

## Electronic Supplementary Information (ESI)

# Shape-tuned surface-active and support-free silver oxygen reduction electrocatalyst enabled high performance fully non-PGM alkaline fuel cell

P. Anandha ganesh<sup>\*a</sup>, AN. Prakrthi<sup>b</sup>, S. Selva Chandrasekaran<sup>c</sup>, D. Jeyakumar<sup>d</sup>

<sup>a</sup>State Key Laboratory for Mechanical Behavior of Materials,  
School of materials science and engineering, Xi'an Jiaotong University,  
No. 28, Xianning West Road, Xi'an, 710049 Shaanxi, China.

<sup>b</sup>Centre for Nanoscience and Engineering,  
Indian Institute of Science,  
Bangalore-560012, Karnataka, India.

<sup>c</sup>Laboratoire de simulation atomistique,  
INAC, CEA Grenoble  
17, Avenue des Martyrs  
38000 Grenoble, FRANCE.

<sup>d</sup>Fuel cell catalysis and nano-materials group,  
Functional Materials Division,  
CSIR-Central Electrochemical Research Institute,  
Karaikudi - 630006, Tamil Nadu, India.

\*Corresponding author. Email: nanoganesh@xjtu.edu.cn.

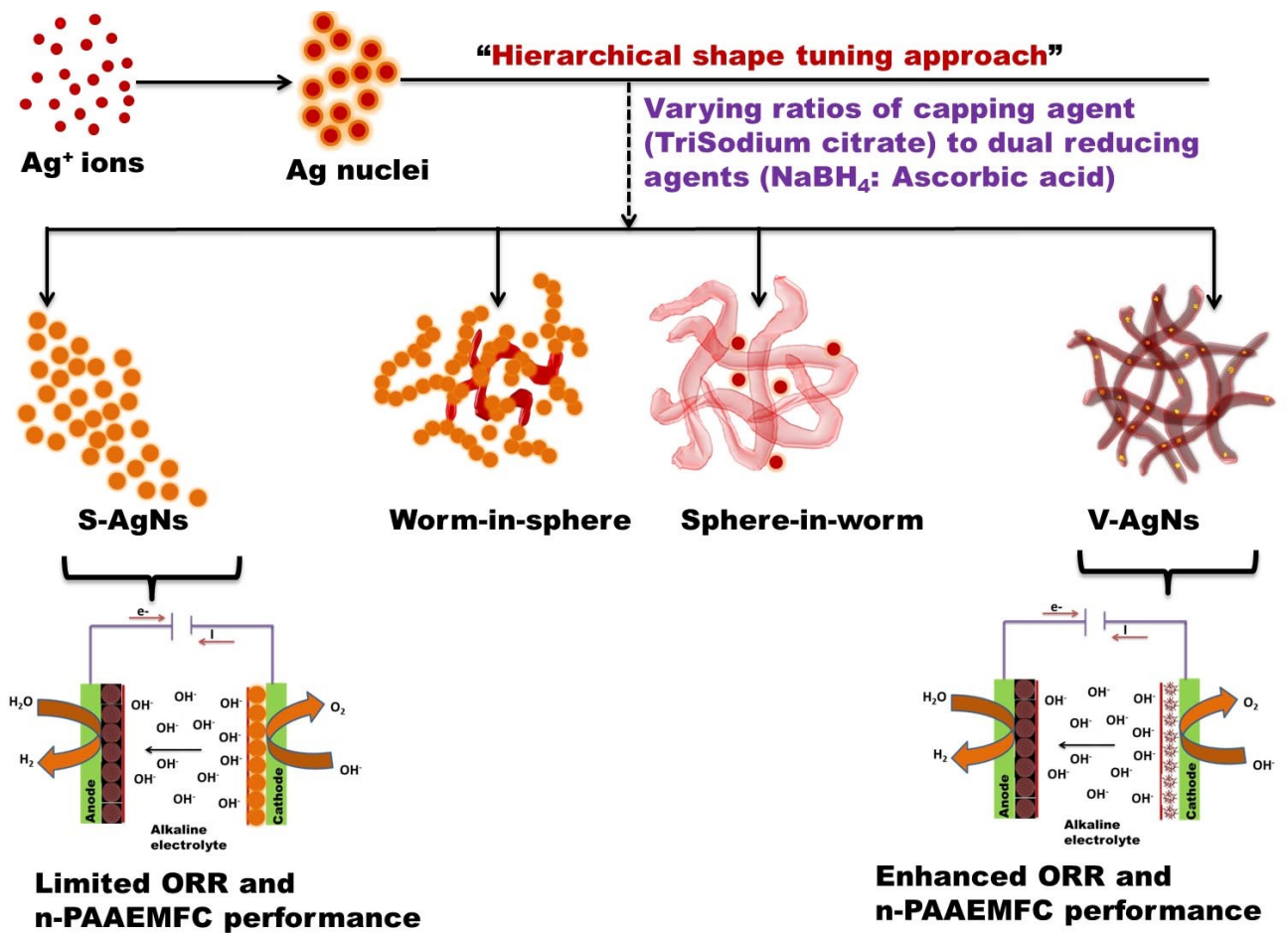
**Materials:** High purity analar grade silver nitrate ( $\text{AgNO}_3$ ) and potassium hydroxide (KOH) pellets were purchased from SISCO Research Laboratories Pvt. Ltd., India. High purity analar grade trisodium citrate decahydrate, sodium borohydride, ascorbic acid, ethanol and isopropyl alcohol were procured from Sigma-Aldrich Pvt. Ltd., India. Commercial carbon supported nickel nanoparticles (Ni/C, Premetek, Ni 80%, ~ 30 nm) was used as anode catalyst. Fumion FAA-3-SOLUT-10 ionomer and FAA-3 anion exchange membranes (70  $\mu\text{m}$  thick, 17  $\text{mScm}^{-1}$  anion (hydroxide) conductivity at 20  $^\circ\text{C}$ ) were from Fumatech. Porous carbon gas diffusion layer (GDL) sheets were purchased from SAI Energy Solutions Pvt. Ltd, India. Hydrogen (99.9%) ( $\text{H}_2$ ), oxygen (99.9%) ( $\text{O}_2$ ) and nitrogen (99.9%) ( $\text{N}_2$ ) were supplied by Indian oxygen limited, India. All glassware was meticulously cleaned using aqua regia and finally rinsed with copious amount of water. Double distilled water (Millipore) was used for all preparative and characterization works.

<b>Silver nitrate (mmol)</b>	<b>Trisodium citrate decahydrate (mmol)</b>	<b>Sodium borohydride (mmol)</b>	<b>Ascorbic acid (mmol)</b>	<b>Shapes attained</b>
8	0.75	0.15	0.1	spheres
“	0.375	0.475	0.15	spheres
“	0.6	0.235	0.165	spheres
“	0.3	0.525	0.175	spheres
“	0.5	0.3	0.2	worm-in-sphere
“	0.25	0.4	0.35	worm-in-sphere
“	0.25	0.35	0.4	sphere-in-worm
“	0.125	0.45	0.425	sphere-in-worm
	0.15	0.435	0.415	worms
“	0.1	0.5	0.4	worms
“	0.05	0.5	0.45	worms

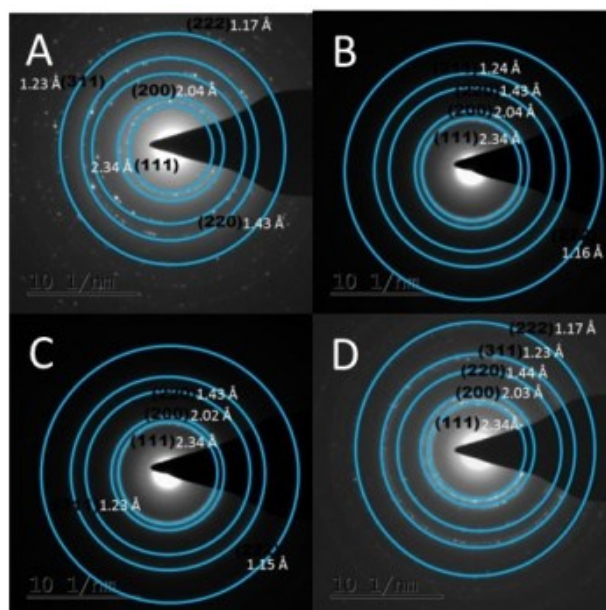
**Table S1:** Details of “hierarchical shape tuning approach” showing the precisely varying ratio of capping agent to dual reducing agents amount and the tuneable shapes obtained from this new approach.

Anode catalyst layer	Cathode catalyst layer	Membrane	Non-PGM loading in MEA	Cell conditions	Power density (mW cm <sup>-2</sup> )	Power density (mWmg <sup>-1</sup> <sub>non-PGM</sub> cm <sup>-2</sup> )	Fuel cell stability test	Year and reference
Cr/Ni (5 mg cm <sup>-2</sup> )	Ag (1 mg cm <sup>-2</sup> )	Quaternary ammonium polysulphone	6 mg cm <sup>-2</sup>	60 °C, H <sub>2</sub> /O <sub>2</sub> (RH = 100%)	50	8.3	NA	2008, 48
Ni/C (5 mg cm <sup>-2</sup> )	Ag (0.5 mg cm <sup>-2</sup> )	TPQPOH152	5.5 mg cm <sup>-2</sup>	70 °C, 200 sccm H <sub>2</sub> /O <sub>2</sub>	76	13.8	NA	2013, 47
NiW (17.5 mg cm <sup>-2</sup> )	Coppy/C (2 mg cm <sup>-2</sup> )	xQAPS	19.5 mg cm <sup>-2</sup>	60 °C, 50 sccm H <sub>2</sub> /O <sub>2</sub> (RH = 100%)	40	2.05	NA	2013, 49
NiW (17.5 mg cm <sup>-2</sup> )	Coppy/C (2 mg cm <sup>-2</sup> )	xQAPS	19.5 mg cm <sup>-2</sup>	60 °C, 50 sccm H <sub>2</sub> /air (RH = 100%)	27.5	1.4	NA	2013, 49
Ni (1.5 mg cm <sup>-2</sup> )	Ag (3mg cm <sup>-2</sup> )	AAEM	4.5 mg cm <sup>-2</sup>	73 °C, 200 sccm dry H <sub>2</sub> / 1000 sccm air	70	15.5	NA	2016, 46
Ni/C (1.25 mg cm <sup>-2</sup> )	V-AgNs (0.25 mg cm <sup>-2</sup> )	AAEM	1.5 mg cm <sup>-2</sup>	60 °C, 500 sccm H <sub>2</sub> / 200 sccm O <sub>2</sub> (RH = 100%)	115.6	75.3	~240 h	Present work
Ni/C (1.25 mg cm <sup>-2</sup> )	S-AgNs (0.25 mg cm <sup>-2</sup> )	AAEM	1.5 mg cm <sup>-2</sup>	60 °C, 500 sccm H <sub>2</sub> / 200 sccm O <sub>2</sub> (RH = 100%)	41.3	20	~136 h	Present work

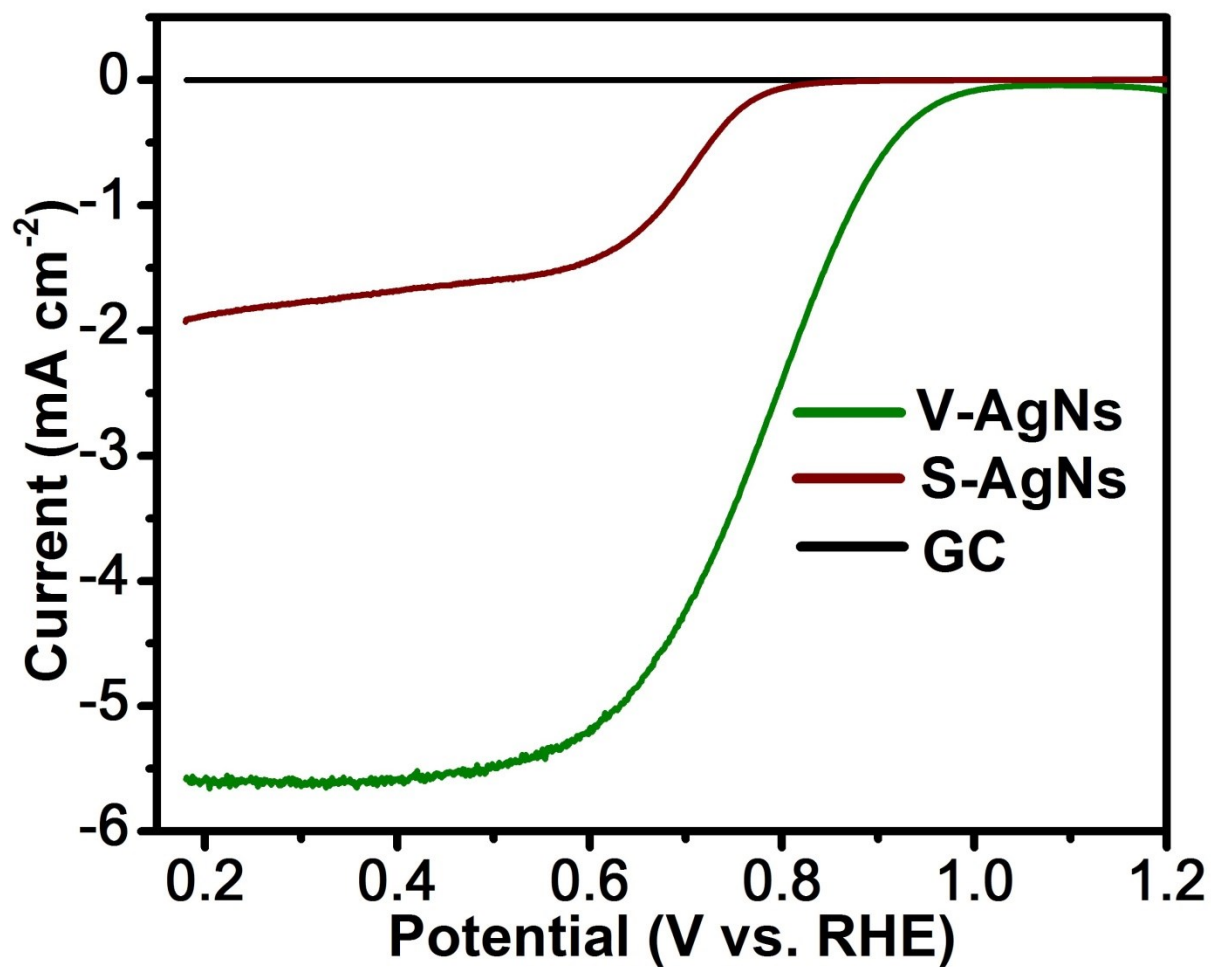
**Table S2:** Fully n-PAAEMFCs performance of the present work comprehensively compared with previous similar literature reports.



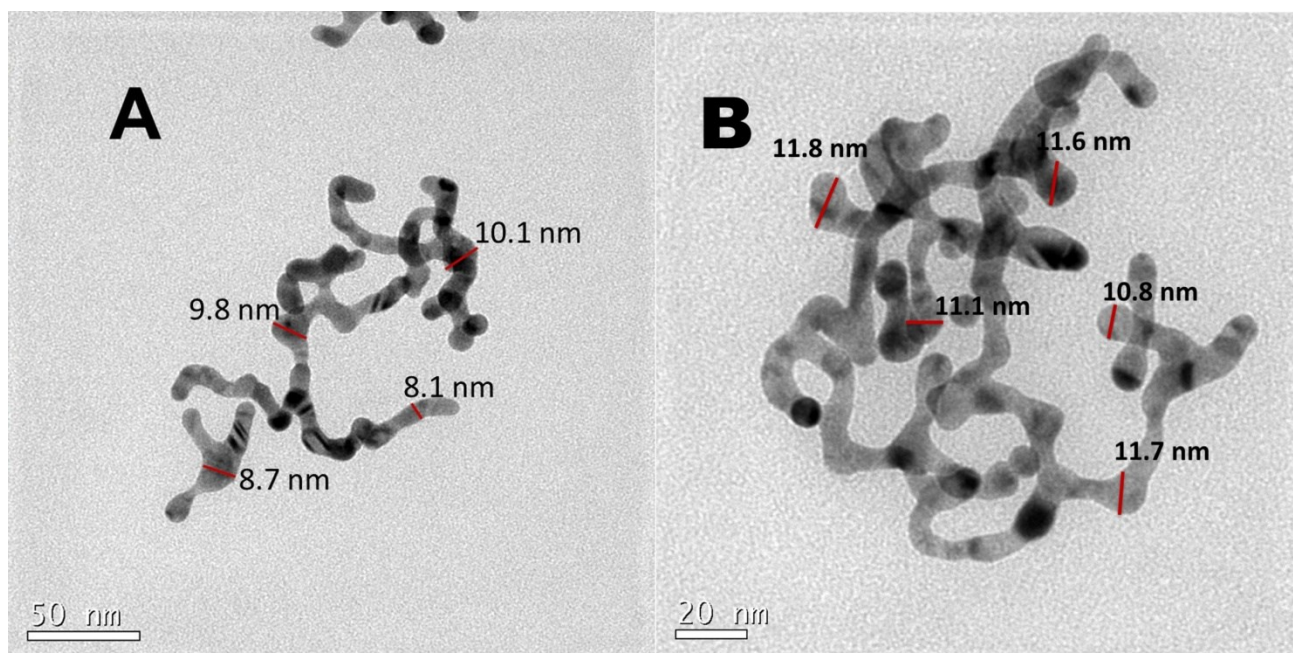
**Fig.S1:** Table of contents graph showing the hierarchical approach of AgNs synthesis and their application in fully non-PGM alkaline fuel cell.



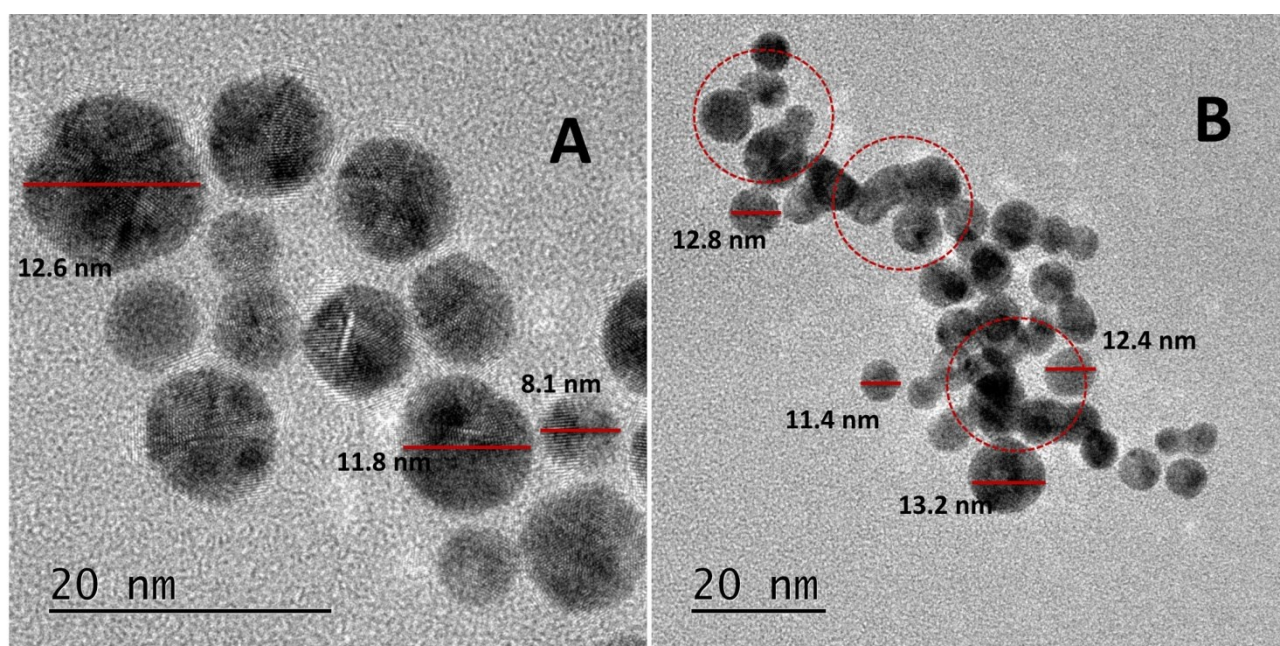
**Fig.S2:** TEM-SAED patterns of (A) S-AgNs, (B) worm-in-sphere, (C) sphere-in-worm, (D) V-AgNs showing their relevant crystallographic planes and d values.



**Fig.S3:** LSV-ORR performance comparison of GC, S-AgNs and V-AgNs electro-catalysts in O<sub>2</sub> saturated 0.5 M KOH recorded at 5 mV s<sup>-1</sup>, 1600 rpm.

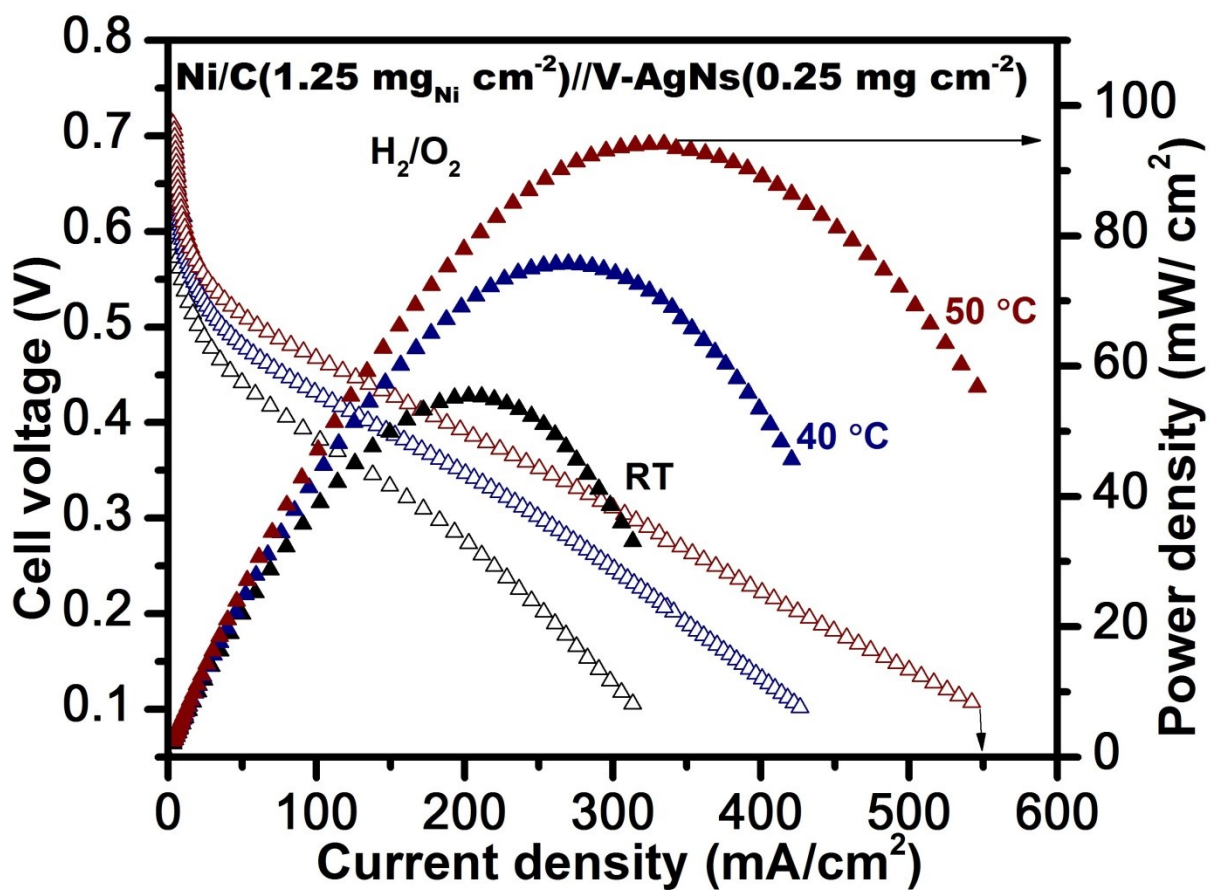


**Fig.S4:** TEM image of V-AgNs (A) before and (B) after AET studies showing the retention of its unique worm like morphology.

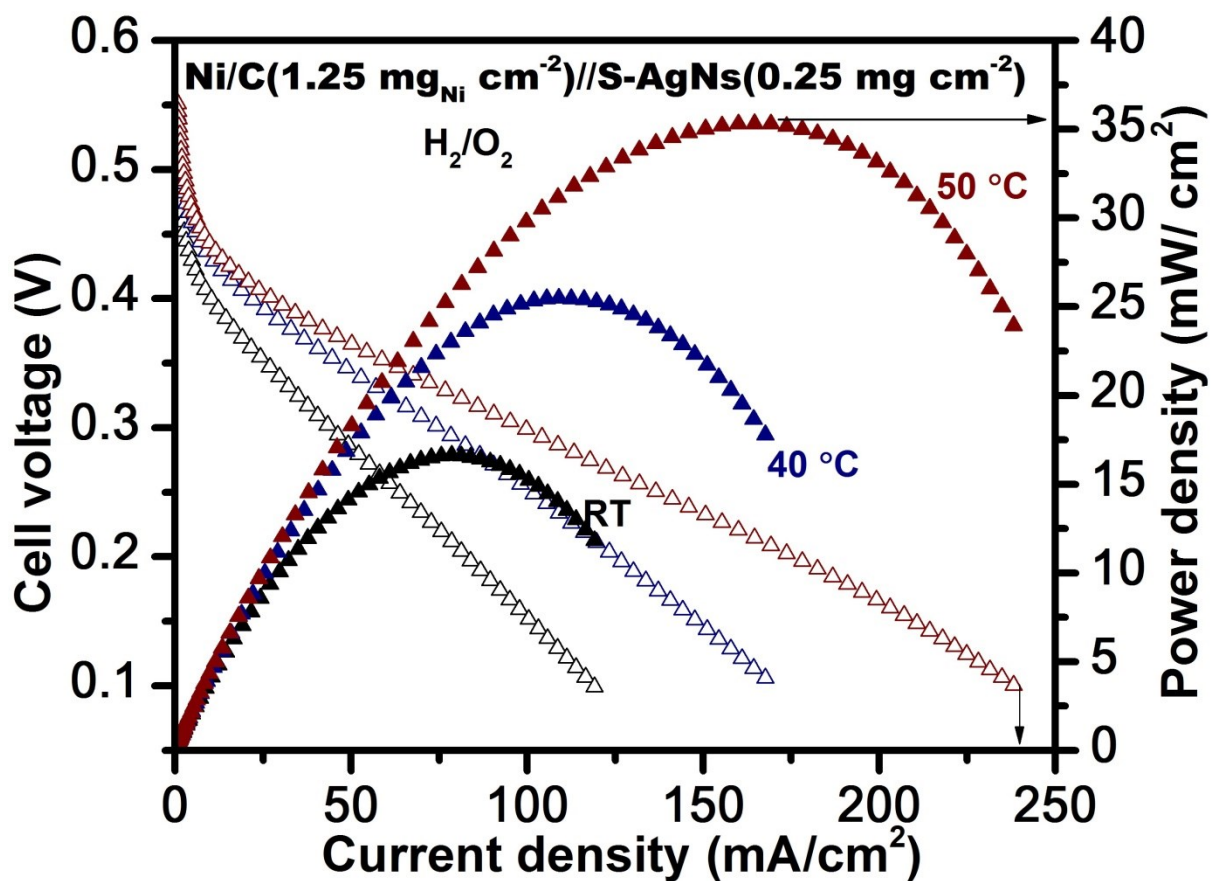


**Fig.S5:** TEM image of S-AgNs (A) before and (B) after AET studies (Red circles showing the agglomerated Ag nano-spheres after AET).





**Fig.S6:** Fully n-PAAEMFC performance at RT (30 °C), 40 °C and 50 °C between MEAs; anode catalyst layer: commercial NiC, cathode catalyst layer: V-AgNs.



**Fig.S7:** Fully n-PAAEMFC performance at RT (30 °C), 40 °C and 50 °C between MEAs; anode catalyst layer: commercial NiC, cathode catalyst layer: S-AgNs.