

Supporting Information

Facile synthesis of nitrogen-defective g-C₃N₄ for superior photocatalytic degradation of rhodamine B

Xiupei Yang,^{*a} Lin Zhang,^a Dan Wang,^a Qian Zhang,^a Jie Zeng, Run Zhang,^{*b}

*^aCollege of Chemistry and Chemical Engineering, Chemical Synthesis and Pollution Control
Key Laboratory of Sichuan Province, China West Normal University, Nanchong 637000,
China. E-mail: xiupeiyang@163.com*

*^bAustralian Institute for Bioengineering and Nanotechnology, The University of Queensland,
Brisbane, Queensland 4072, Australia. E-mail: r.zhang@uq.edu.au*

*Corresponding author

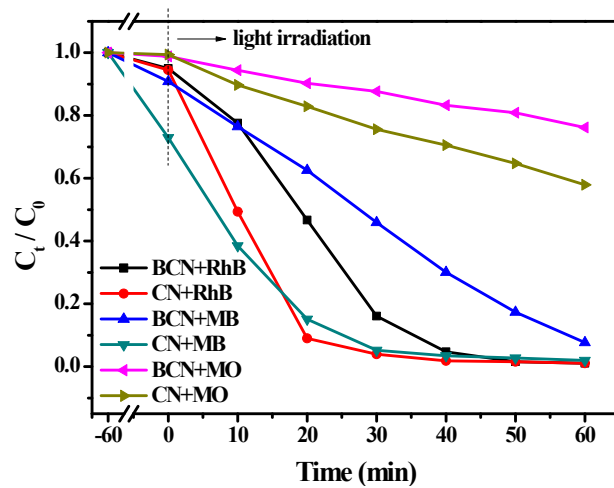


Fig. S1. Degradation curves of RhB, MO and MB by BCN and CN.

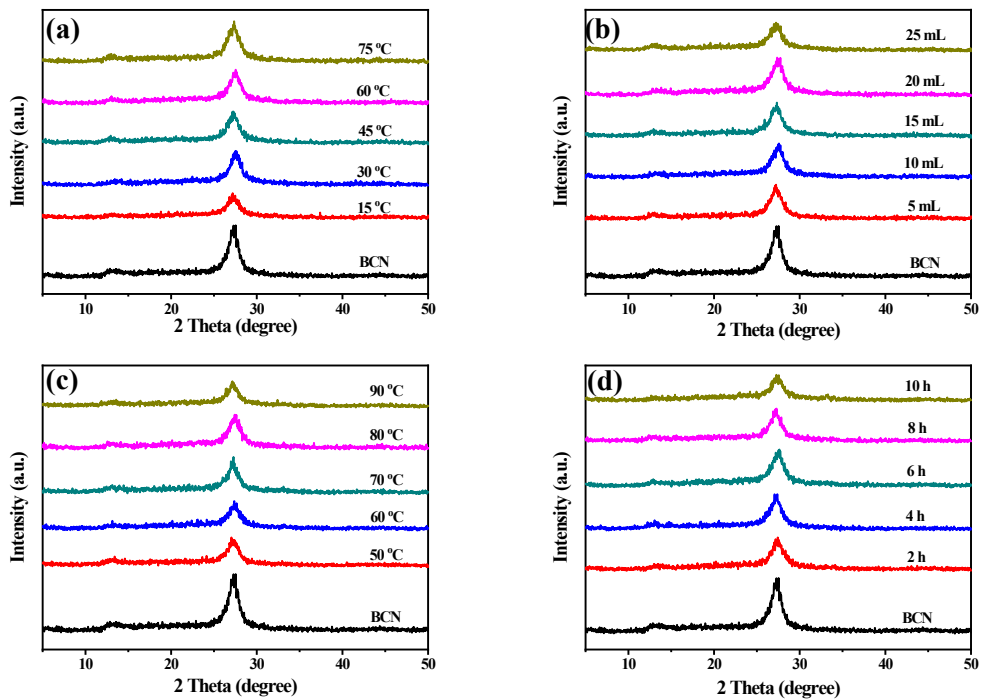


Fig. S2. XRD patterns of g-C₃N₄ synthesized under different modification conditions: (a) different water treatment temperature, (b) different water treatment consumption, (c) different precursor drying temperature, and (d) different precursor drying time.

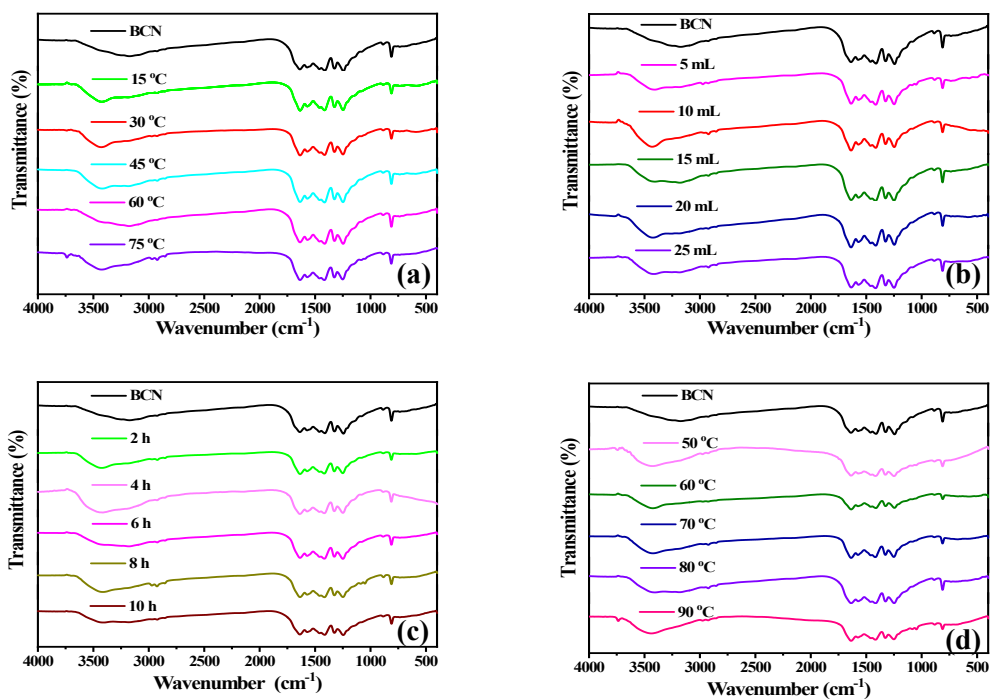


Fig. S3. FT-IR spectra of g-C₃N₄ under different modification conditions: (a) different water treatment temperature, (b) different water treatment consumption, (c) different precursor drying temperature, and (d) different precursor drying time.

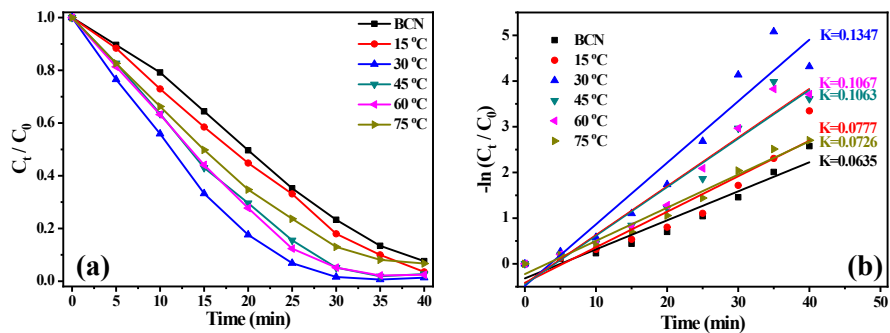


Fig. S4. The g-C₃N₄ material prepared by optimizing the precursor melamine different water treatment temperature for RhB degradation: (a) the degradation curve, (b) the first-order kinetics fitting curve.

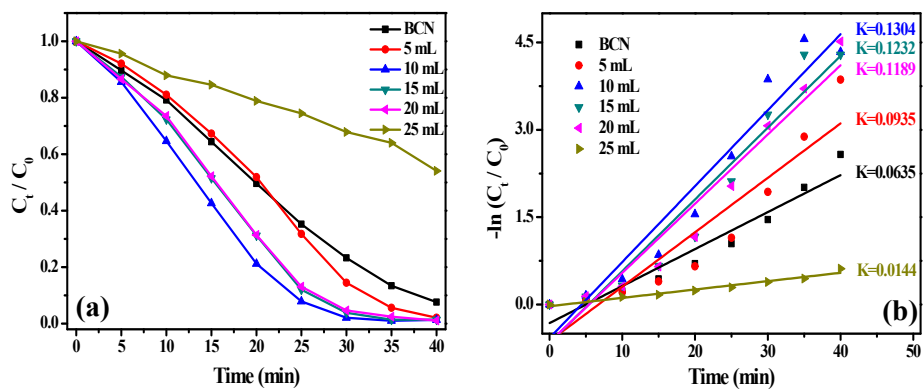


Fig. S5. The g-C₃N₄ material prepared by optimizing the precursor melamine different water treatment consumption for RhB degradation: (a) the degradation curve, (b) the first-order kinetics fitting curve.

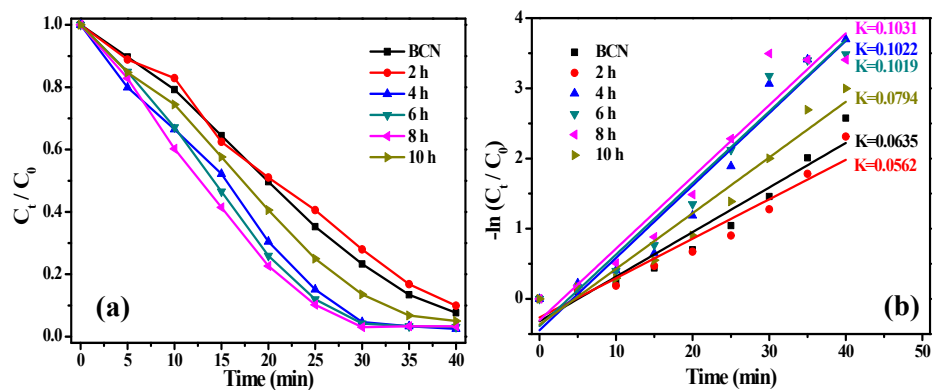


Fig. S6. The g-C₃N₄ material prepared by optimizing the precursor melamine different drying time for RhB degradation: (a) the degradation curve, (b) the first-order kinetics fitting curve.

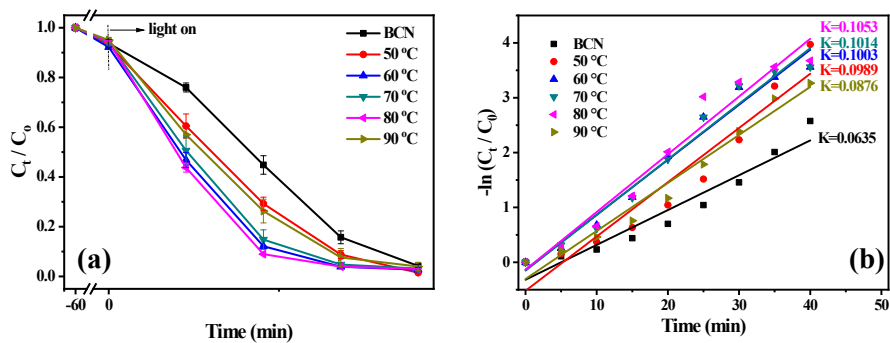


Fig. S7. The g-C₃N₄ material prepared by optimizing the precursor melamine different drying temperature for RhB degradation: (a) the degradation curve, (b) the first-order kinetics fitting curve.

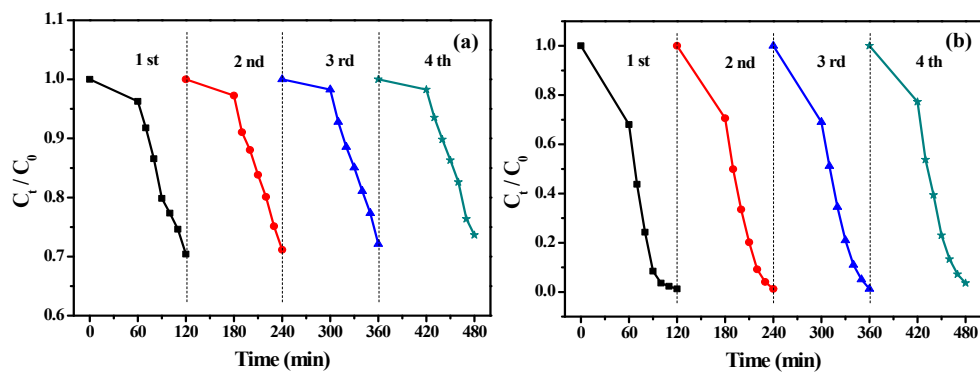


Fig. S8. Degradation cycle curves of MO(a) and MB(b) by CN.

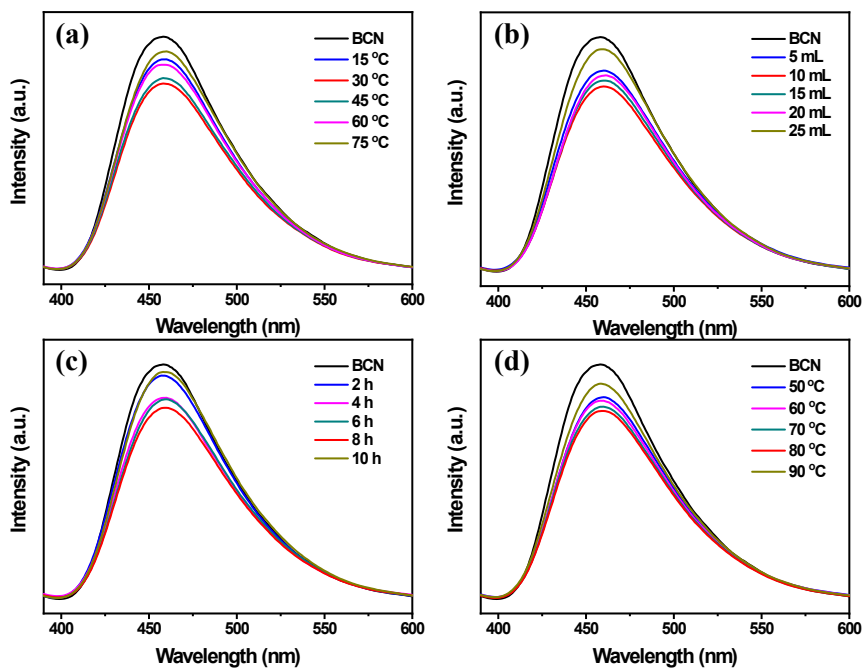


Fig. S9. PL spectrum of g-C₃N₄ synthesized under different modification conditions: (a) Different water treatment temperature, (b) Different water treatment consumption, (c) Different precursor drying time, (d) Different precursor drying temperature.

Table S1. EDS elemental analysis of BCN and CN.

Element	C	N	C/N
BCN	51.30 %	48.70 %	1.053
CN	63.41 %	35.59 %	1.782