

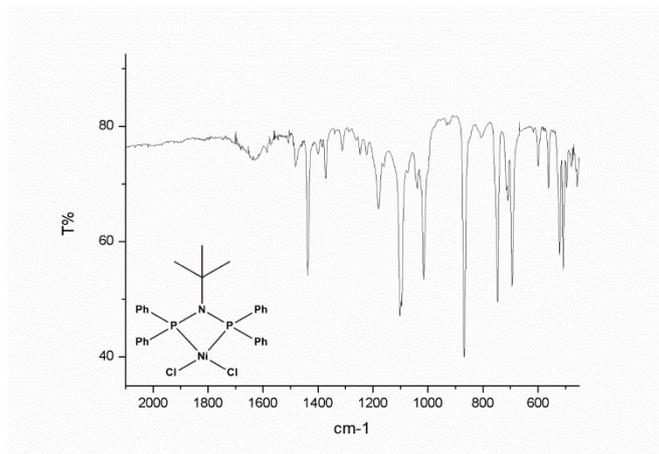
## SUPPORTING INFORMATION

### Effects of the halogenido ligands to Kumada-coupling catalytic activity of [Ni{*t*-BuN(PPh<sub>2</sub>)<sub>2</sub>-κ<sup>2</sup>P}X<sub>2</sub>], X = Cl, Br, I, complexes

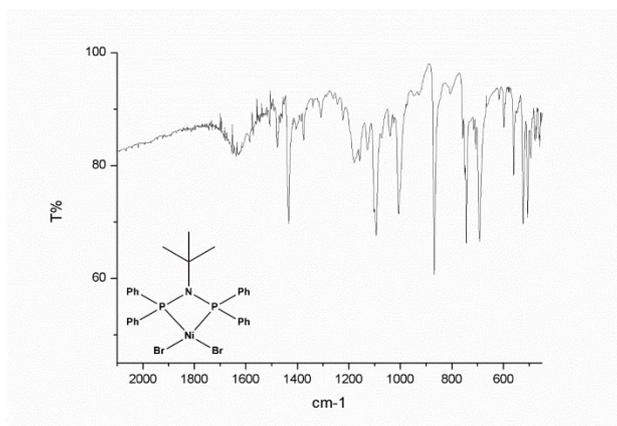
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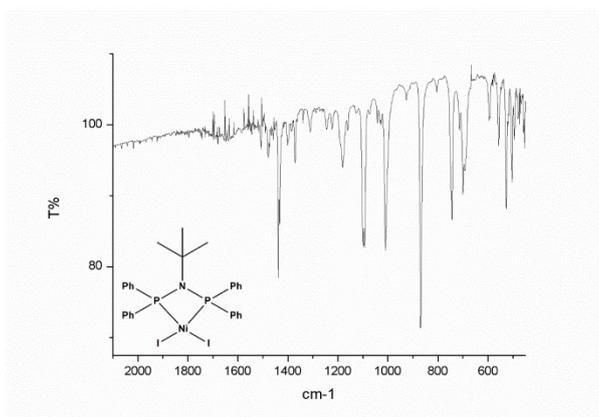
## IR spectra



**Figure S1.** FT-IR spectrum of [Ni(P,P)Cl<sub>2</sub>]. Selected bands (KBr, cm<sup>-1</sup>): 1437 ν(P-Ph), 1180, 1101, 1015, 868 ν(P-N-P), 746, 694.

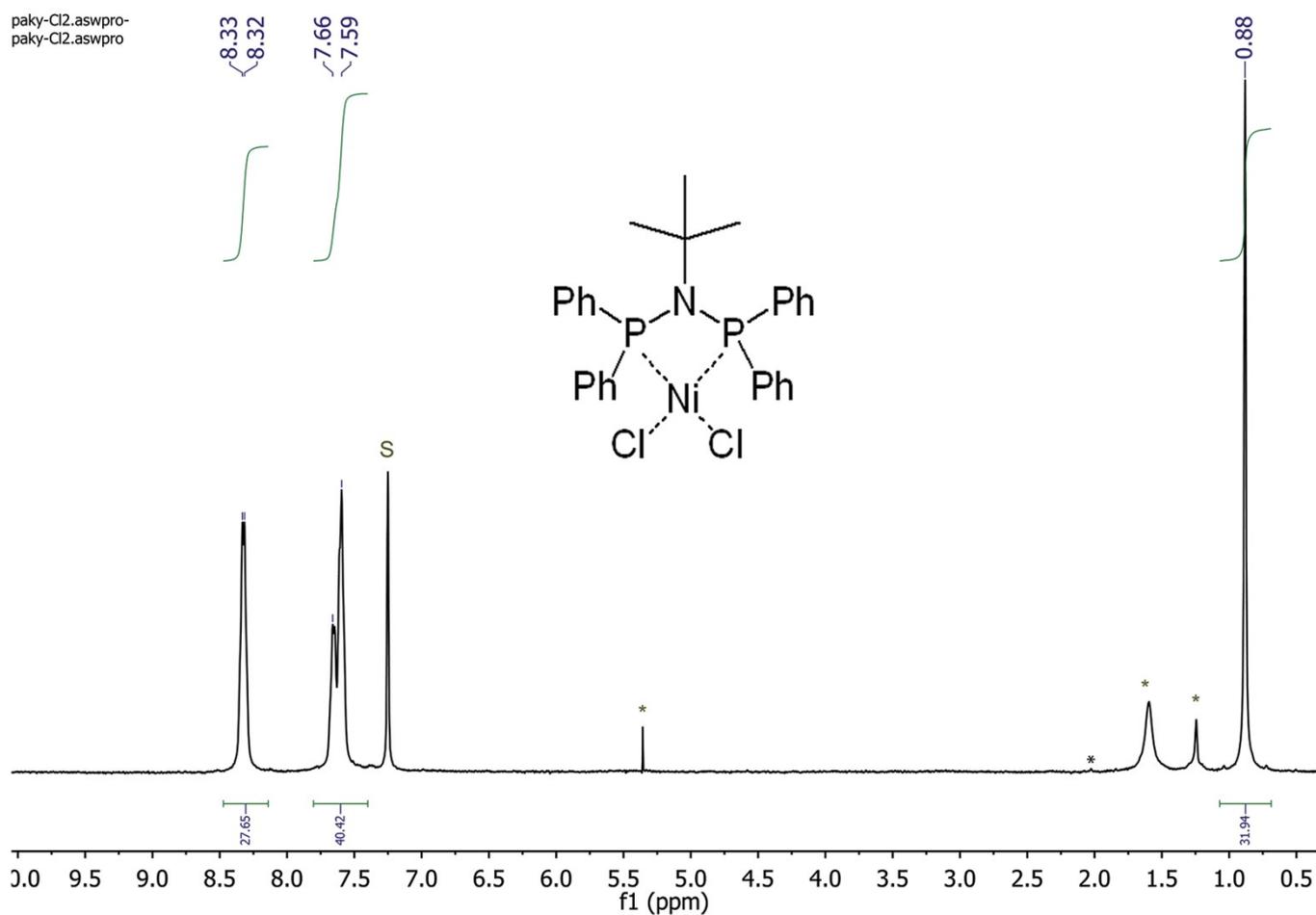


**Figure S2.** FT-IR spectrum of [Ni(P,P)Br<sub>2</sub>]. Selected bands (KBr, cm<sup>-1</sup>): 1477, 1433 ν(P-Ph), 1173, 1093, 1007, 868 ν(P-N-P), 744, 692.

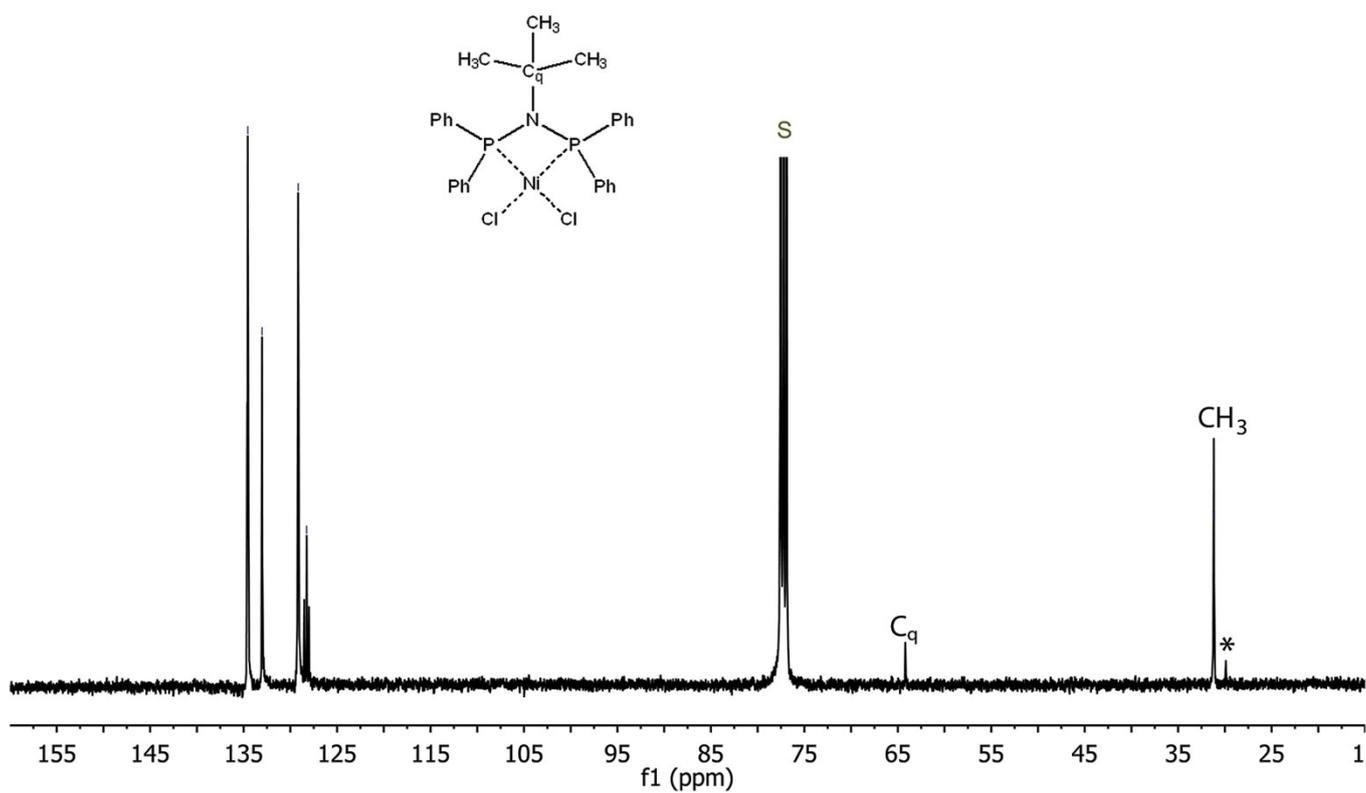


**Figure S3.** FT-IR spectrum of [Ni(P,P)I<sub>2</sub>]. Selected bands (KBr, cm<sup>-1</sup>): 1481, 1437 ν(P-Ph), 1370, 1180, 1098, 1009, 868 ν(P-N-P), 744, 700.

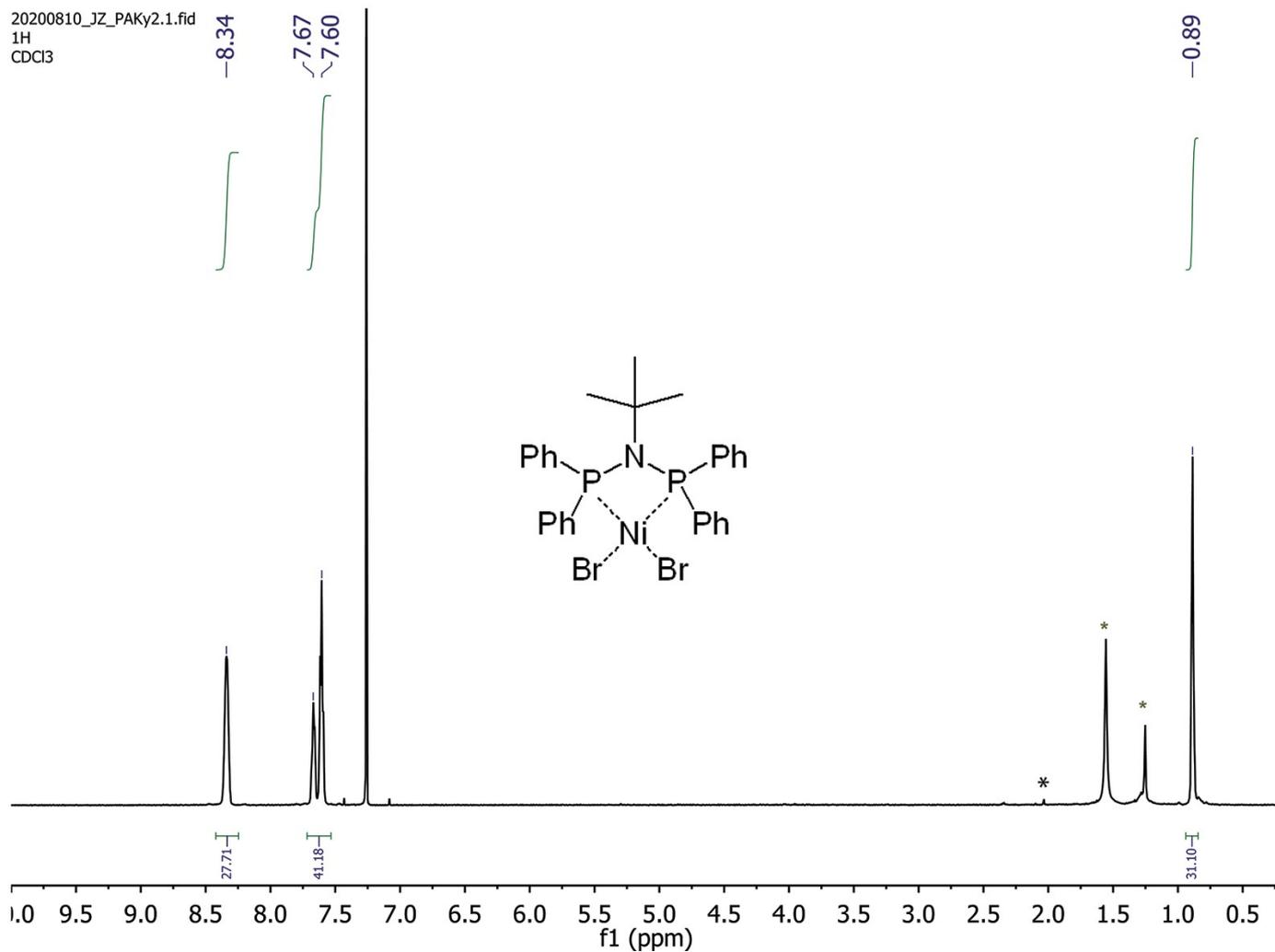
# $^1\text{H}$ and $^{13}\text{C}$ NMR spectra



**Figure S4.**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ni}(\text{P},\text{P})\text{Cl}_2]$ . S – 7.25 ppm  $\text{CH}_3\text{Cl}$  (solvent signal), 5.27 ppm trace of dichloromethane, 2.10 ppm trace of acetone, 1.50 ppm trace of moisture, 1.25 ppm trace of hexane, all impurities are marked by an asterisk.



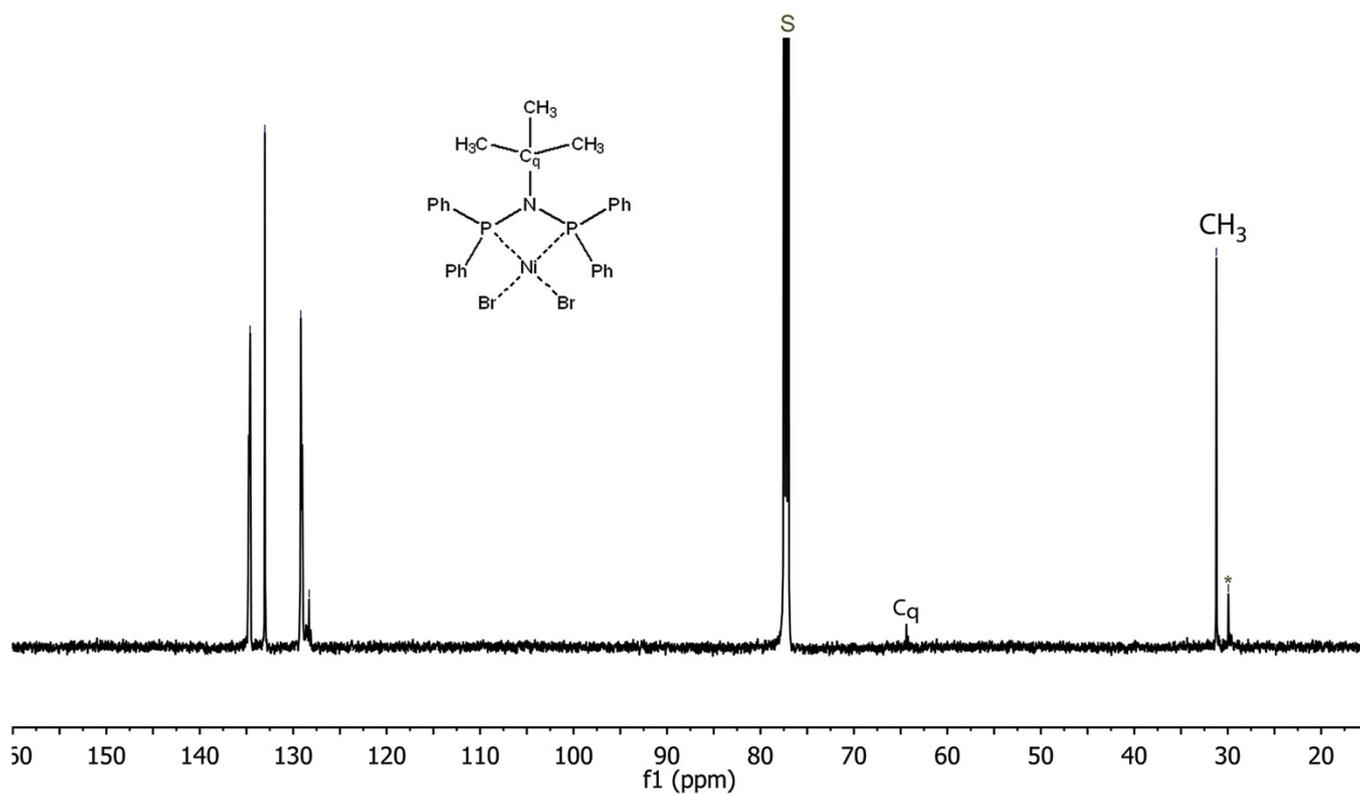
**Figure S5.** <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of [Ni(P,P)Cl<sub>2</sub>], S – solvent peak, residual trace of acetone is marked by an asterisk.



**Figure S6.**  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ni}(\text{P},\text{P})\text{Br}_2]$ . S – 7.25 ppm  $\text{CH}_3\text{Cl}$  (solvent signal), 2.1 ppm – acetone, 1.50 ppm trace of moisture, 1.25 ppm trace of hexane, all impurities are marked by an asterisk.

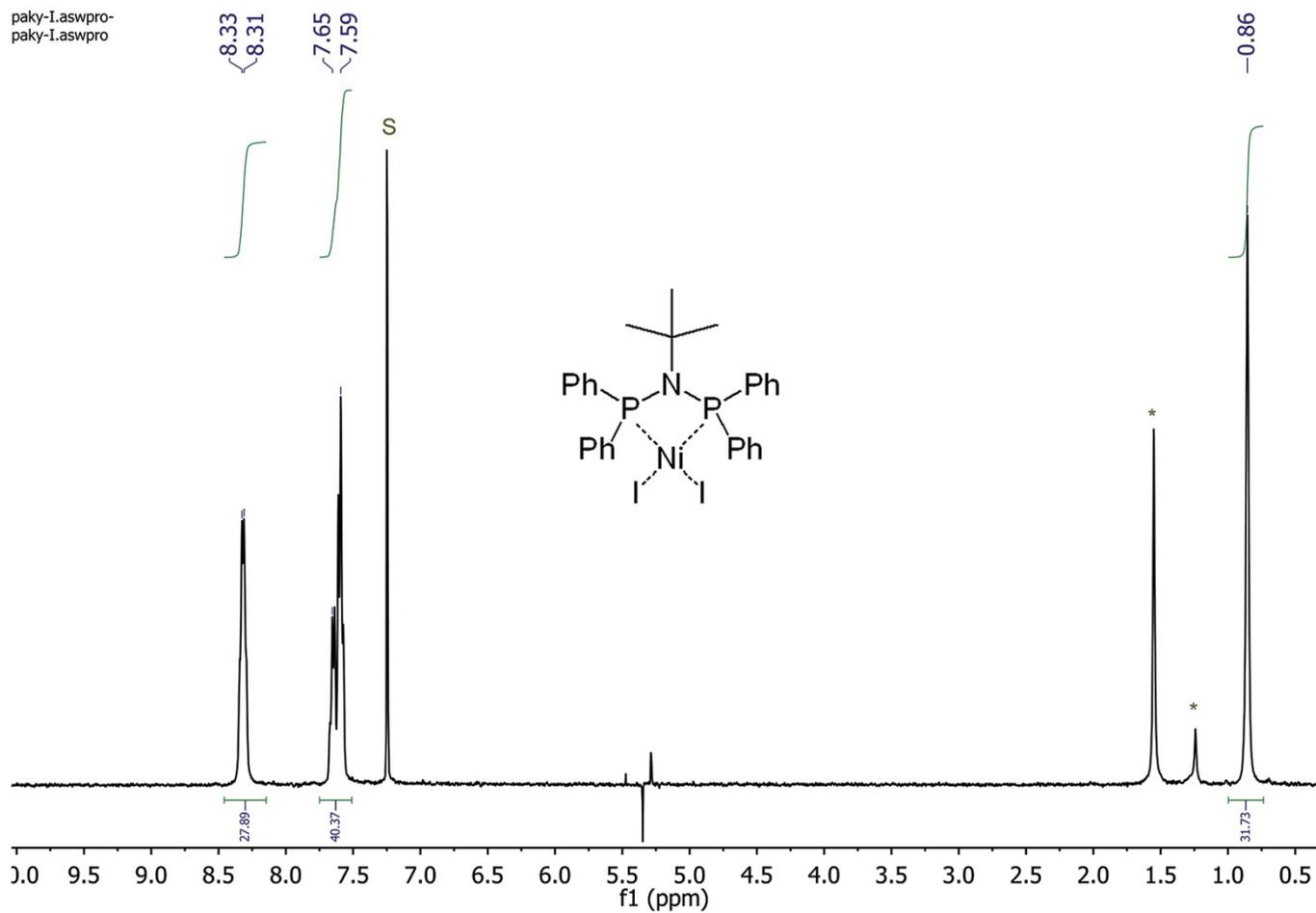
134.6  
133.0  
129.2  
128.3

31.2



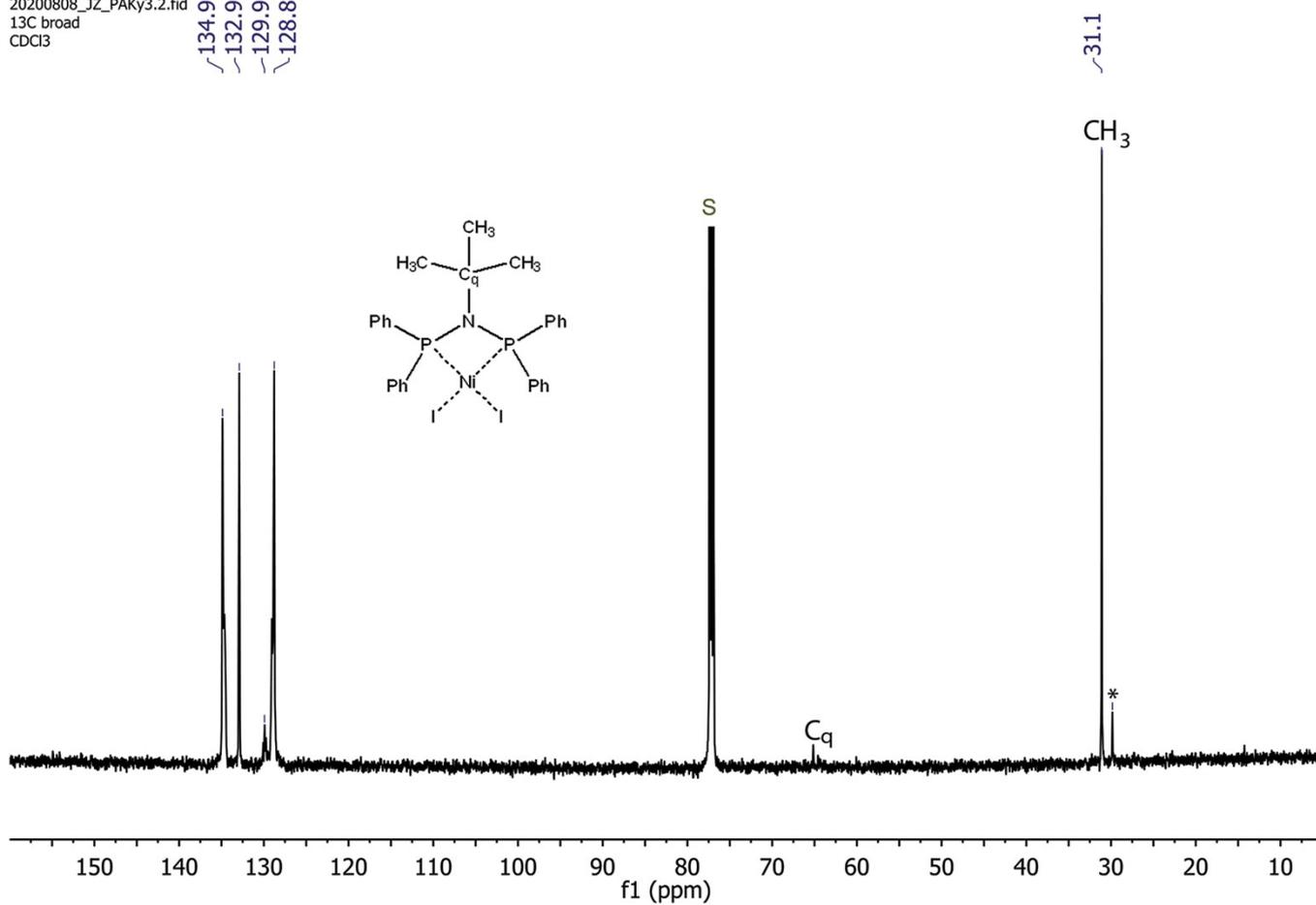
**Figure S7.** <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) spectrum of [Ni(P,P)Br<sub>2</sub>]. S – solvent peak, residual trace of acetone (29.9 ppm) is marked by an asterisk.

paky-I.aswpro  
paky-I.aswpro



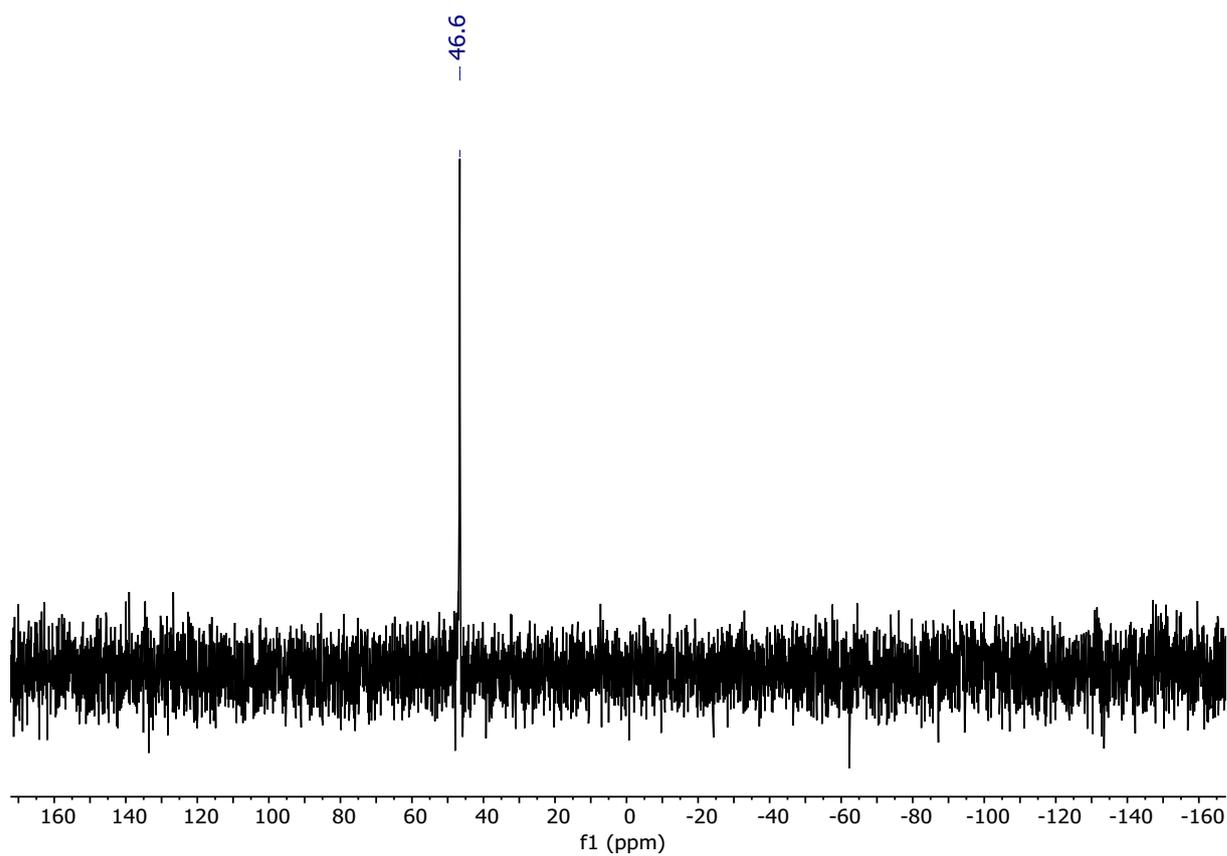
**Figure S8.** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of [Ni(P,P)I<sub>2</sub>]. S – 7.25 ppm CH<sub>3</sub>Cl (solvent signal), 1.50 ppm moisture, 1.25 ppm trace of hexane, all impurities are marked by an asterisk.

20200808\_JZ\_PAKy3.2.fid  
13C broad  
CDCl3

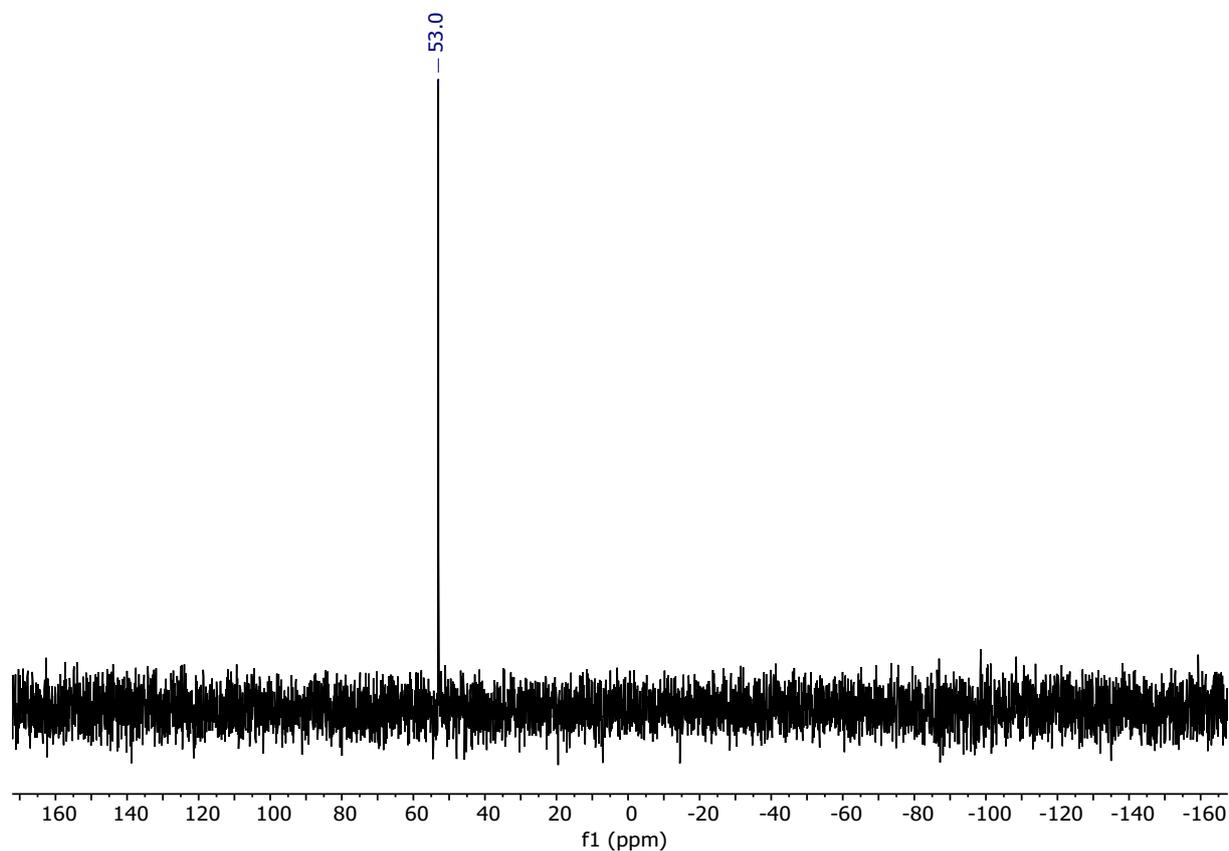


**Figure S9.**  $^{13}\text{C}$  NMR (151 MHz,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ni}(\text{P,P})\text{I}_2]$ , S – solvent peak, trace of acetone (29.9 ppm) is marked by an asterisk.

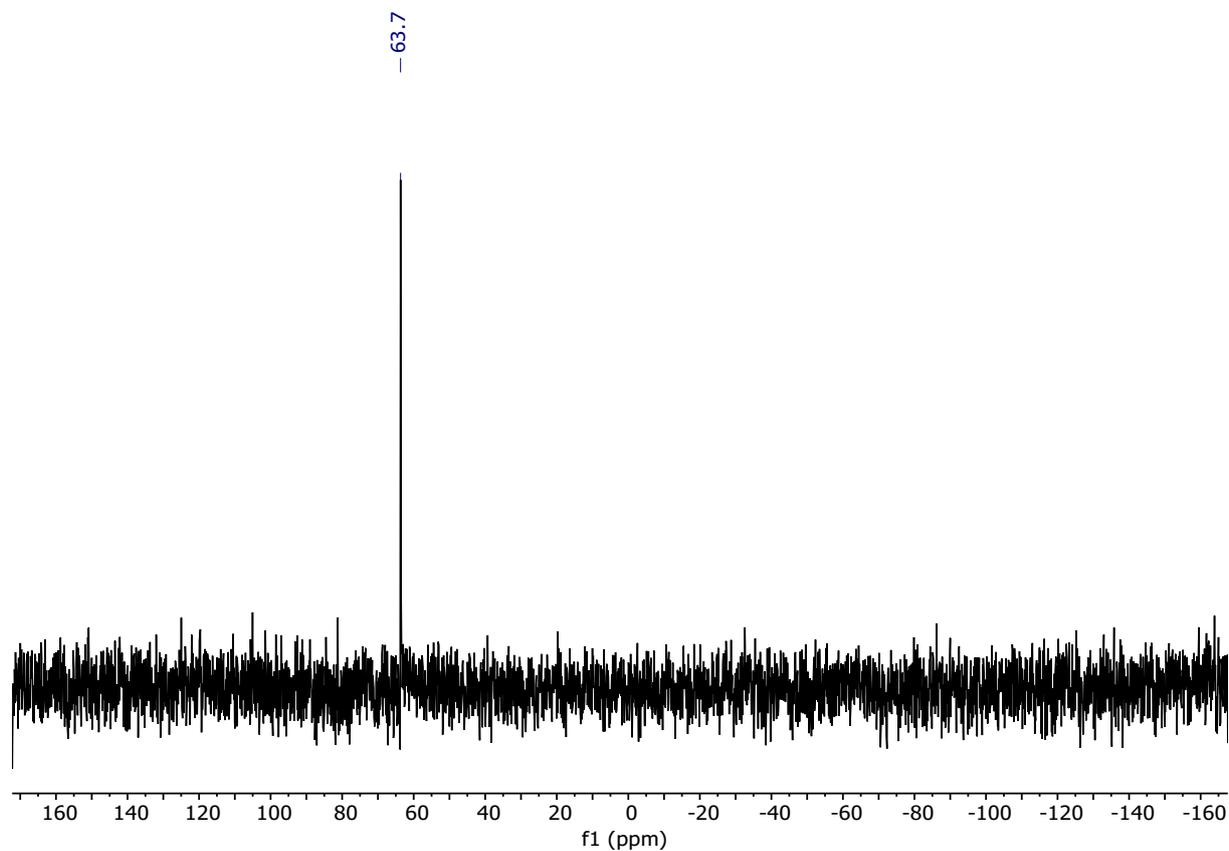
## $^{31}\text{P}$ NMR spectra



**Figure S10.**  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ni}(\text{P},\text{P})\text{Cl}_2]$ . The spectral window was extended to the full scale.

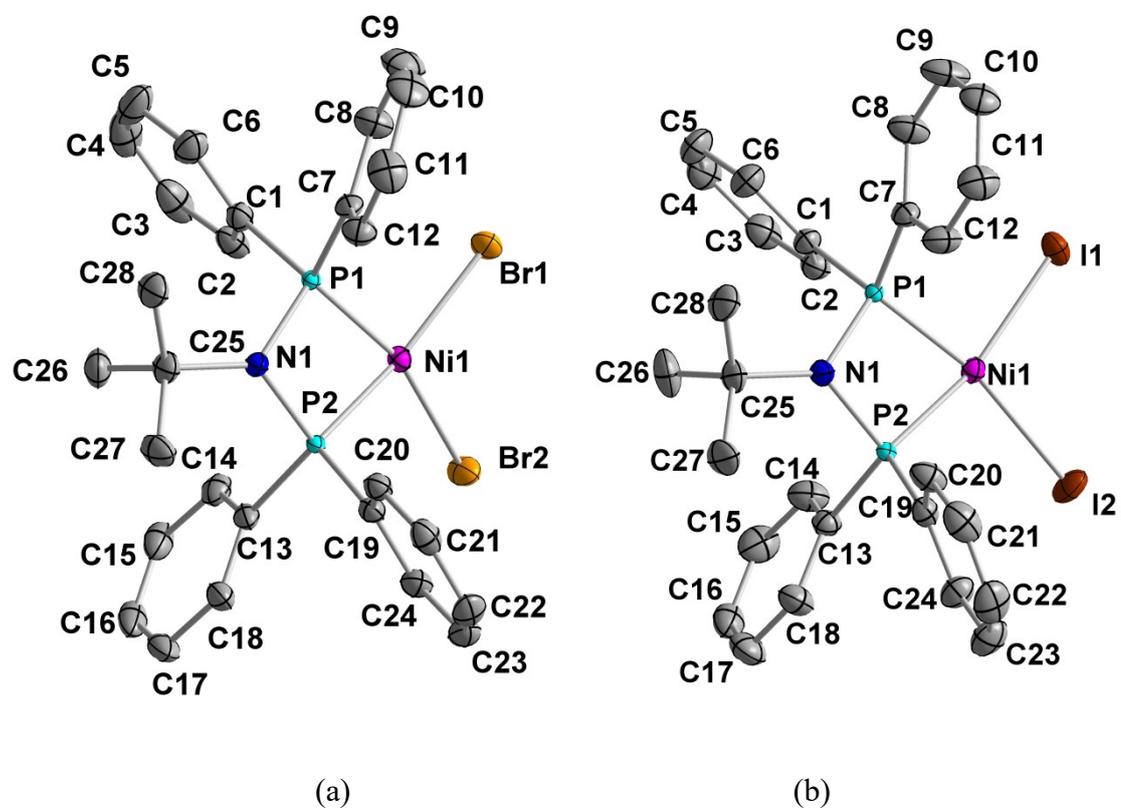


**Figure S11.**  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ni}(\text{P},\text{P})\text{Br}_2]$ . The spectral window was extended to the full scale.



**Figure S12.**  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ni}(\text{P},\text{P})\text{I}_2]$ . The spectral window was extended to the full scale.

## X-ray crystallography



**Figure S13.** Molecular structures and atom numbering of  $[\text{Ni}(\text{P},\text{P})\text{X}_2]$ , X = Br, I: (a) and (b), respectively. Thermal ellipsoids are presented at a level 50% probability. Hydrogen atoms are omitted for clarity.

## MS spectra

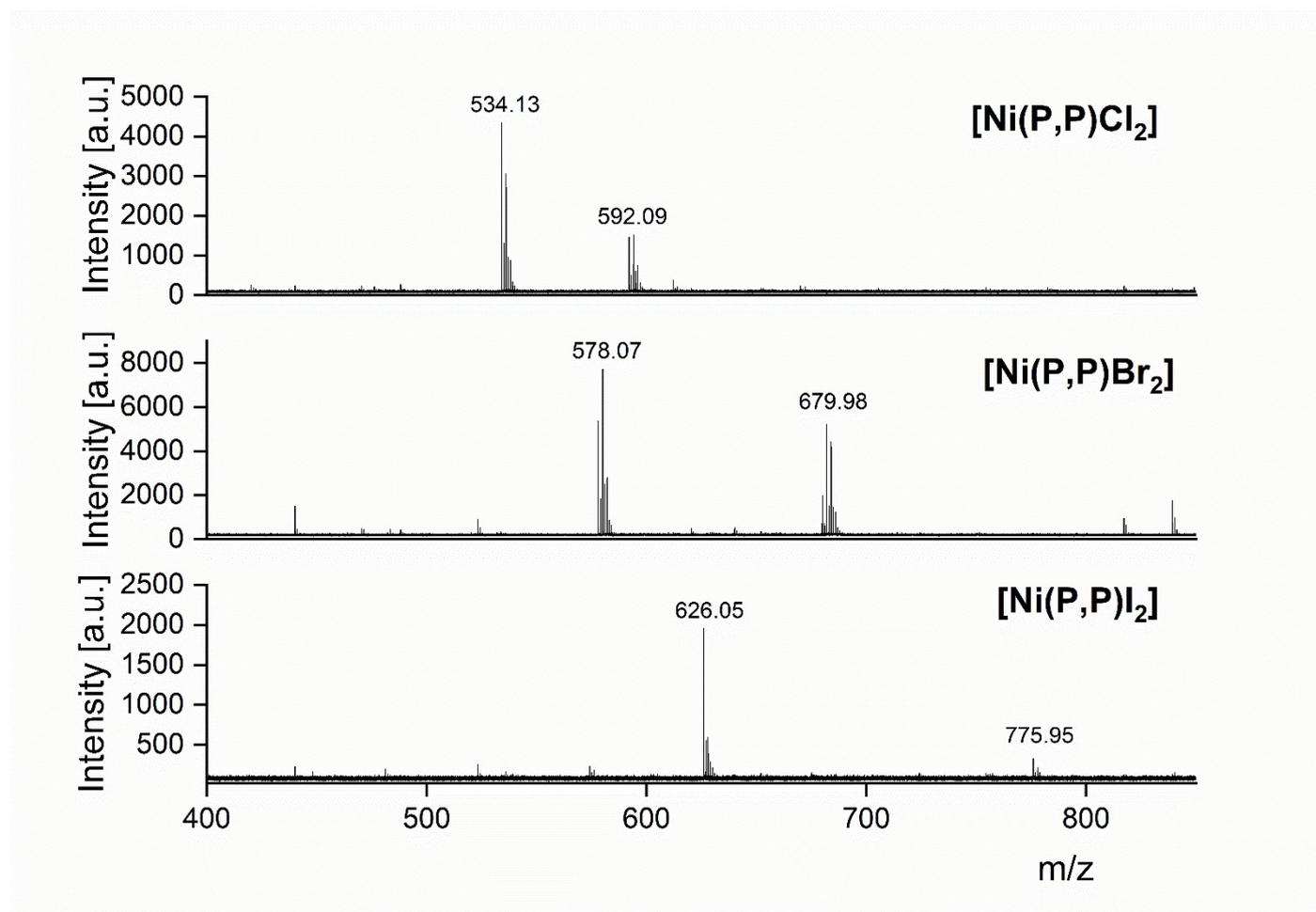


Figure S14. LDI-MS spectra of  $[\text{Ni}(\text{P},\text{P})\text{X}_2]$ .

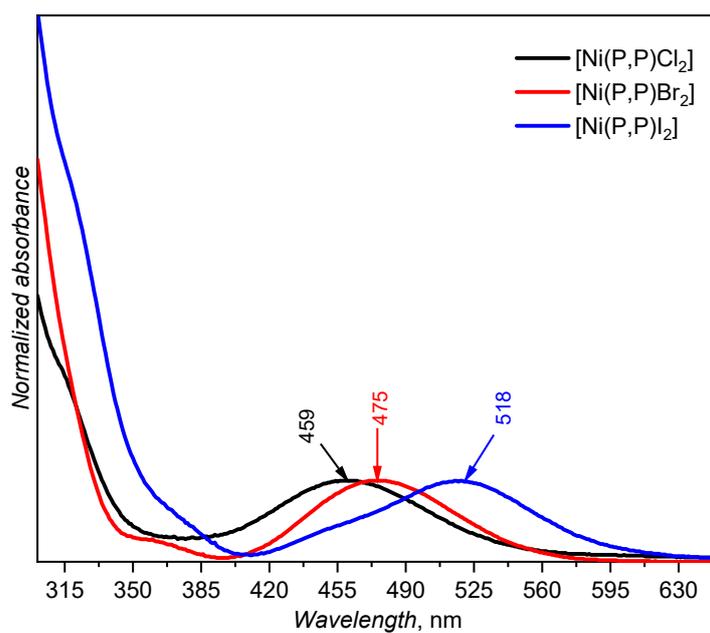


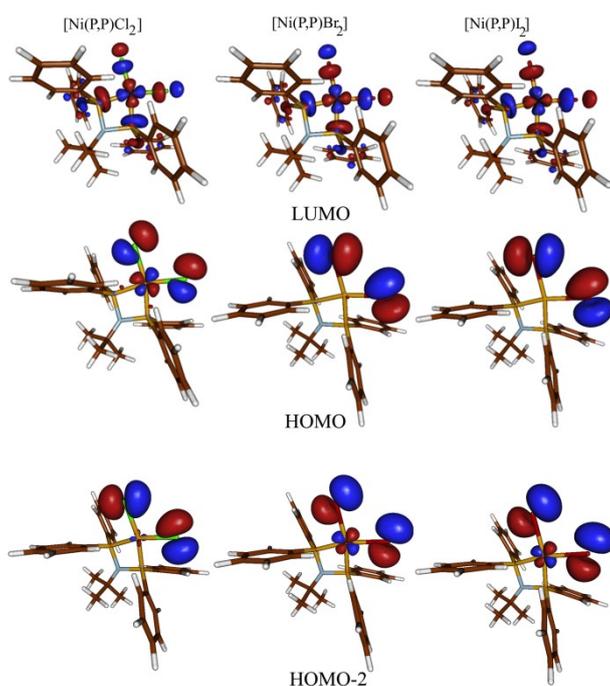
Figure S15. UV-vis spectra of complexes  $[\text{Ni}(\text{P},\text{P})\text{X}_2]$ , X = Cl (1), Br (2), I (3), in THF solution ( $\sim 10^{-5}$  M)

**Table S1.** Data for reactions of **4-Bu** substrate with Grignard reagent **T** catalyzed by complexes  $[\text{Ni}(\text{P},\text{P})\text{X}_2]$ .  $Y_{\text{CROSS}}$  and  $Y_{\text{HOMO}}$  stand for the substrate conversion by cross-coupling and homo-coupling, respectively (Scheme 2).

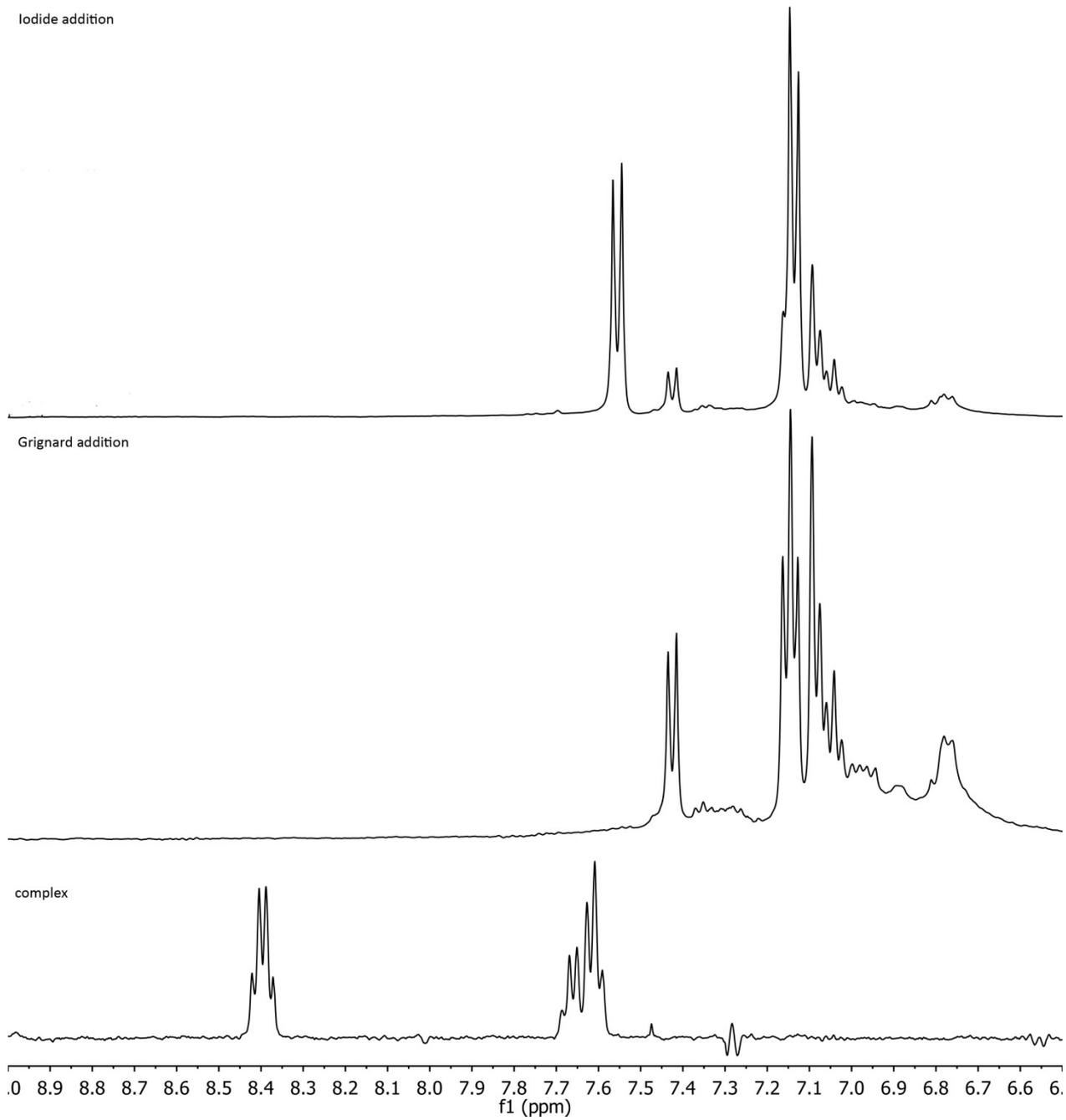
Reaction	$t$ , min	X = Cl		X = Br		X = I	
		$Y_{\text{CROSS}}$ %	$Y_{\text{HOMO}}$ %	$Y_{\text{CROSS}}$ %	$Y_{\text{HOMO}}$ %	$Y_{\text{CROSS}}$ %	$Y_{\text{HOMO}}$ %
<b>4-Bu + T</b>	20	76	< 2	1.0	< 2	1	< 2
	90	85	< 2	51	< 2	22	< 2
	1080	90	< 2	61	< 2	65	< 2

**Table S2.** Data for Kumada coupling reactions between substrates with shielded iodine atom and Grignard reagent **T** catalyzed by  $[\text{Ni}(\text{P},\text{P})\text{Cl}_2]$ .  $Y_{\text{CROSS}}$  and  $Y_{\text{HOMO}}$  stand for the substrate conversion by cross-coupling and homo-coupling, respectively (Scheme 2).

Catalyst	$t$ , min	<b>4-Bu</b>		<b>2-Me</b>		<b>2,6-diMe</b>	
		$Y_{\text{CROSS}}$ %	$Y_{\text{HOMO}}$ %	$Y_{\text{CROSS}}$ %	$Y_{\text{HOMO}}$ %	$Y_{\text{CROSS}}$ %	$Y_{\text{HOMO}}$ %
$[\text{Ni}(\text{P},\text{P})\text{Cl}_2]$	20	76	< 2	20	8	57	18
	90	85	< 2	73	12	58	24
	1080	90	< 2	76	15	62	26



**Figure S16.** The DFT-calculated LUMO, HOMO and HOMO-2 orbitals of complexes  $[\text{Ni}(\text{P},\text{P})\text{X}_2]$  showing contours at  $\pm 0.05$  a.u.



**Figure S17.** <sup>1</sup>H NMR spectra upon addition of first **T** and then **4-Bu** to a THF solution of [Ni(**P,P**)Cl<sub>2</sub>].

Reagent T - *p*-tolylMgBr  
4-tBu - 4-iodo-*t*-butylbenzene

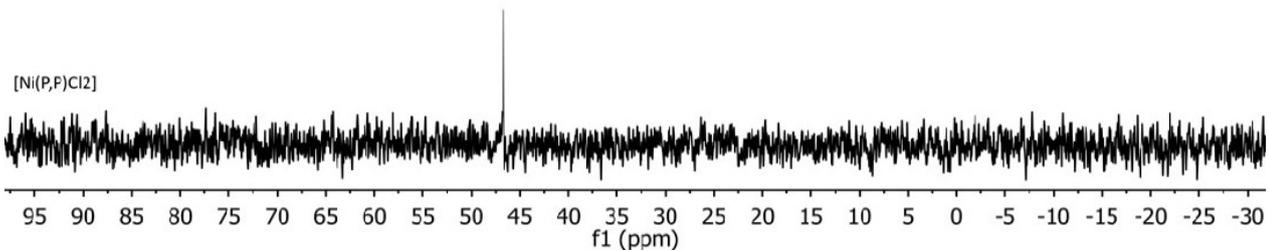
[Ni(P,P)Cl<sub>2</sub>] + Reagent T + 4-tBu added 2nd



[Ni(P,P)Cl<sub>2</sub>] + Reagent T added 1st



[Ni(P,P)Cl<sub>2</sub>]



**Figure S18.** <sup>31</sup>P NMR spectra upon upon addition of first T and then 4-**Bu** to a THF solution of [Ni(P,P)Cl<sub>2</sub>].