

Supplementary Information

Viscosity effect on strategic kinetic overgrowth of molecular crystals in various morphologies: concave and octapod fullerene crystals

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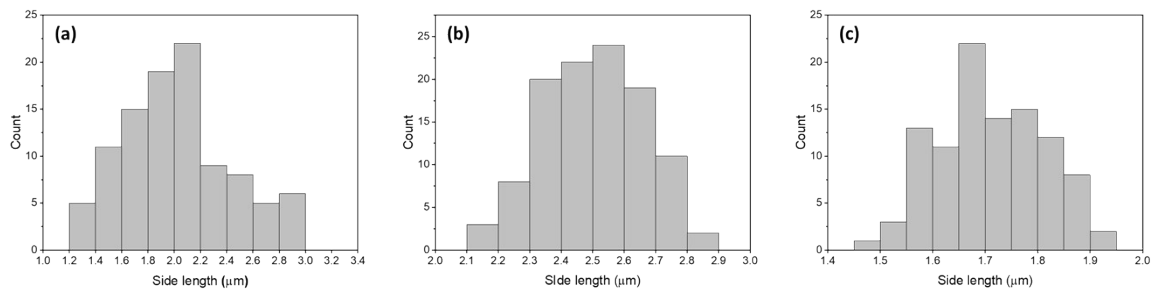


Fig. S1 Size distributions of C₇₀ (a) cubes, (b) concave cubes, and (c) octapods.

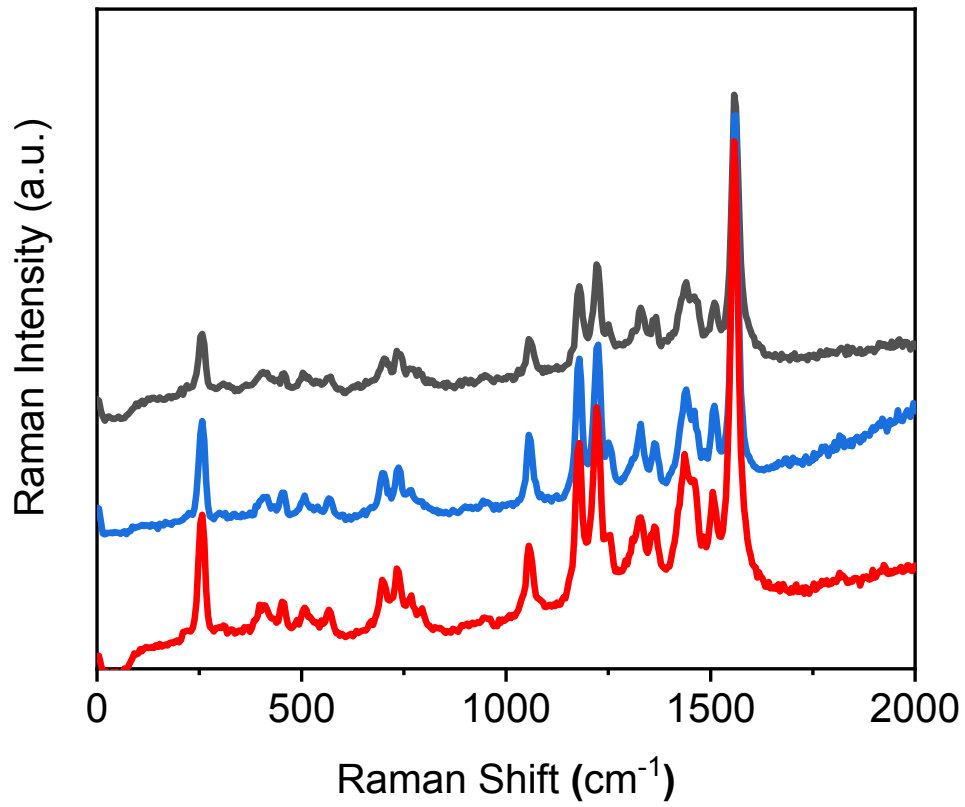


Fig. S2 Raman spectra of C₇₀ cubes (black), concave cubes (blue), and octapods (red).

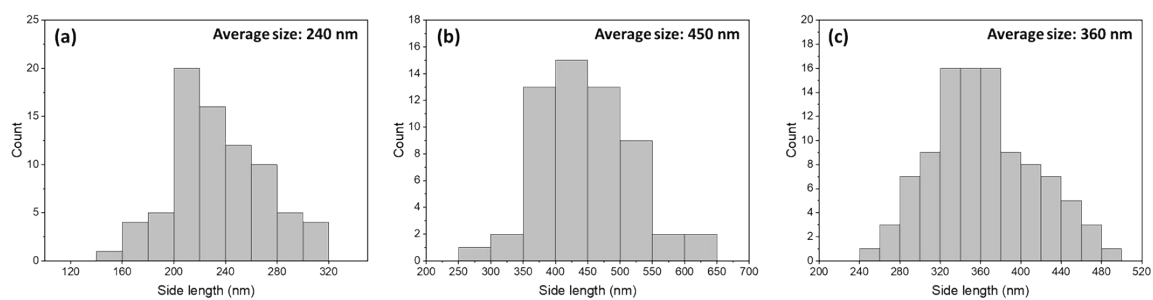


Fig. S3 Size distributions and the average sizes of C_{70} seed crystals obtained at (a) 25°C, (b) -16°C, and (c) -78°C.

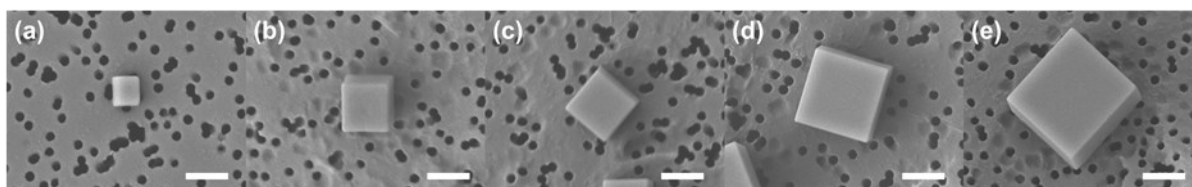


Fig. S4 Time-dependent SEM images of C_{70} cube crystals obtained at 25°C. Growth time for each image is (a) 0 min, (b) 1 min, (c) 5 min, (d) 30 min, and (e) 3 h, respectively. (Scale bar: 1 μm)

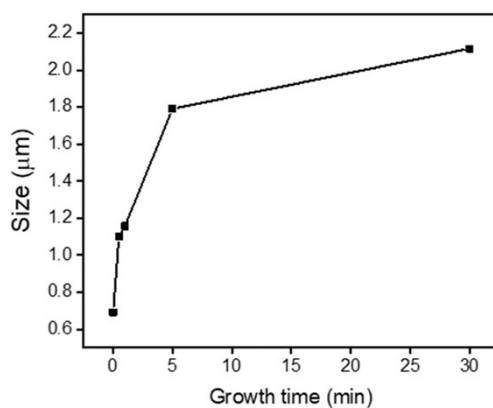


Fig. S5 The size change of C_{70} cube crystals formed at 25°C as a function of growth time.

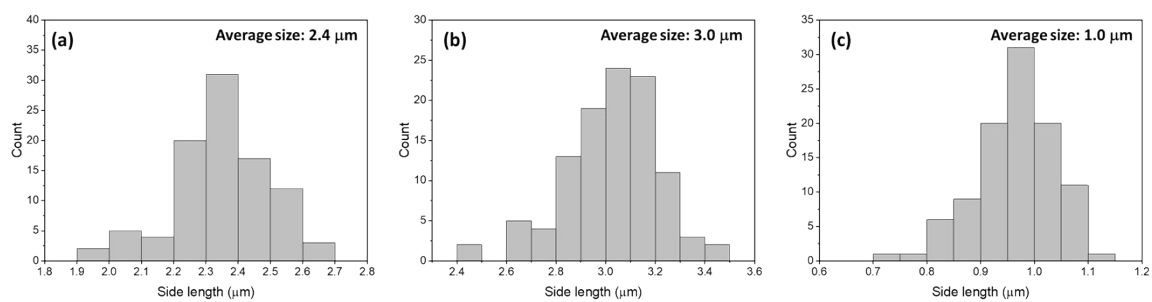


Fig. S6 Size distributions and the average sizes of C_{70} crystals obtained using acetone as antisolvent at (a) 25°C , (b) -16°C , and (c) -78°C .

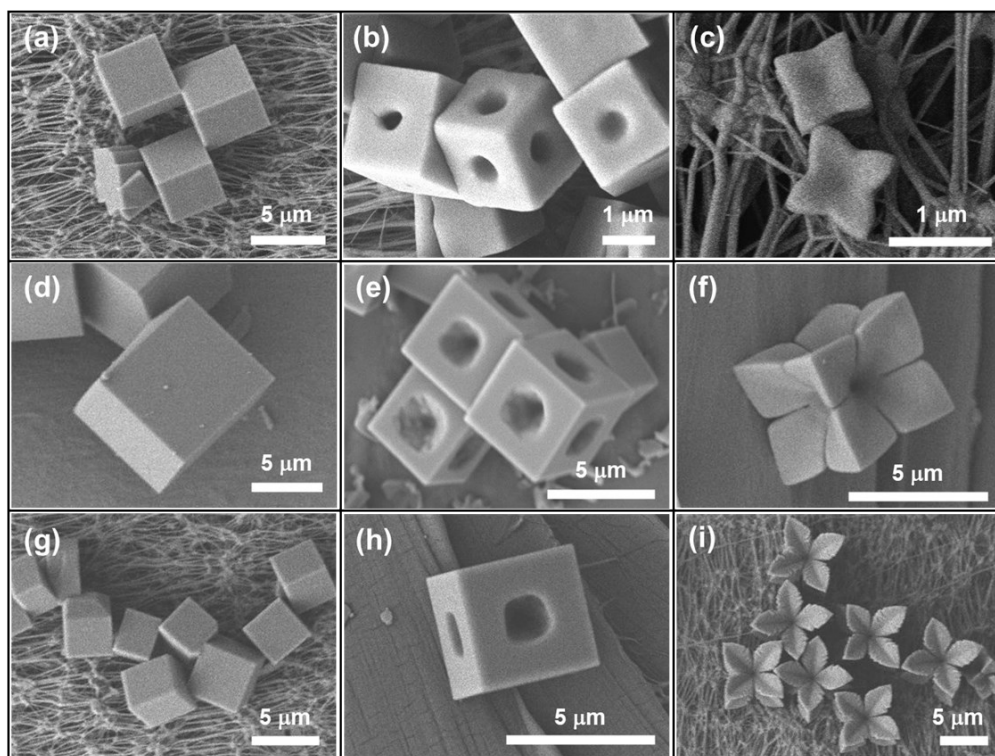


Fig. S7 Temperature-dependent morphology of C_{70} crystals using (a-c) ethanol, (d-f) 1-propanol, and (g-i) 1-butanol as antisolvent. ASC was performed at 25°C (left), -16°C (middle), and -78°C (right), respectively.

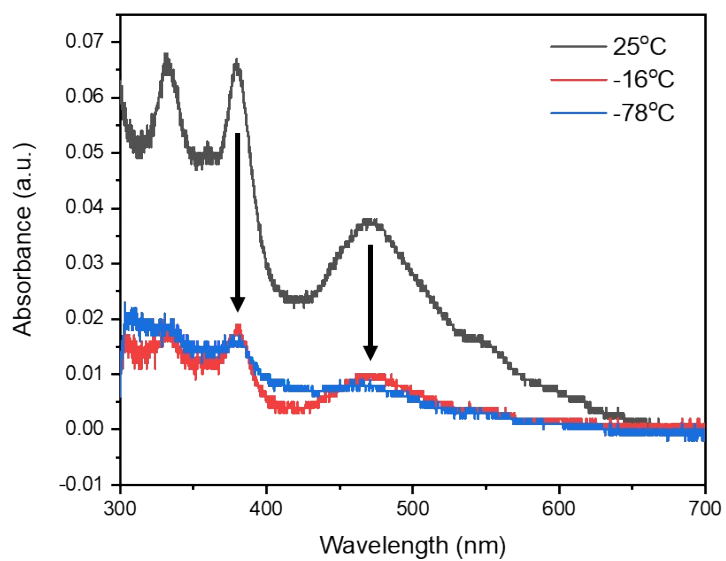


Fig. S8 UV-VIS absorption spectra of saturated crystallization solutions at 25°C (black), -16°C (red) and -78°C (blue).