

Supplementary Information

Structure, phase transition and properties of the one-dimensional antiferromagnet Cu(2,6-dimethylpyrazine)Br₂

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1. Thermogravimetric analysis (TGA) data

To verify the constitution and purity of **1**, we have performed thermogravimetric Analysis (TGA) was performed by NETZSCH STA 409 PC/PG instrument in N₂ flow, the heating rate is 10 °C/min. The result is shown in Fig. S1. The first step corresponds to the decomposition of 2,6-dimethylpyrazine organic ligand (calculated: 32.62%, found: 34.51%)

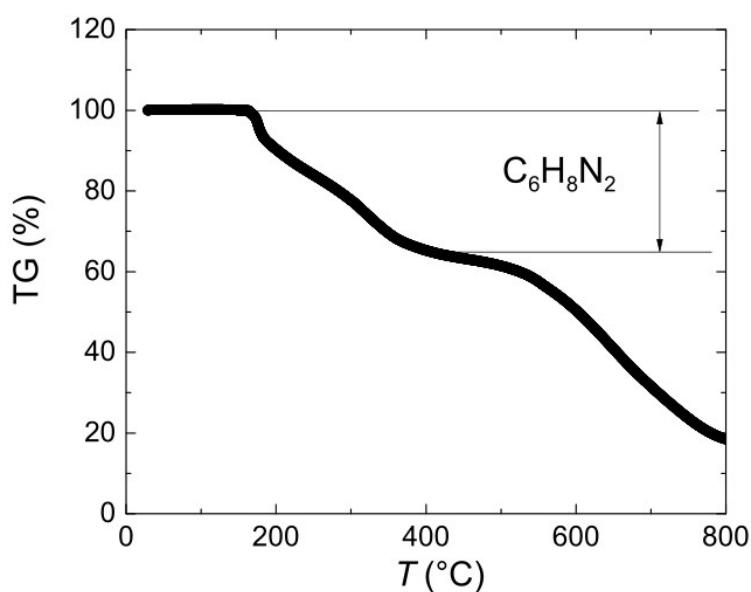


Figure S1. TGA data of **1**. The first step corresponds to the decomposition of 2,6-dimethylpyrazine organic ligand.

2. IR and Raman measurements

We performed both infrared (IR) and Raman spectroscopy measurements and the results are presented in Fig. S2. IR spectroscopy was performed using a Magna 550 FT-IR spectrometer (Nicolet Instruments, France) in the range from 400 to 2000 cm⁻¹.

Raman spectroscopy was measured by HORIBA LabRAM HR Evolution. The positions of characteristic peaks are listed in Table S1. The spectra are dominated by the vibrational mode of 2,6-dimethylpyrazine molecules. By comparison with the results of density functional theory (DFT) calculations for 2,6-dimethylpyrazine molecules, we assigned the observed peaks to the corresponding vibrations (Table S1).

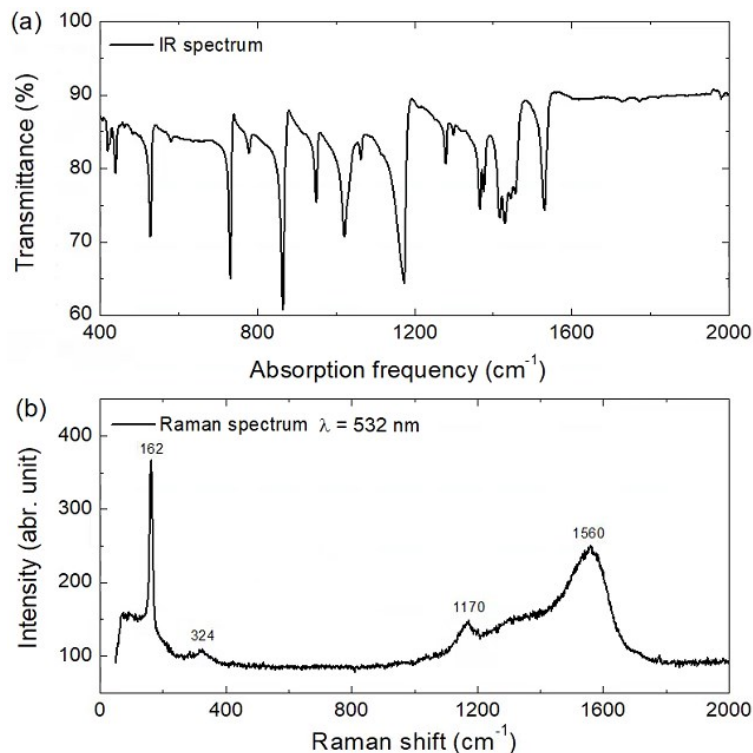


Figure S2. (a) IR and (b) Raman spectroscopy of **1**. The incident photon wavelength for Raman measurement is 532 nm. Four characteristic Raman shift peaks appear at 162, 324, 1170, and 1560 cm⁻¹.

Table S1 Observed and calculated frequencies (cm⁻¹).

IR	Raman	Assignments*
	162	Ring torsion
	324	Ring torsion
419		C-CH ₃ bending
439		Ring torsion
528		C-CH ₃ wagging
731		C-CH ₃ wagging
865		C-H wagging
950		C-H wagging
1021		CH ₃ rocking
1173	1170	C-H bending
1280		Ring stretching
1366		CH ₃ bending symmetric
1430		CH ₃ bending asymmetric
1530		Ring stretching

* From the DFT calculations by A. Pawlukoć et al. Journal of Molecular Structure, 2008, **892**, 261–267