

## Supplementary Information

### **Low temperature scalable synthetic approach enabling high bifunctional electrocatalytic performance of $\text{NiCo}_2\text{S}_4$ and $\text{CuCo}_2\text{S}_4$ thiospinels**

Ginena Bildard Shombe,<sup>1,2</sup> Shumaila Razzaque,<sup>3</sup> Malik Dilshad Khan,<sup>1,4\*</sup> Tebello Nyokong,<sup>5</sup> Philani Mashazi,<sup>5,6</sup> Jonghyun Choi,<sup>7</sup> Sanket Bhoyate,<sup>7</sup> Ram K. Gupta,<sup>7</sup> and Neerish Revaprasadu<sup>1\*</sup>

<sup>1</sup>Department of Chemistry, University of Zululand, Private Bag X1001, KwaDlangezwa 3880, South Africa.

<sup>2</sup>Chemistry Department, University of Dar-es-Salaam, P.O. Box 35061, Dar-es-Salaam, Tanzania.

<sup>3</sup>Key Laboratory of Material Chemistry for Energy Conversion and Storage, Ministry of Education, Hubei Key Laboratory of Material Chemistry and Service Failure, School of Chemistry and Chemical Engineering, Huazhong University of Science and Technology, Luoyu Road No. 1037, Wuhan, China.

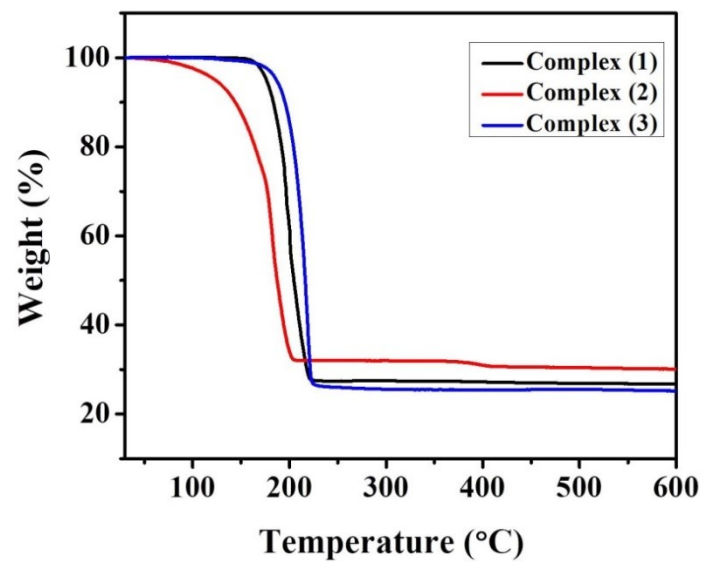
<sup>4</sup>Institute of Physical Chemistry, Polish Academy of Sciences, Kasprzaka 44/52, 01-224 Warsaw, Poland.

<sup>5</sup>Institute for Nanotechnology Innovation, P.O. Box 94, Rhodes University, Makhanda 6140, South Africa.

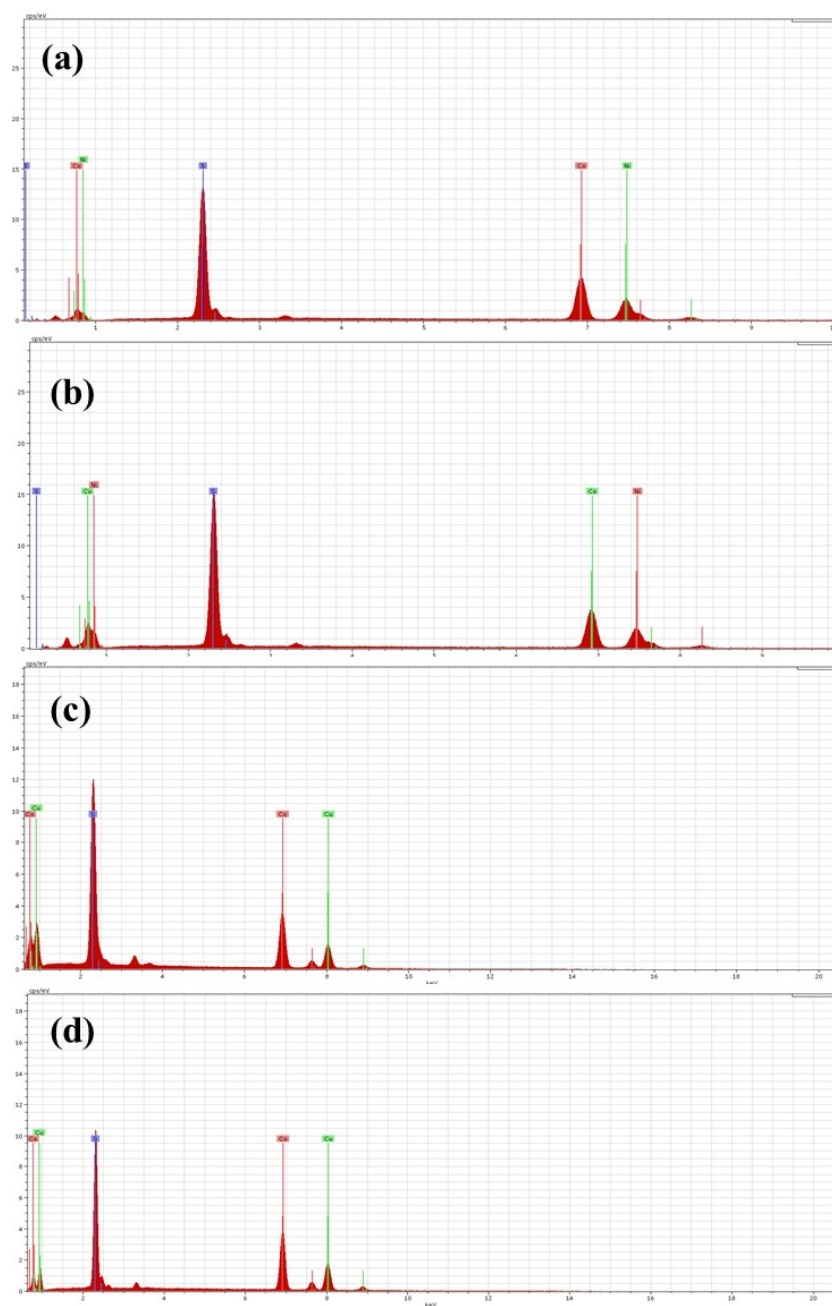
<sup>6</sup>Department of Chemistry, P. O. Box 94, Rhodes University, Makhanda 6140, South Africa.

<sup>7</sup>Department of Chemistry, Pittsburg State University, Pittsburg, KS 66762, USA.

\*Email: [RevaprasaduN@unizulu.ac.za](mailto:RevaprasaduN@unizulu.ac.za); [malikdilshad@hotmail.com](mailto:malikdilshad@hotmail.com)



**Figure S1.** TGA curves of complexes nickel ethyl xanthate (**1**), copper ethyl xanthate (**2**), and cobalt ethyl xanthate (**3**).



**Figure S2.** EDX spectra of NiCo<sub>2</sub>S<sub>4</sub> synthesized at (a) 200 °C and (b) 300 °C. EDX spectra of CuCo<sub>2</sub>S<sub>4</sub> synthesized at (c) 200 °C and (d) 300 °C

**Table S1.** Comparison chart with previously reported supercapacitor electrodes.

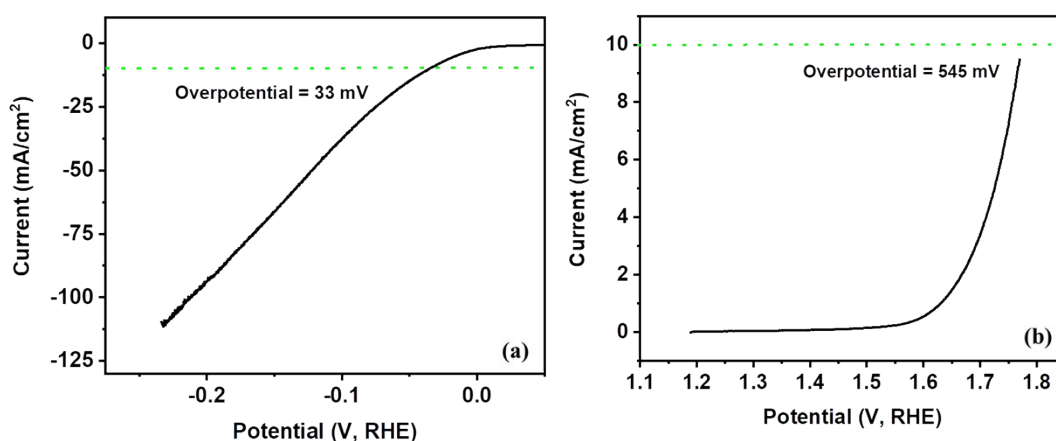
Electrode Material	Synthetic Method	Specific Capacitance (F/g)	Current Density (A/g)	Reference
NiCo <sub>2</sub> S <sub>4</sub>	Sacrificial template method	437	1	1
NiCo <sub>2</sub> S <sub>4</sub>	Solvothermal	519.51	2	2
NiCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	744	1	3
NiCo <sub>2</sub> S <sub>4</sub> @MnO <sub>2</sub>	Hydrothermal	520.7	1	4
AMC@ NiCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	651.1	0.6	5
CuCo <sub>2</sub> S <sub>4</sub> /polyacrylonitrile		385	1	6
CuCo <sub>2</sub> S <sub>4</sub> @NiCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	539.2 C/g	1	7
CuCo <sub>2</sub> S <sub>4</sub>	Hydrothermal and sulfuration process	373.4	1	8
CuCo <sub>2</sub> S <sub>4</sub>	Anion-exchange and post-annealing	424	1	9
CuCo <sub>2</sub> S <sub>4</sub> /Graphene	Hydrothermal	525.4	1	10
CuCo <sub>2</sub> S <sub>4</sub>	Two-step hydrothermal approach	~300	0.17 (mA/cm <sup>2</sup> )	11
CuCo <sub>2</sub> O <sub>4</sub>	Hydrothermal	285.5	0.5	12
Co <sub>3</sub> O <sub>4</sub>	Hydrothermal	272.86	0.5	13
NiCo <sub>2</sub> O <sub>4</sub>	Hydrothermal	225	1	14
CoFe <sub>2</sub> O <sub>4</sub>	Electrodeposition	768	0.5	15
CuCo <sub>2</sub> S <sub>4</sub>	hydrothermal and sulfuration process	373.4	1	8
AgBiS <sub>2</sub>	Melt method	175	0.5	16
<b>CuCo<sub>2</sub>S<sub>4</sub> (CCS-300)</b>	<b>Solventless approach</b>	<b>475</b>	<b>0.5</b>	<b>This work</b>
<b>NiCo<sub>2</sub>S<sub>4</sub> (NCS-300)</b>	<b>Solventless approach</b>	<b>1200</b>	<b>0.5</b>	

\*AMC-Activated multiporous carbon

**Table S2.** Comparison chart with previously reported electrocatalyst electrodes.

Catalyst Materials	Synthetic Method	$\eta_{\text{HER}}$ (mV) at 10 mA/cm <sup>2</sup>	$\eta_{\text{OER}}$ (mV) at 10 mA/cm <sup>2</sup>	Reference
CuCo <sub>2</sub> S <sub>4</sub> /NiCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	206	271	17
CuCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	-	310	18
CuCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	158	290 (20 mA/cm <sup>2</sup> )	12
NiCo <sub>2</sub> O <sub>4</sub>	Hydrothermal	-	346	19
NiCo <sub>2</sub> S <sub>4</sub>	Solvothermal	226	-	20
NiCo <sub>2</sub> S <sub>4</sub>	Hydrothermal followed by sulfidation	-	309	19
NiCo <sub>2</sub> O <sub>4</sub> /NF	Solvothermal	~250	~350	21

NiCo <sub>2</sub> Se <sub>4</sub>	Hydrothermal		270	19
NiCo <sub>2</sub> S <sub>4</sub> /NF	Solvothermal	~200	~330	21
Ni <sub>3</sub> S <sub>2</sub> /NF	Hydrothermal and thermal sulfurization process	271	-	22
NiCo-LDH/NF	Hydrothermal	231	-	22
NiCo <sub>2</sub> S <sub>4</sub> /NF	Hydrothermal	210	260	23
NiCo <sub>2</sub> S <sub>4</sub> /NF	Two-step hydrothermal method	191	<251 (40 mA/cm <sup>2</sup> )	24
NiCo <sub>2</sub> S <sub>4</sub> @N/S-rGO	Solvothermal	-	470	25
NiCo <sub>2</sub> O <sub>4</sub> NS/CC	Hydrothermal	-	368	26
NiCo <sub>2</sub> S <sub>4</sub> NS/CC	Hydrothermal	-	316	26
CuCo <sub>2</sub> S <sub>4</sub>	Solution-based chemical route	-	395	27
FeO@CuCo <sub>2</sub> S <sub>4</sub>	Hydrothermal	-	240	28
<b>CCS-300</b>	<b>Solventless approach</b>	<b>224</b>	<b>269</b>	<b>This Work</b>
<b>NCS-300</b>	<b>Solventless approach</b>	<b>209</b>	<b>318</b>	



**Figure S3.** Electrocatalytic activity of commercial platinum for HER (a) and OER (b) processes.

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