Supporting Information

1,2,3-Triazole Framework: A Strategic Structure for C-H•••X Hydrogen Bonding and Practical Design of an Effective Pd-Catalyst for Carbonylation and Carbon-Carbon Bond Formation

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1) General Remarks

All reagents were purchased from commercial suppliers (Merck and Aldrich) and used without further purification. The FT-IR spectra of products were measured as KBr disc by Jasco-680 FTIR spectrophotometer. The X-ray diffraction (XRD) patterns were recorded by using a Philips Xpert MPD diffractometer with Cu Ka radiation (1 = 0.15418 nm). Transmission Electron Microscopy (TEM) was carried out on Philips CM120 microscope 100 KV. Scanning electron microscopy (FE-SEM) images were obtained using a Hitachi S-4160 instrument. Analysis of Pd content was measured by inductively coupled plasma spectroscopy (ICP) using Perkin Elmer Optima 7300 DV .Melting points were determined by a Gallenkamp melting apparatus. Gas chromatography (GC) analyses were performed on a BEIFIN 3420 gas chromatograph equipped with a Varian CP SIL 5CB column: 30 m, 0.32 mm, 0.25 mm for consideration of reactions conversions and yields. The ¹H and ¹³C NMR spectra were recorded on a Bruker AVANCE spectrometer at 400, 300 and 100 MHz using TMS as an internal standard in CDCl₃ and DMSO at RT.

2) Elemental EDX mapping of the catalyst Pd@click-MNPs/CS (Figure 1S.)



C Ka



O Ka





C N Fe



3) Characteristic spectra of the products



Figure 2S. ¹HNMR spectra of 1-benzyl-4-phenyl-1-H-1,2,3-triazole in DMSO and CDCl₃ solvent



200 190 180 170 160 150 140 130 120 110 10 90 80 70 60 50 40 30 20 10 0 f1 (ppm)



































