

## Supporting Information

### Self-supported Cu<sub>3</sub>P nanowires electrode as an efficient electrocatalyst for oxygen evolution reaction

Xin Zhou,<sup>a</sup> Xiaoliang Zhou<sup>\*a</sup>, Limin Liu<sup>\*a</sup>, Hanyu Chen,<sup>a</sup> Xingguo Hu<sup>a</sup>, Jiaqi Qian<sup>a</sup>, Di Huang<sup>a</sup>, Bo Zhang<sup>b</sup> and Junlei Tang<sup>\*a</sup>

<sup>a</sup> College of Chemistry and Chemical Engineering, Southwest Petroleum University, Chengdu, 610500, PR China.

<sup>b</sup> Hydrogen energy division, DONG FANG BOILER GROUP CO., LTD., Chengdu, 611731, PR China.

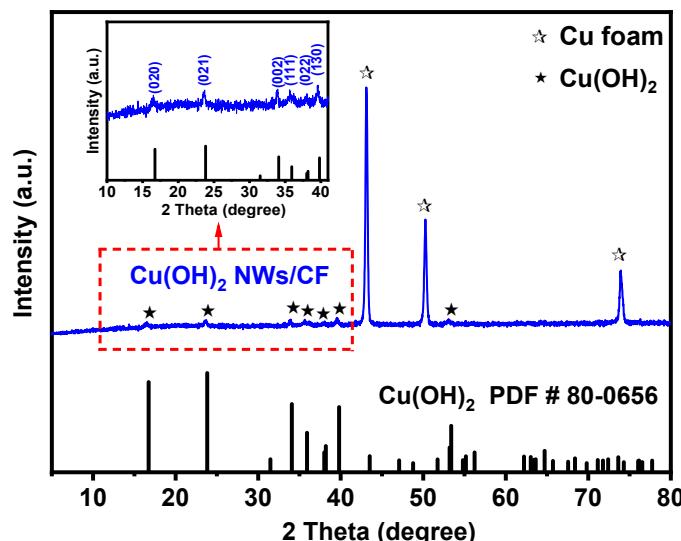


Figure S1. XRD patterns of Cu(OH)<sub>2</sub> NWs/CF.

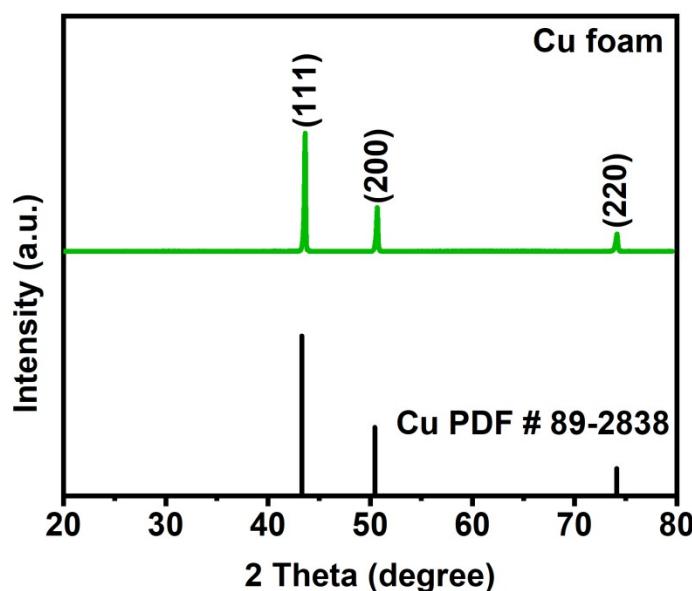
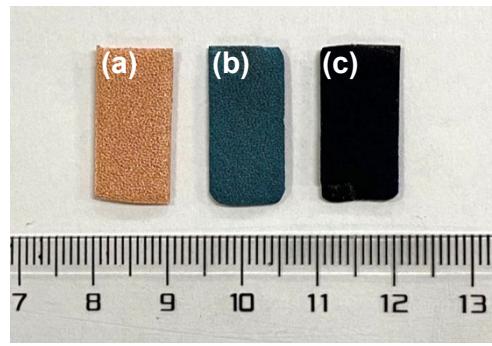
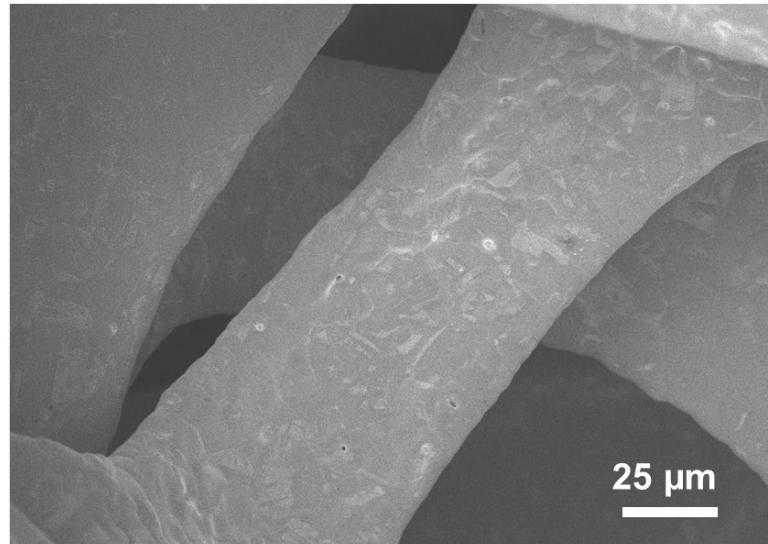


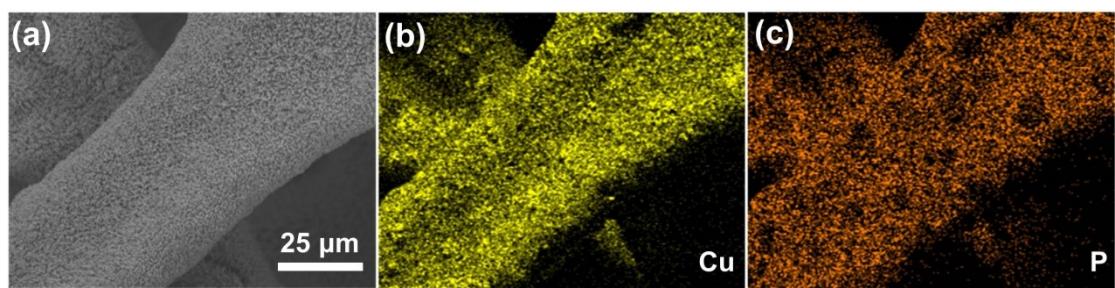
Figure S2. XRD patterns of bare Cu foam.



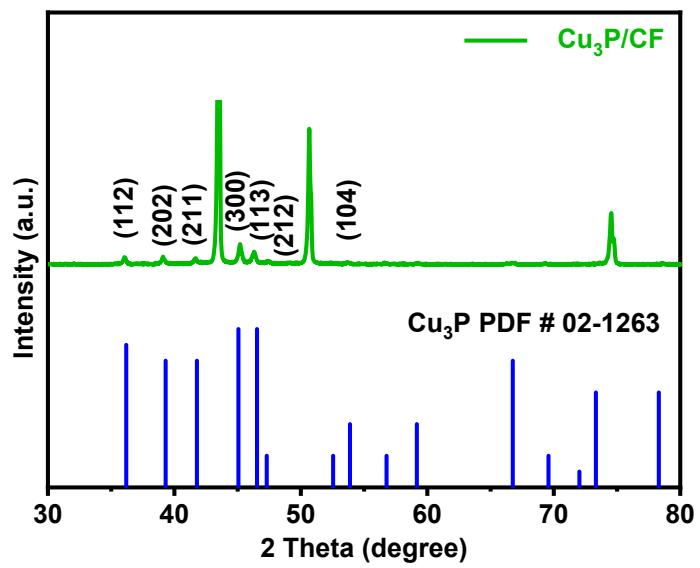
**Figure S3.** Optical photograph of (a) Cu foam, (b)  $\text{Cu}(\text{OH})_2$  NWs/CF, and (c)  $\text{Cu}_3\text{P}$  NWs/CF.



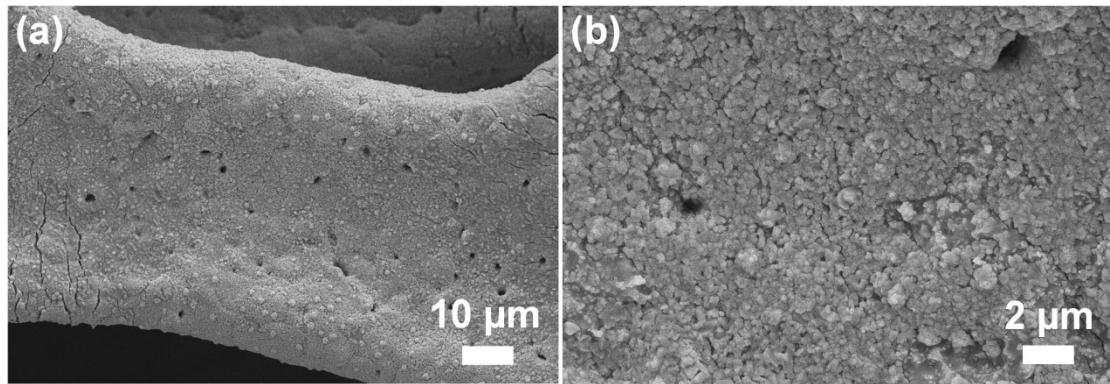
**Figure S4.** SEM image of bare Cu foam.



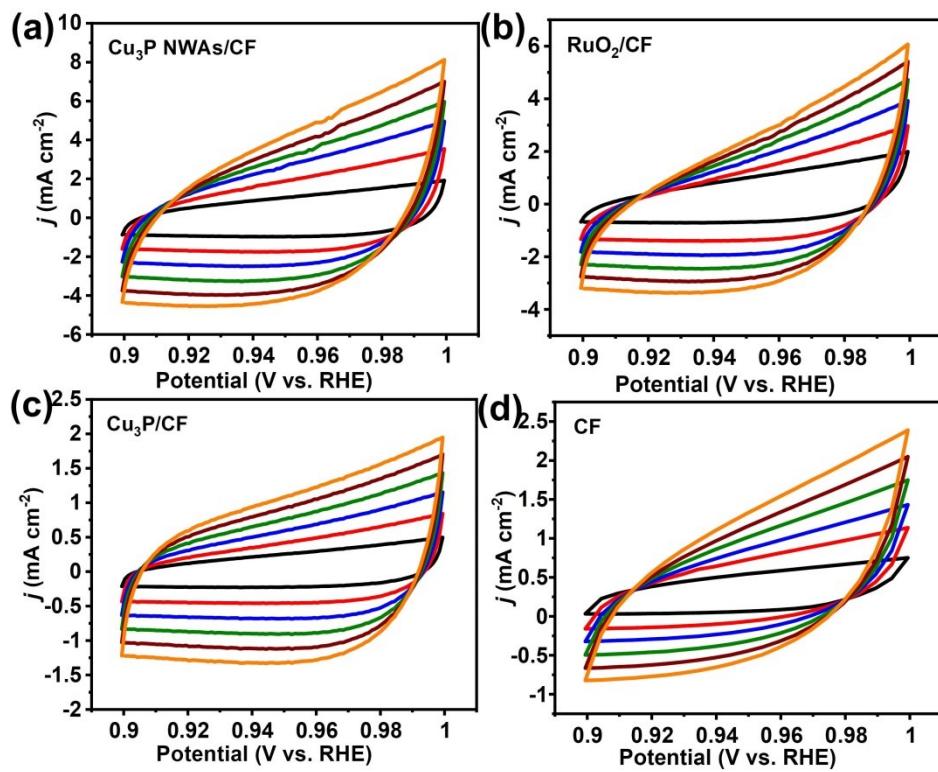
**Figure S5.** SEM-EDS elemental mapping images of  $\text{Cu}_3\text{P}$  NWs/CF.



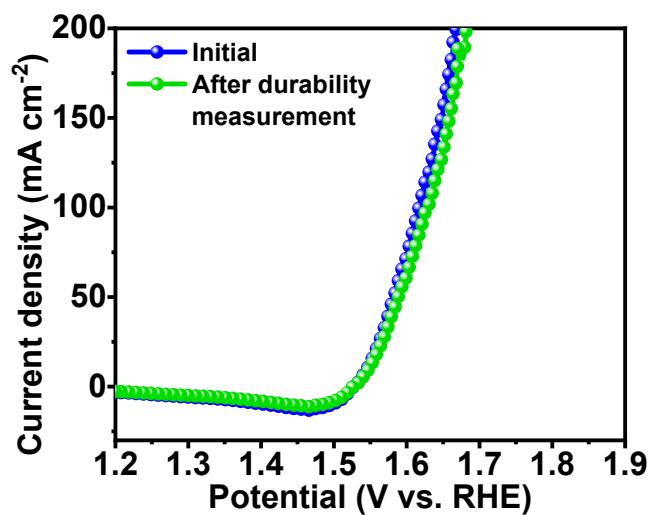
**Figure S6.** XRD patterns of  $\text{Cu}_3\text{P/CF}$ .



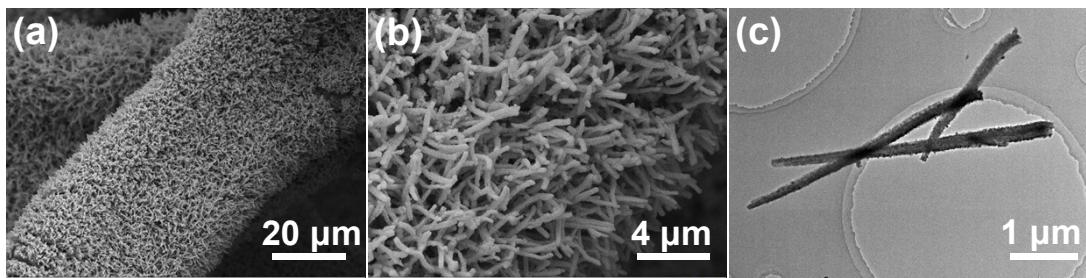
**Figure S7.** (a) Low-magnification and (b) high-magnification SEM images of  $\text{Cu}_3\text{P/CF}$ .



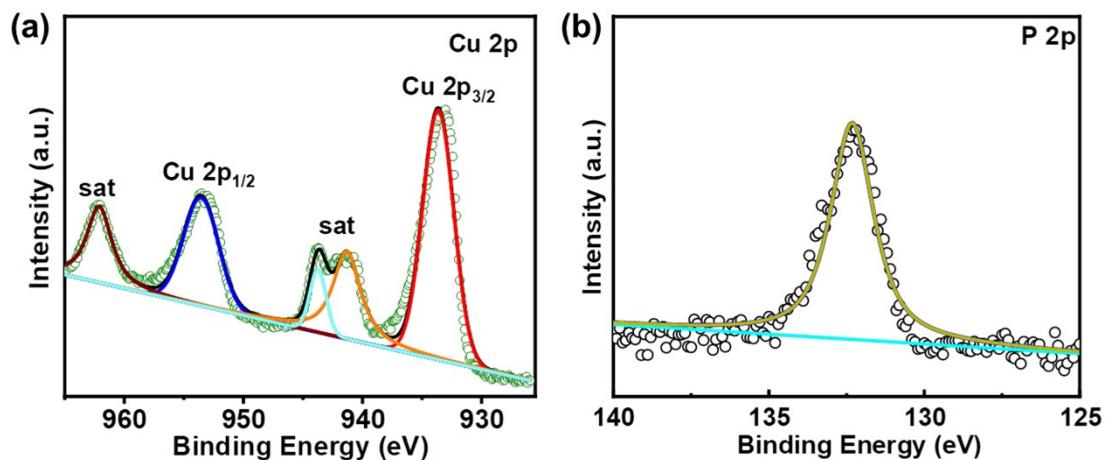
**Figure S8.** Cyclic voltammograms of (a)  $\text{Cu}_3\text{P}$  NWs/CF, (b)  $\text{RuO}_2/\text{CF}$ , (c)  $\text{Cu}_3\text{P}/\text{CF}$ , and (d) Cu foam.



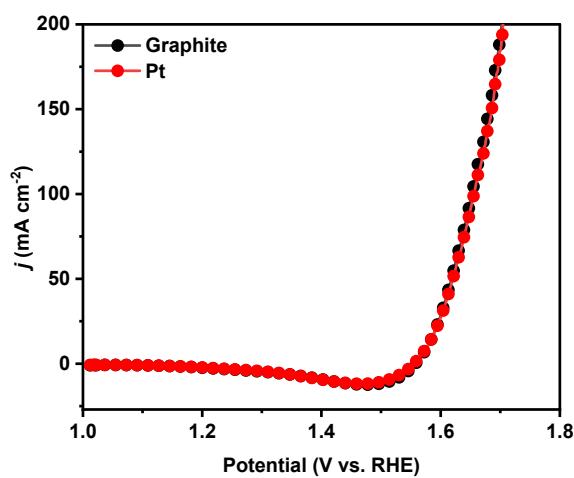
**Figure S9.** LSV curves of  $\text{Cu}_3\text{P}$  NWs/CF before and after long-term durability measurement.



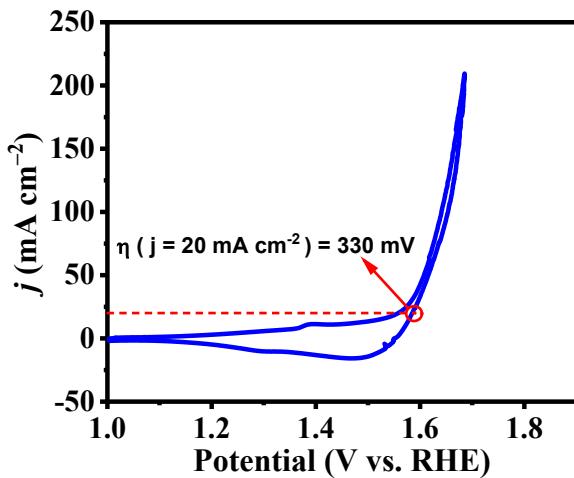
**Figure S10.** (a-b) SEM images and (c) TEM image of  $\text{Cu}_3\text{P}$  NWs/CF after long-term durability measurement.



**Figure S11.** High-resolution XPS spectra of (a) Cu 2p and (b) P 2p for  $\text{Cu}_3\text{P}$  NWs/CF after long-term durability measurement.



**Figure S12.** The comparison of LSV curves with platinum and graphite rods as counter electrodes for  $\text{Cu}_3\text{P}$  NWs/CF.



**Figure S13.** CV curve at a scan rate of  $1 \text{ mV s}^{-1}$  for  $\text{Cu}_3\text{P}$  NWs/CF.

**Table S1.** Comparison of OER performances of  $\text{Cu}_3\text{P}$  NWs/CF and other reported electrocatalysts.

Catalyst	Substrate	Electrolyte	$j (\text{mA cm}^{-2})$	Onerpotential (mV)	Reference
$\text{Cu}_3\text{P}$ NWs/CF	Cu foam	1.0 M KOH	10	316	This work
			20	327	
			50	352	
$\text{Cu}_3\text{P}$ NB/Cu	Cu mesh	1.0 M KOH	10	380	1
			20	390	
$\text{NiCoP}@\text{Cu}_3\text{P}$	Cu foam	1.0 M KOH	10	309	2
$\text{Cu}_3\text{P}$ nanosheets	Ni foam	1.0 M KOH	10	320	3
$\text{Ni}_3\text{S}_4$ nanosheets	Ni foam	1.0 M KOH	20	310	4
$\text{NiMoP}_2$ nanowire	Carbon cloth	1.0 M KOH	10	330	5
$\text{NiCo-LDH}$ nanosheets	Carbon paper	1.0 M KOH	20	390	6
$\text{NiO}/\text{NiFe}_2\text{O}_4$	Carbon paper	1.0 M KOH	10	303	7
NiCo-Mixed Oxide	GCE	1.0 M KOH	10	380	8
CoMnP nanoparticles	GCE	1.0 M KOH	10	330	9
$\text{IrO}_2$	GCE	1.0 M KOH	10	320	10

Note: GCE stands for glassy carbon electrode.

**Table S2.** The electrolyte resistance ( $R_s$ ) and charge transfer resistance ( $R_{ct}$ ) of various samples.

Catalyst	$R_s$ (ohm)	$R_{ct}$ (ohm)
<b>Cu<sub>3</sub>P NWs/CF</b>	1.6	3.5
<b>RuO<sub>2</sub>/CF</b>	1.6	4.5
<b>Cu<sub>3</sub>P/CF</b>	1.6	9.5
<b>Cu foam</b>	1.7	14.5

## References

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