transform Infrared Spectroscopy (FTIR). Nitrogen isotherms were measured at 77K by Autosorb-iQ automatic specific surface and pore size distribution analyzer, Quanta-chrome. Powder X-ray diffraction (PXRD) was investigated on a D-MAX X-ray diffractometer (Rigaku, Japan). Scanning electron microscopy (SEM) images were operated on Apreo S instrument, (FEI American). Transmission electron microscopy (TEM) images were performed on TECNAI G² TF20 at an accelerating voltage of 200 kV, (FEI American). The powder was dispersed in EtOH solution and a drop mixture was dropped onto the copper network. TGA curves were obtained under argon protection by STA 449 F5 (NETZSCH), and the temperature was elevated from 50°C to 800°C with a heating rate of 10°C min⁻¹. DSA100 was used in measuring the water contact angle, (KRUSS, Germany).

Test of thermal conductivity

Multifunction-rapid thermal conductivity tester (DRE-III, Xiang-yi) was used in measurement of CMP foams under different humidity conditions(50%/70%). Reference standard is ISO22007-2.



Figure S1 The image of CMP-PT foam



Figure S2 Powder X-ray diffraction patterns of CMP-ED and CMP-PT



Figure S3 Nitrogen adsorption and desorption isotherms (a), and distribution of pore size (b) of SiO₂ aerogel.



Figure S4 Water contact angle of SiO₂ aerogel

Samples	The ratio of content of H:	The ratio of content of Br:	
	C (mol: mol)	C (mol: mol)	
CMP-ED	69.13:100	1.73:100	
CMP-PT	61.30:100	6.49:100	
CMP-ED after TGA	11.07:100	0	
CMP-PT after TGA	9.1:100	0	

Table S1 The elemental an	alysis of CMP-ED and CMP-PT
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Figure S5 the FTIR spectra of CMP-PT and CMP-ED before and after TGA



Figure S6 Thermographic images of SiO₂ aerogel, CMP-ED and CMP-PT at 298.15 K

43.1°C ⊡100.7°C	e=1.00	47.7°C	e=1.00 (III) 98.3°C[]	43.0°C ⊡101.3°C	e=1.00
41.8°C55	-‡43.1°C	€;43.9°C -E	47.7°C		- 43.0°C 40.7°C€3
MAX:100.7°C1 87.2°C	e=1.00	6 MAX:98.3°C MIN 79.8°C	e=1.00 20.7°C 3	MAX:101.3°C 79.8°C	MIN:40.7°C 09:53 e=1.00 = 21.2°C@
	21.8°C	€::89.6°¢t		5	85.4°C53.e
MAX:89.5°C	MIN:21.8°C 15:2	7 MAX:89.6°C MIN	1:20.7°C 15:28	MAX:85.4°C	MIN:21.2°C 09:54
SiO ₂	aerogel	CMI	P-ED	CN	AP-PT

Figure S7 Thermographic images of SiO₂ aerogel, CMP-ED and CMP-PT at 398.15 K

Table S2 Thermal conductivity datas of CMP-ED, CMP-PT and commercial SiO $_2$ gel

Test environmental	Therma	(W m ⁻¹ K ⁻¹)	
humidity	CMP-ED	CMP-PT	commercial SiO ₂ gel
50%	0.03404	0.03565	0.03409
70%	0.03408	0.03705	0.03655
Value changed (%)	0.12	3.93	7.22



Figure S8 Comparison of thermal conductivity between CMP foams and other polymers ¹⁻³

Reference

- J. Li, A. Zhang, S. Zhang, Q. Gao, W. Zhang and J. Li, *Composites Part B: Engineering*, 2019, 156, 368-377.
- 2. L. Wang, Y.-K. Wu, F.-F. Ai, J. Fan, Z.-P. Xia and Y. Liu, *Polymers*, 2018, 10, 1310.
- 3. P. Mu, W. Bai, Z. Zhang, J. He, H. Sun, Z. Zhu, W. Liang and A. Li, *Journal of Materials Chemistry A*, 2018, **6**, 18183-18190.