

Supporting information for:

Physical and photocatalytic properties of sprayed Dy doped ZnO thin films under sunlight irradiation for degrading methylene blue

G. El Fidha ^{a,b}, N. Bitri ^b, F. Chaabouni ^b, S. Acosta ^c, F. Güell ^d, C. Bittencourt ^c, J. Casanova-Chafer ^e, E. Llobet ^e

^a Université de Tunis, École nationale supérieure d'ingénieurs de Tunis, Avenue Taha Hussein Montfleury, 1008 Tunis, Tunisie

^b Université de Tunis El Manar, Ecole Nationale d'Ingénieurs de Tunis, Laboratoire de Photovoltaïque et matériaux semi-conducteurs, 1002 Tunis, Tunisie

^c Chimie des Interactions Plasma-Surface (ChIPS), Research Institute for Materials Science and Engineering, Université de Mons, 7000, Mons, Belgium

^d ENFOCAT-IN2UB, Universitat de Barcelona, C/Marti i Franquès 1, 08028 Barcelona, Spain

^e MINOS, Universitat Rovira i Virgili, Avda. Països Catalans, 26, 43007, Tarragona, Spain

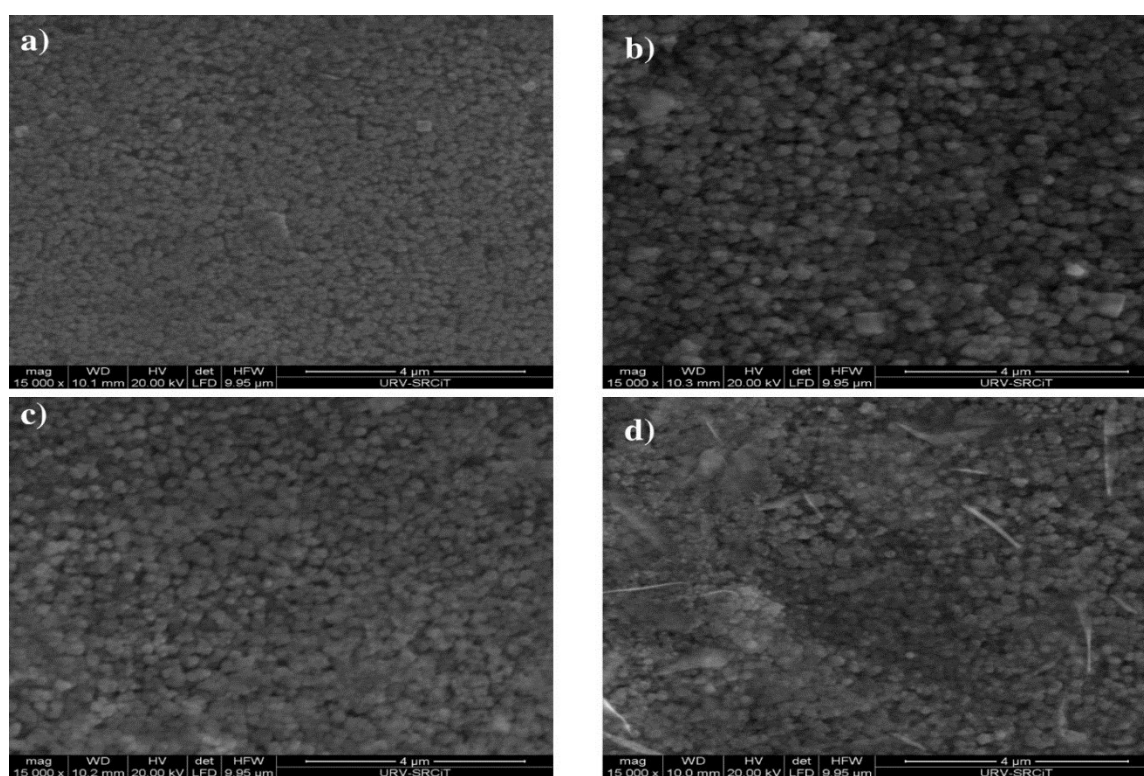


Figure S.1: SEM images of ZnO doped and undoped thin films: (a): Pure, (b) ZnO:Dy 2%,(c) ZnO : Dy4%, (d) ZnO : Dy 6%. All the films show dense and spherical structure strongly agglomerated. These results are consistent with the results of AFM.

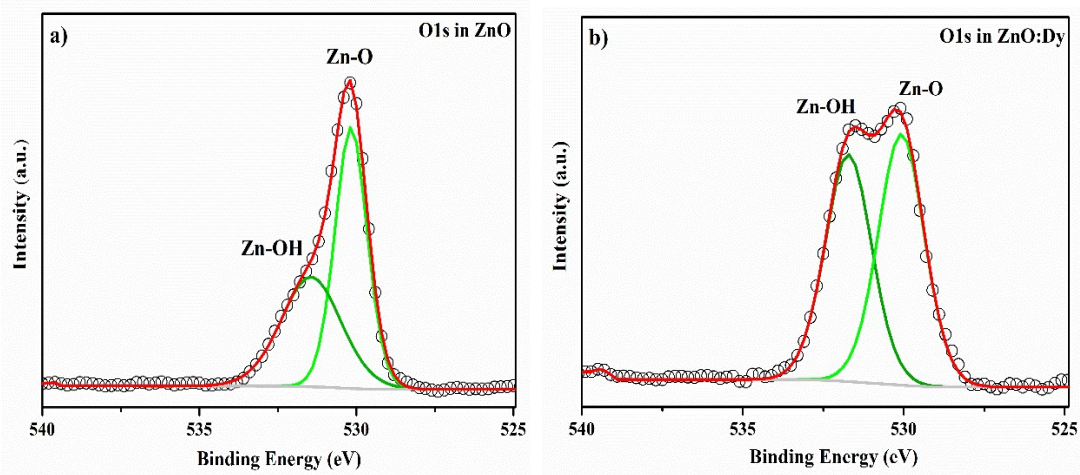


Figure S.2: O 1s peak analysis for the pure ZnO sample (a) and 6% doped Dy-ZnO (b).

Two main peaks are observed at 530.0 eV and 531.8 eV which are attributed to lattice oxygen of Zn-O and Zn-OH groups (defective ZnO), respectively.

Table S.1: XPS quantification expressed in at.%.

Sample	Oxygen	Zinc	Dysprosium
Pure ZnO	48.7	51.3	-
6% doped Dy-ZnO	57.0	38.0	6.0

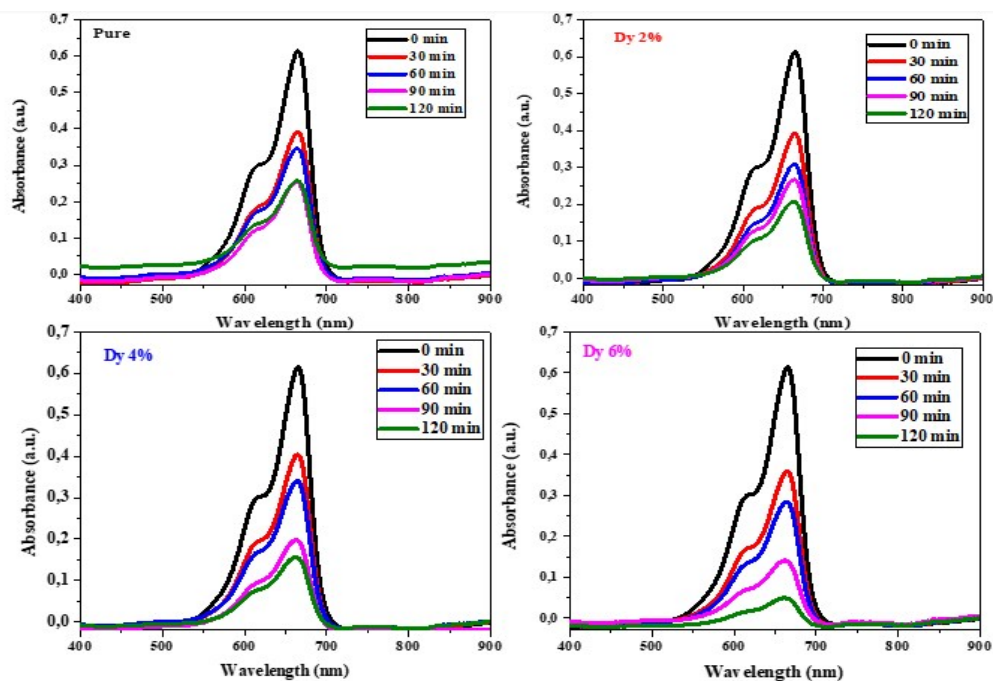


Figure S.3: Absorption spectra of MB after solar irradiation for 2 hours using the Dy doped ZnO thin films contents (first run)

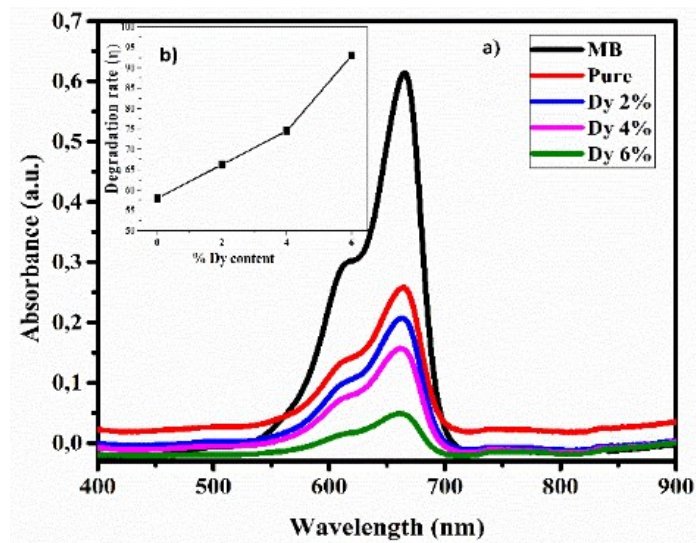


Figure S.3: Absorption spectra of MB after solar irradiation for 2 hours using the Dy doped ZnO thin films at various Dy contents.

Table S. 2. Comparison of photocatalytic activities of ZnO-doped rare earths materials for degrading dyes.

Photocatalyst	Method of elaboration	Sources of Irradiation	Dye	Degradation efficiency (%) -Time (min)	Ref.
Sm-ZnO	chemical precipitation	Visible lamp	MB	94.94% - 60 min	1
ZnO-Sm ZnO-Er ZnO-La	electrospinning	UV-light	Congo-Red dye	84.59- 240 min 78.86 - 240 min 73.23- 240 min	2
ZnO-Gd	Co-precipitation	Visible light	MB	93% - 90min	3
Dy-ZnO	sonochemical method.	UV light	MB	98% - 300min	4
Dy- ZnO	Combustion method	UV light	DR-31	99.6 - 60min	5
Dy-ZnO	Spray Pyrolysis	Solar irradiation	MB	92% - 120min	This work

References

- [1]: Sukriti, P. Chand and V. Singh, J. Rare Earths, 2020, **38**, 29-38.
- [2] : P. Pascariu, C. Cojocaru, N. Olaru, P. Samoila, A. Airinej, M.Lgnat, L. Sacarescu and D.Timpu, J. Environ. Manage, 2019, **239**, 225-234.
- [3]: S. Selvaraj, M. K. Mohan, M. Navaneethan, S. Ponnusamy and C. Muthamizhchelvan, Mater. Sci. Semicond. Process., 2019, **103**, 104622.
- [4]: O. Yayapao, T. Thongtem, A. Phuruangrat and S. Thongtem, J. Alloys Compd., 2013, **576**, 72-79.
- [5]: S. Bhatia, N. Verma and R. Kumar, J. Alloys Compd. , 2017, **726**, 1274-1285.