

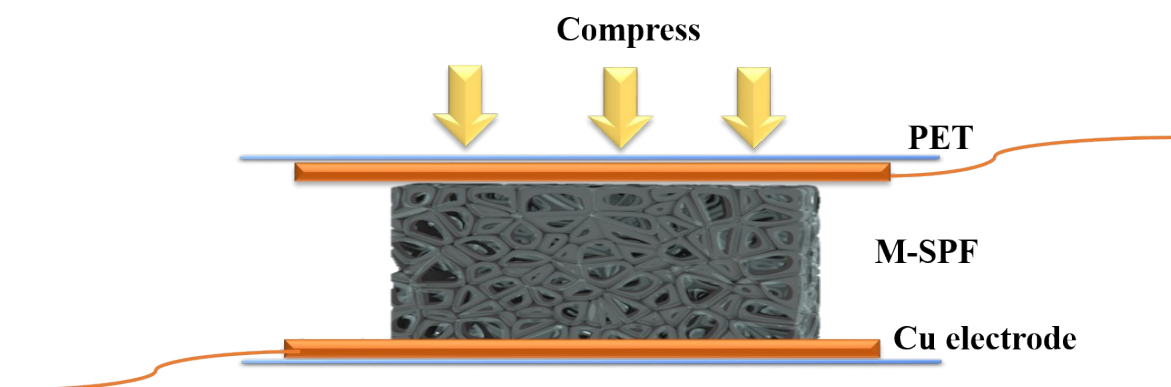
Supporting online materials for

**Ultrahigh compressibility and superior elasticity carbon framework derived from shaddock peel for high-performance pressure sensing**

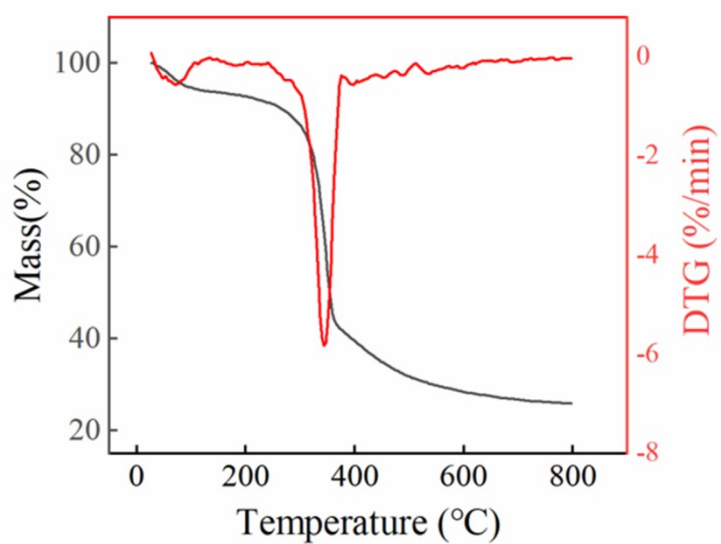
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Douyong Min<sup>1,2</sup>**

### *Optimization of PDMS concentration*

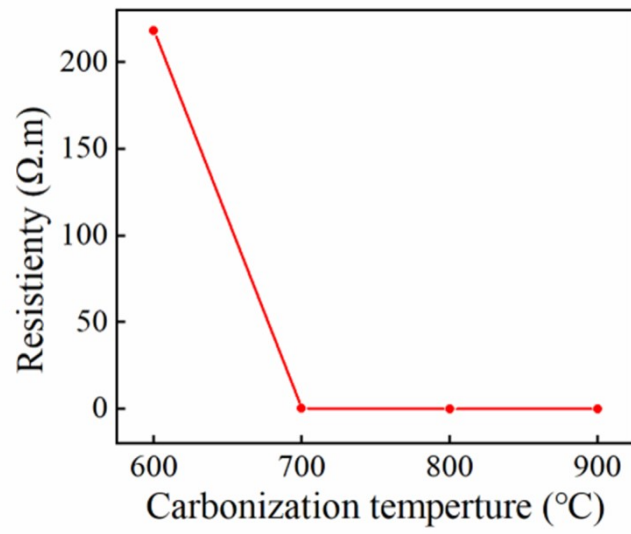
PDMS and N-hexane mixed solvent with different concentration of PDMS was prepared by adding PDMS base agent (0.5 g, 1.0 g, 1.5 g and 2.0 g) and curing agent (base agent and curing agent in a mass ratio of 10:1) into 10 mL N-hexane. The detailed preparation of M-SPF samples has been described in section 2.3 and the corresponding products were named as M-SPF-0.5, M-SPF-1.0, M-SPF-1.5 and M-SPF-2.0. We found that the M-SPF-0.5 was fragile and non-elastic, which is not available for flexible sensing material. The densities of M-SPF-1.0, M-SPF-1.5 and M-SPF-2.0 were calculated to be 0.358 g/cm<sup>3</sup>, 0.571 g/cm<sup>3</sup>, 0.726 g/cm<sup>3</sup>, respectively, suggesting the thickness of PDMS layer on C-SPF increased as the concentration of PDMS increased. Due to the PDMS is a non-conductive medium, the conductivity of M-SPF will decrease with the increased amount of PDMS. The conductivities of M-SPF-1.0, M-SPF-1.5 and M-SPF-2.0 samples were tested to be 1.47 s/m, 0.61 s/m and 0.19 s/m, respectively. Thus, M-SPF-1.0 with the best conductivity was used as the pressure sensing material in this study.



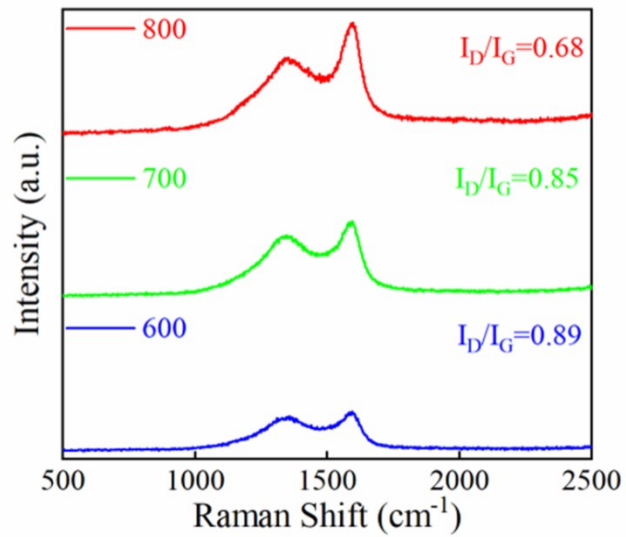
**Fig. S1.** The composition of the prepared pressure sensor.



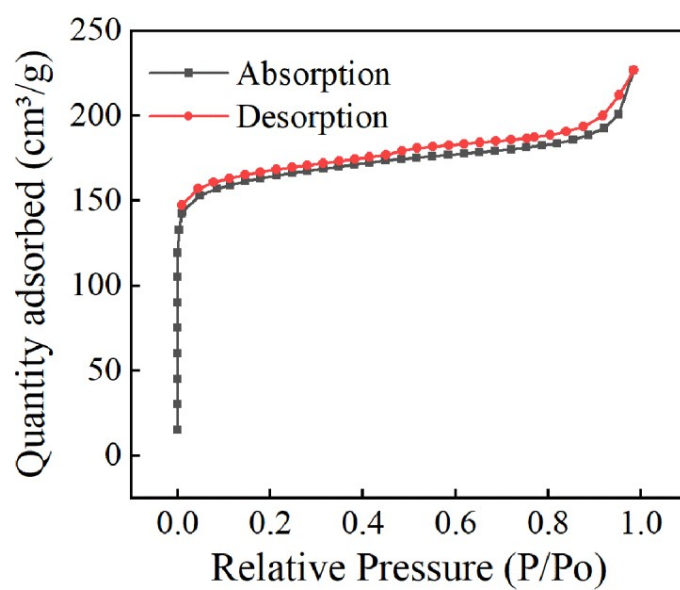
**Fig. S2.** TG-DTG curves of SPF.



**Fig. S3.** The conductivity of C-SPFs obtained at different temperatures.



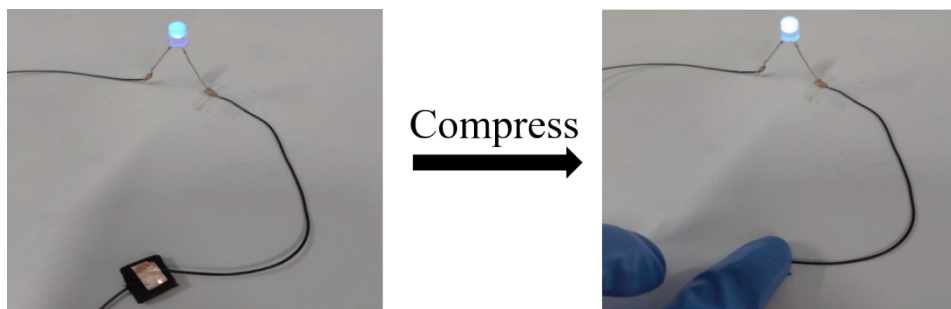
**Fig. S4.** Raman spectra of samples at different temperatures.



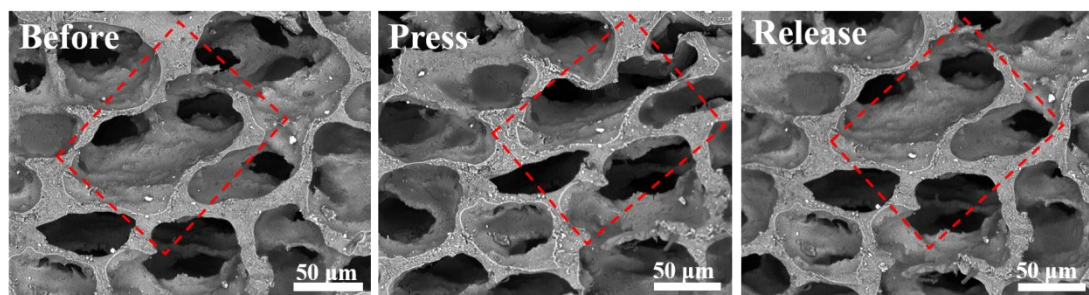
**Fig. S5.** The nitrogen adsorption and desorption isotherms of C-SPF in Brunauer-Emmett-Teller (BET) measurement.



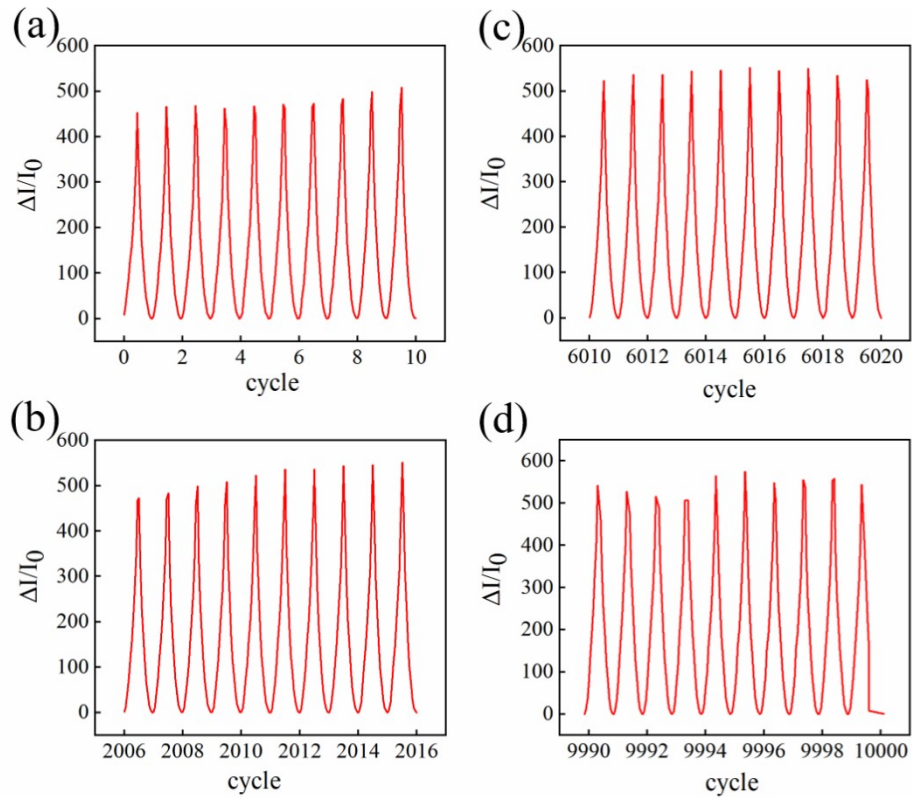
**Fig. S6.** Photos of C-SPF before and after compression.



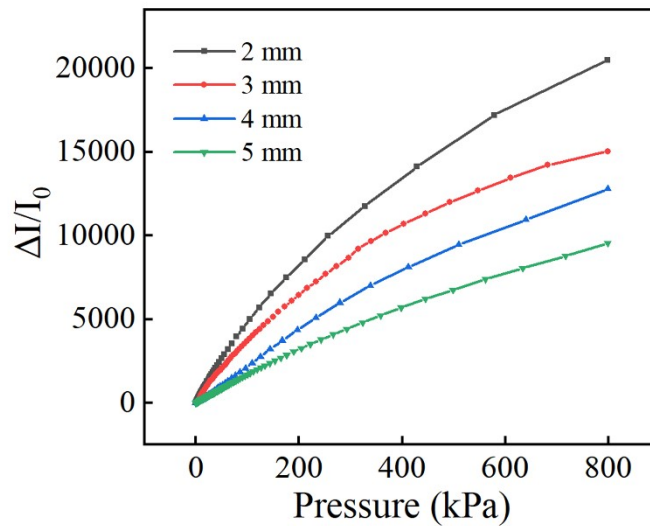
**Fig. S7.** Changes in brightness of LED lights without pressing and pressing.



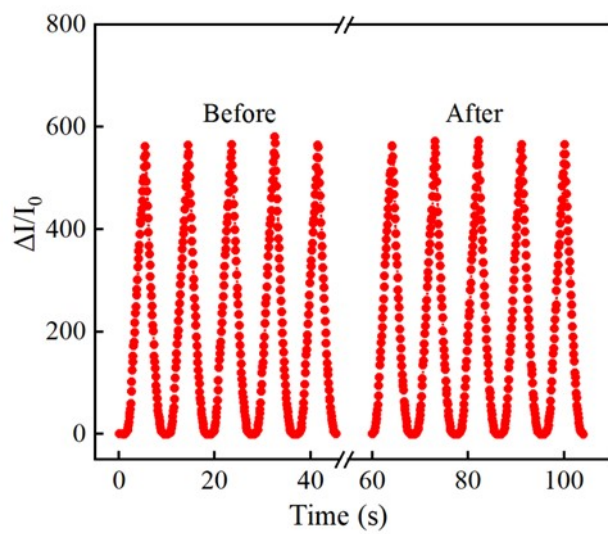
**Fig. S8.** SEM images of M-SPF in the process of compression and release.



**Fig. S9.** Partial electrical response curves for 10,000 cyclic of M-SPF at 50% strain.

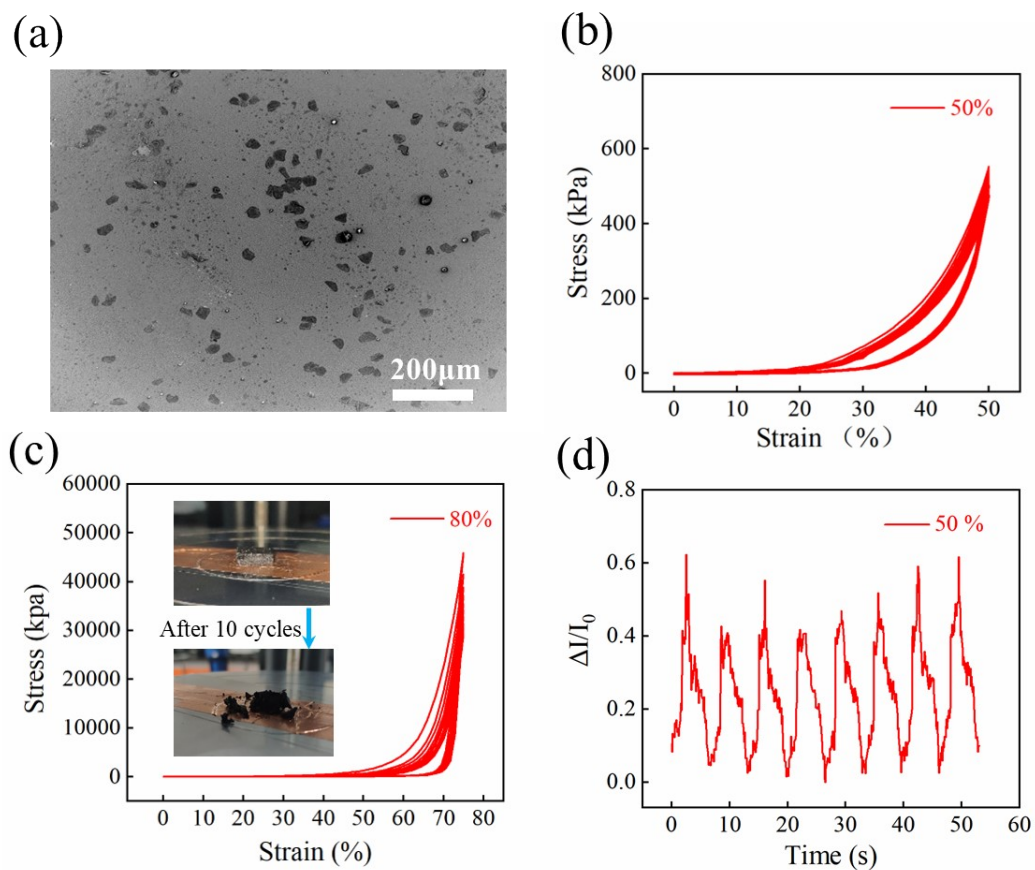


**Fig. S10.** The effect of thickness on the sensitivity of M-SPF-based sensor.



**Fig. S11.** Response of the same pressure sensor to electrical signals at 50% strain before and after three months.





**Fig. S12.** (a) SEM image of CSPP-PDMS (b) 100 cyclic loading–unloading curves under 50% compression strain. (c) 10 cyclic loading–unloading curves under 80% compression strain. (The inset represents the appearance of CSPP-PDMS before and after compression) (d) The RCR of the CSPP-PDMS based pressure sensor under 50% strain.

**Table S1** The bulk density, porosity and the ratio of carbon and oxygen content of the samples.

<b>Samples</b>	<b>Bulk density (g/cm<sup>3</sup>)</b>	<b>Porosity (%)</b>	<b>C/O</b>
SP	0.0841	89	2.10
SPF	0.0342	94	2.53
C-SPF	0.0257	93	9.59
M-SPF	0.3584	67	1.76

**Table S2** Material properties of each component in the simulation work.

<b>Samples</b>	<b>Bulk density (g/cm<sup>3</sup>)</b>	<b>Elastic modules(kPa)</b>	<b>Poisson ratio</b>
C-SPF	0.0257	0.65	0.25
M-SPF	0.3584	21.9	0.35

**Table S3** Comparison of the detection range and sensitivity for various pressure sensing materials.

<b>Materials</b>	<b>Detection range and Sensitivity</b>	<b>Ref.</b>
2D WSe <sub>2</sub> Nanosheets	1 -100 kPa (S=29.24 kPa <sup>-1</sup> ) 0.001-0.5 kPa (S=18.42 kPa <sup>-1</sup> )	14
2D MoSe <sub>2</sub> Nanosheets	1-35 kPa (S=7.28 kPa <sup>-1</sup> ) 40-100 kPa (S=2.63 kPa <sup>-1</sup> ) 5 kPa (S=5 kPa <sup>-1</sup> )	15
Alginate and graphene Sponge	50 kPa (S=1 kPa <sup>-1</sup> ) 500-1000 kPa (S=0.005 kPa <sup>-1</sup> )	17
rGO and Polyaniline Sponge	0-27 kPa (S=0.152 kPa <sup>-1</sup> )	19
PANI Paper	2-90 kPa (S=2.23 kPa <sup>-1</sup> )	20
PPy and Cotton	0.1-5 kPa (S=4.5 kPa <sup>-1</sup> )	21
Graphene–Polyurethane Sponge	0-2 kPa (S=0.26 kPa <sup>-1</sup> ) 2-10 kPa (S=0.03 kPa <sup>-1</sup> )	50
GO- Hybridized CNTs Aerogels	0-1 kPa (S=1.22 kPa <sup>-1</sup> ) 1-8 kPa (S=0.39 kPa <sup>-1</sup> )	51
Silver Nano-flower Decorated Graphene Oxide-Sponges	0-10 kPa (S=0.572 kPa <sup>-1</sup> )	52
Carbon Black and Polyurethane Sponge	0-2 kPa (S=0.068 kPa <sup>-1</sup> ) 2-10 kPa (S=0.023 kPa <sup>-1</sup> ) 10-16 kPa (S=0.036 kPa <sup>-1</sup> )	53
Nano-fibrous Aerogels	0-3.5 kPa (S=0.43 kPa <sup>-1</sup> ) 3.5-5 kPa (S=1.02 kPa <sup>-1</sup> ) 0-0.1 kPa (S=22.05 kPa <sup>-1</sup> )	62
CNT/rGO–CNF carbon aerogel	0.1-1 kPa (S=11.82 kPa <sup>-1</sup> ) 1-5 kPa (S=0.44 kPa <sup>-1</sup> )	64
Carbide Nano-sheets and Bacterial Cellulose Carbon Aerogels	0-10 kPa (S=12.5 kPa <sup>-1</sup> )	S1

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Aligned carbon nanotubes/graphene	0-0.3 kPa ( $S=19.8 \text{ kPa}^{-1}$ )	S2
	0.3-6 kPa ( $S=0.27 \text{ kPa}^{-1}$ )	
Graphene Sponges	0.3-10 kPa ( $S=0.046 \text{ kPa}^{-1}$ )	S3
	10-40 kPa ( $S=0.007 \text{ kPa}^{-1}$ )	
Graphene/Polyimide Nanocomposite Foam	0-1.5 kPa ( $S=0.18 \text{ kPa}^{-1}$ )	S4
	1.5-7 kPa ( $S=0.023 \text{ kPa}^{-1}$ )	
M-SPF	1-10 kPa ( $S=48.5 \text{ kPa}^{-1}$ )	This work
	10-100 kPa ( $S=63.4 \text{ kPa}^{-1}$ )	
	100-800 kPa ( $25.6 \text{ kPa}^{-1}$ )	

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## References

- S1. Z. Chen, Y. Hu, H. Zhuo, L. Liu, S. Jing, L. Zhong, X. Peng and R. C. Sun, *Chem. Mater.*, 2019, **31**, 3301-3312.
- S2. M. Jian, K. Xia, Q. Wang, Z. Yin, H. Wang, C. Wang, H. Xie, M. Zhang and Y. Zhang, *Adv. Funct. Mater.*, 2017, **27**, 1606066.
- S3. L. Zhu, Y. Wang, D. Mei and X. Wu, *J. Microelectromech. S.*, 2019, **28**, 154-163.
- S4. Y. Qin, Q. Peng, Y. Ding, Z. Lin, C. Wang, Y. Li, J. Li, Y. Yuan, X. He and Y. Li, *ACS Nano*, 2015, **9**, 8933-8941.