

Electronic Supplementary Information

**High-performance electrocatalyst based on polyazine derived mesoporous
nitrogen-doped carbon for oxygen reduction reaction**

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S1. BET surface areas and pore volume variation with the pyrolytic temperature

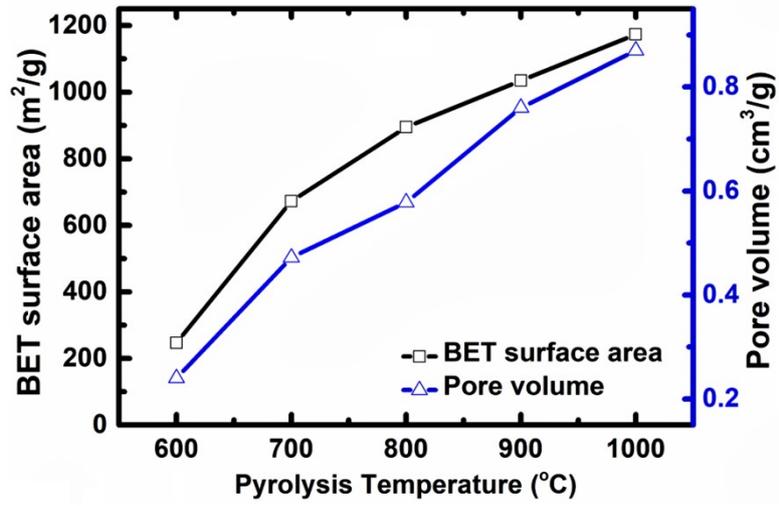


Figure S1 the BET surface areas and pore volume variation with the pyrolytic

temperature

S2 Pyrolysis behavior of F127/PAZ composite

Figure S1 shows the thermal decomposition of F127, PAZ and their composite from 25 °C to 998 °C in a nitrogen atmosphere, respectively. As shown, a major weight reduction of above 99% for F127 takes place as the temperature reached 410 °C, suggesting that the template can be totally destroyed at this stage. For the PAZ without the F127 template, a rapid weight loss was observed at the temperature range of 150–300 °C (around 45% weight loss), which indicates the cleavage of small molecules from the further condensation of the linear polymer to form a framework structure and facilitate carbonization. In the range of 400–995 °C, the carbonization is improved, accompanied by the gradual decomposition of carbonaceous residue with approximately 40% of further weight loss. In contrast, the profile of the 127/PAZ composite colligated the pyrolytic behaviour of both the PAZ and F127. Two significant weight loss stages can be observed in the temperature range of 150–280 °C and 300–410 °C, which are attributed to the partial destruction of PAZ and the complete decomposition of F127, respectively. The final carbon yield (atom efficiency toward the desired carbon structure) is about 12% of the composite when it

is heated up to 995 °C.

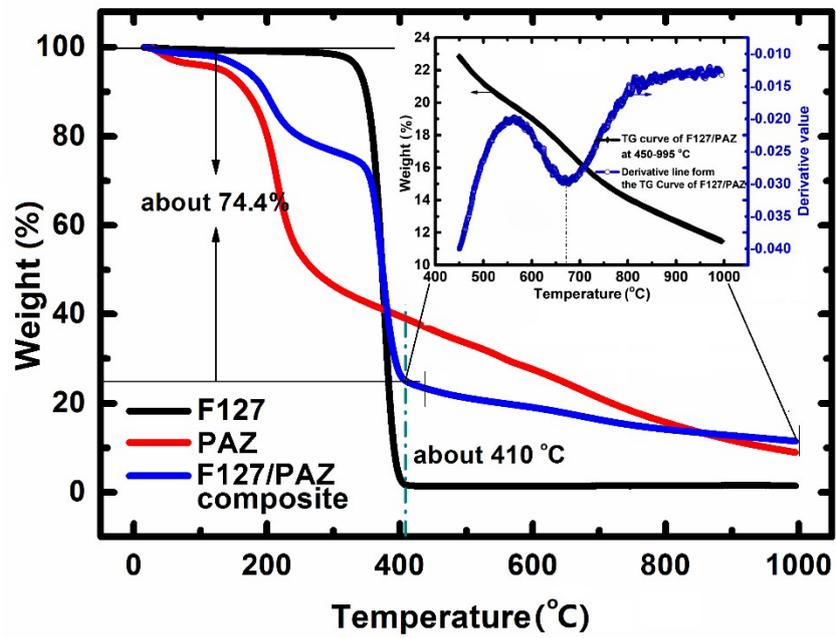


Figure S2 Thermogravimetric behaviors of the F127/PAZ composite compared to F127 and PAZ in N₂ atmosphere

Table S1 Comparison NMC-700 with previous reported metal free ORR catalytic in 0.1 M KOH at a scan rate of 5 or 10 mV/s with the RDE rotation rate of 1600 rpm.

Materials	Precursors	Onset potential	^a Relative half-wave potential	Tafel slopes	H ₂ O ₂ yields	References
N-doped graphene	ammonia	ca. 25 mV positive	43 mV positive	-	ca. 10%	1
N-doped Carbon/graphene composite	polydopamine	ca. 10 mV positive	40 mV positive	-	4.3-16.5%	2
N-doped carbon	ionic liquids	ca. 25 mV negative	-	-	-	3
N, S-dual-doped graphene	melamine, disulfide	benzyl ca. 30 mV negative	-	-	-	4
N-doped carbon	ZIF-8	-0.13 V vs. SCE (not compare with Pt catalyst)	-	-	-	5
N-doped carbon/CNT composite	PANI	ca. 30 mV negative	26 mV positive	66 mV decade ⁻¹		6
N-doped carbon	Polyazine	ca. 6 mV negative	90 mV negative	87 mV decade ⁻¹	<5%	This work

a. Compared with the results from commercial Pt (20 at.%) catalyst

S3 Normalization of kinetic current at 0.875 V to the exposure indexes of nitrogen species for ORR.

Table S2 Physical and chemical characteristic parameter of materials prepared under different pyrolysis temperature (PT)

PT (°C)	SSA (m ² g ⁻¹)	Kinetic current densities at 0.875 V (mA cm ⁻²)	N-species containing (at.%) / EI ^a	
			pyridinic-N	pyrrolic-N
600	247	0.29	8.21/430	1.74/2027
700	672	1.87	4.08/652	0.97/2742
800	895	1.70	2.46/438	0.49/2022
900	1034	0.62	0.76/300	0.29/786
1000	1174	0.60	0.56/235	0.2/657

a. The exposure indexes (*EI*) of the proposed active sites can be calculated as follows:

$$EI = \text{SSA} \times \text{Nitrogen species containing (pyridinic-N or pyrrolic-N)}.$$

Normalization of $J_{0.875 \text{ V}}$ to *EI* can be carried out as follows:

$$\text{nominal } J_{0.875 \text{ V}} = \frac{J_{0.875 \text{ V}} \times S_{\text{we}}}{EI} \quad (\mu\text{A g m}^{-2})$$

Where S_{we} refers to the area of work electrode, the value is 0.19625 cm².

S4 RRDE measurement

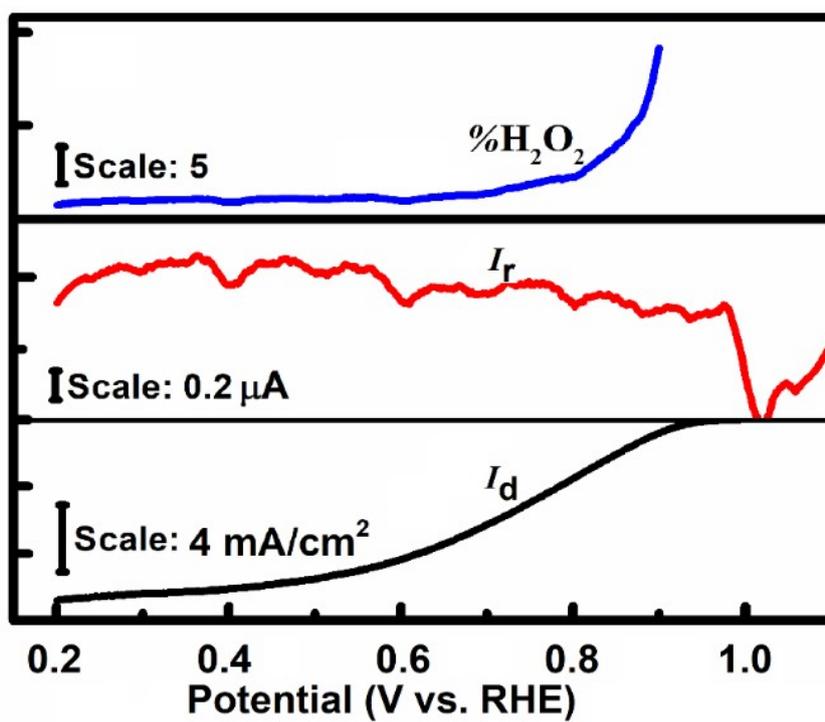


Figure S3 Rotating ring-disk electrode measurement combined with H₂O₂ yield over

NMC-700 in O₂-saturated 0.1 M KOH solution at the rotation rate of 1600 rpm.

References:

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