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## **Supplemental materials**

## **Figures**

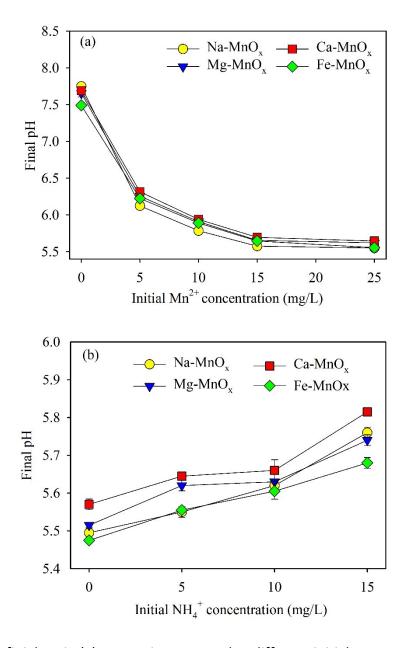


Fig.S1 The finial pH in (a) ammonium removal at different initial concentration of  $Mn^{2+}$  and (b)  $Mn^{2+}$  removal at different initial concentration of  $NH_4^+$ .

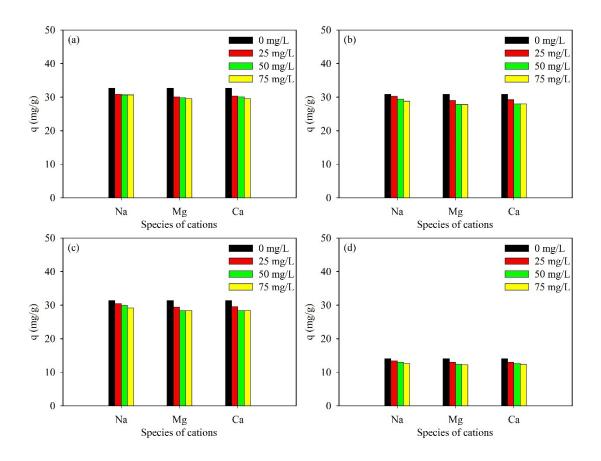


Fig.S2 Effects of different cations on  $Mn^{2+}$  adsorption of (a)Na-MnO<sub>x</sub>, (b)Mg-MnO<sub>x</sub>, (c)Ca-MnO<sub>x</sub> and (d)Fe-MnO<sub>x</sub>.

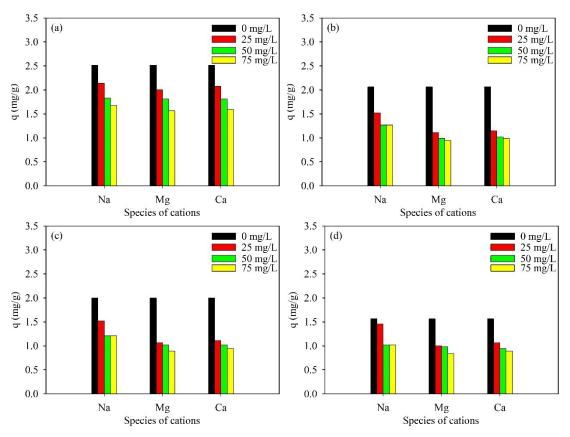
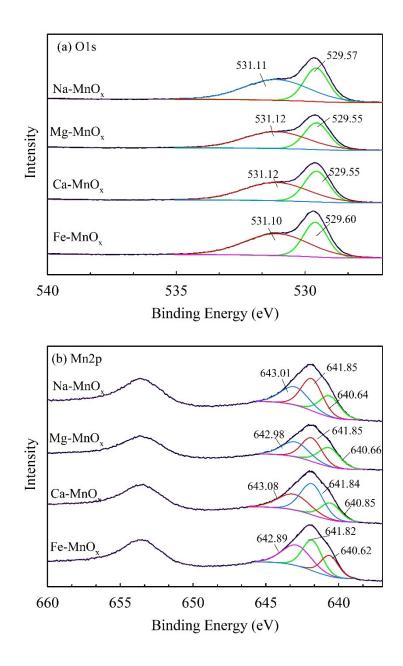


Fig.S3 Effects of cations on  $NH_4^+$  adsorption of (a)Na-MnO<sub>x</sub>, (b)Mg-MnO<sub>x</sub>, (c)Ca-MnO<sub>x</sub> and (d)Fe-MnO<sub>x</sub>.



 $\begin{tabular}{ll} \textbf{Fig.S4} XPS profiles of O1s and Mn2p in manganese oxides after the reaction: (a) O1s, (b)Mn2p \end{tabular}$ 

Table S1 Atomic percentage of different species of O and Mn in mangang

**Table S1** Atomic percentage of different species of O and Mn in manganese oxides after the reaction

Species of	Species of O (%)			Species of Mn (%)		
the oxides	O <sub>latt</sub>	$O_ad$	Os	IV	III	II
Na-MnO <sub>x</sub>	37.33	62.67	/	31.21	41.99	26.8
Mg-MnO <sub>x</sub>	37.70	62.30	/	31.36	36.80	31.84
Ca-MnO <sub>x</sub>	38.90	61.10	/	32.31	45.85	21.83
Fe-MnO <sub>x</sub>	36.76	63.24	/	38.31	38.06	23.63