

Electronic Supplementary Material (ESI) for RSC Advances.

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Does electrophilic activation of nitroalkanes in polyphosphoric acid involves formation of nitrile oxides?

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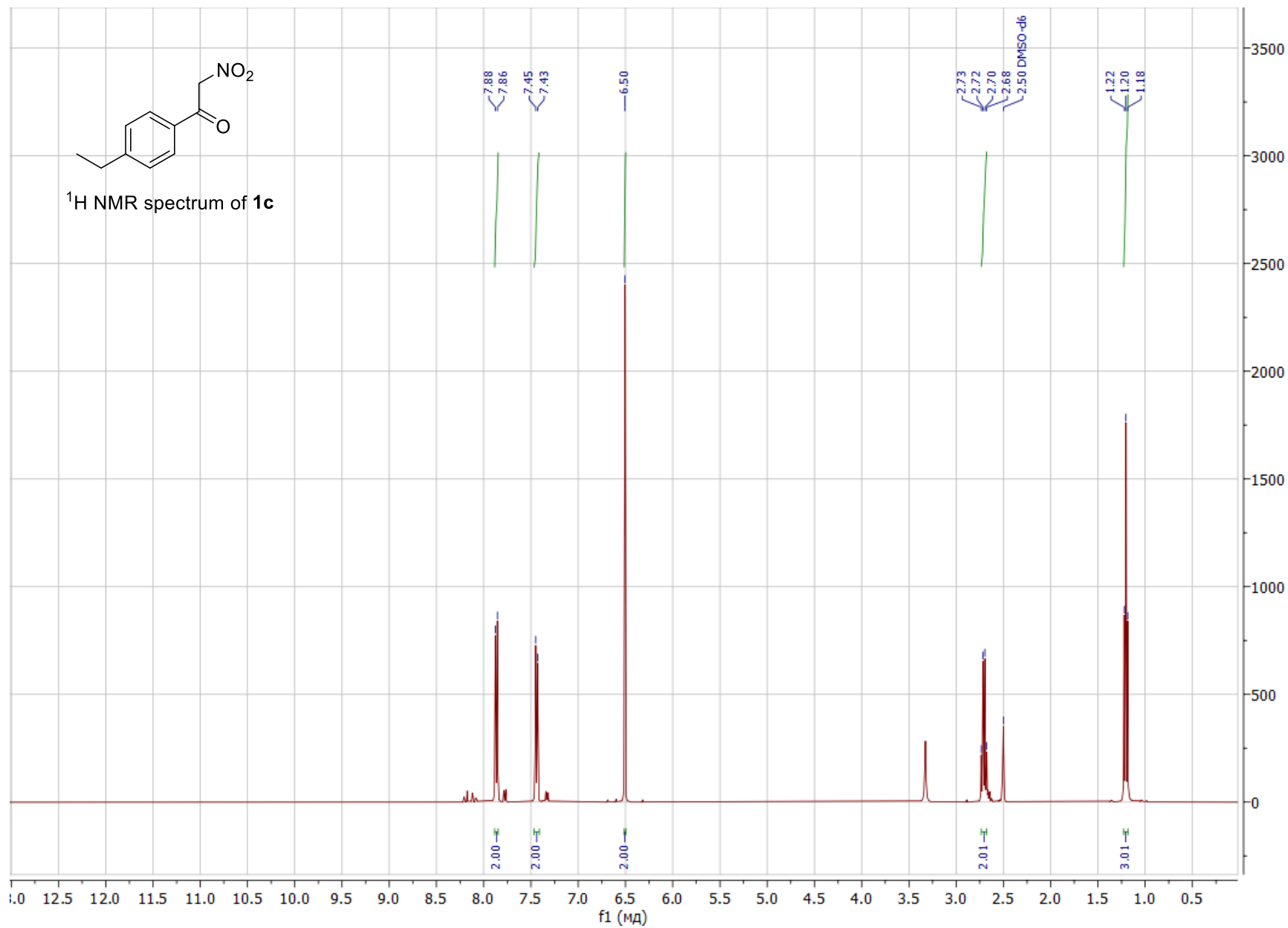
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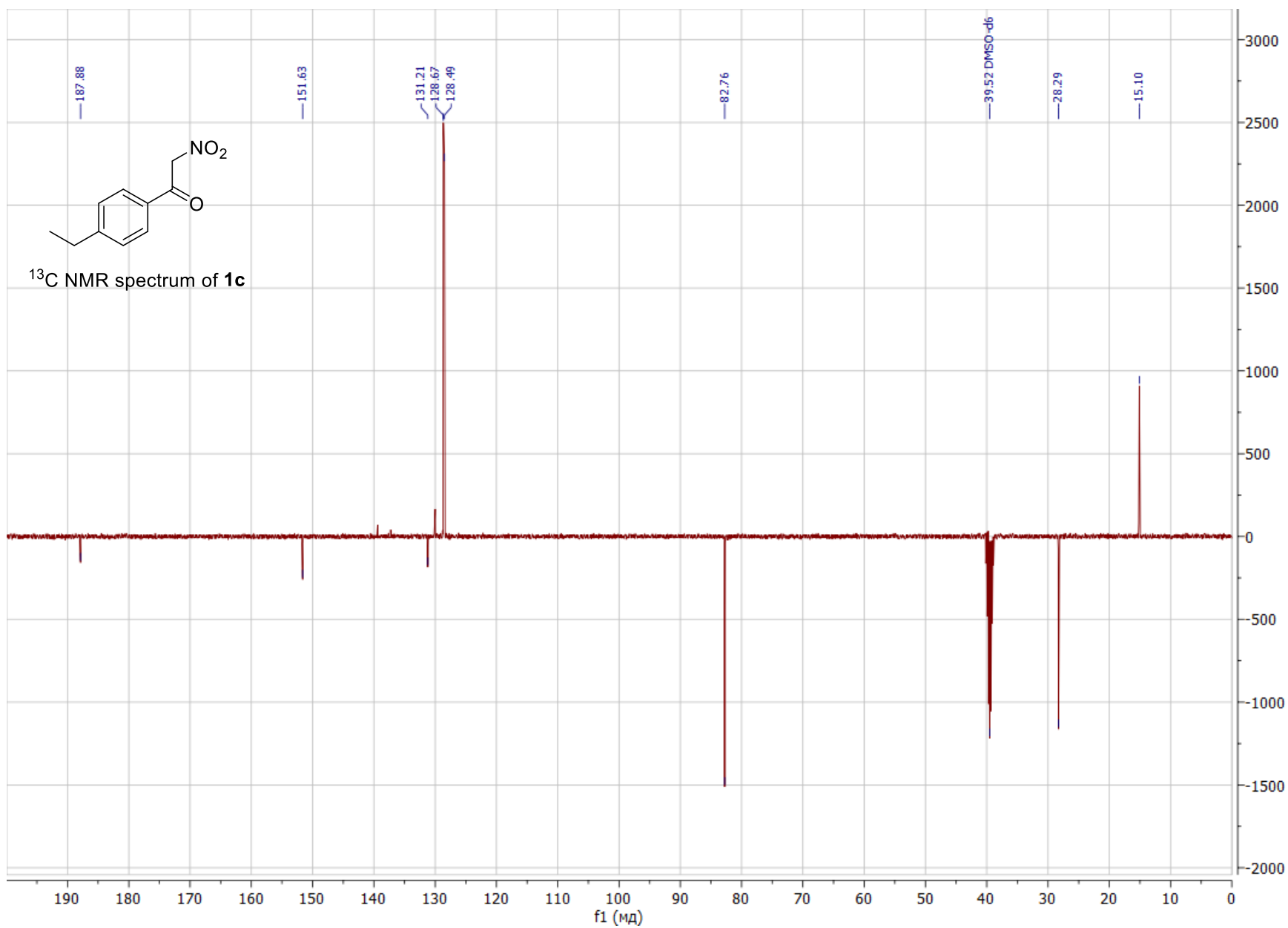
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Supporting Information

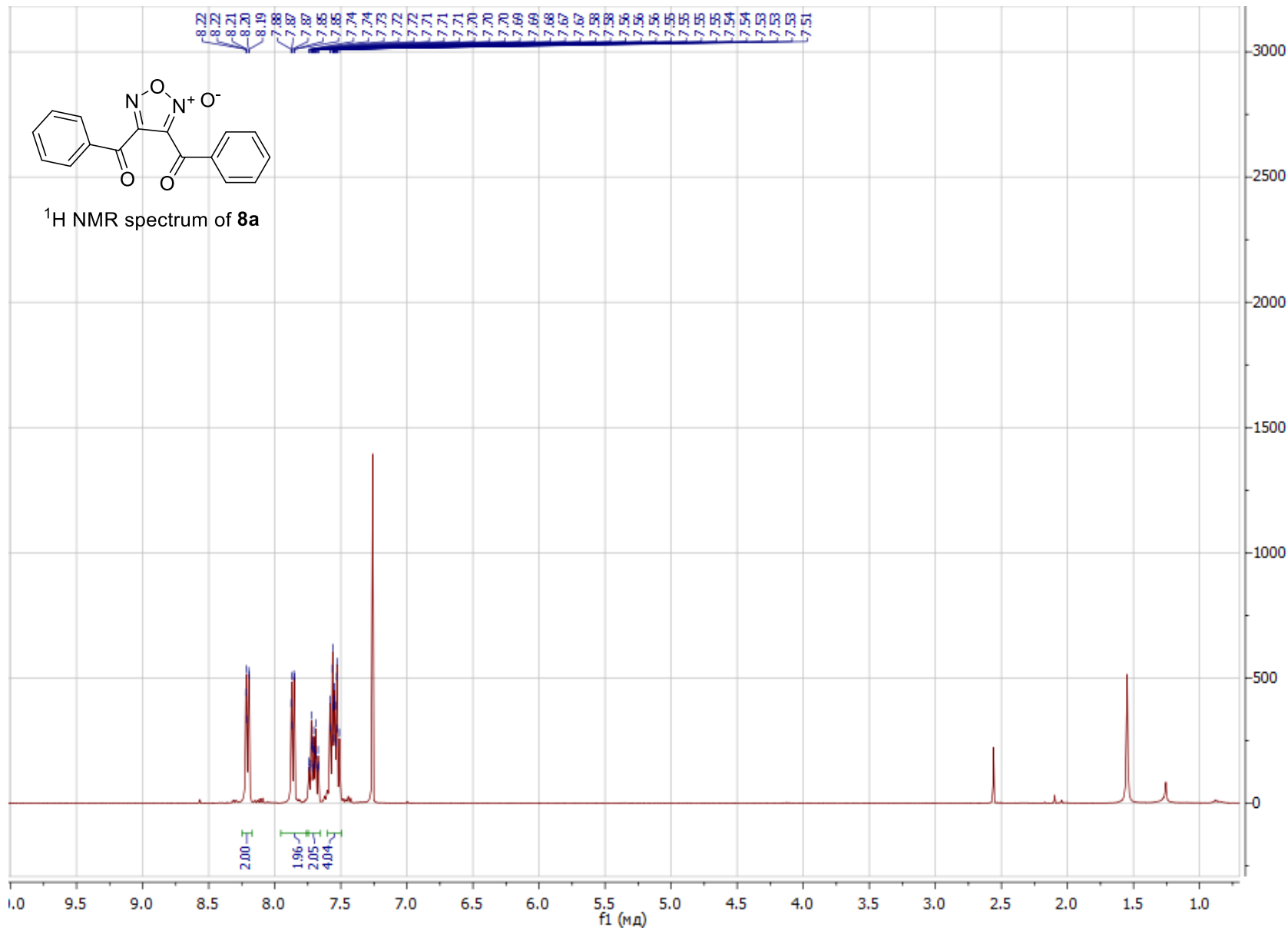
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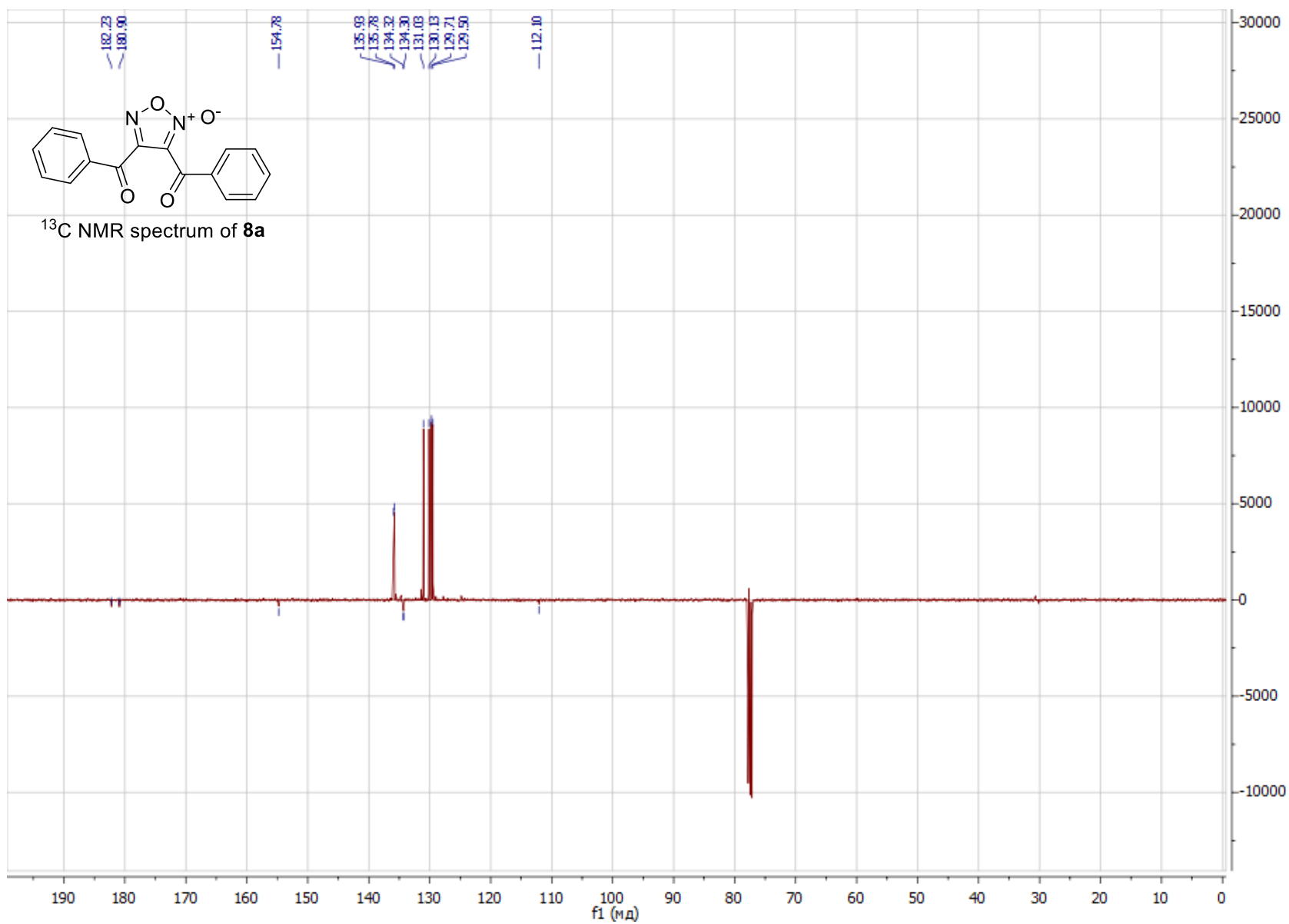
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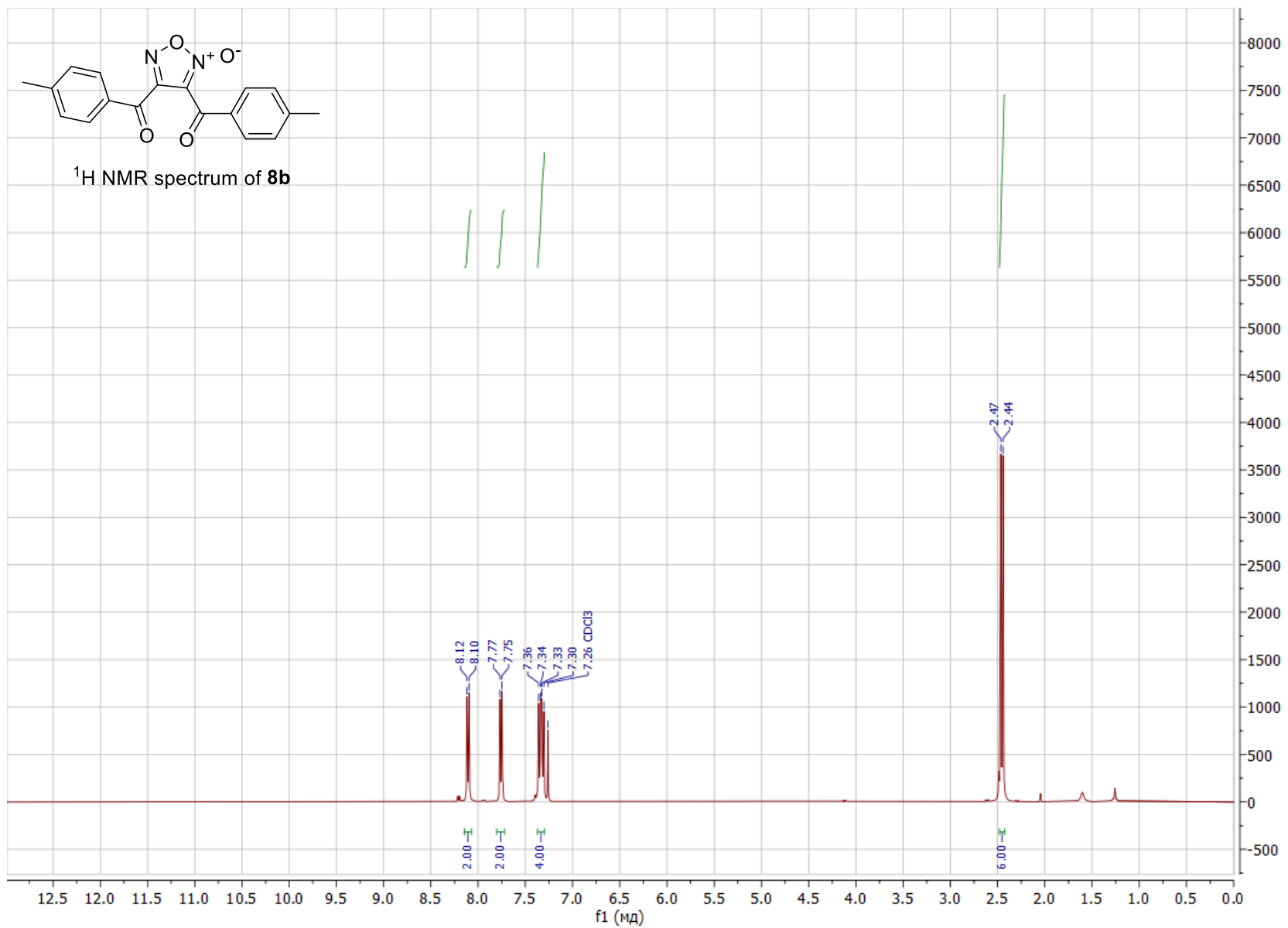


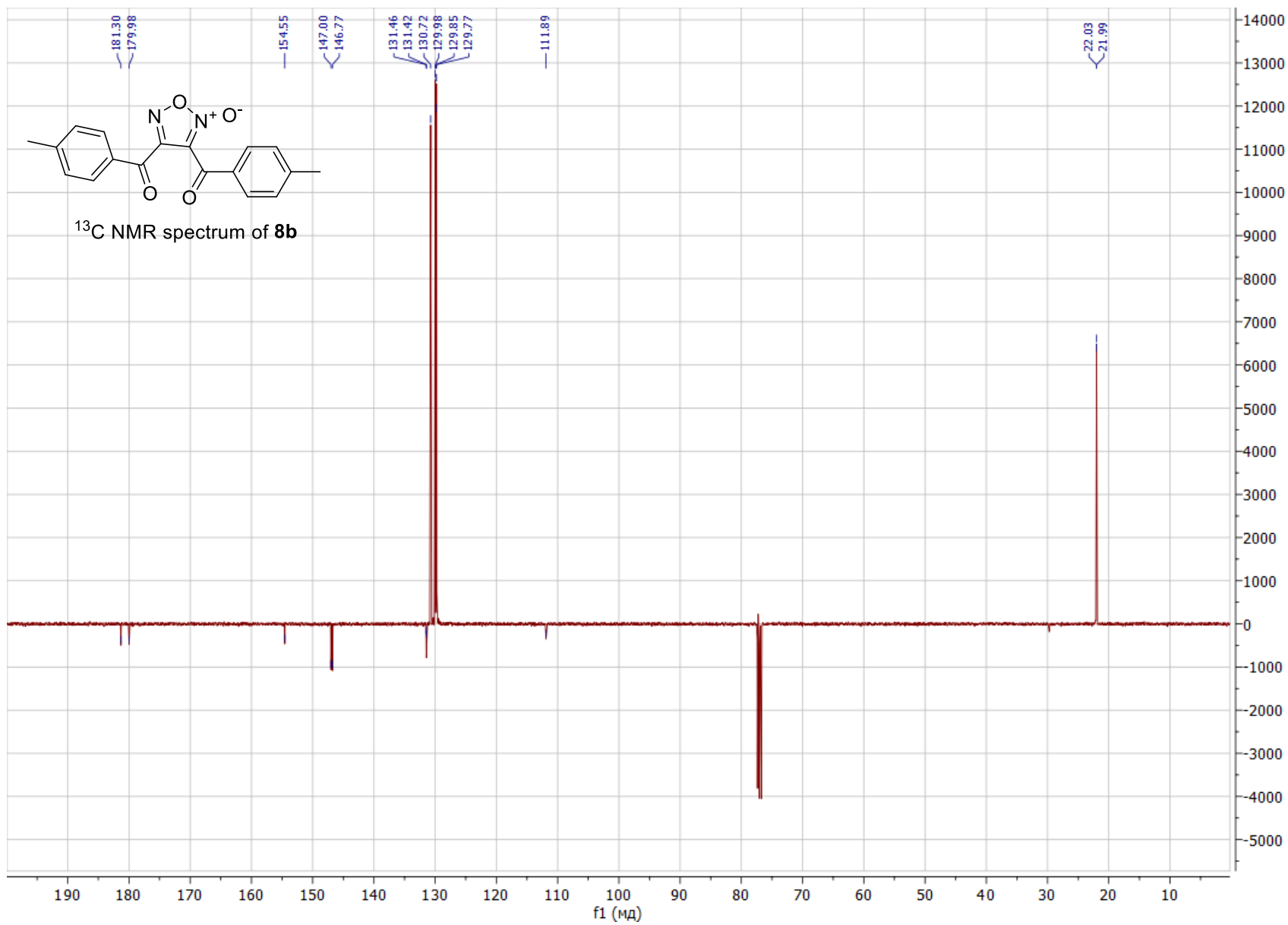


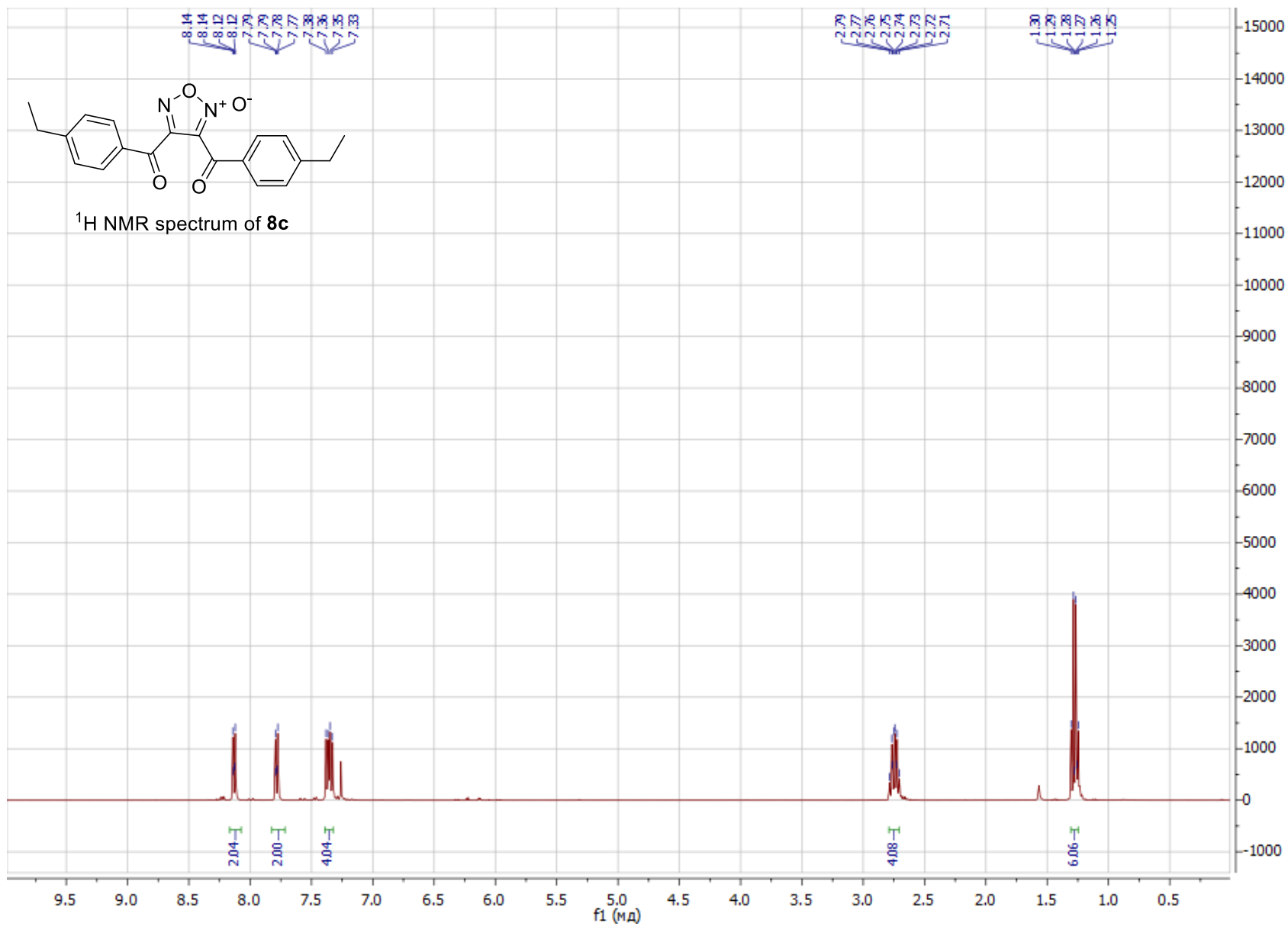
¹H and ¹³C NMR spectral charts for furoxans 8

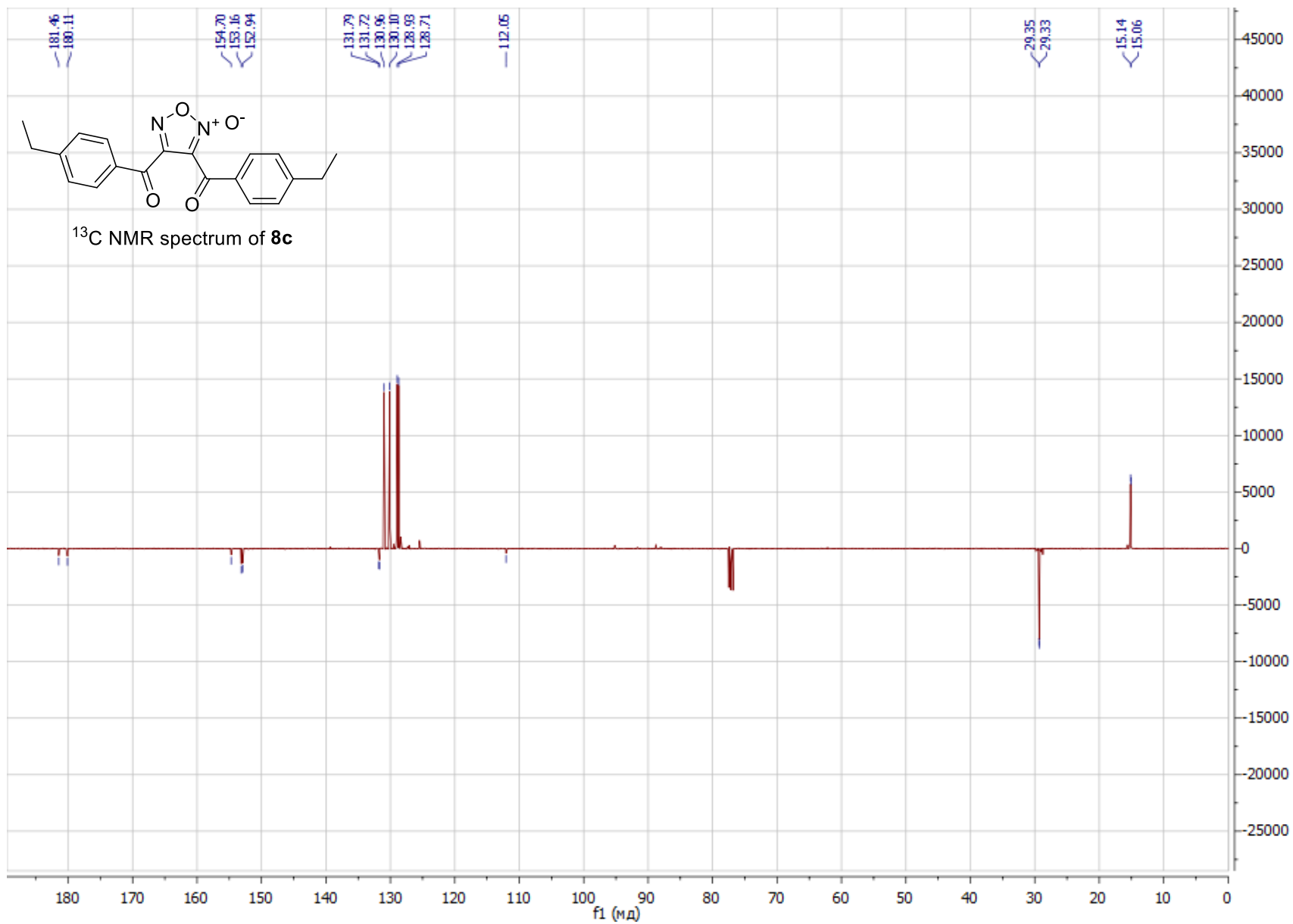


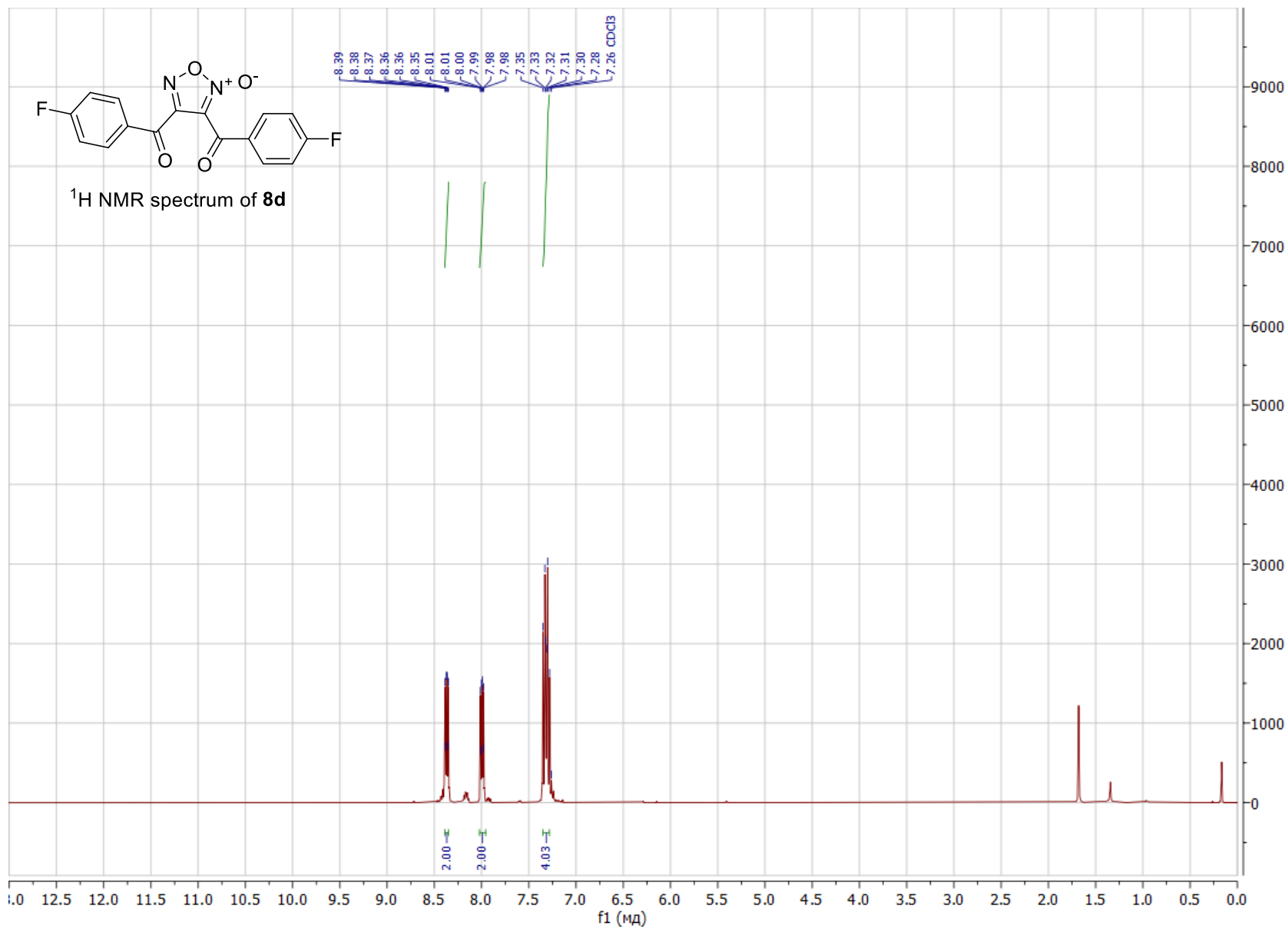


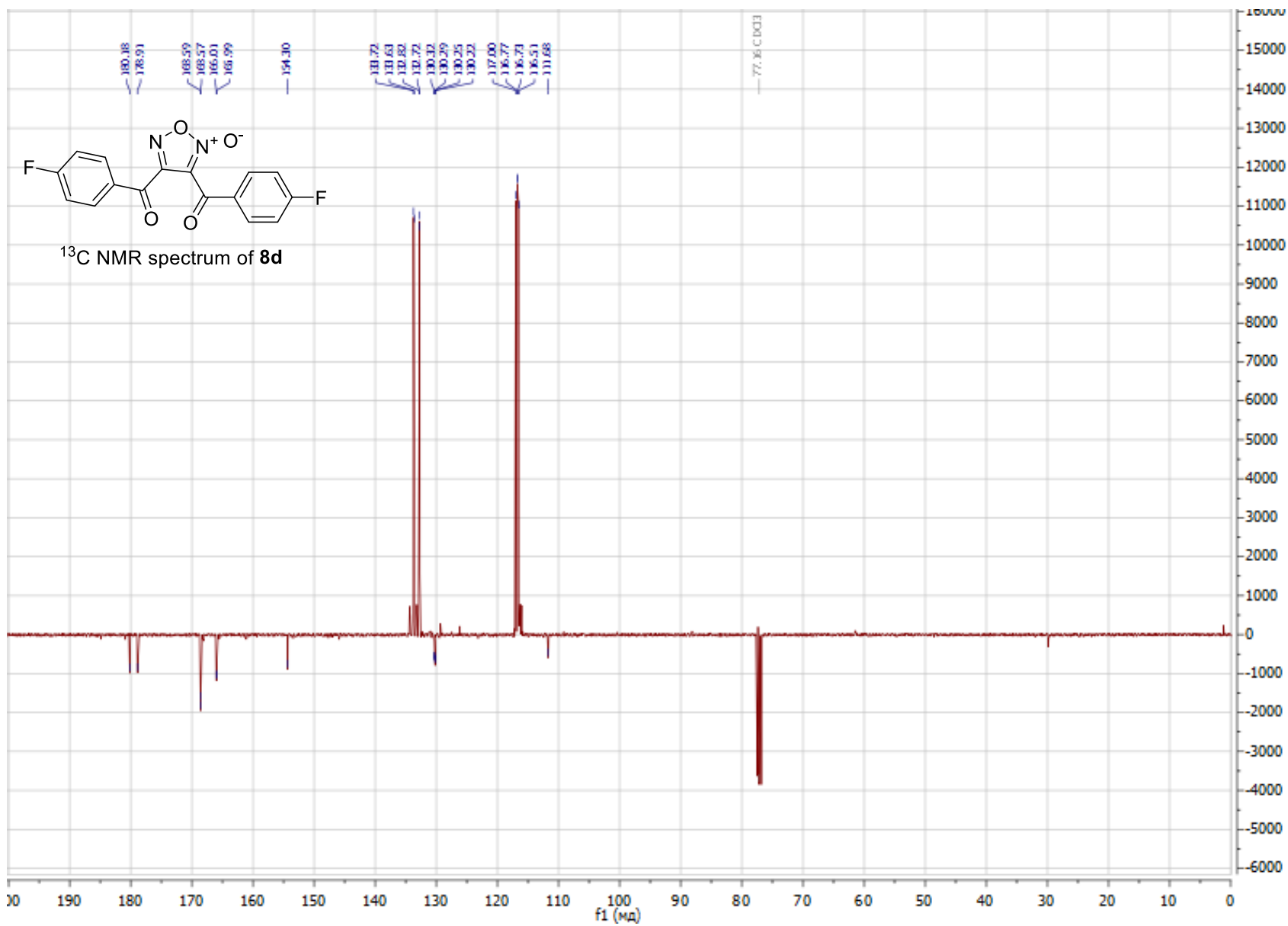


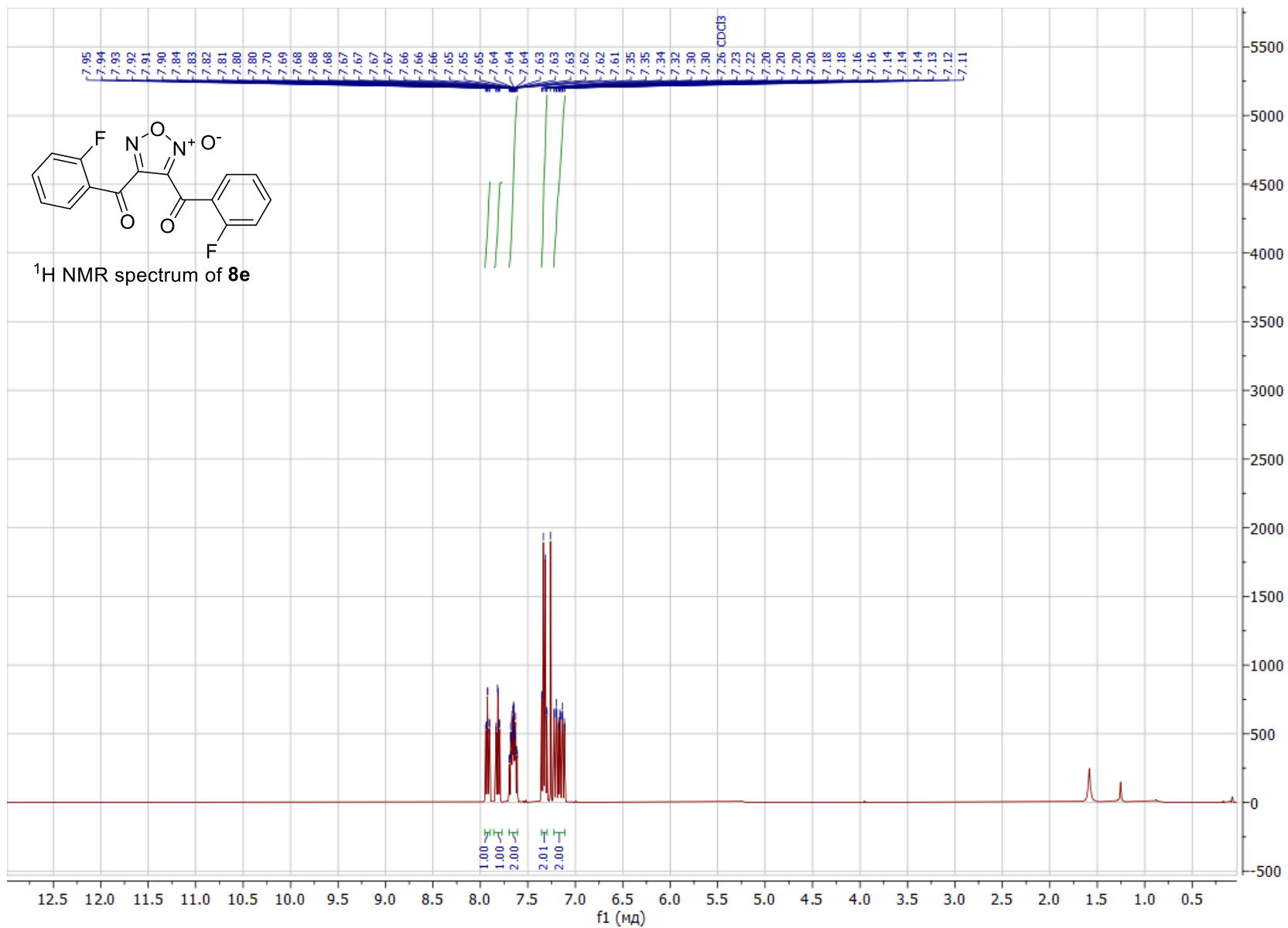


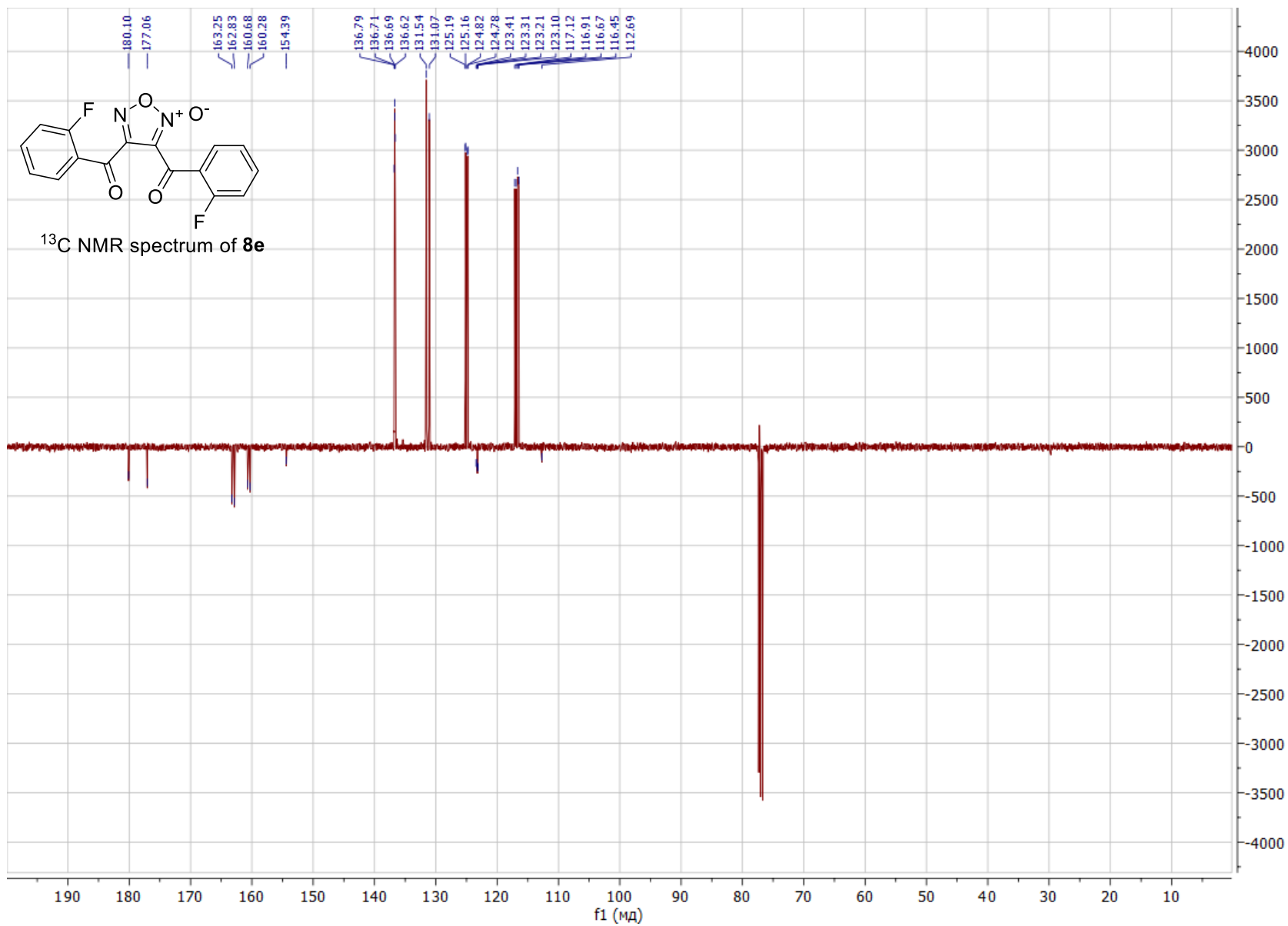


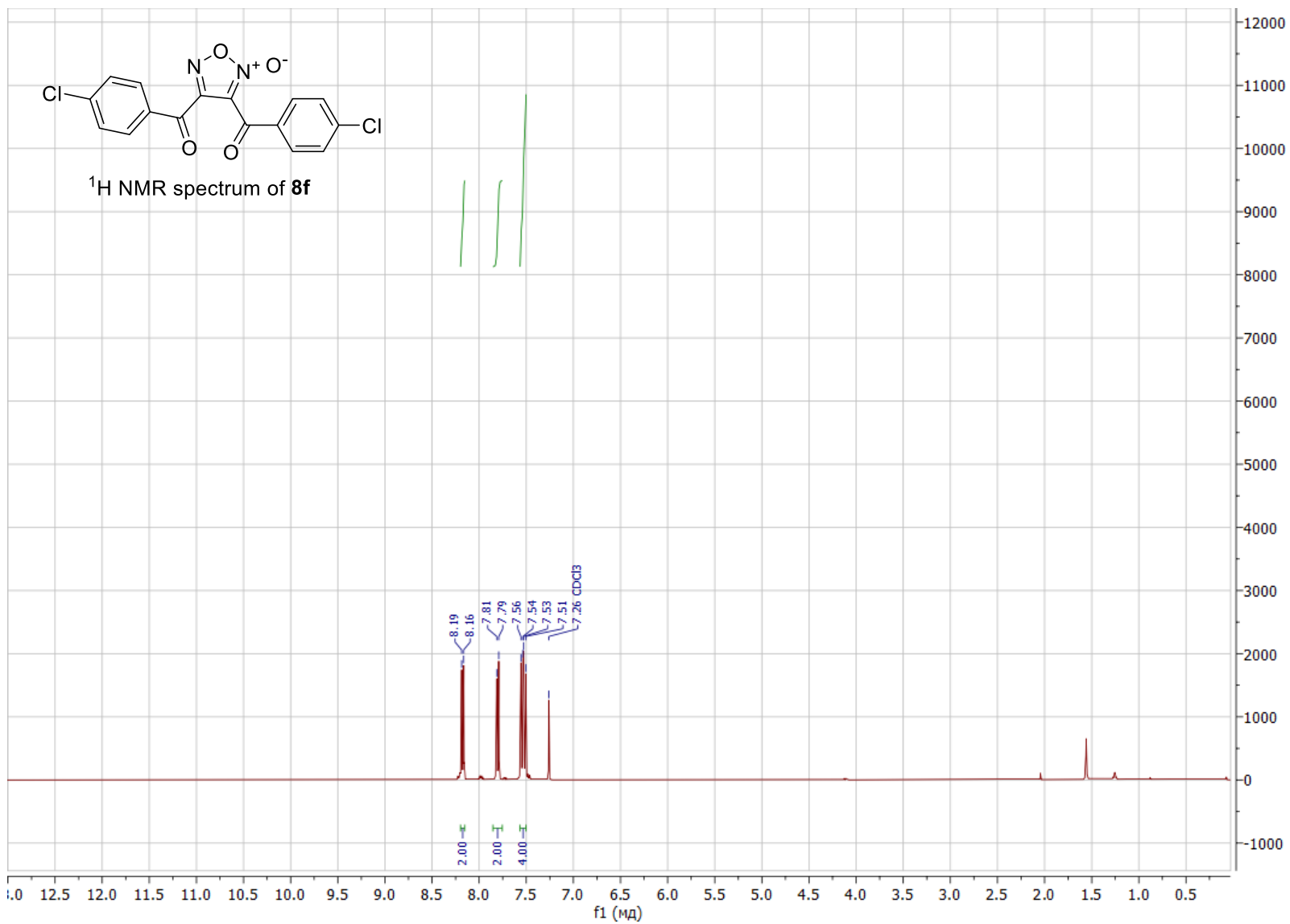


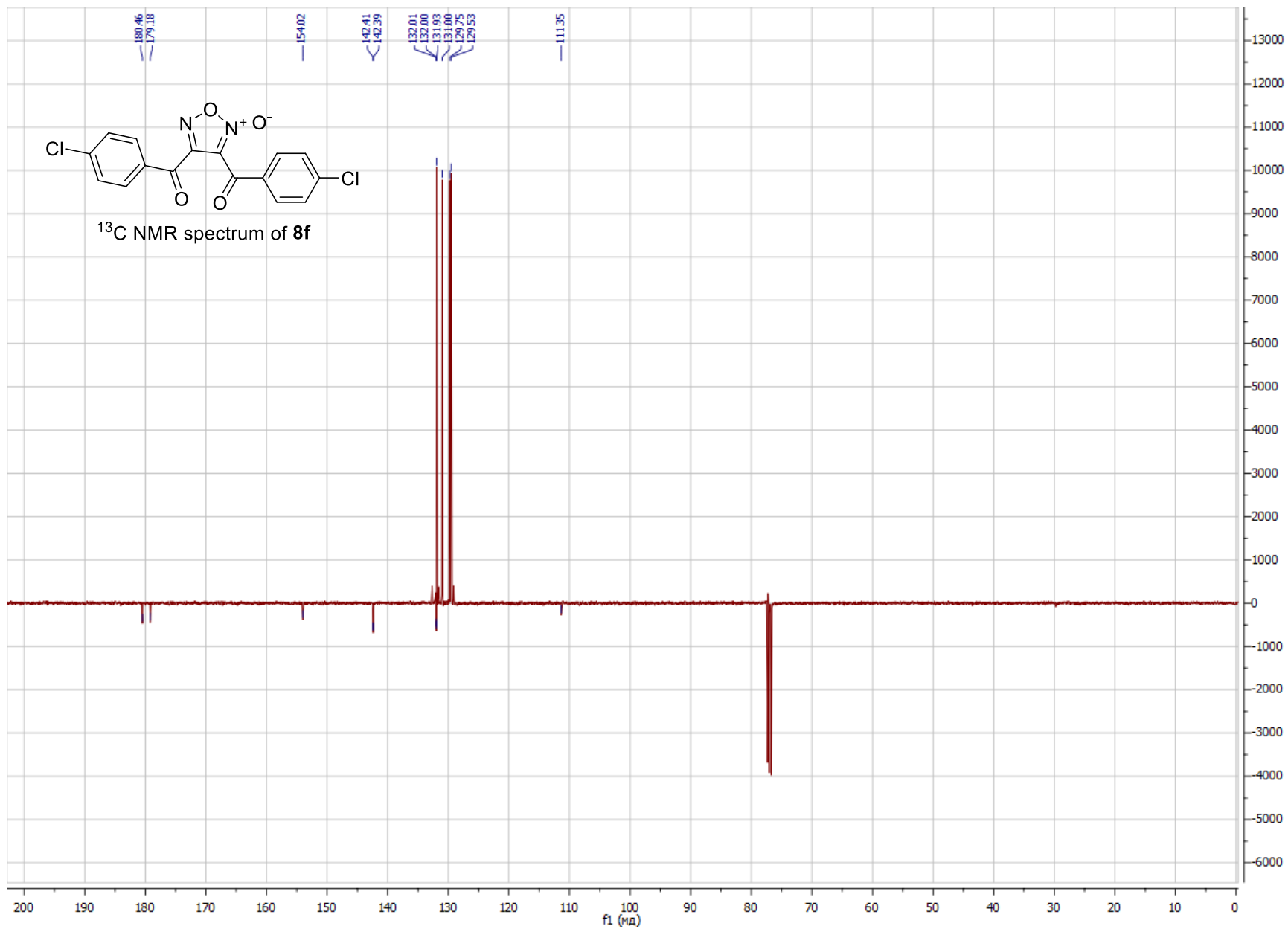


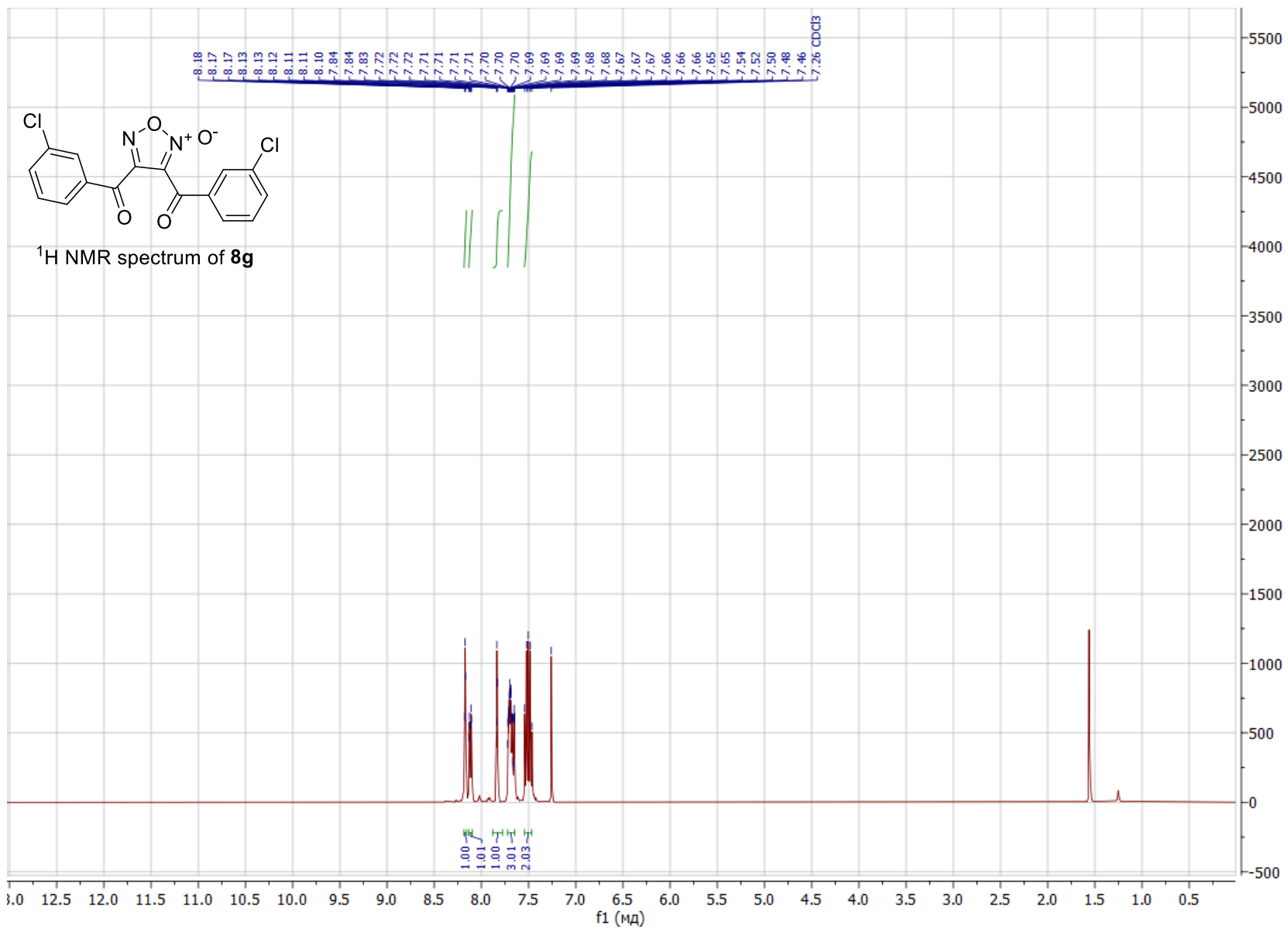


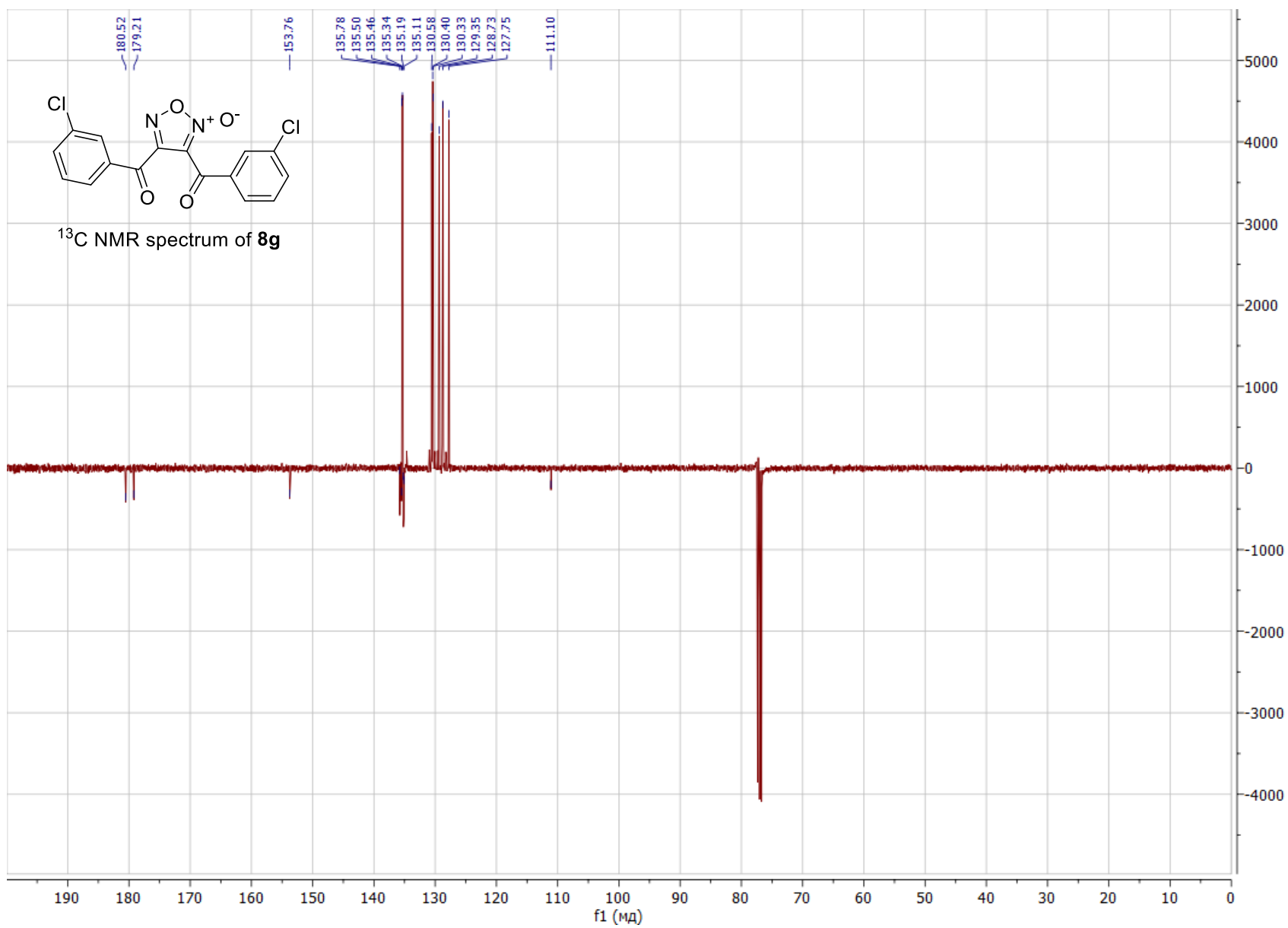


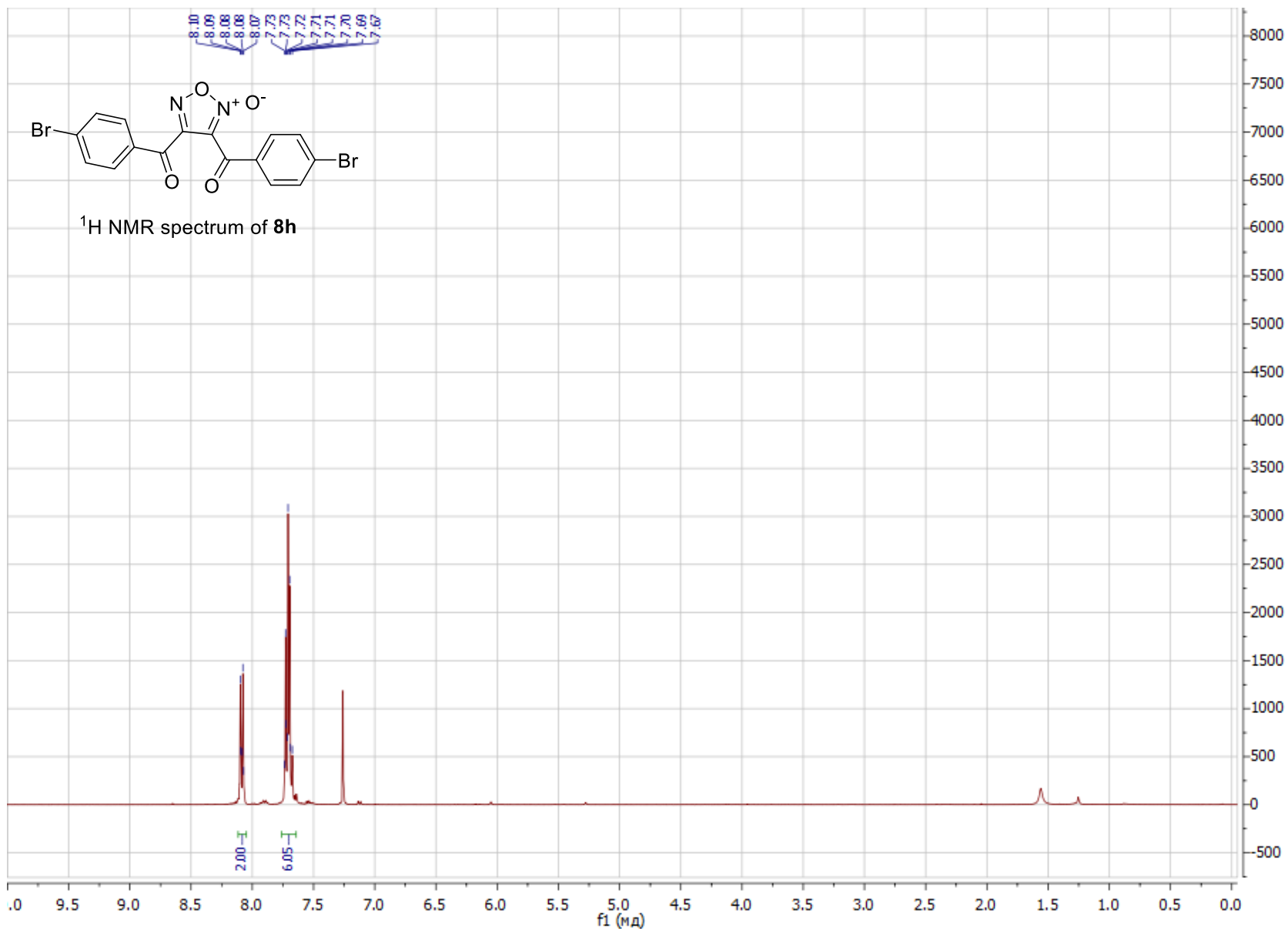


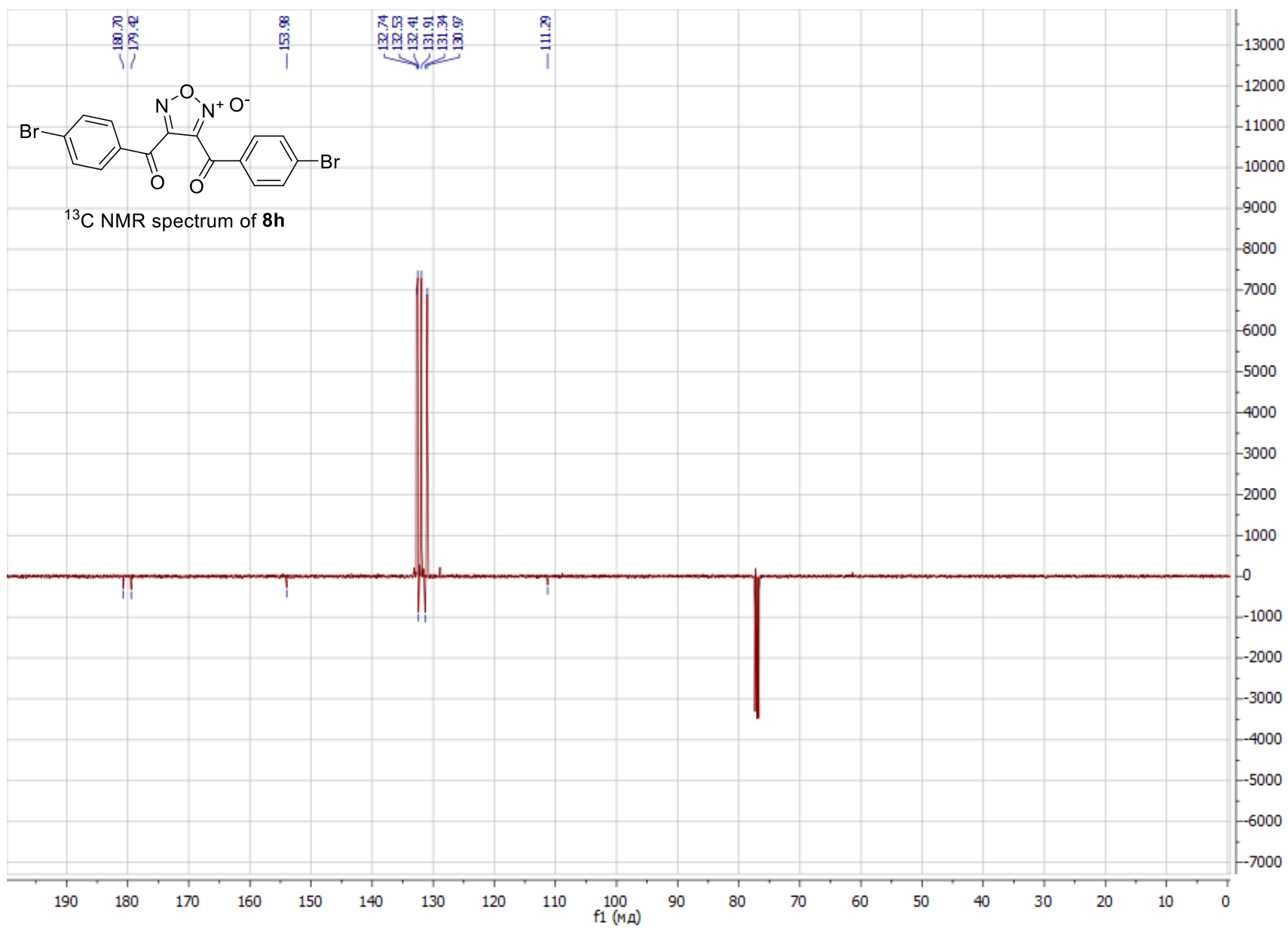


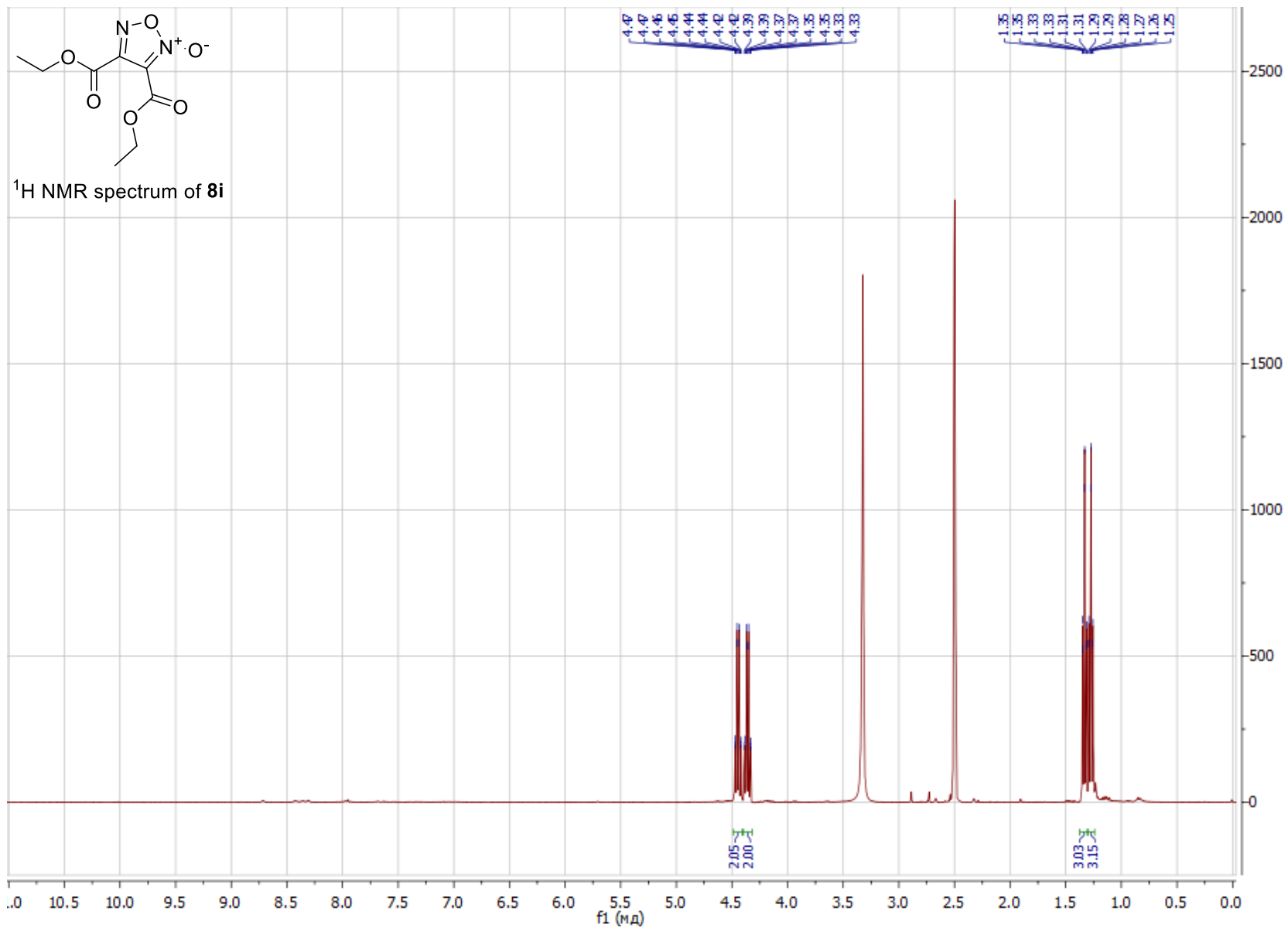


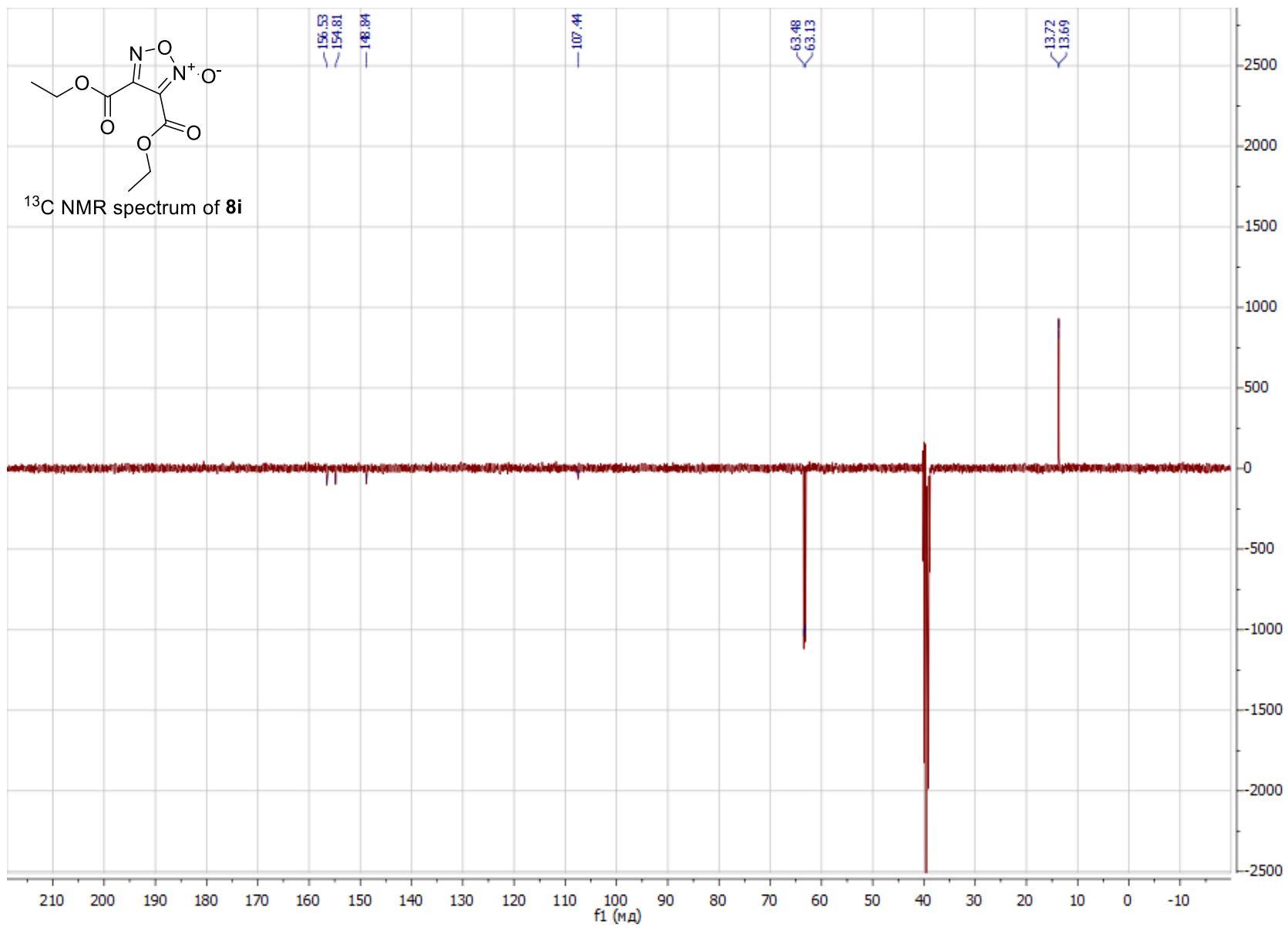




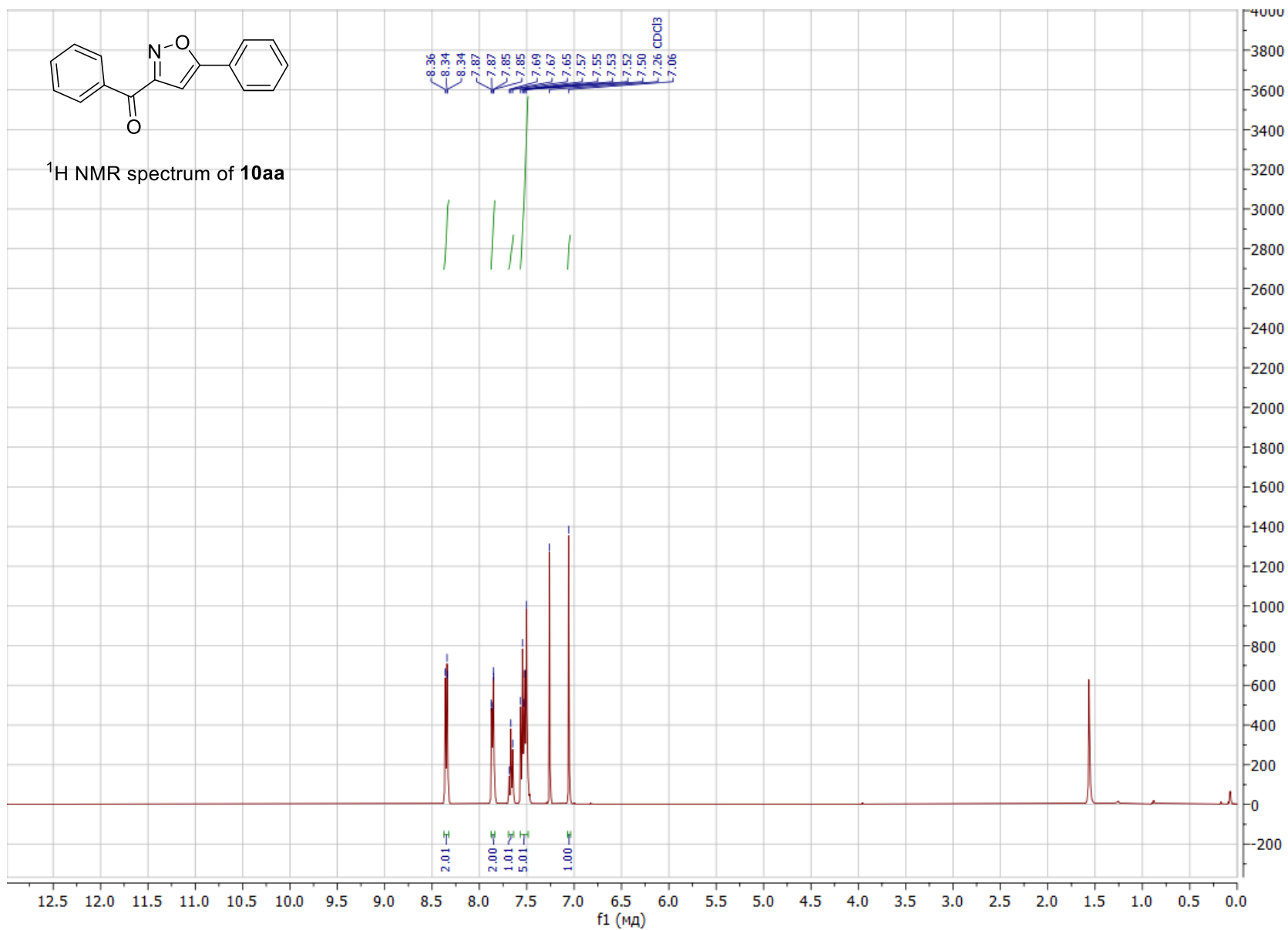


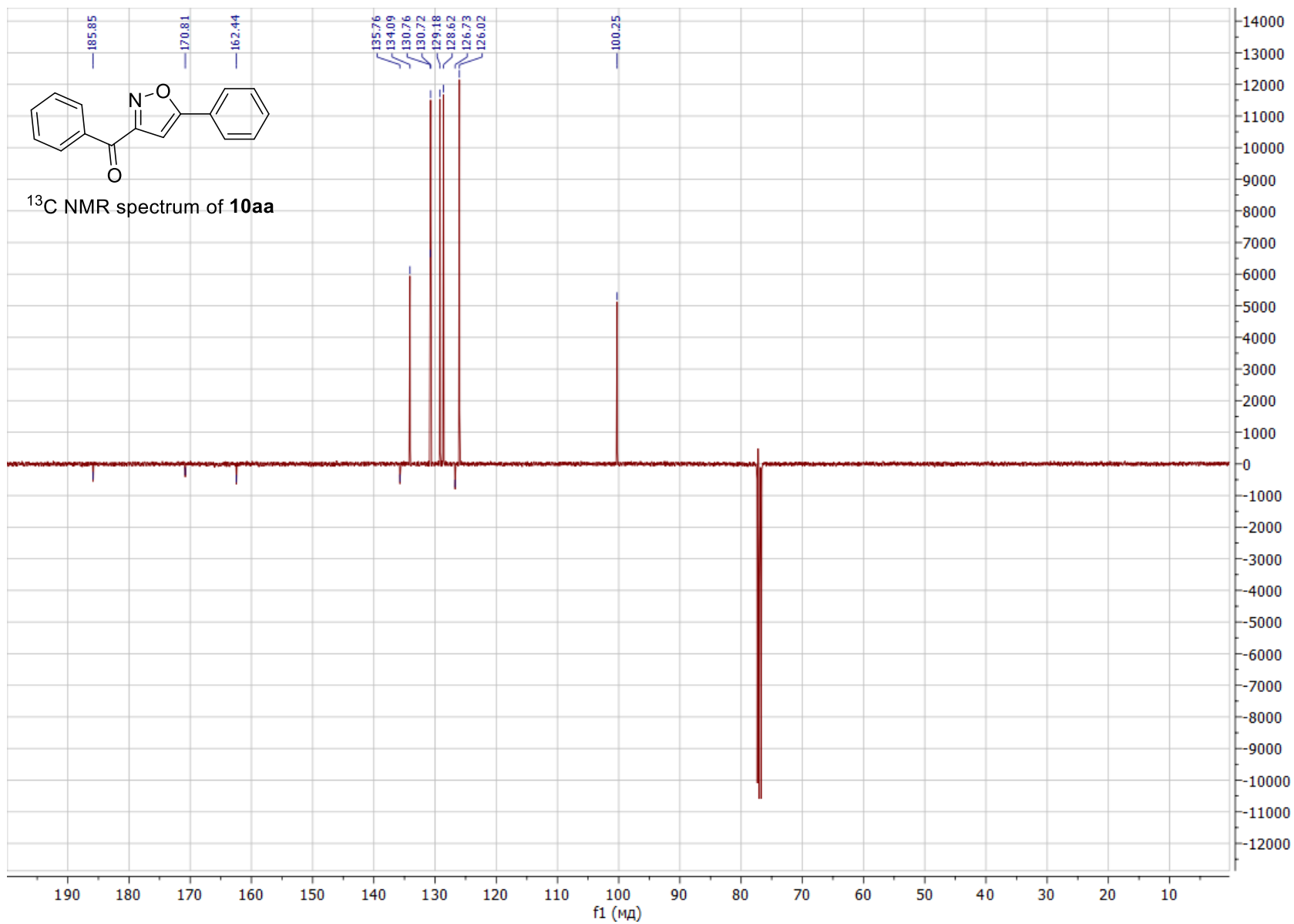


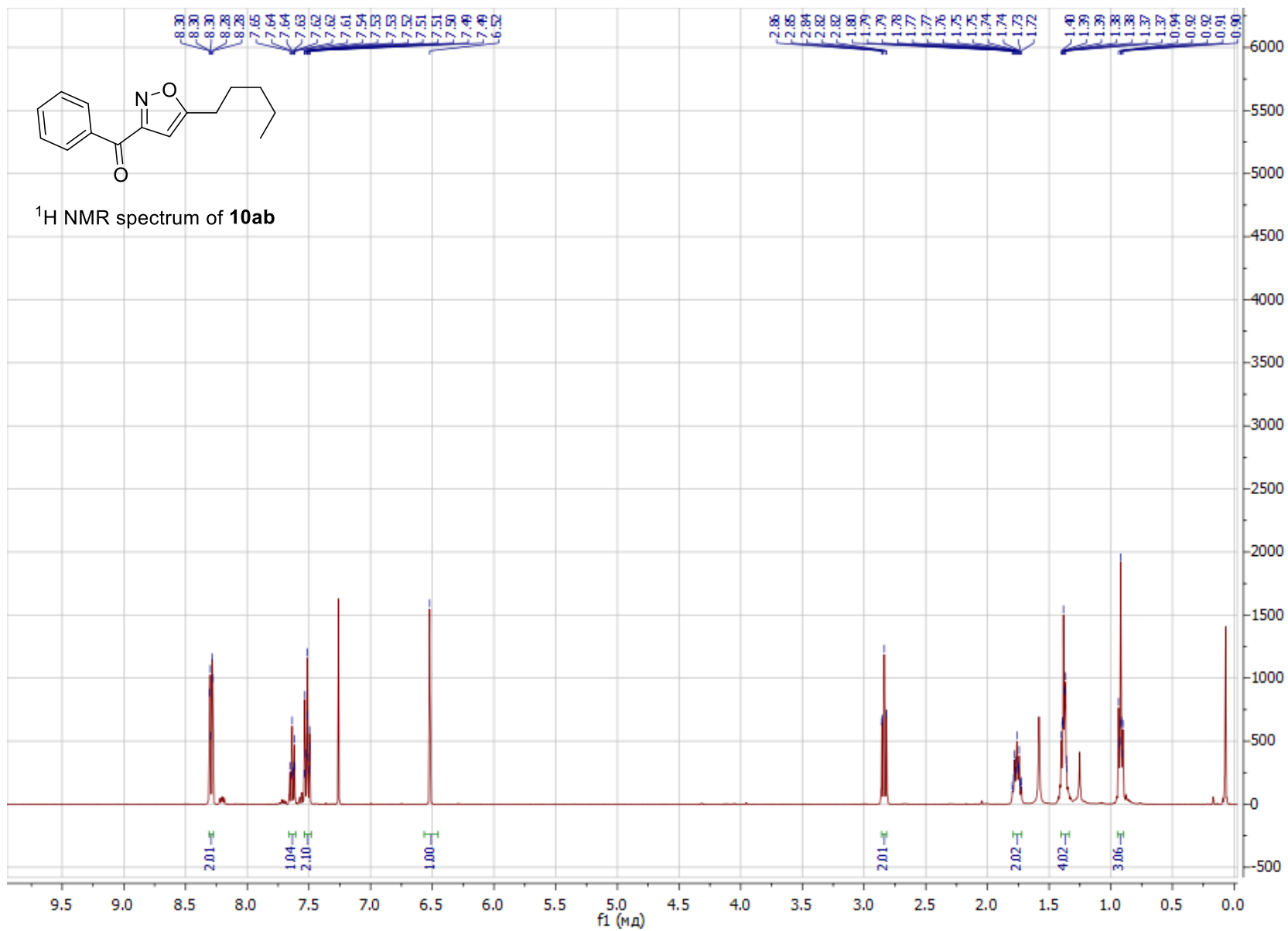


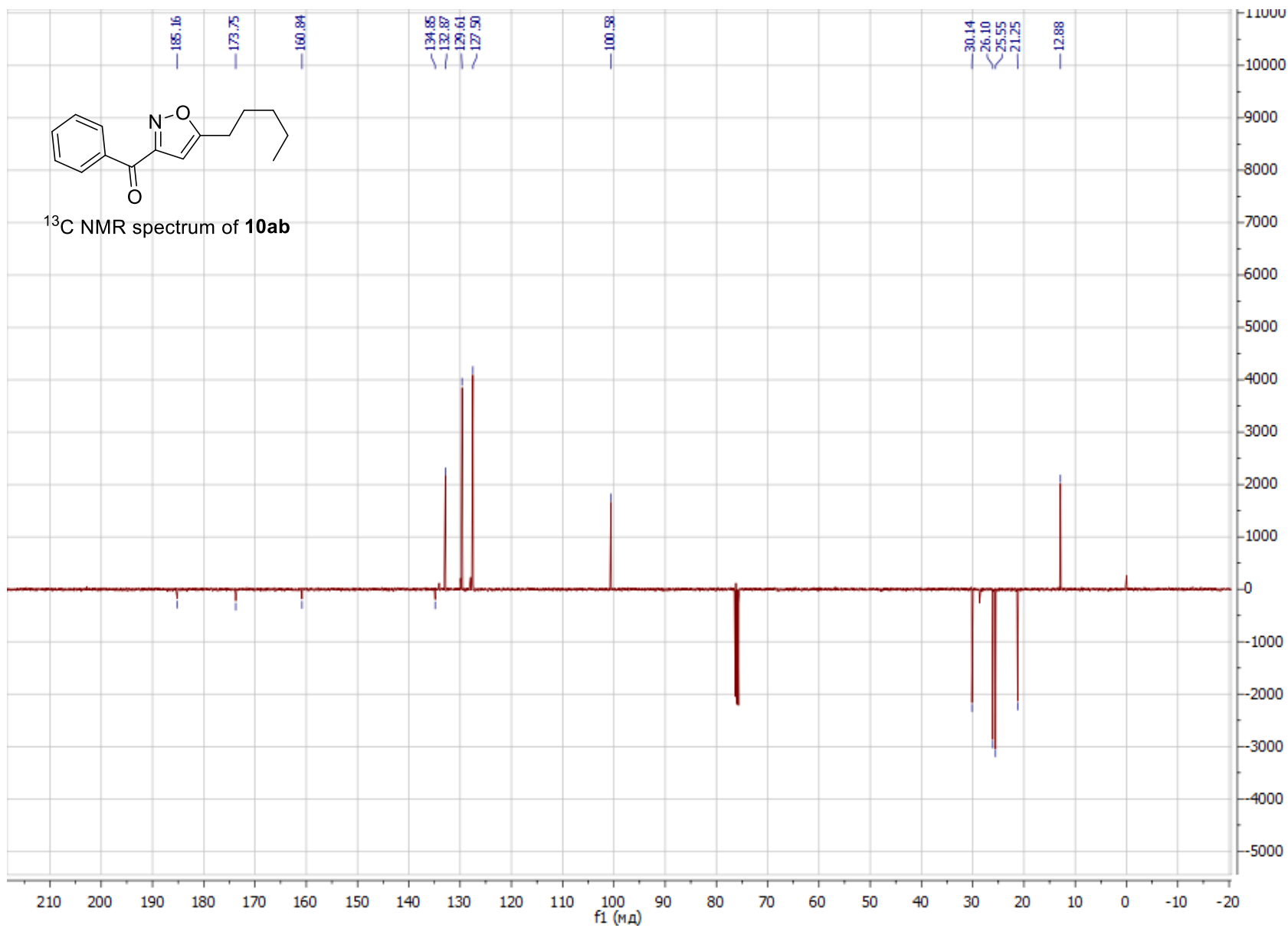


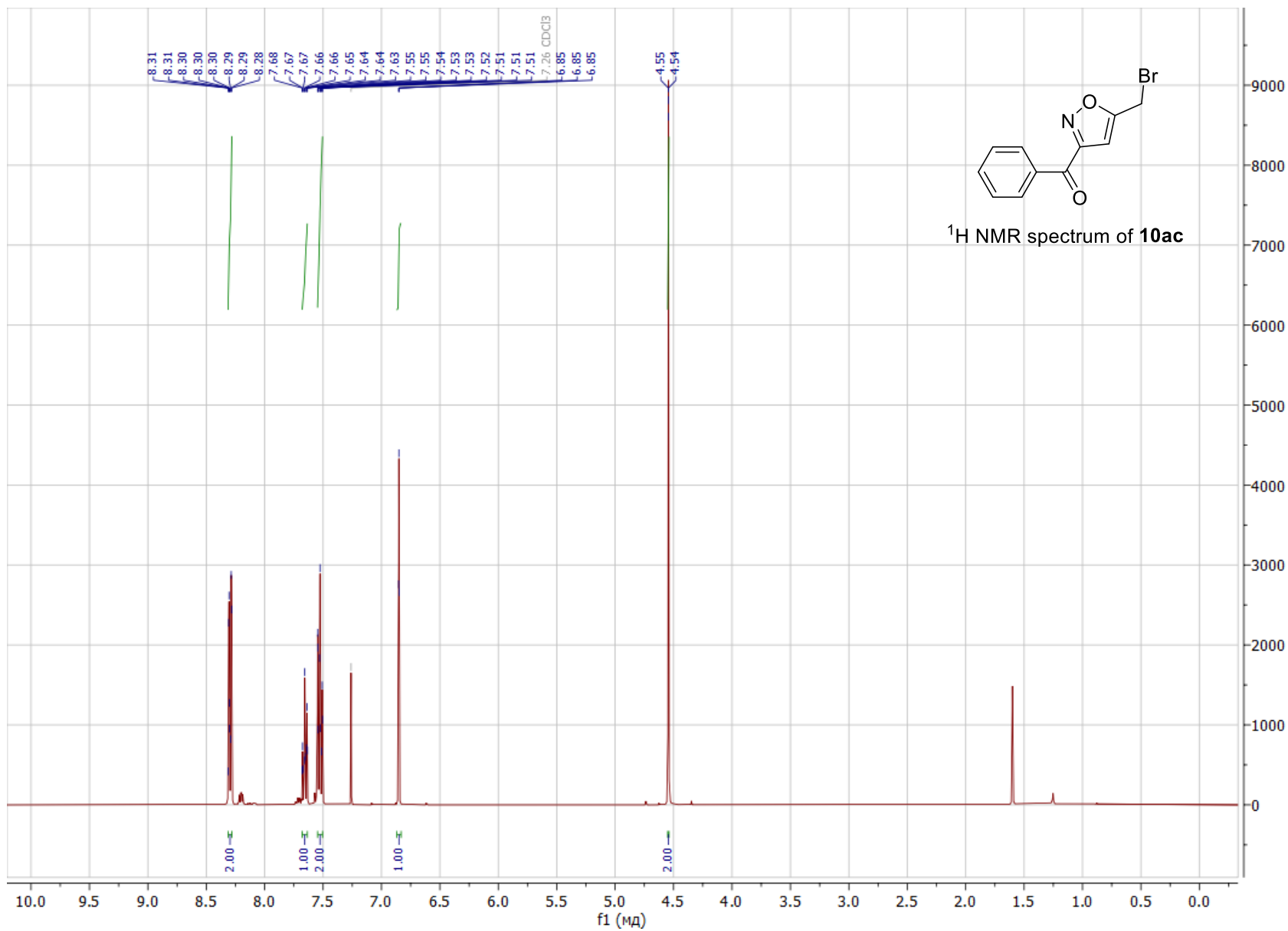
¹H and ¹³C NMR spectral charts for isoxazoles 10

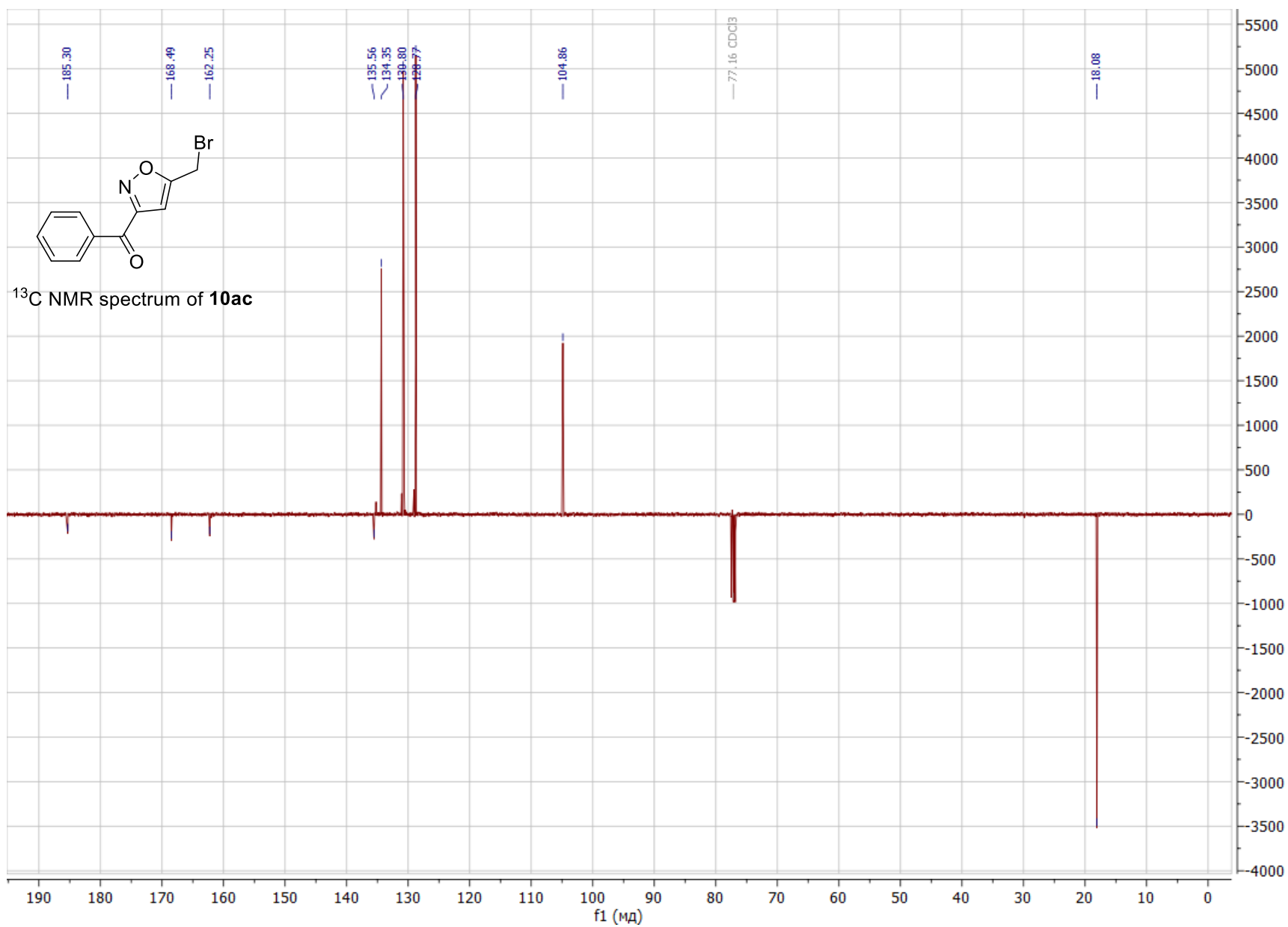


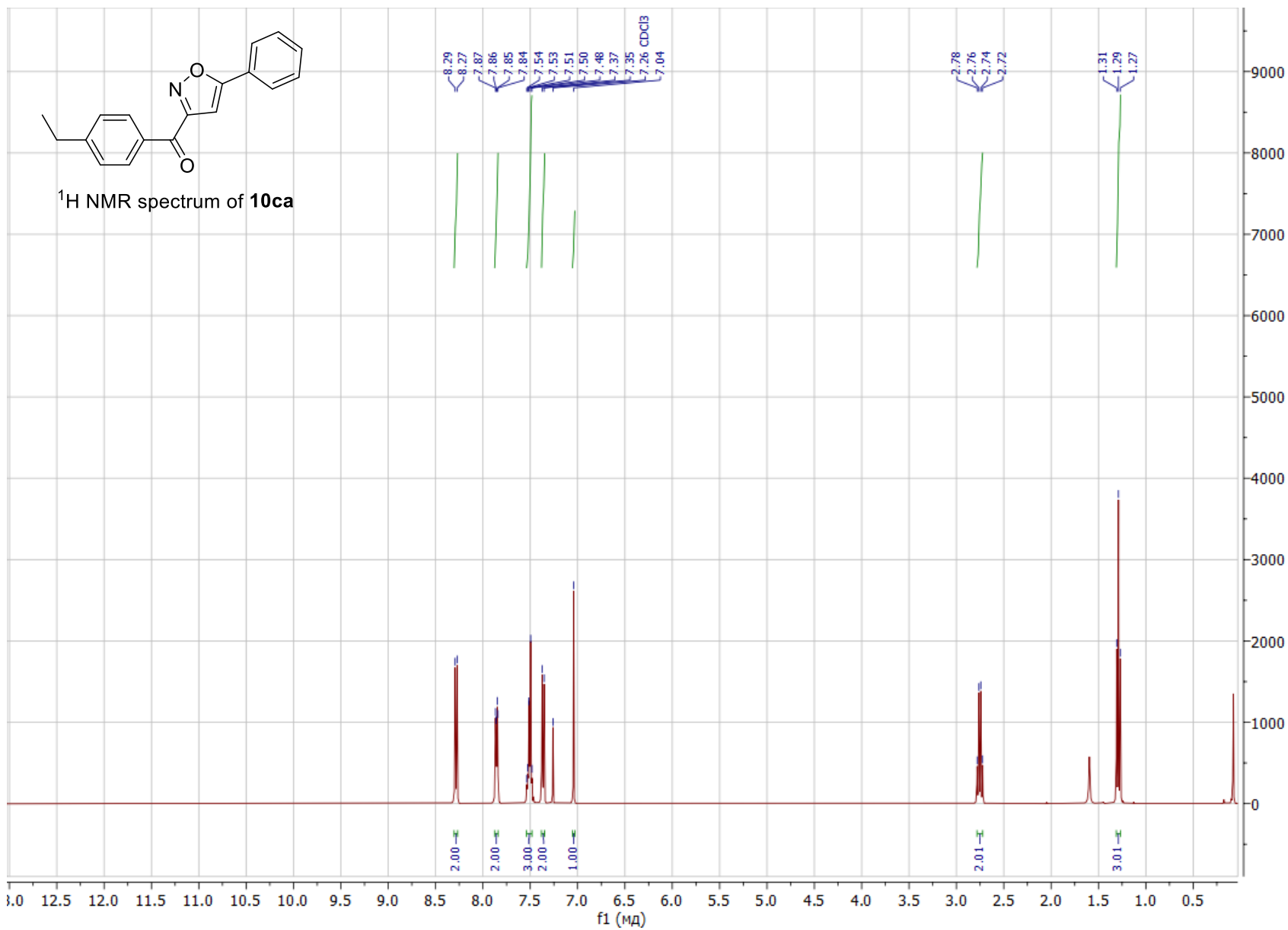


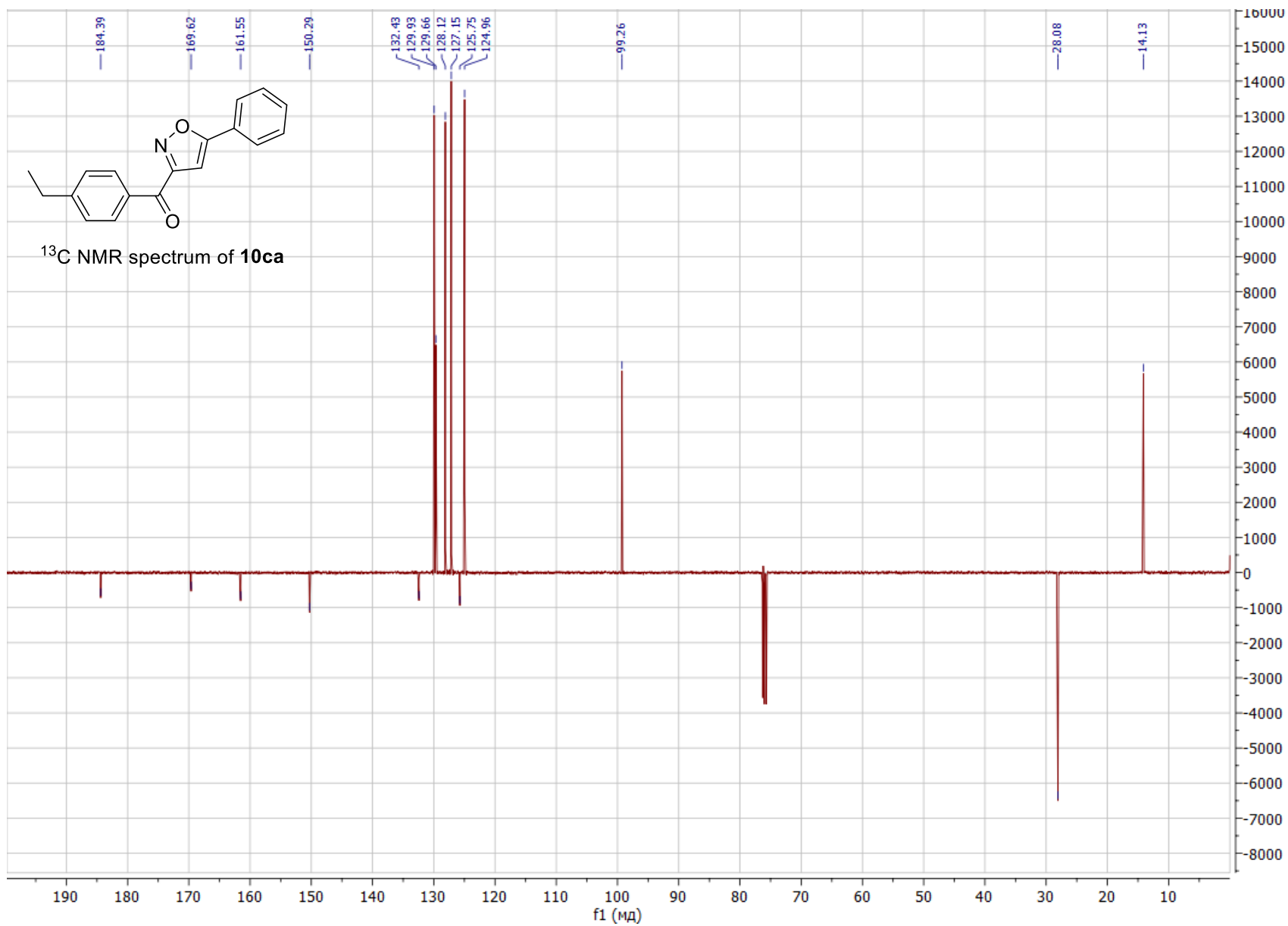


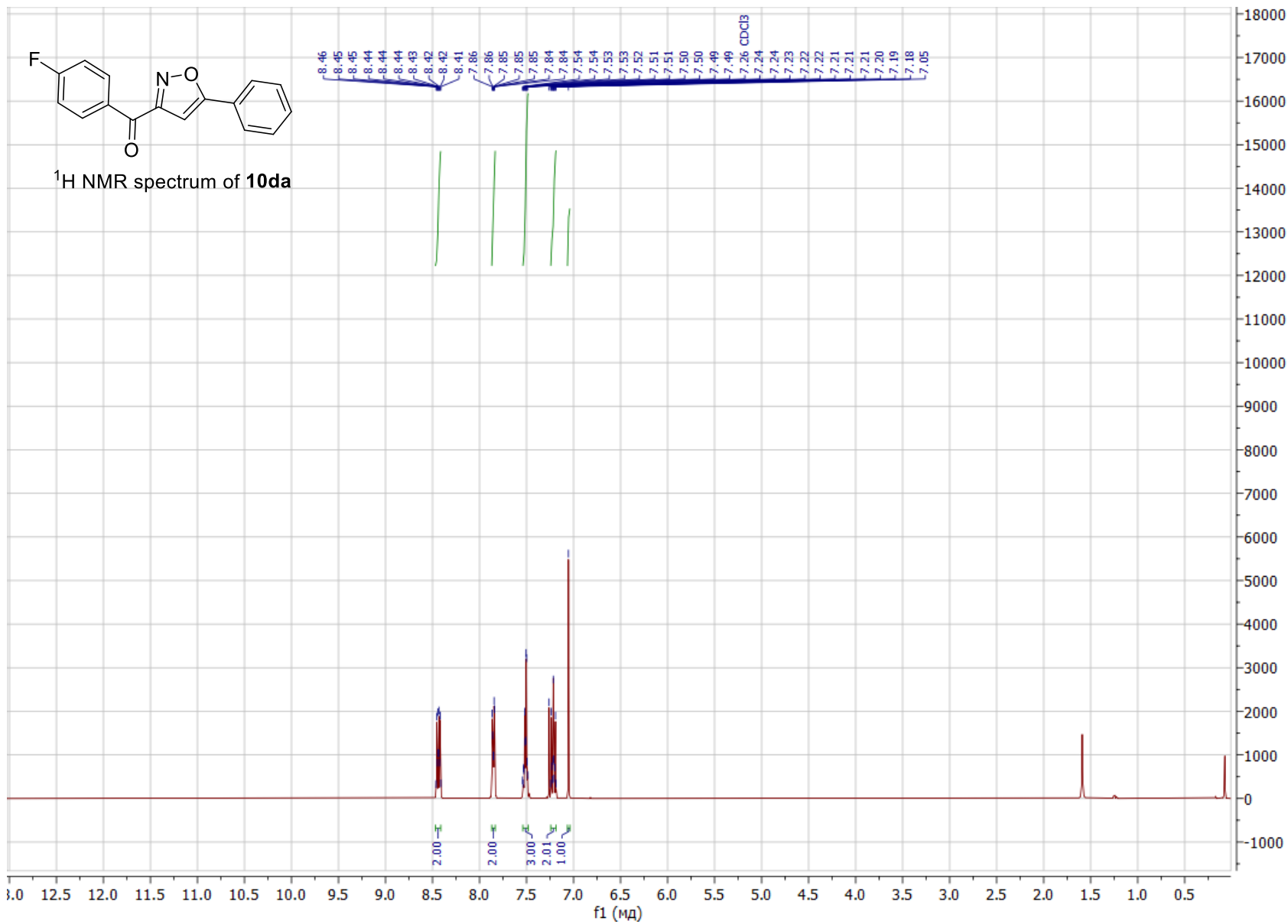


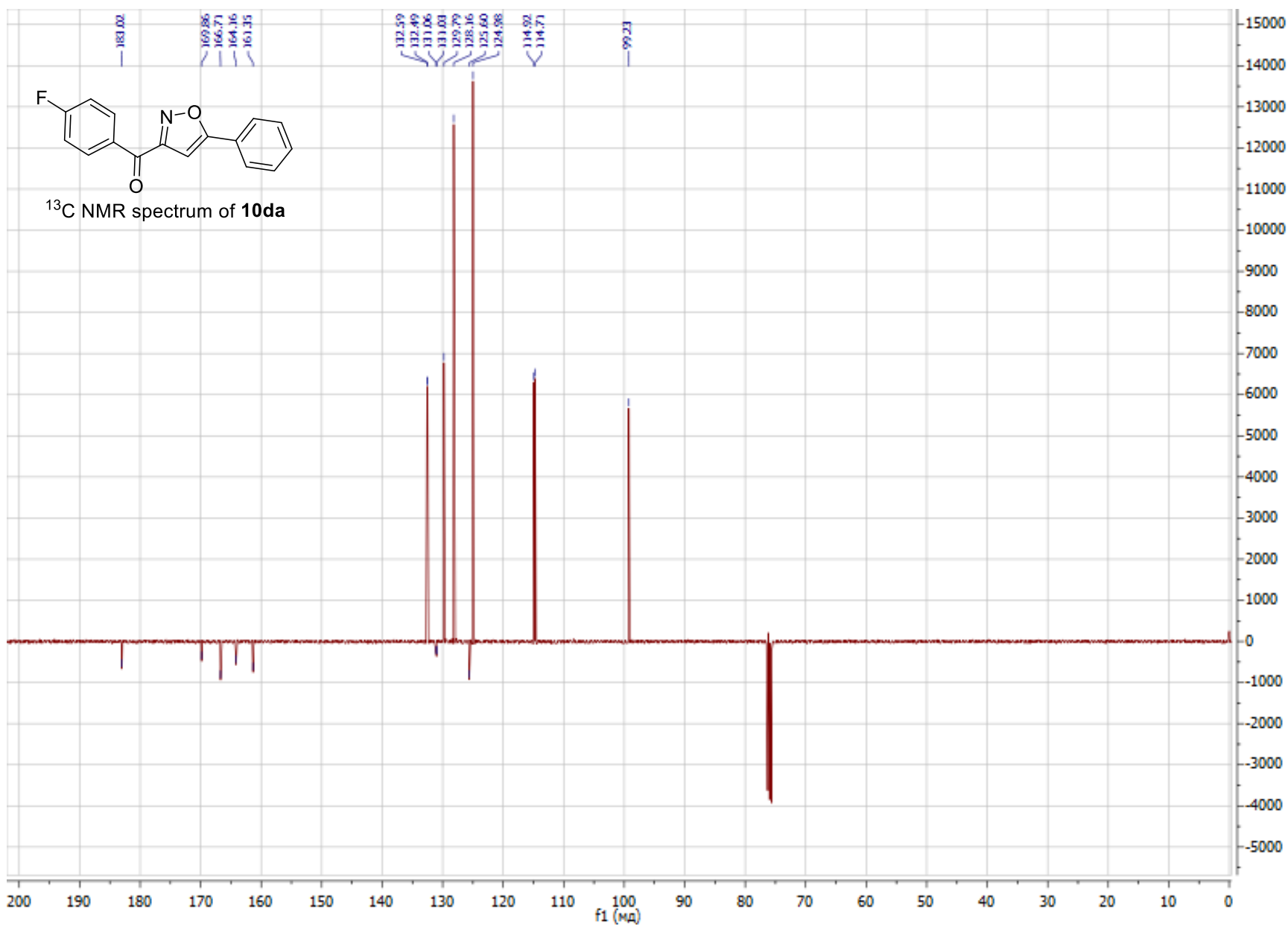


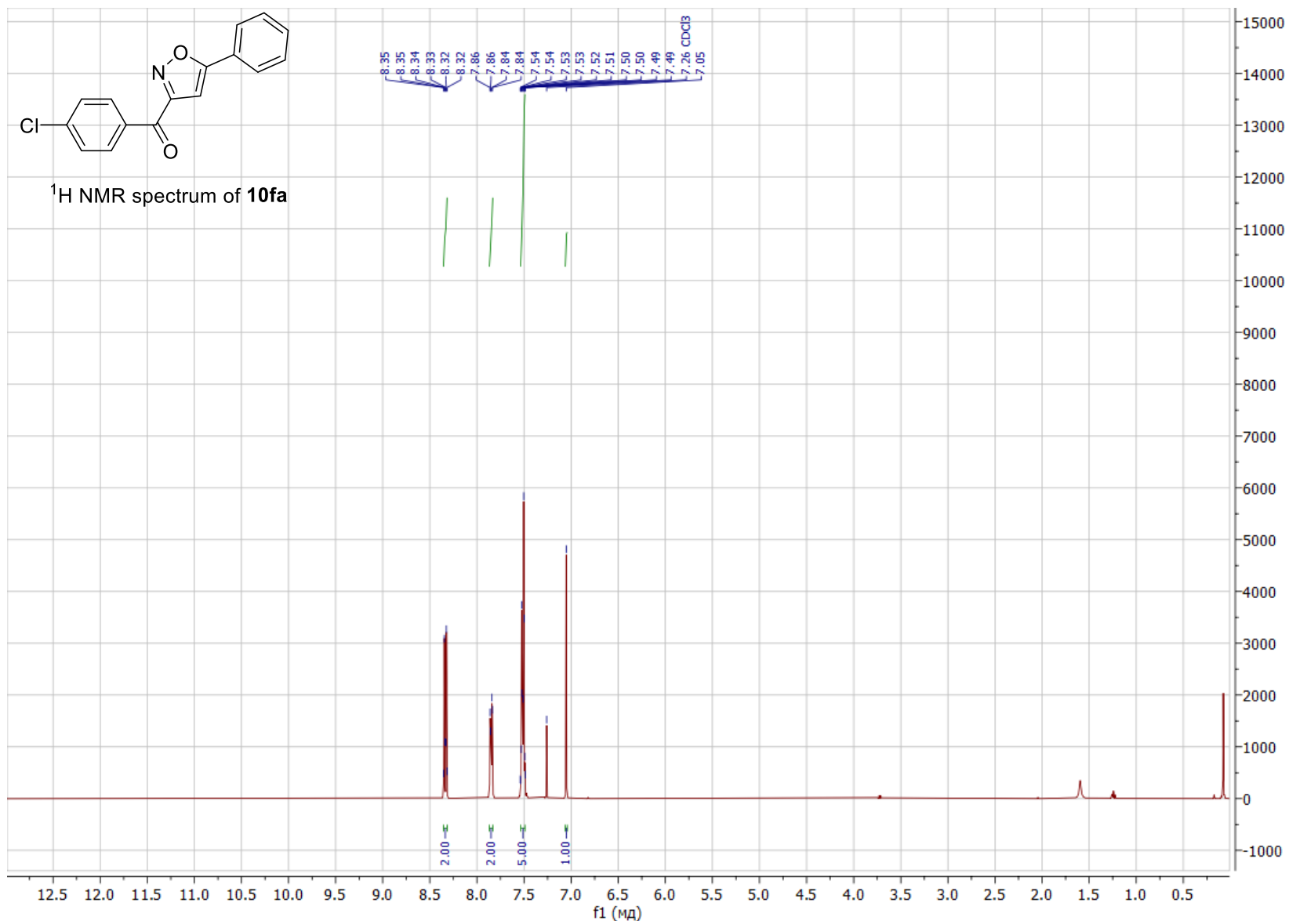


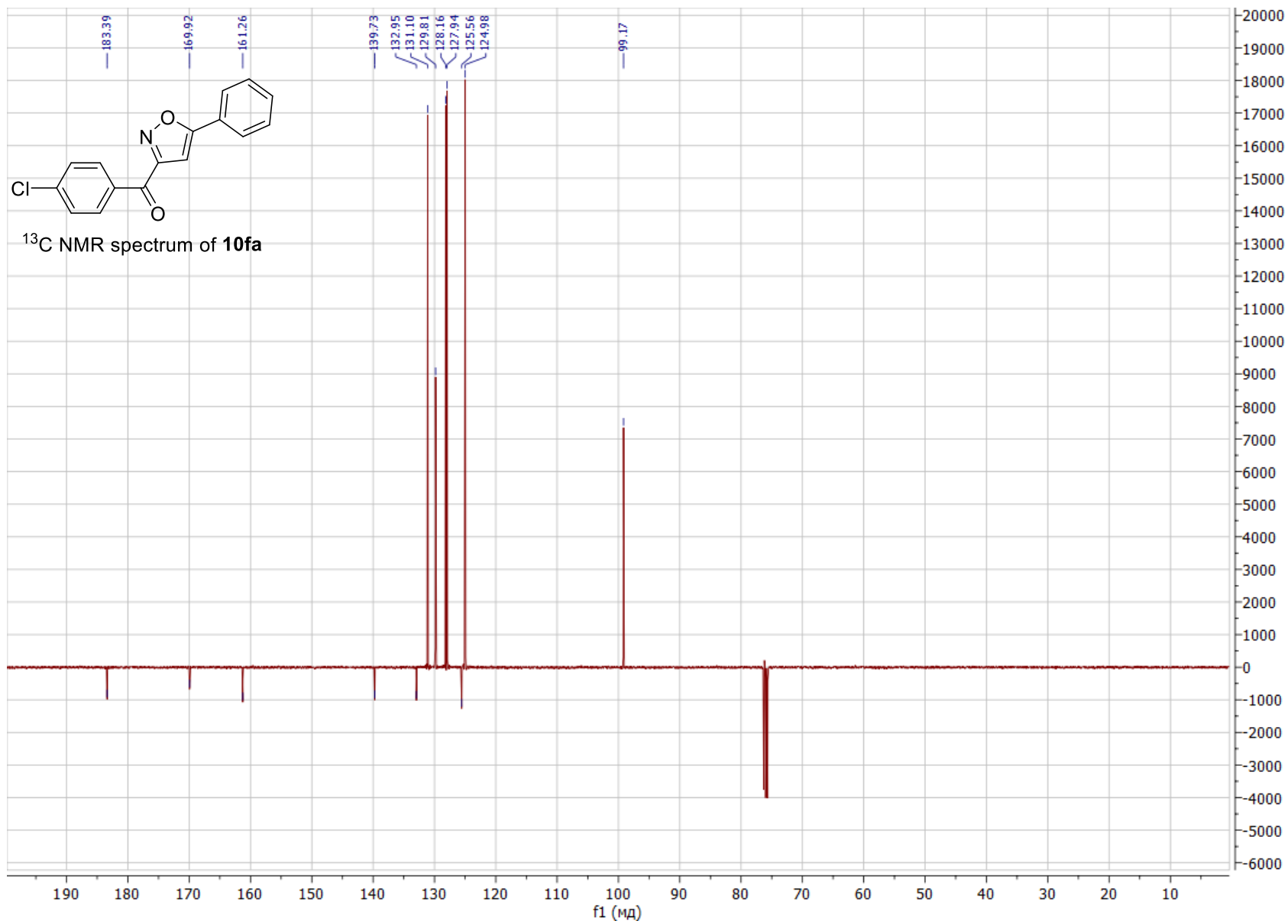


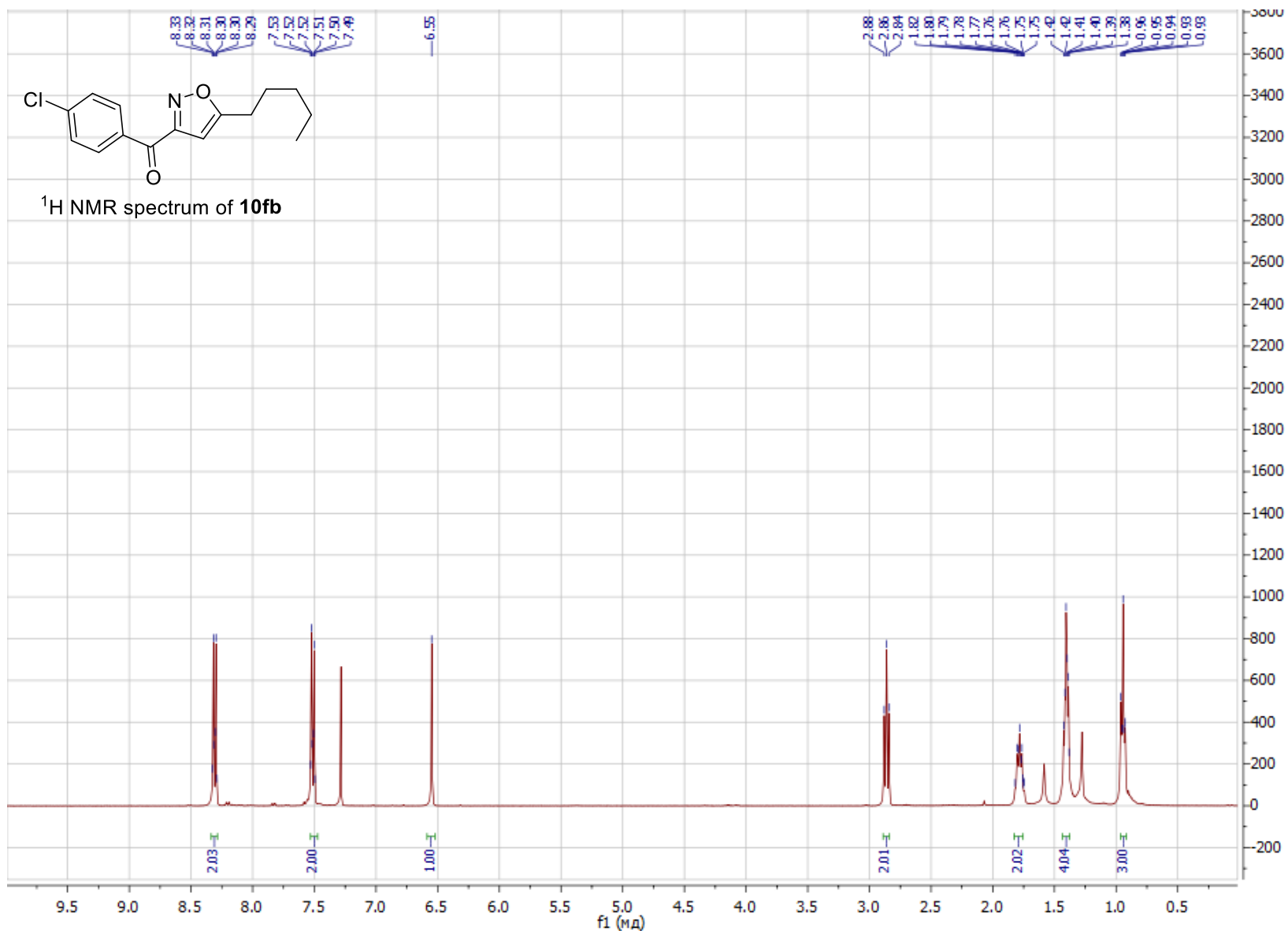


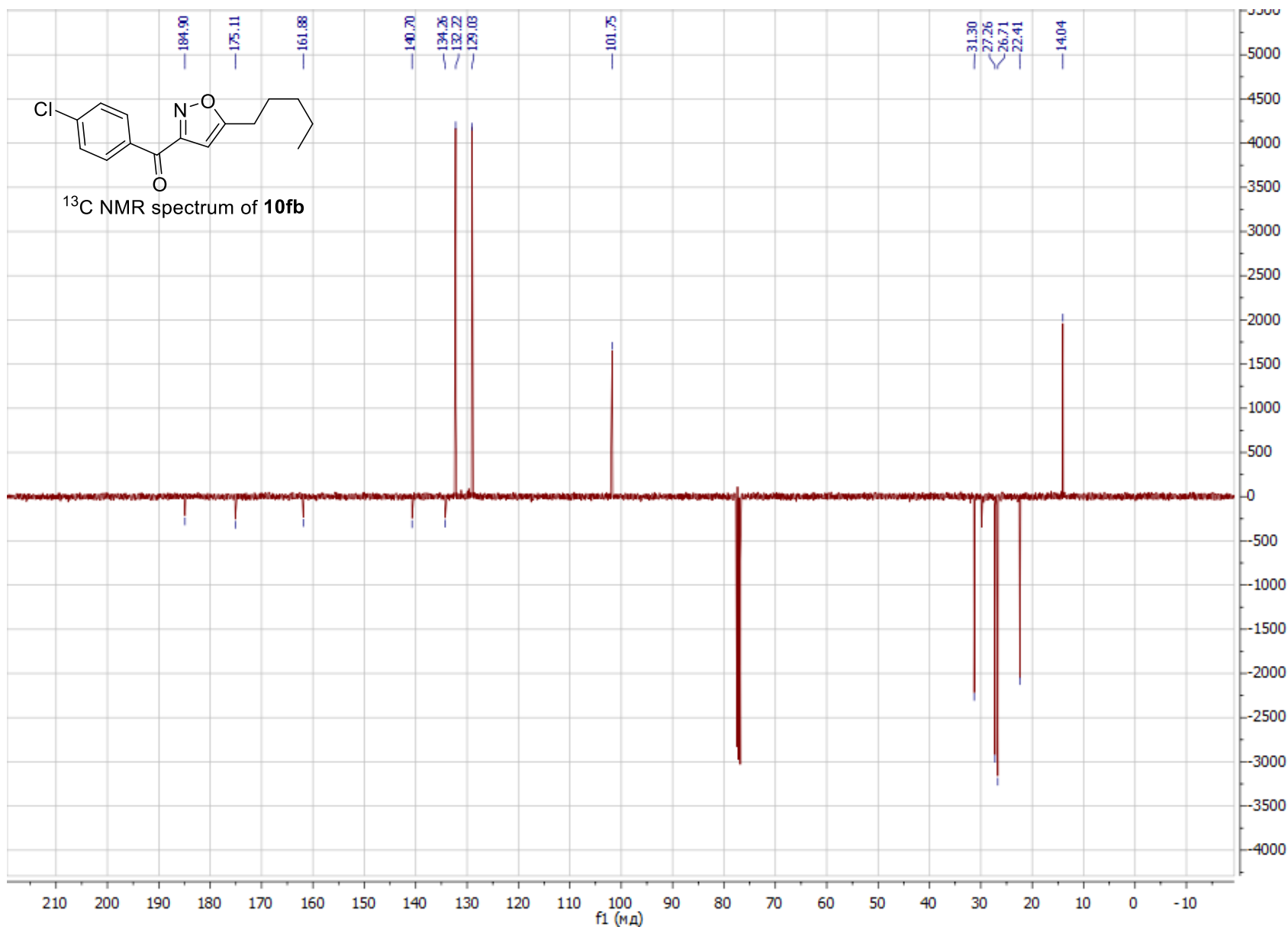


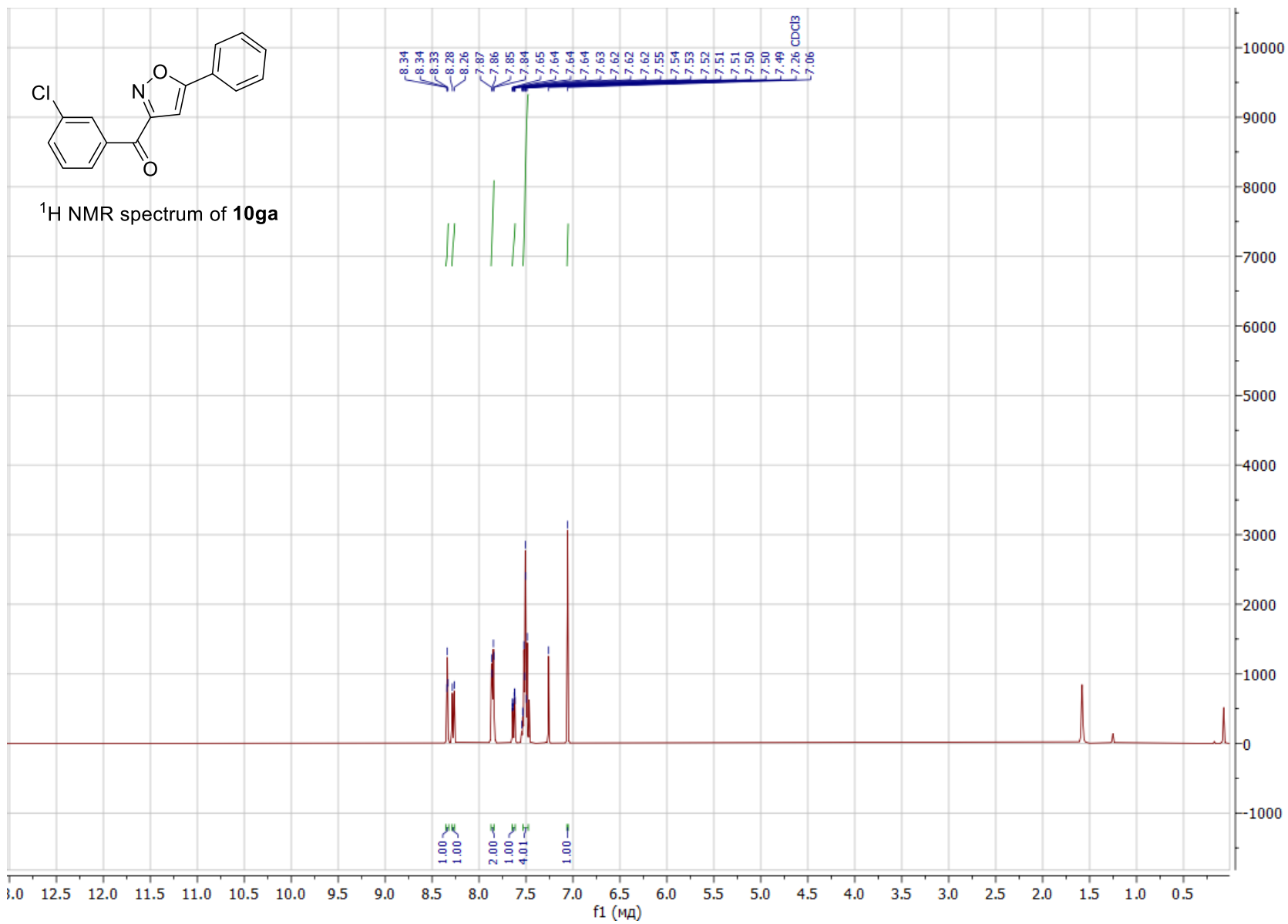


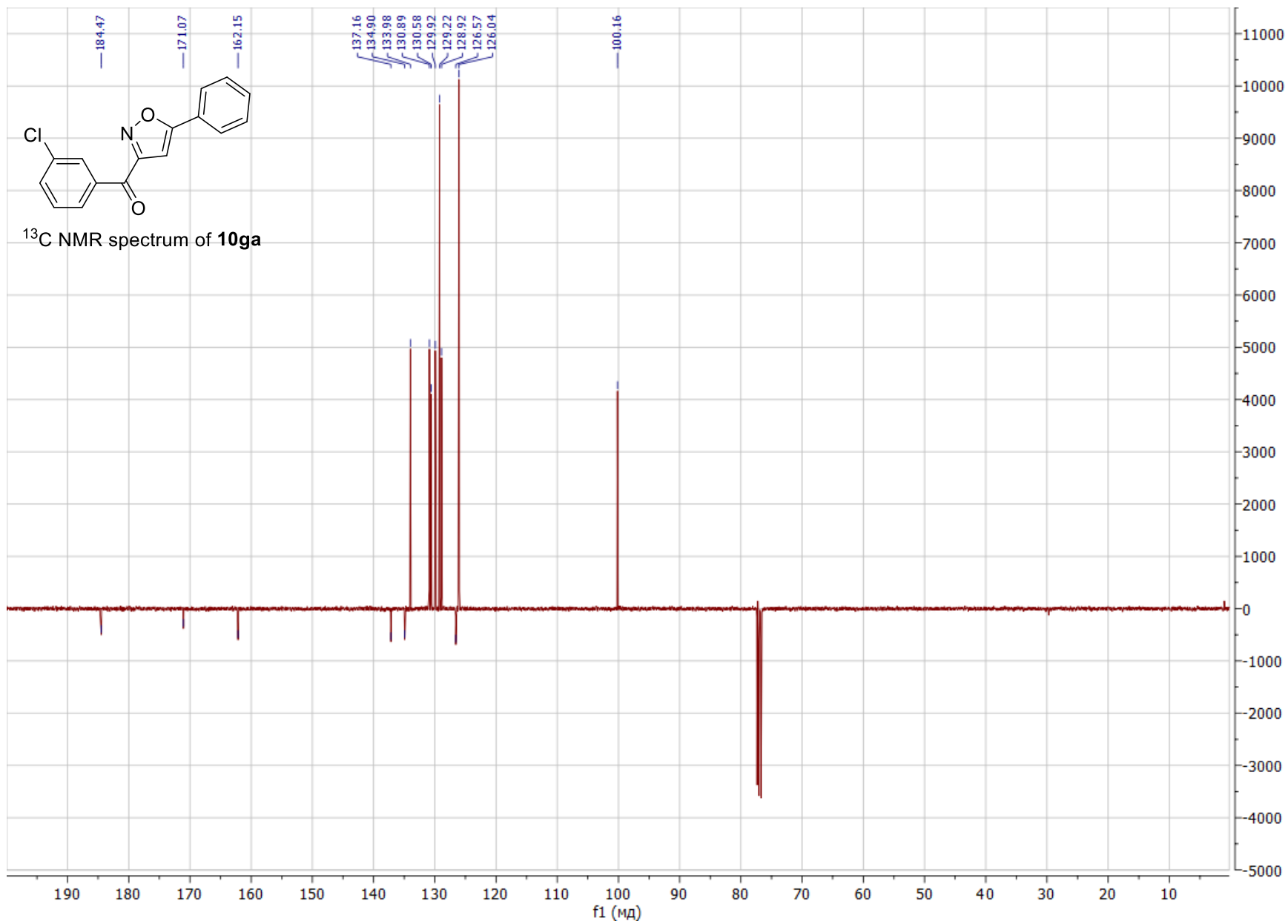


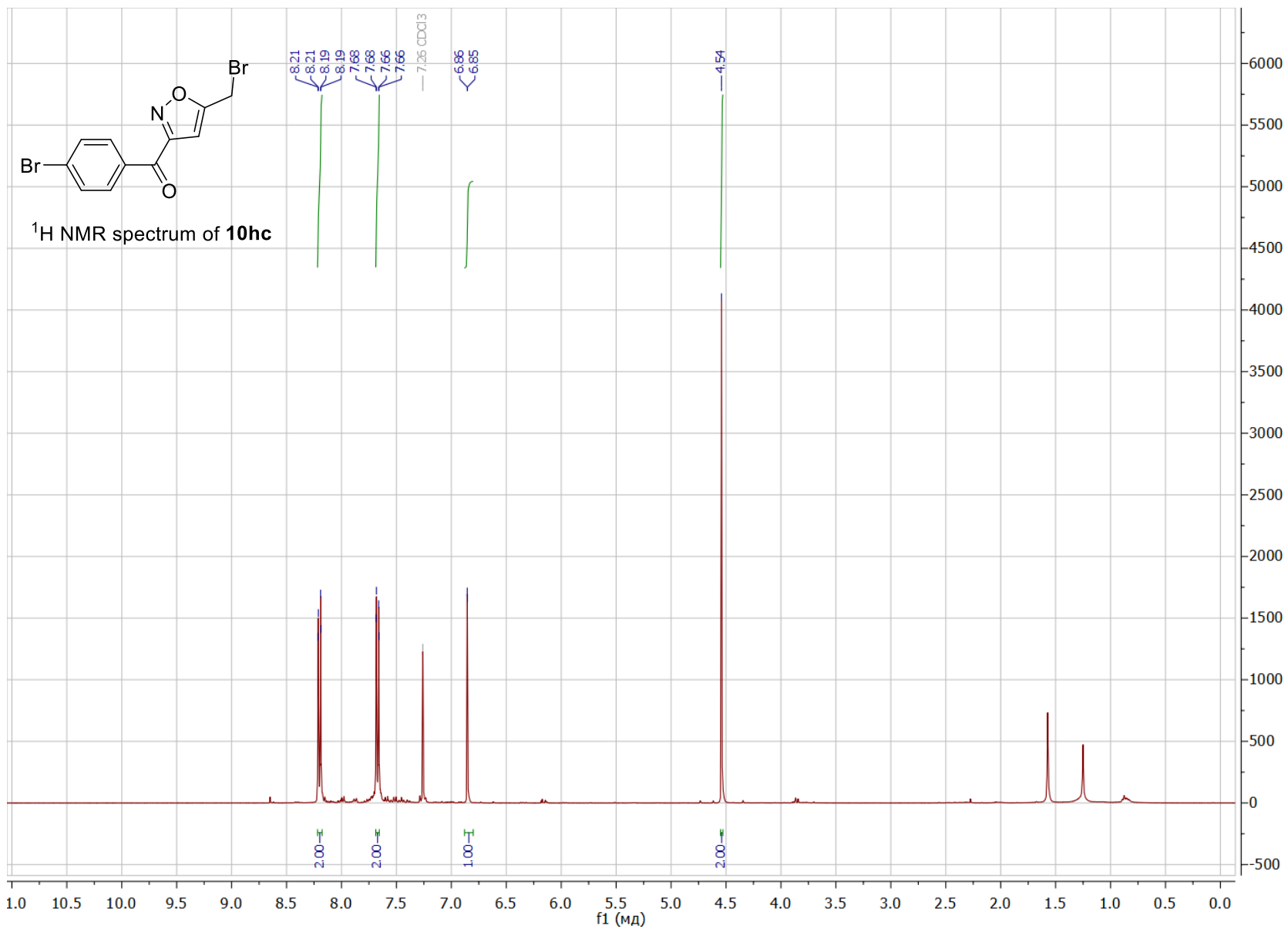


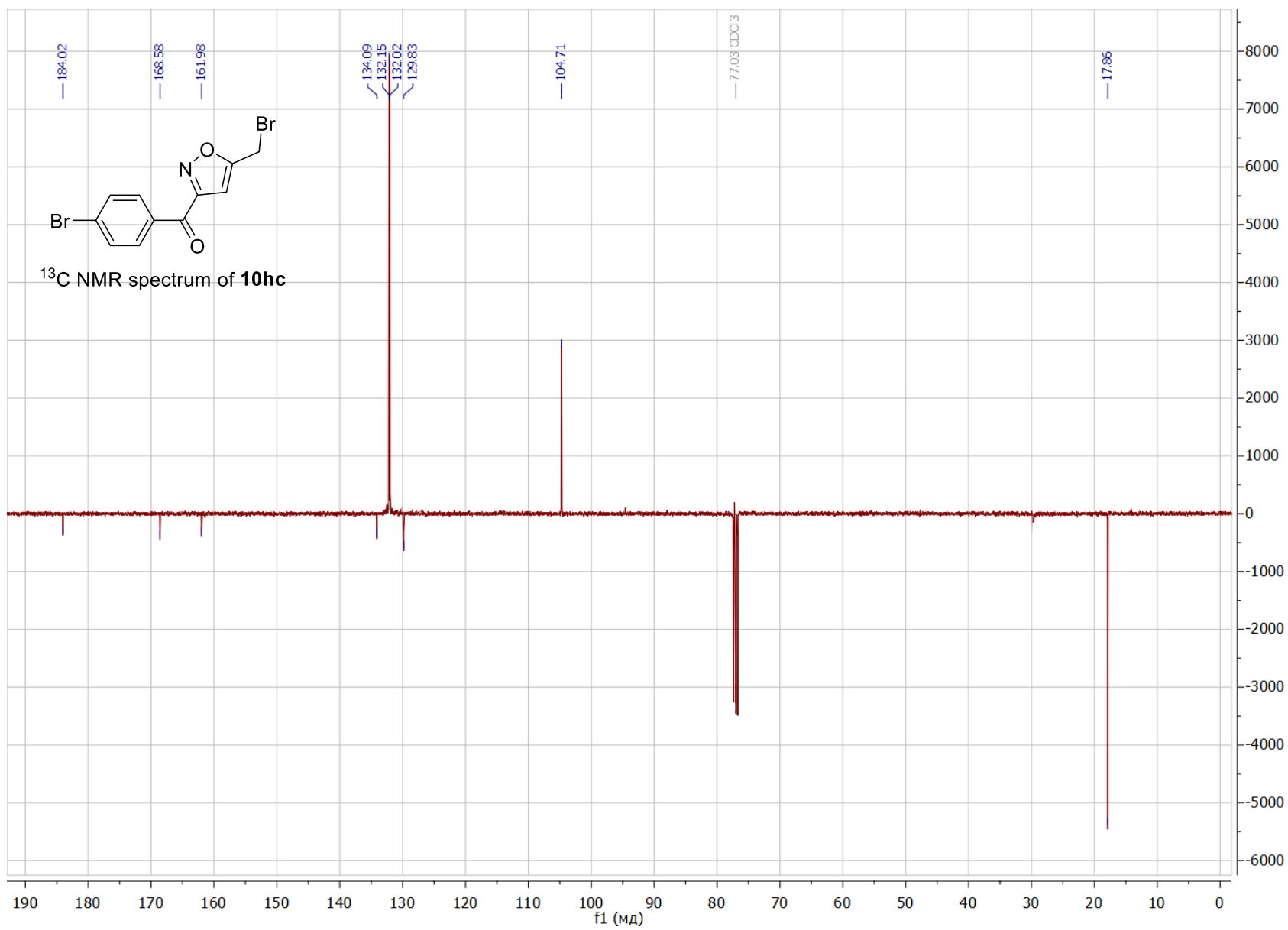




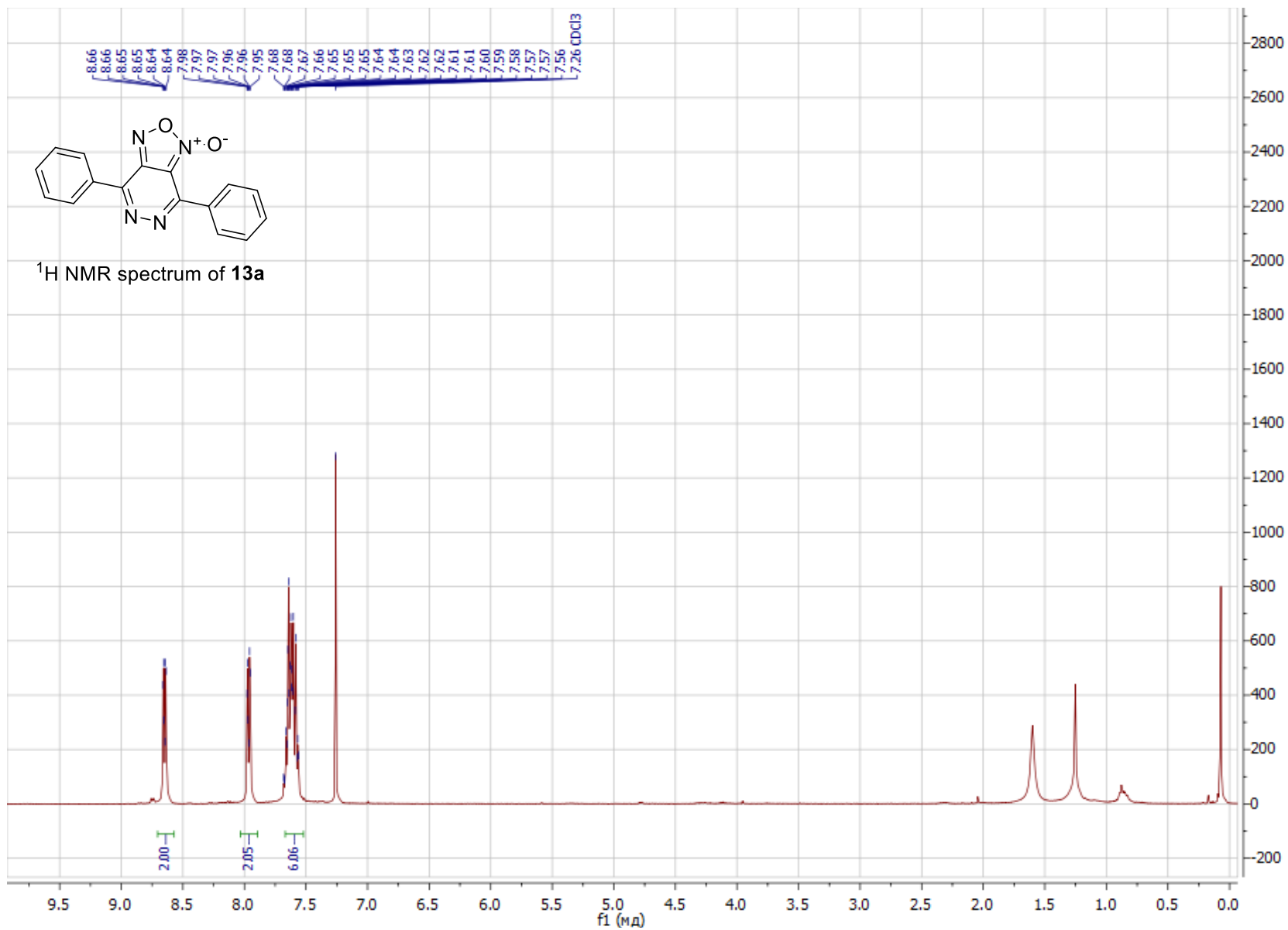


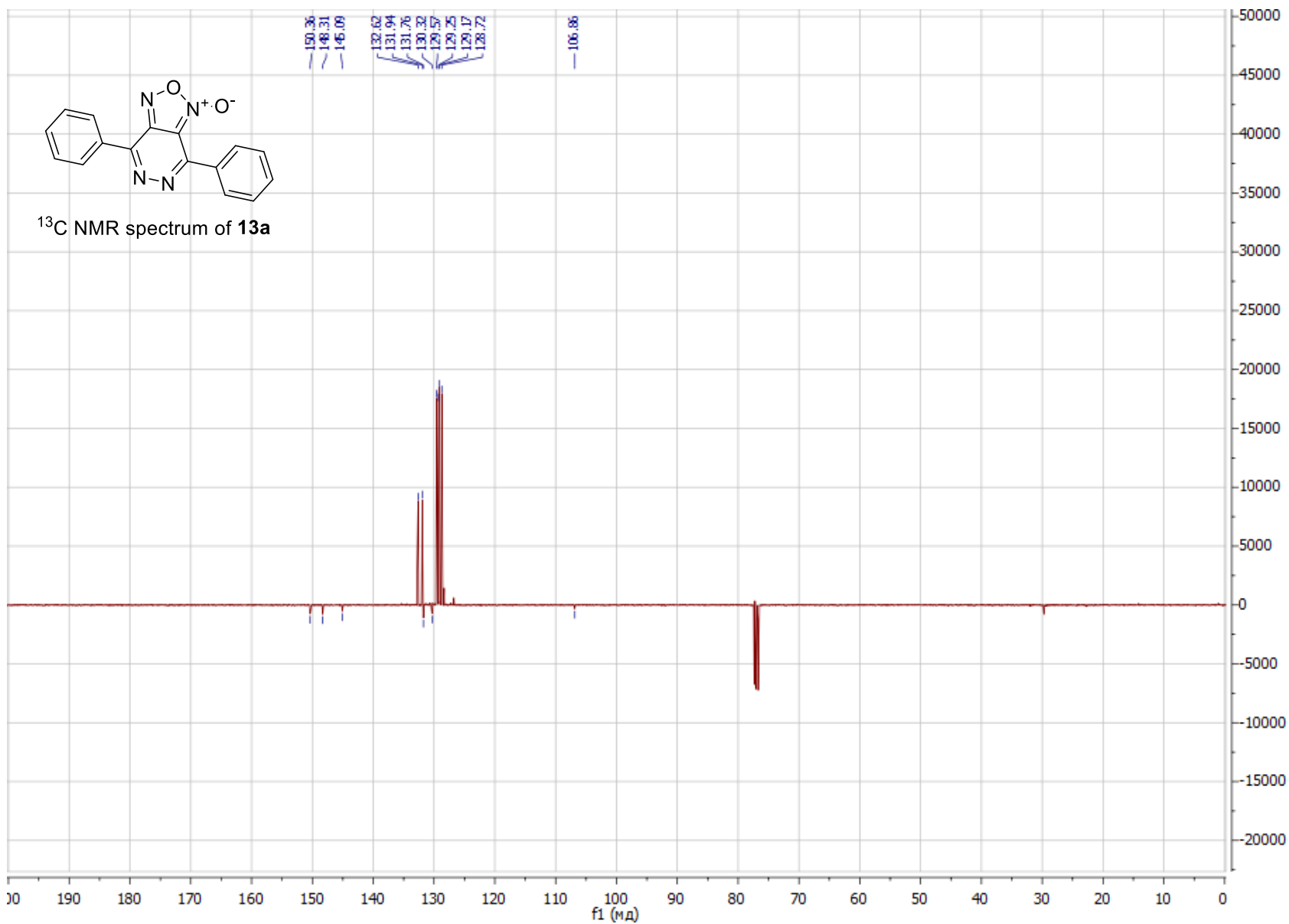




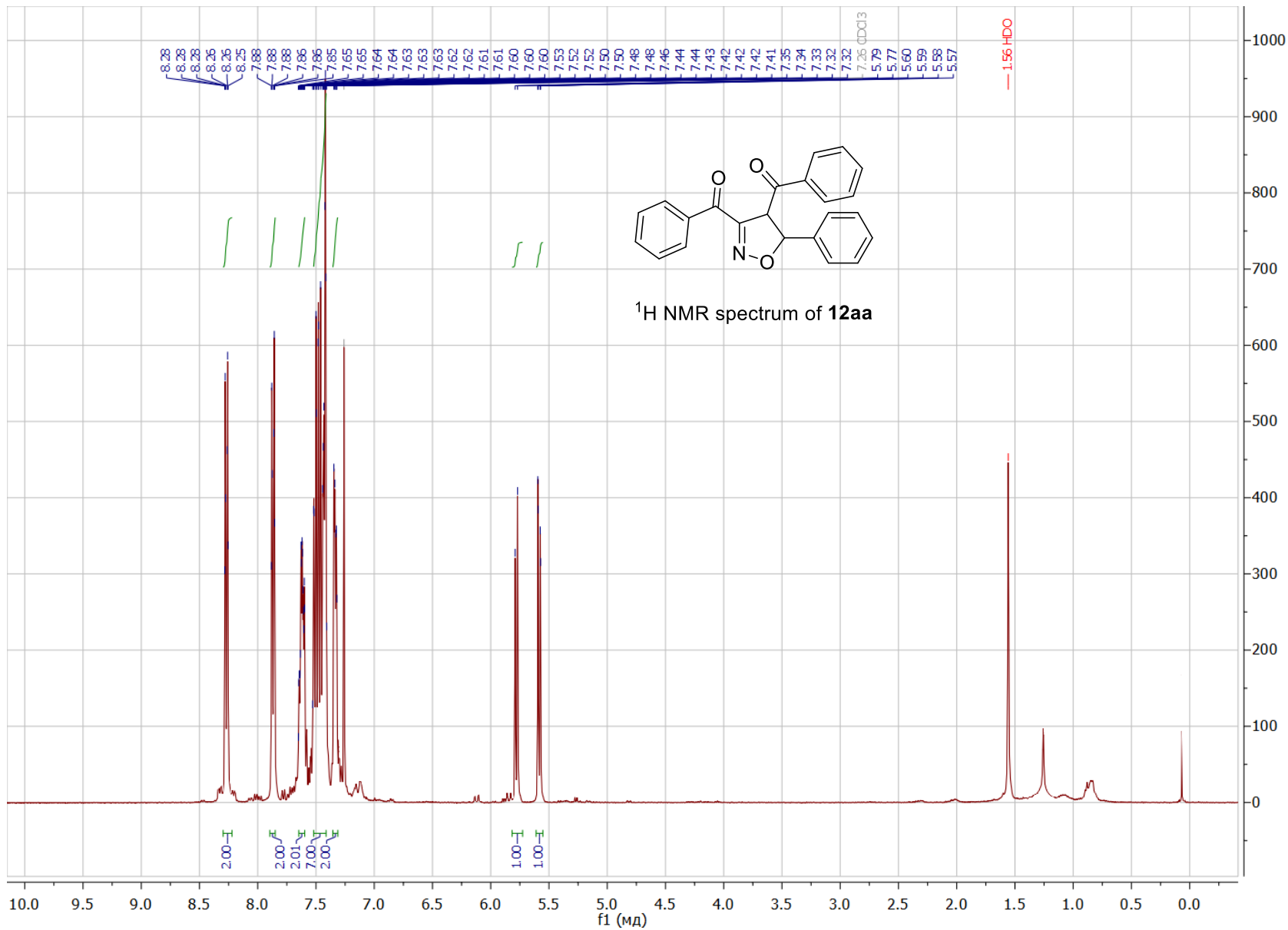


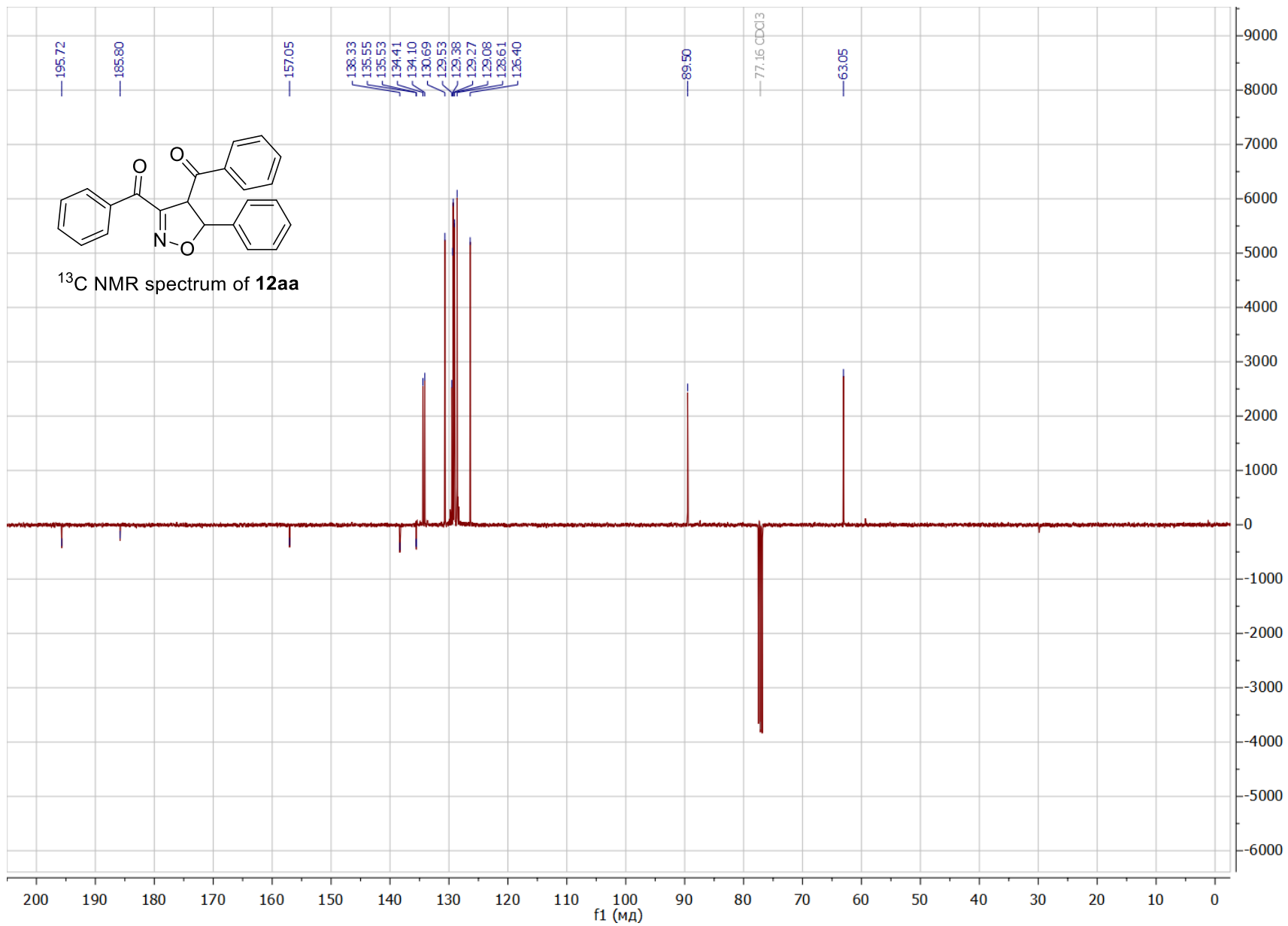
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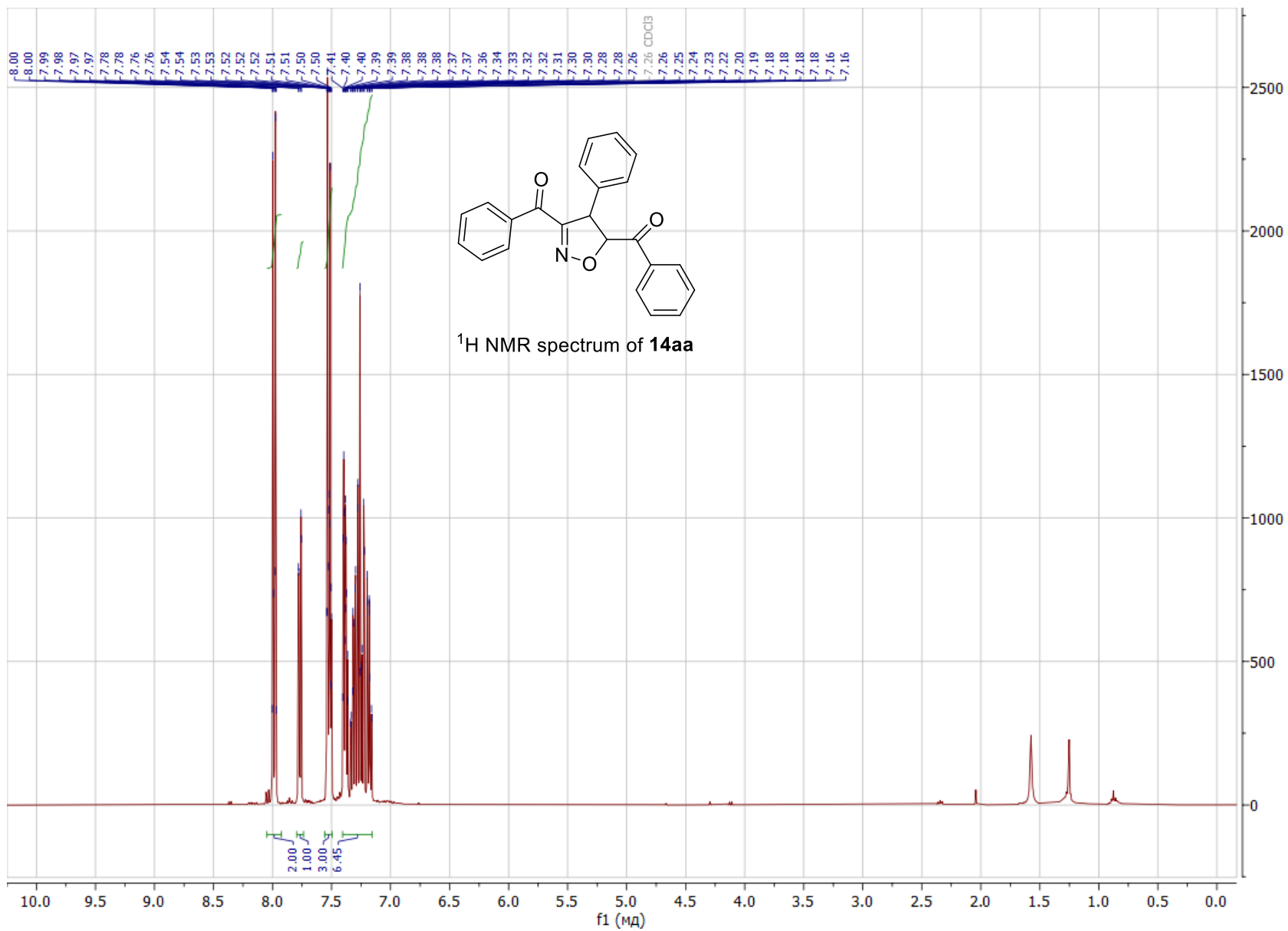


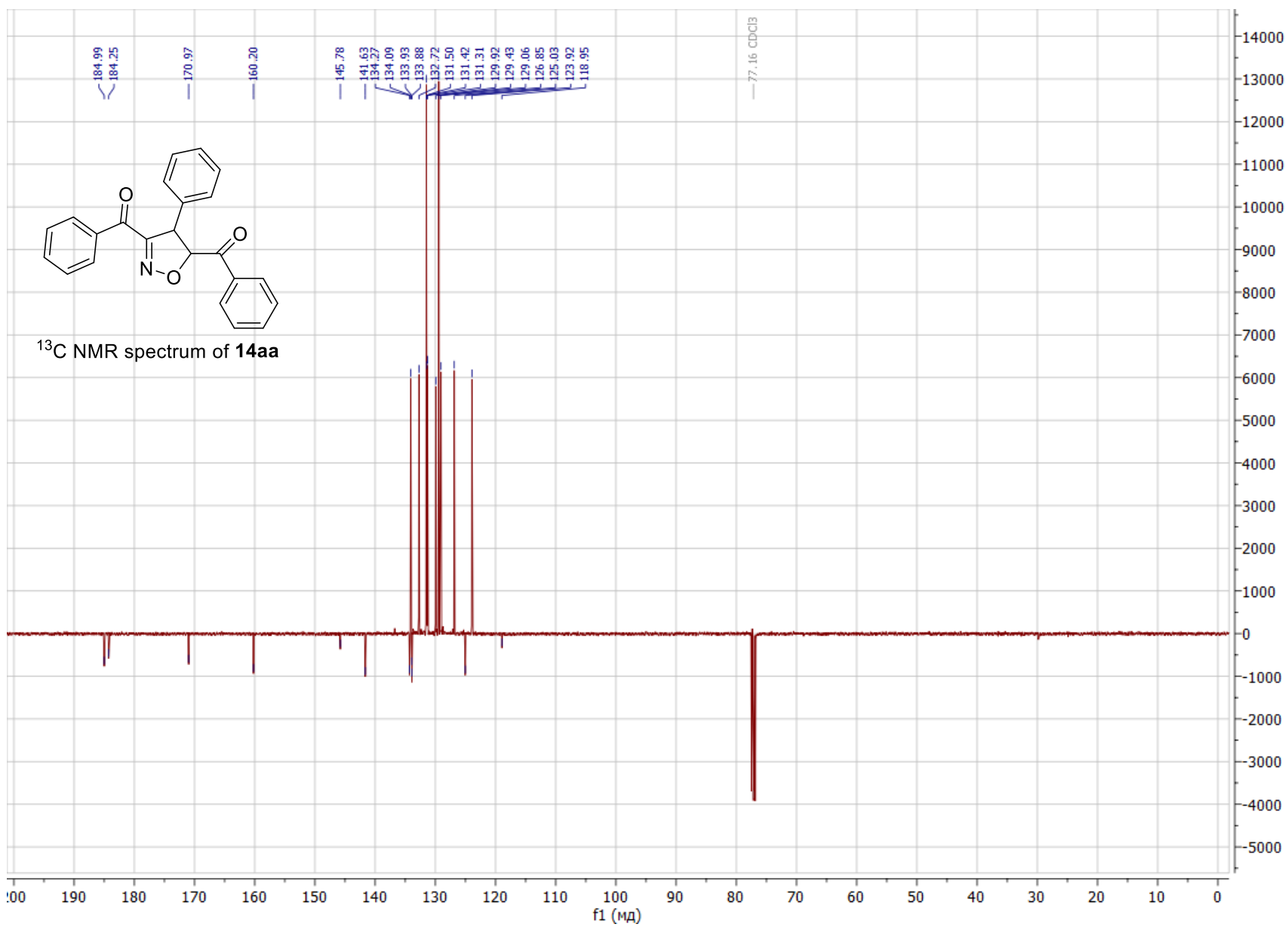


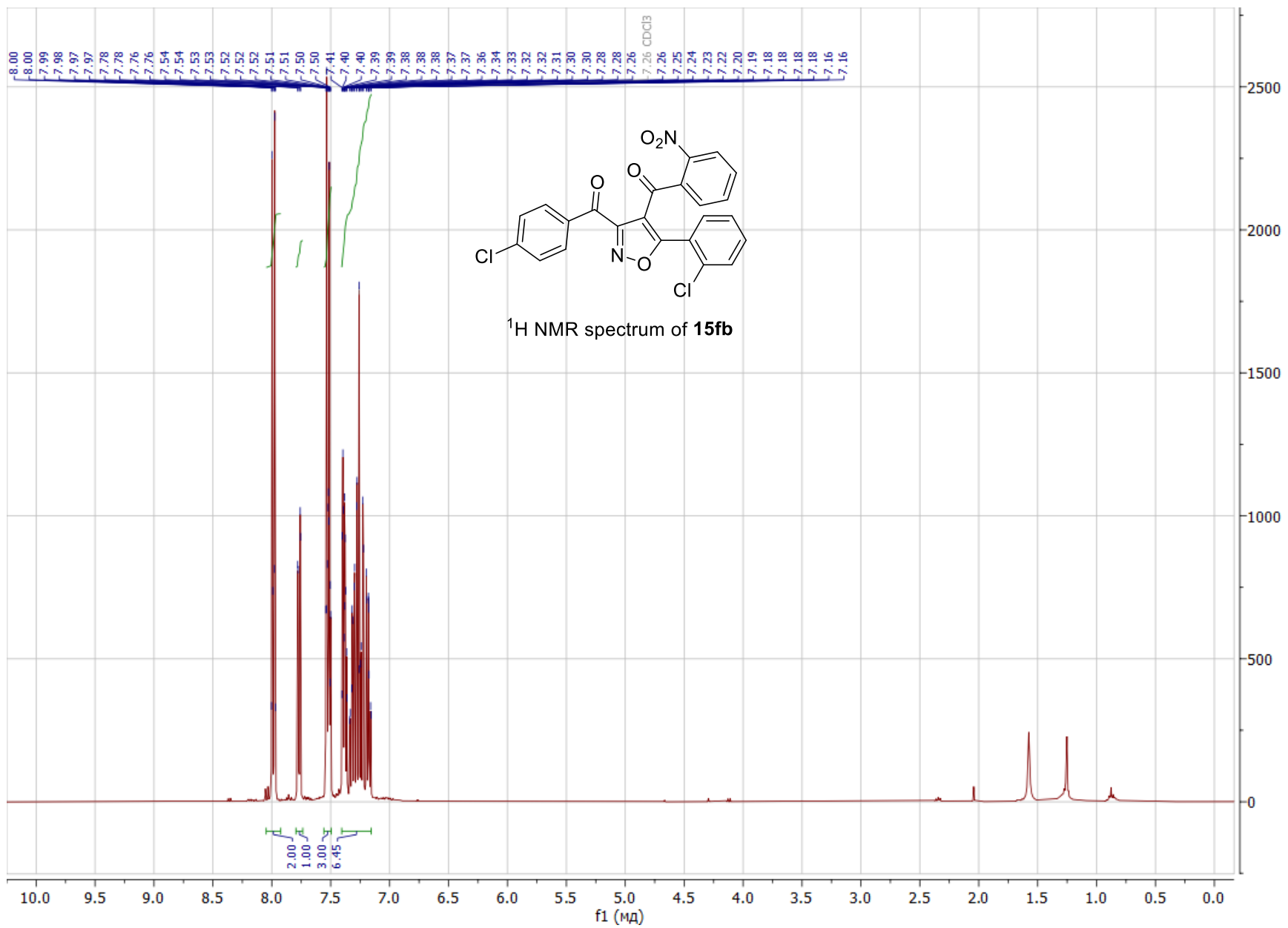
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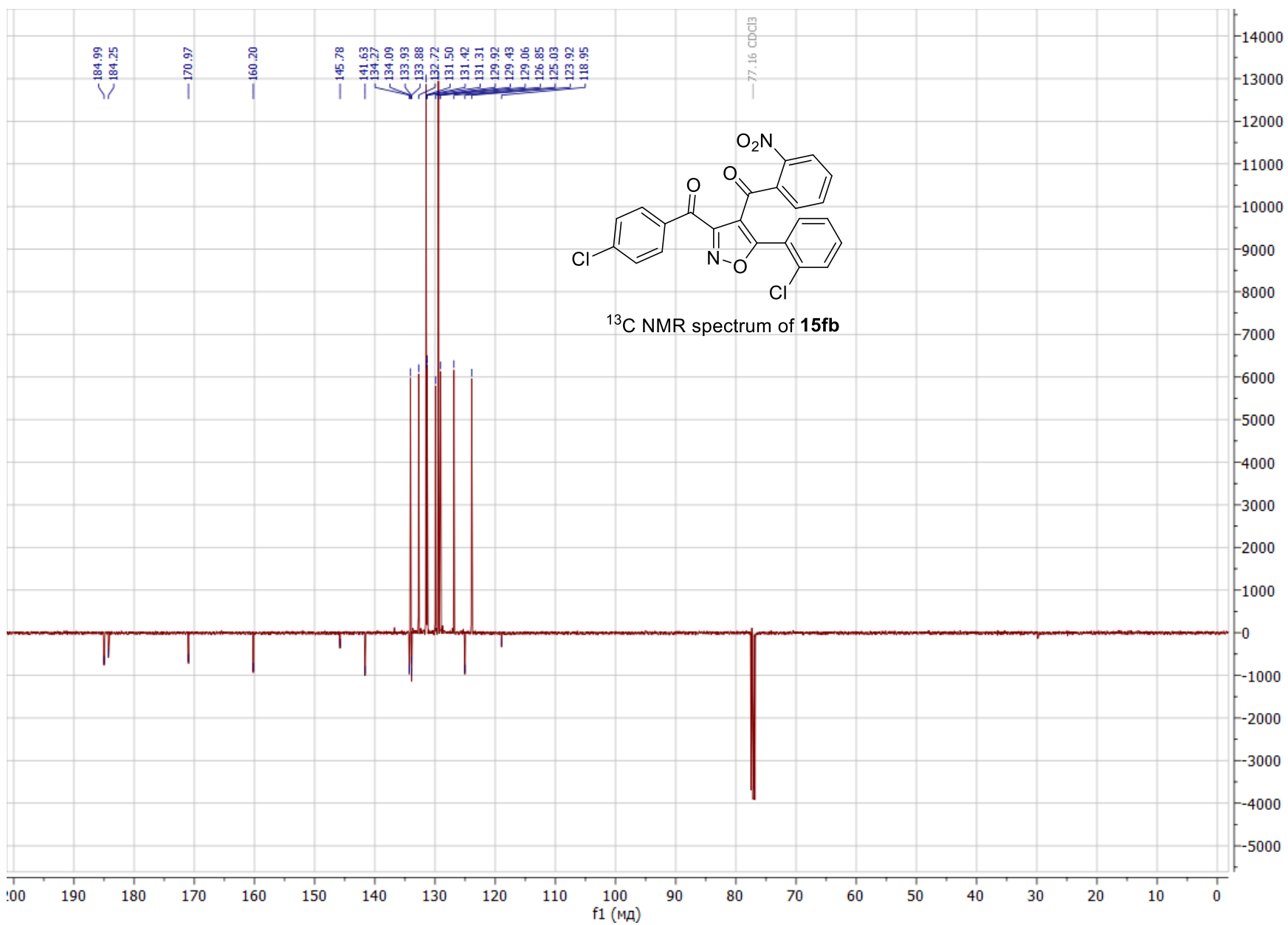




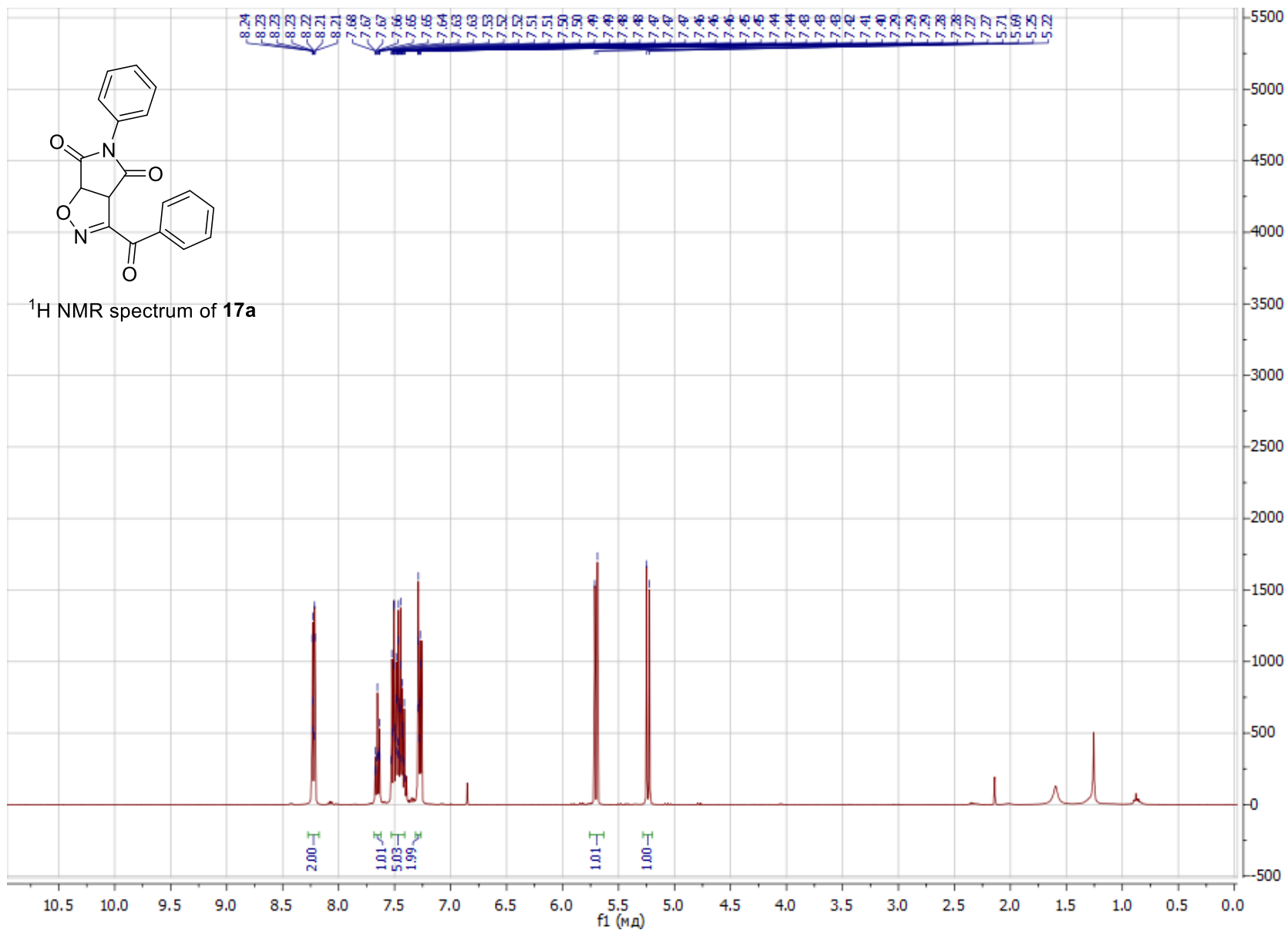


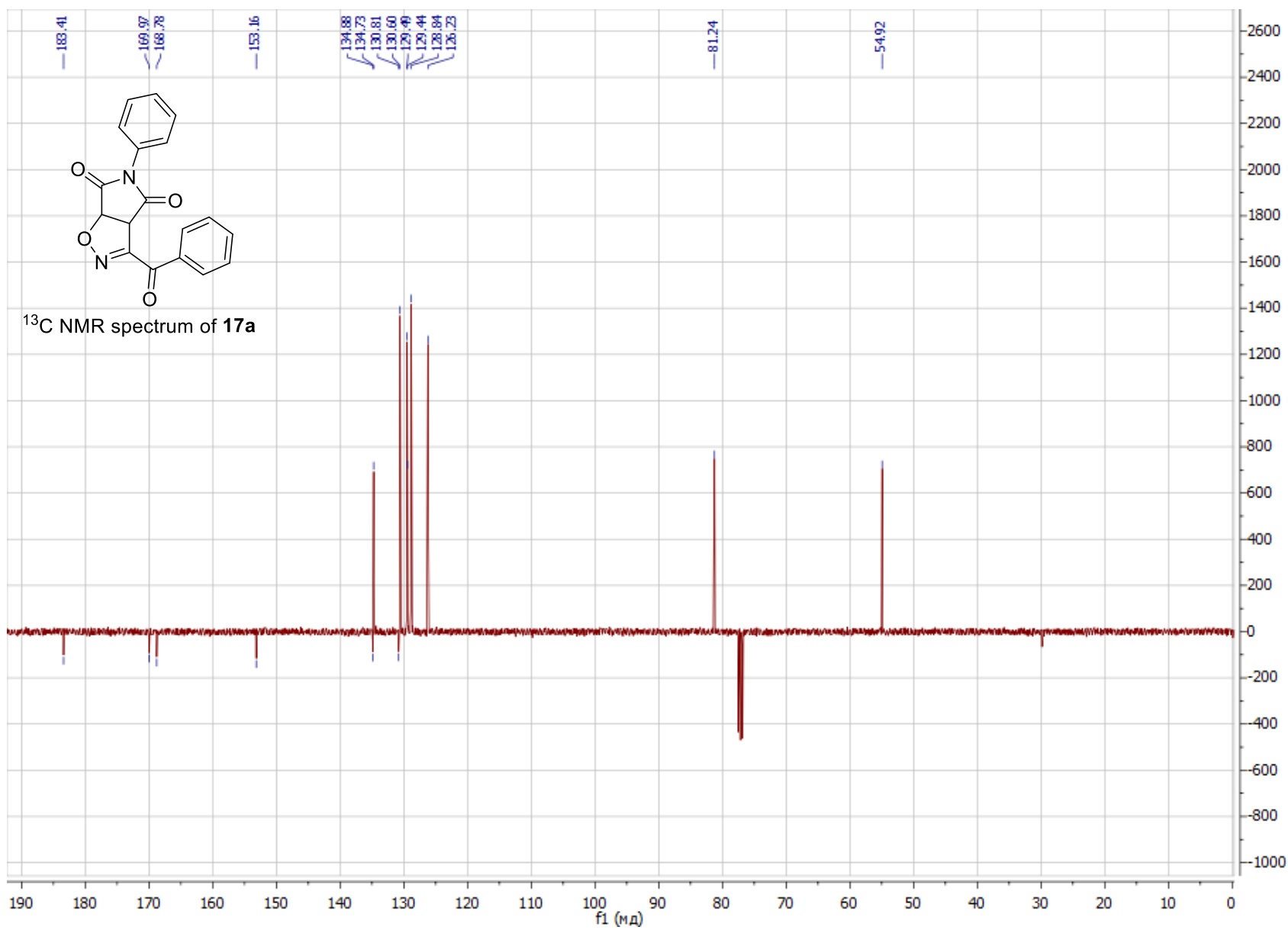




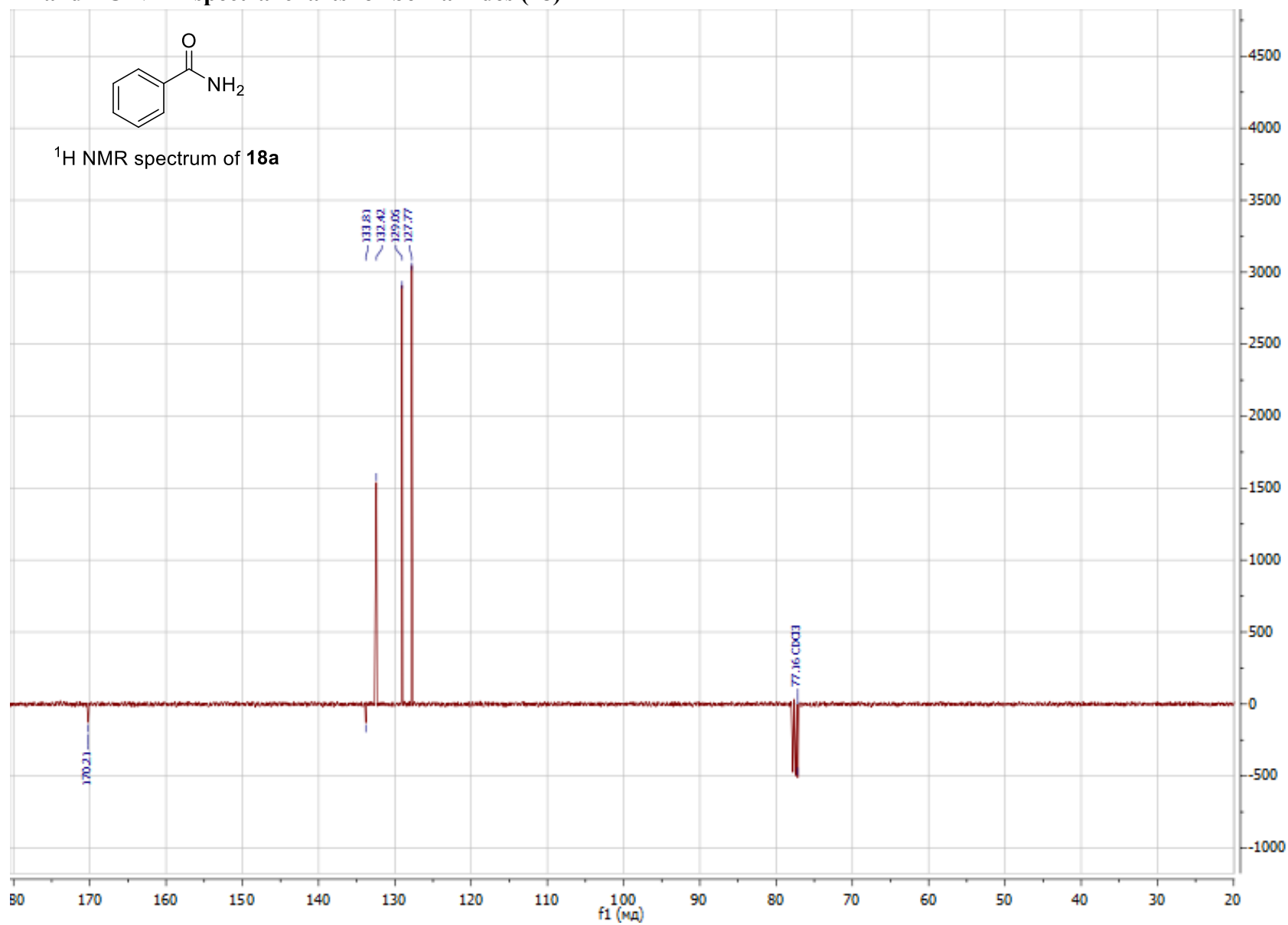


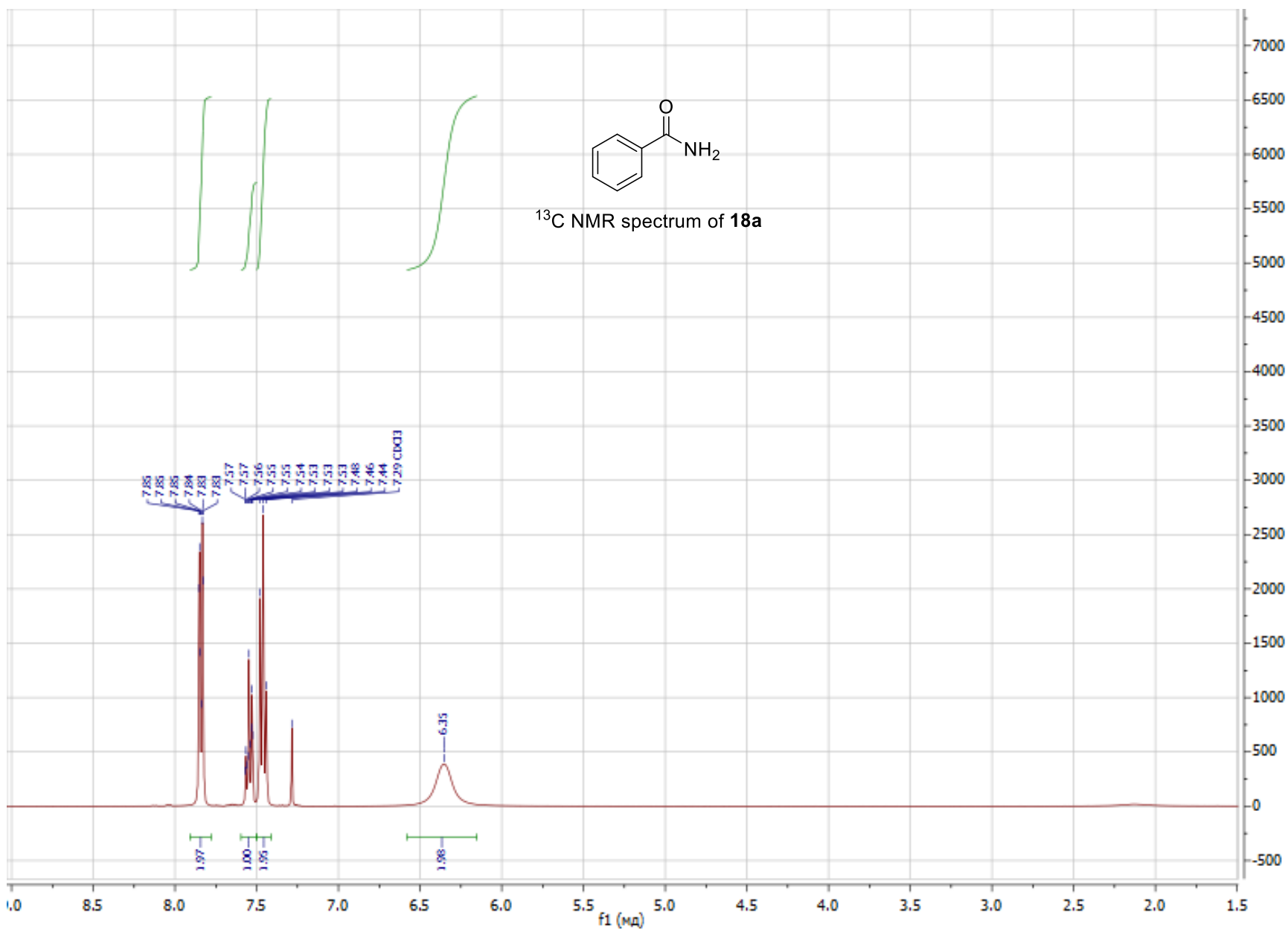
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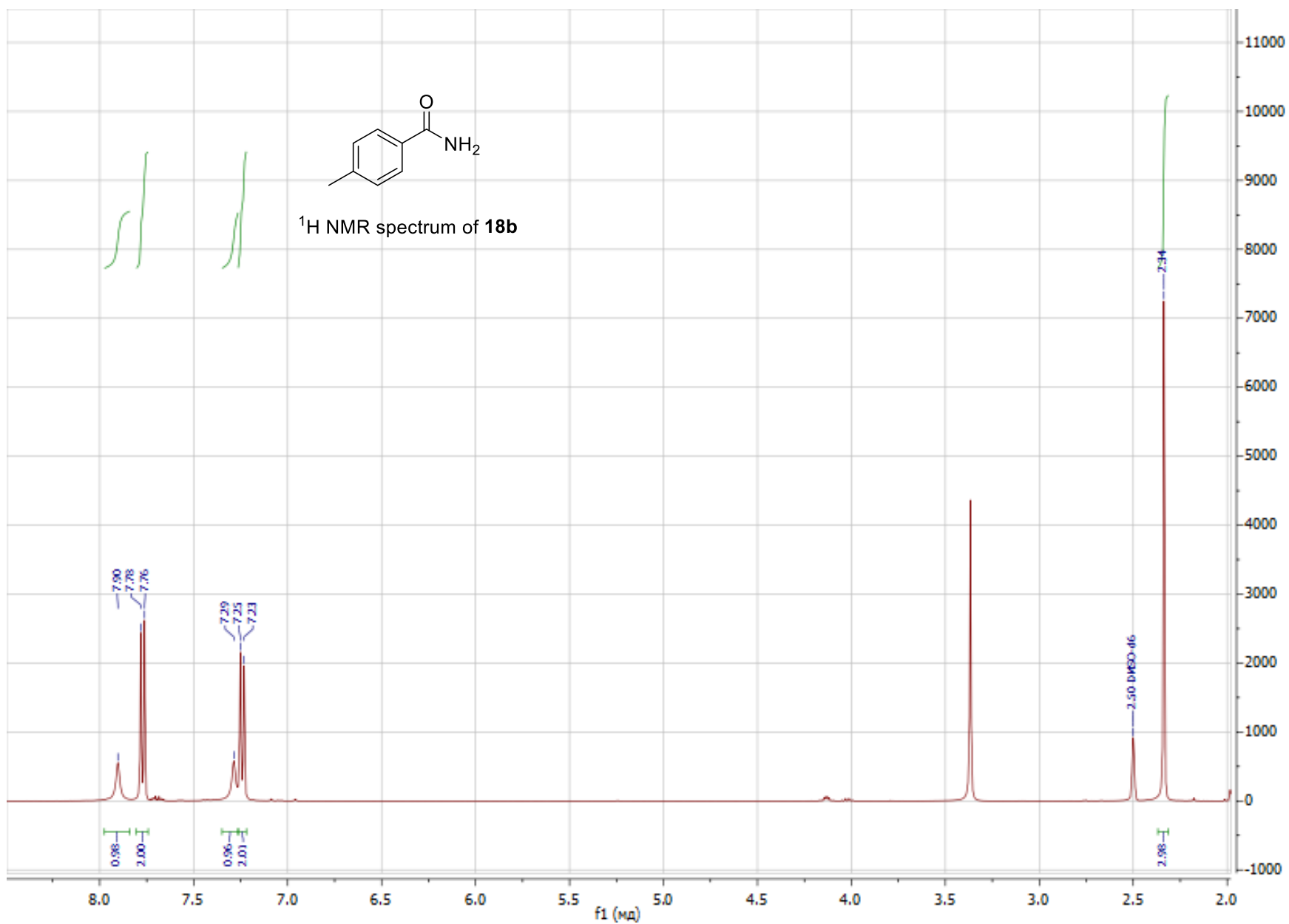


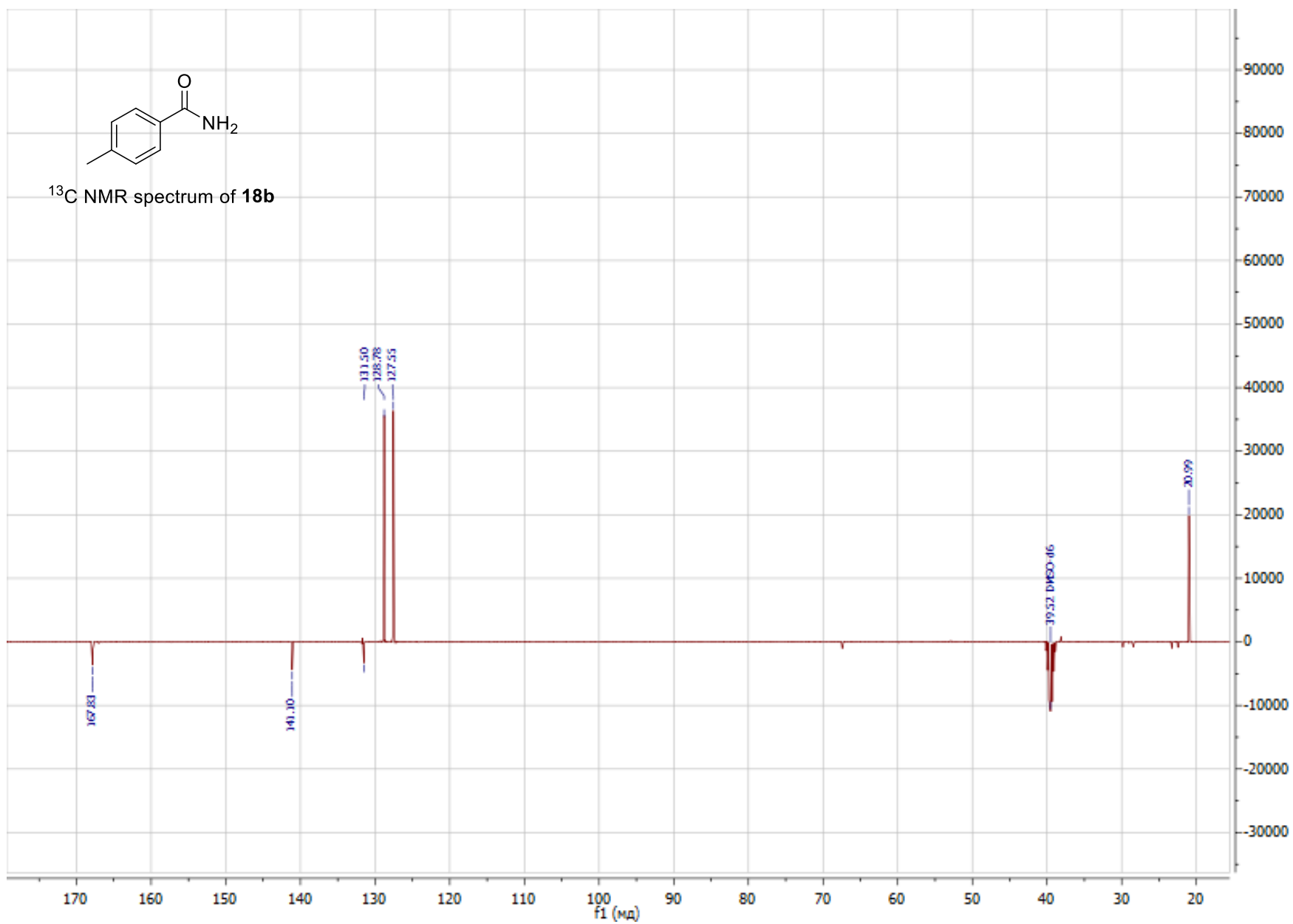


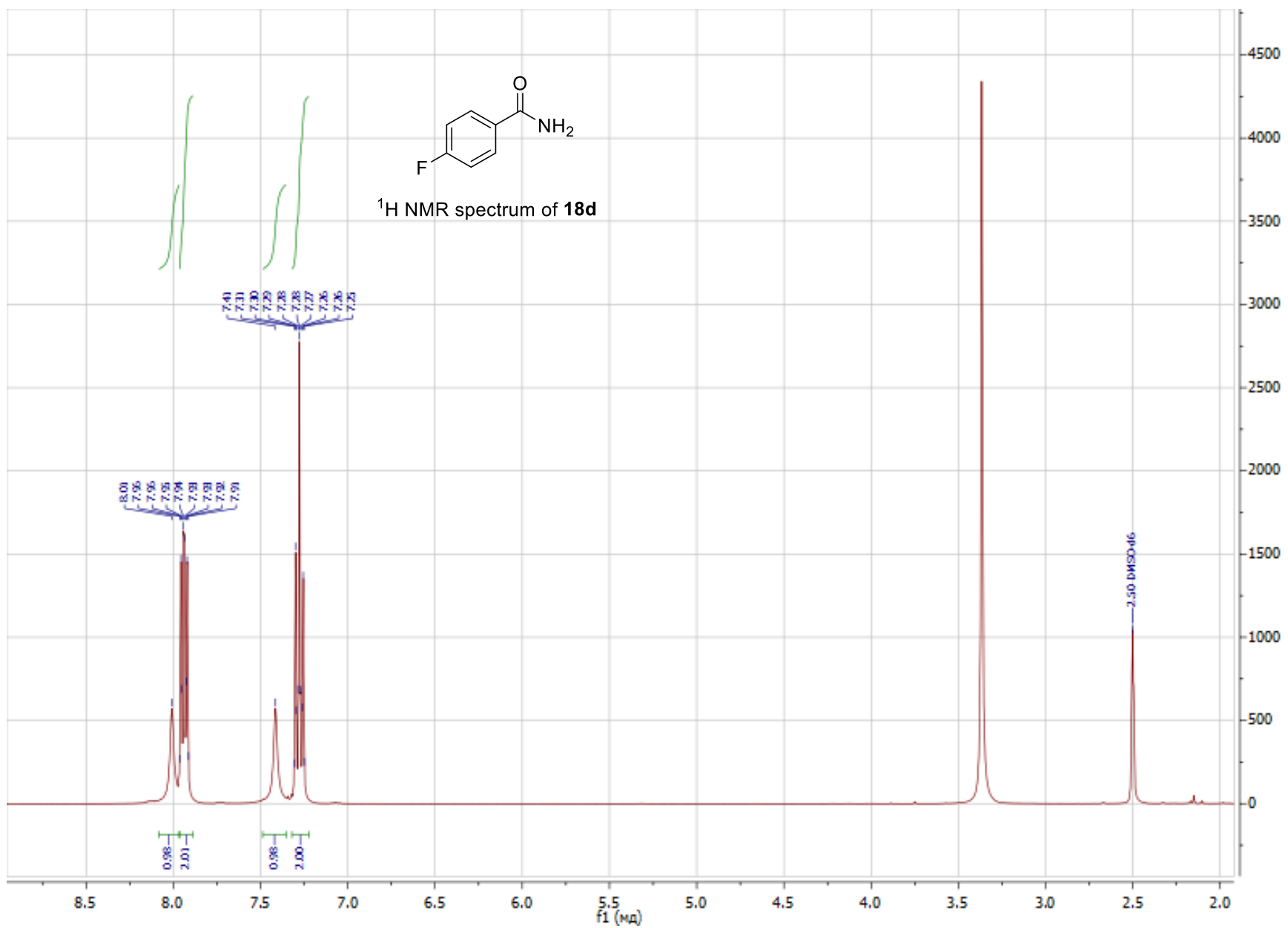
¹H and ¹³C NMR spectral charts for benzamides (18)

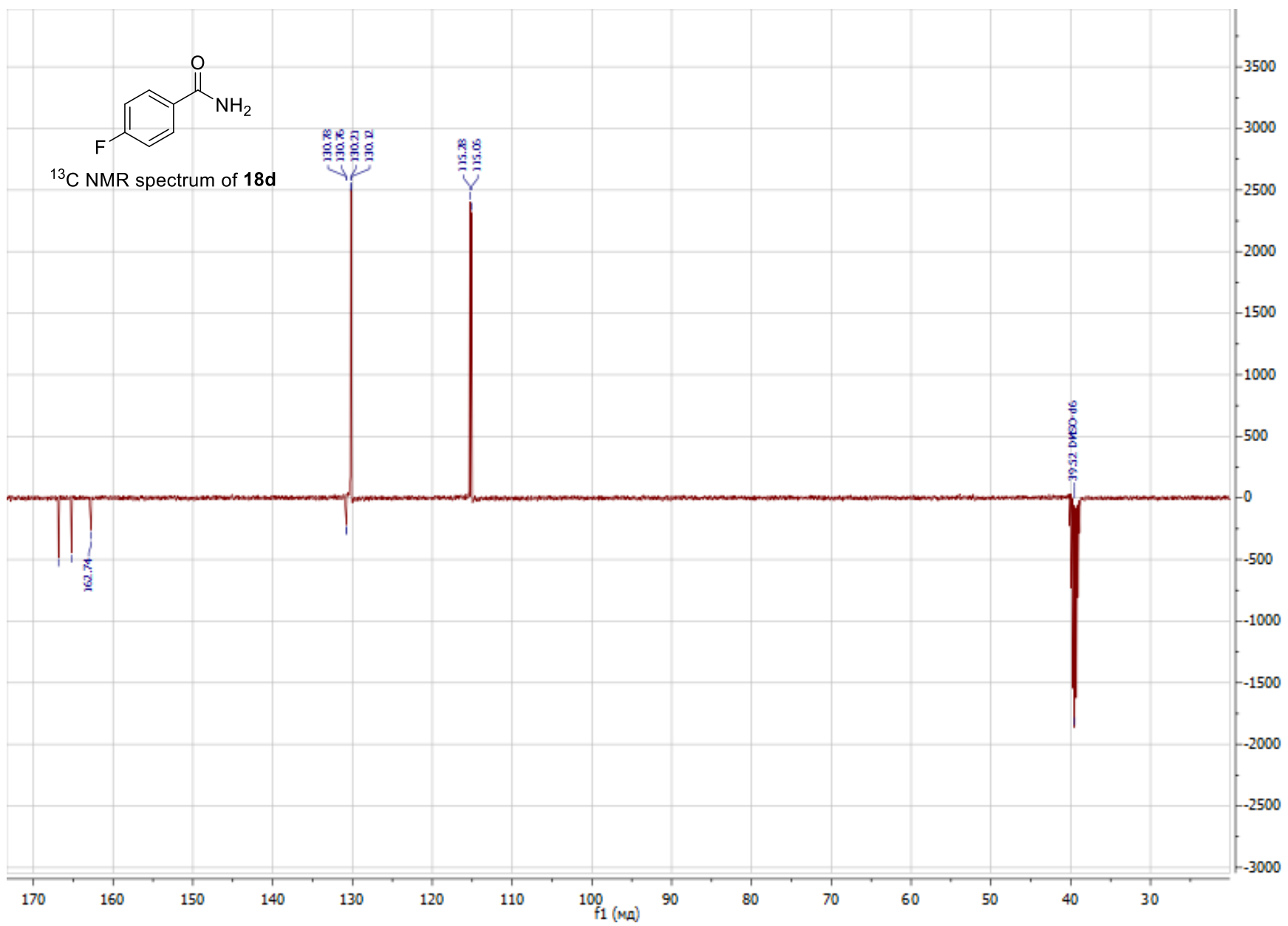


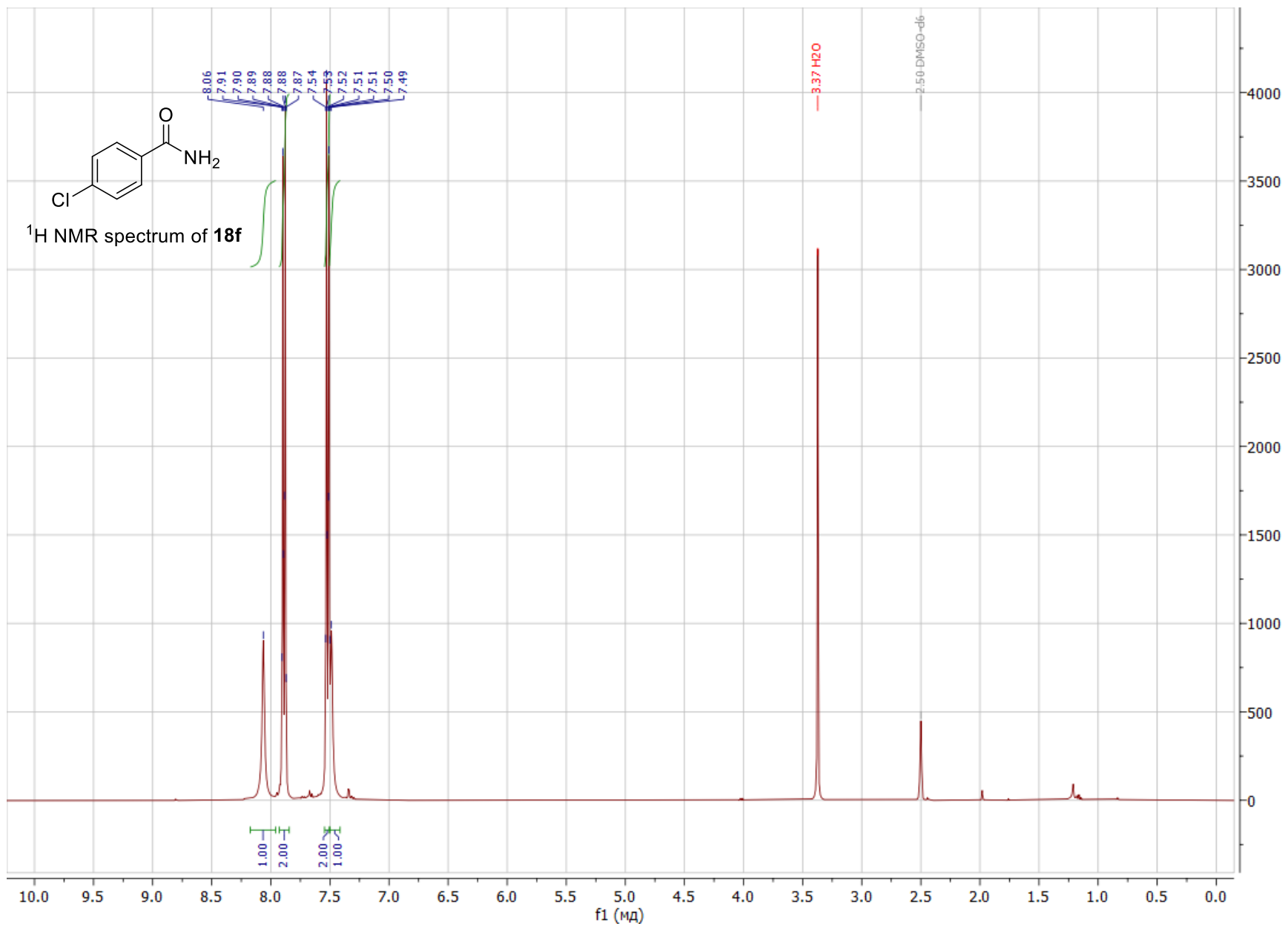


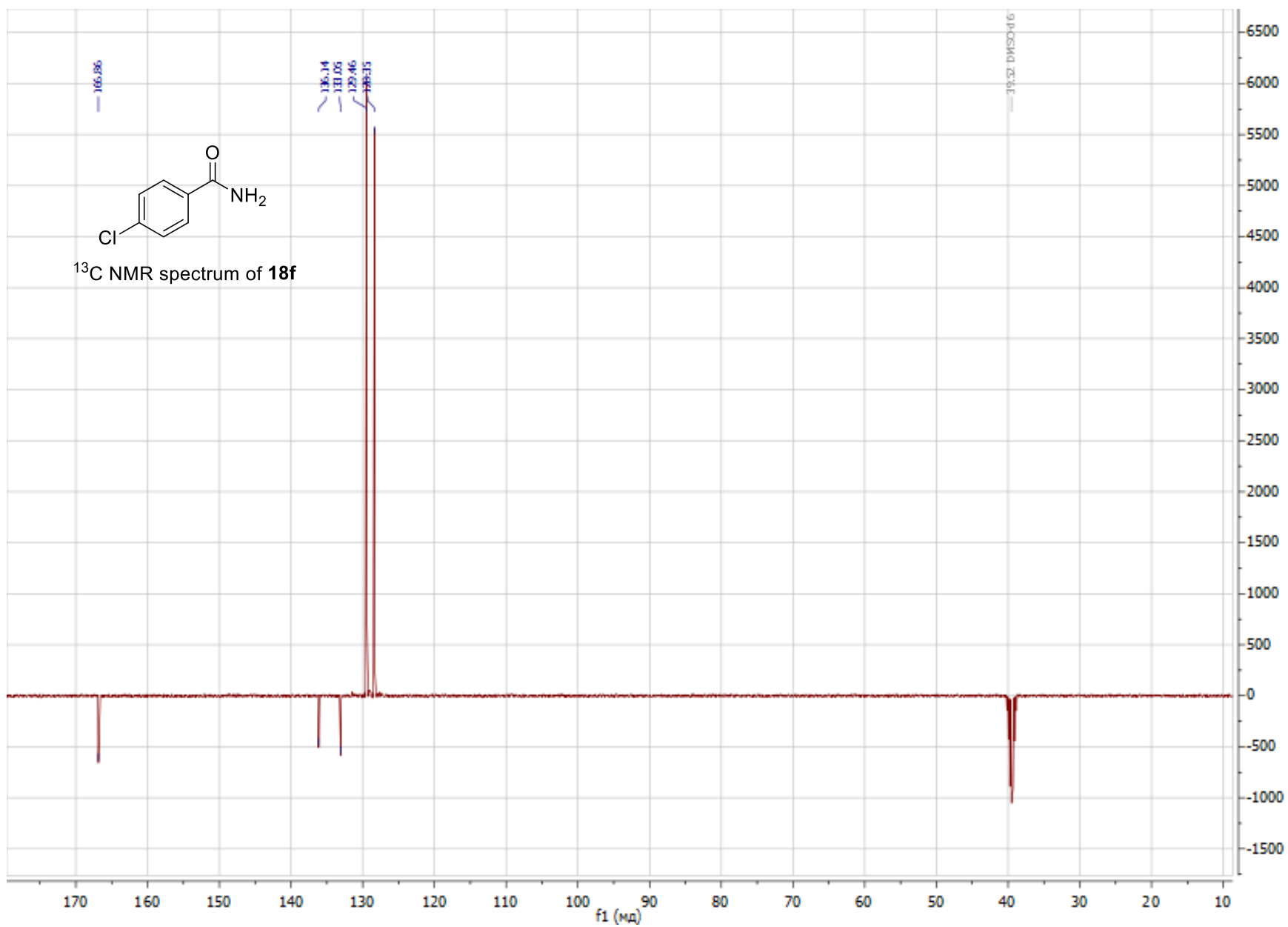


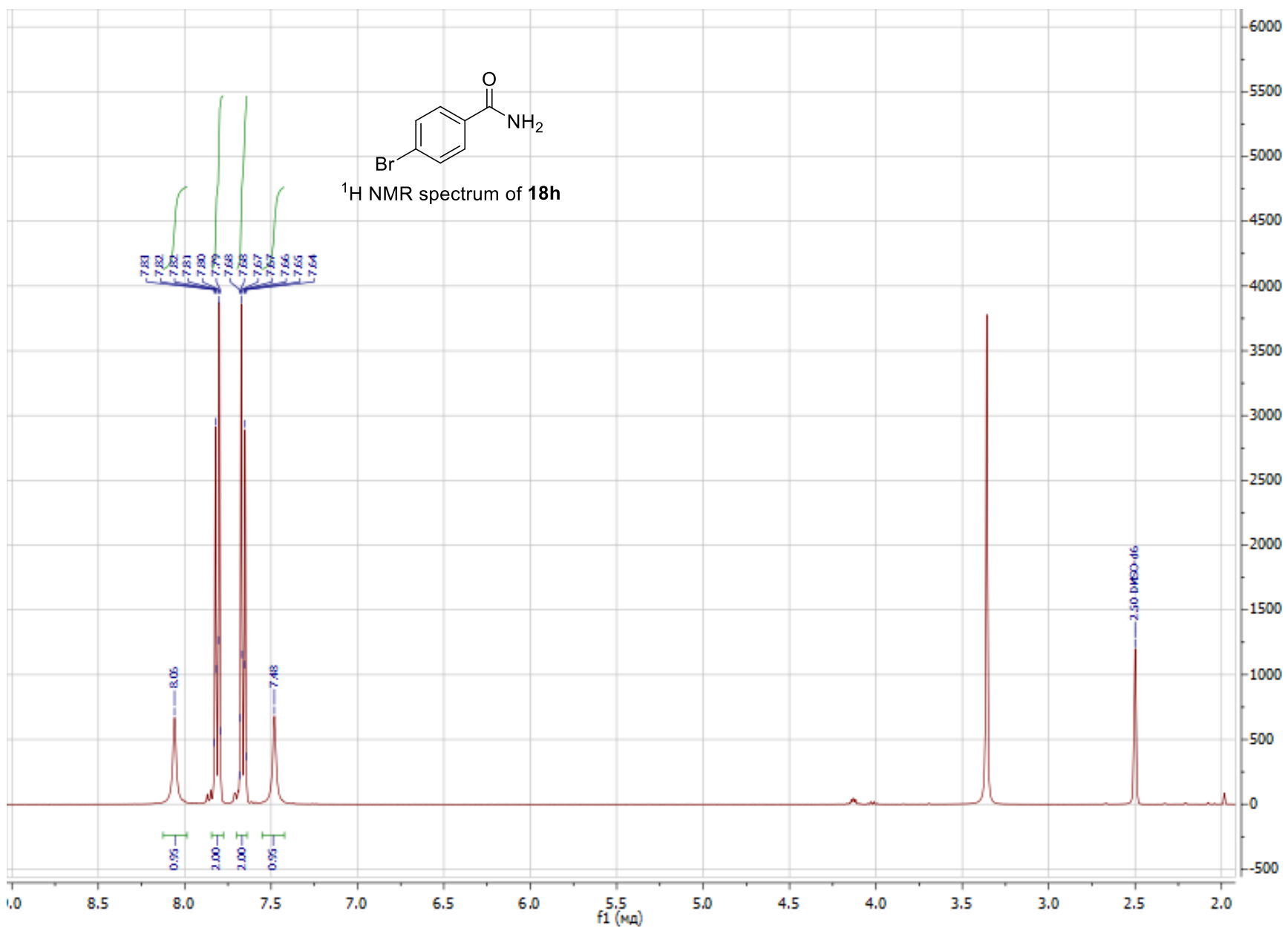


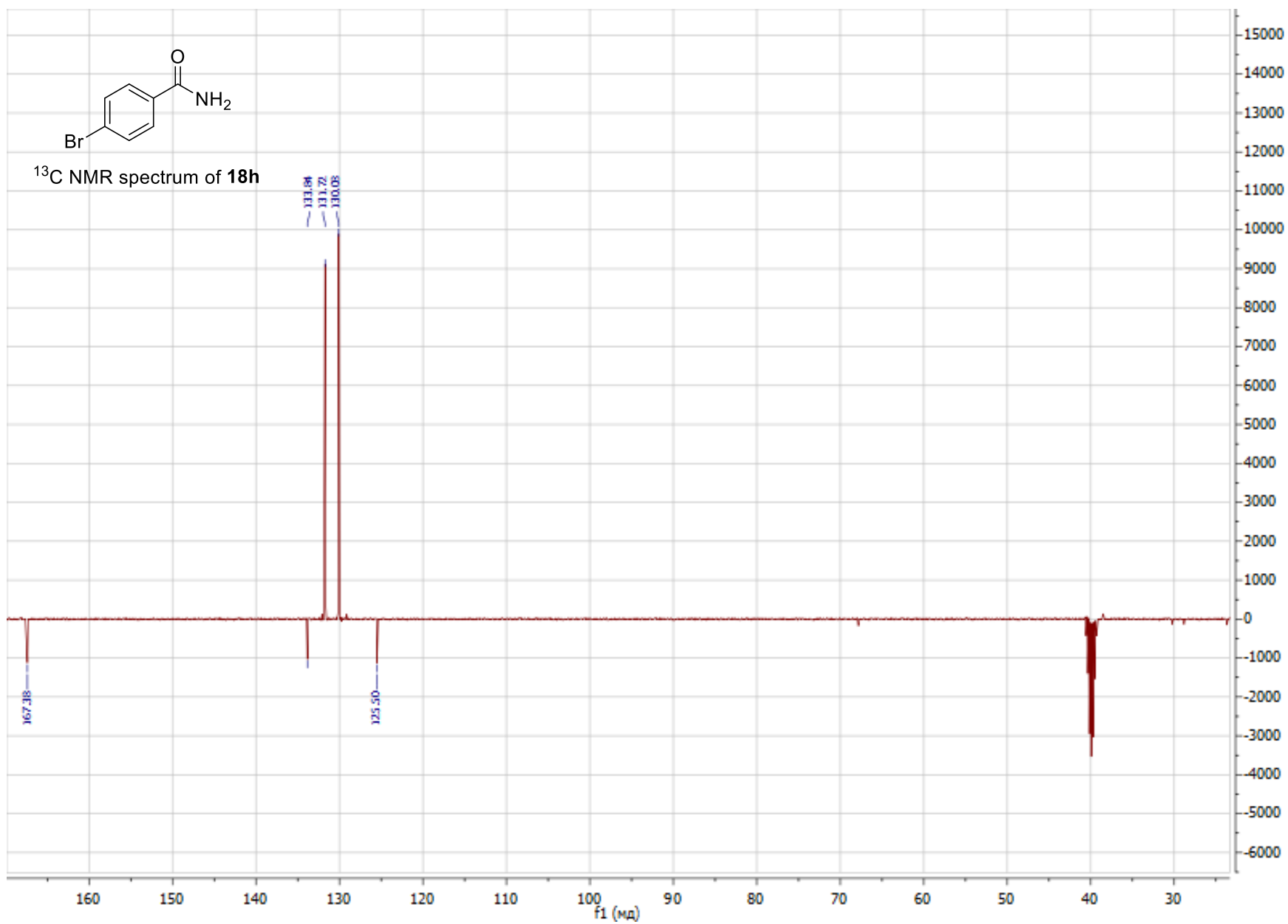




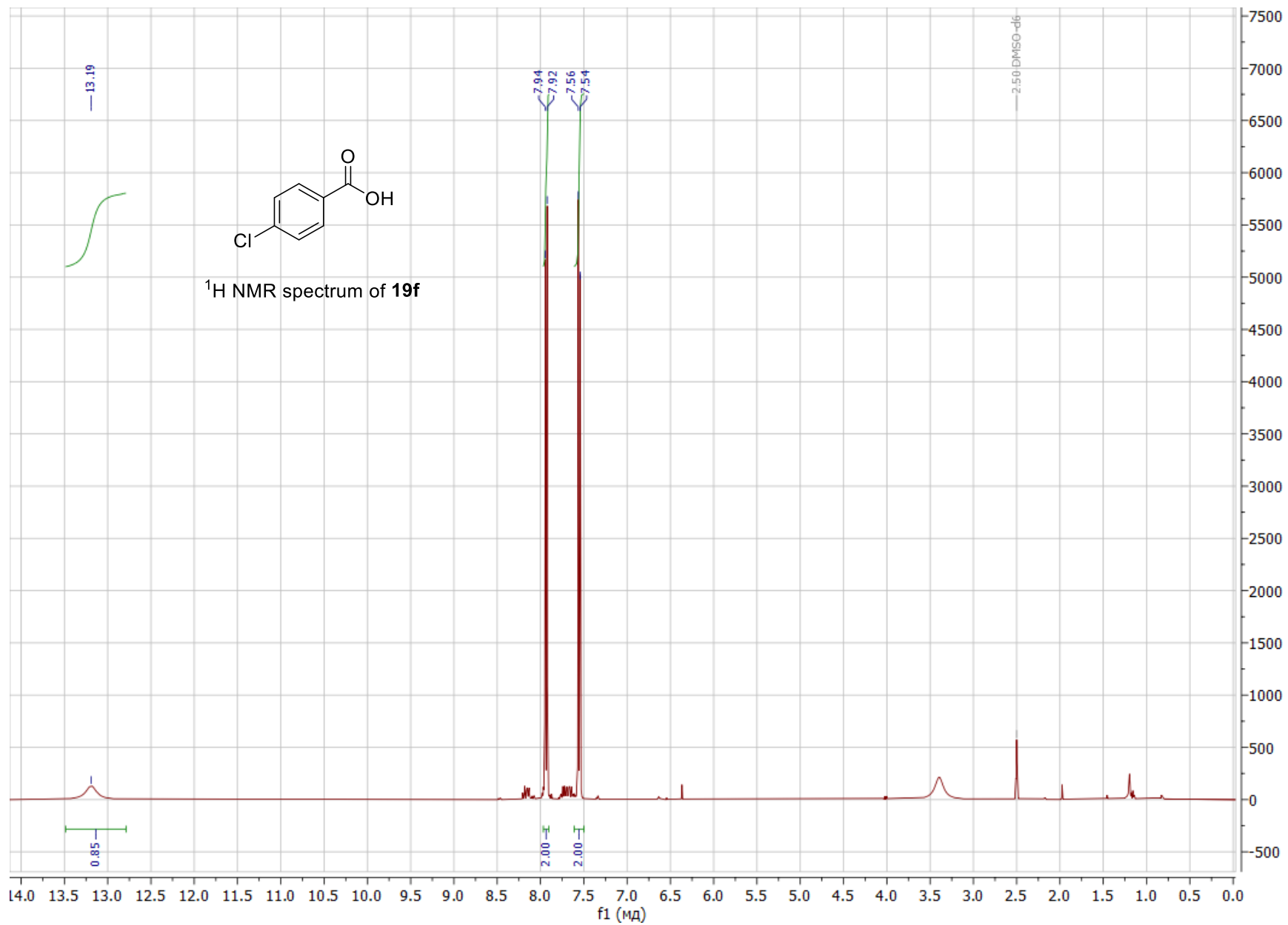


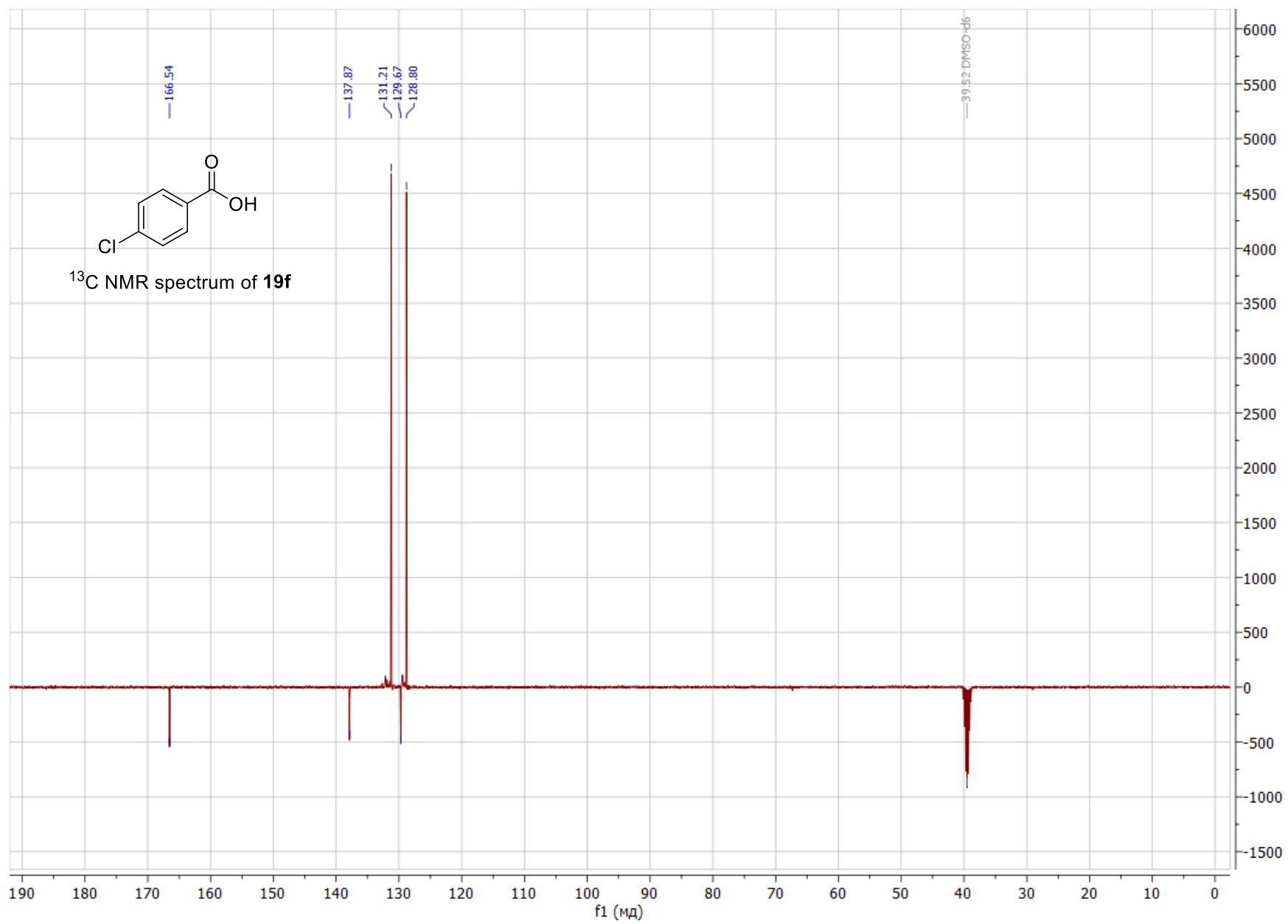






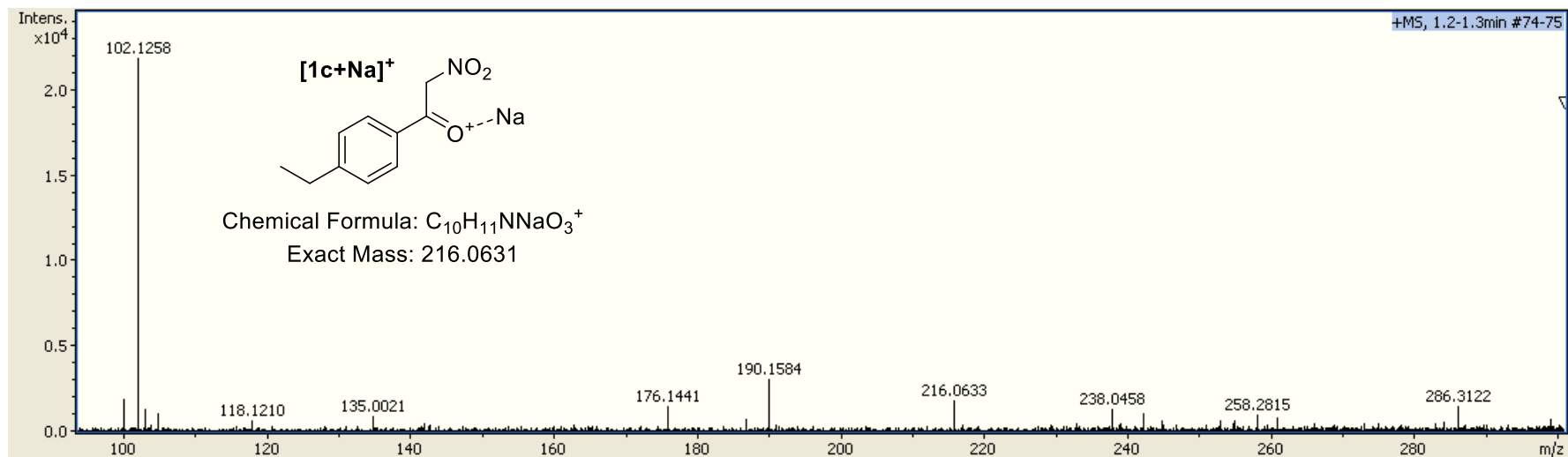
^1H and ^{13}C NMR spectral charts for starting 4-chlorobenzoic acid (19f)



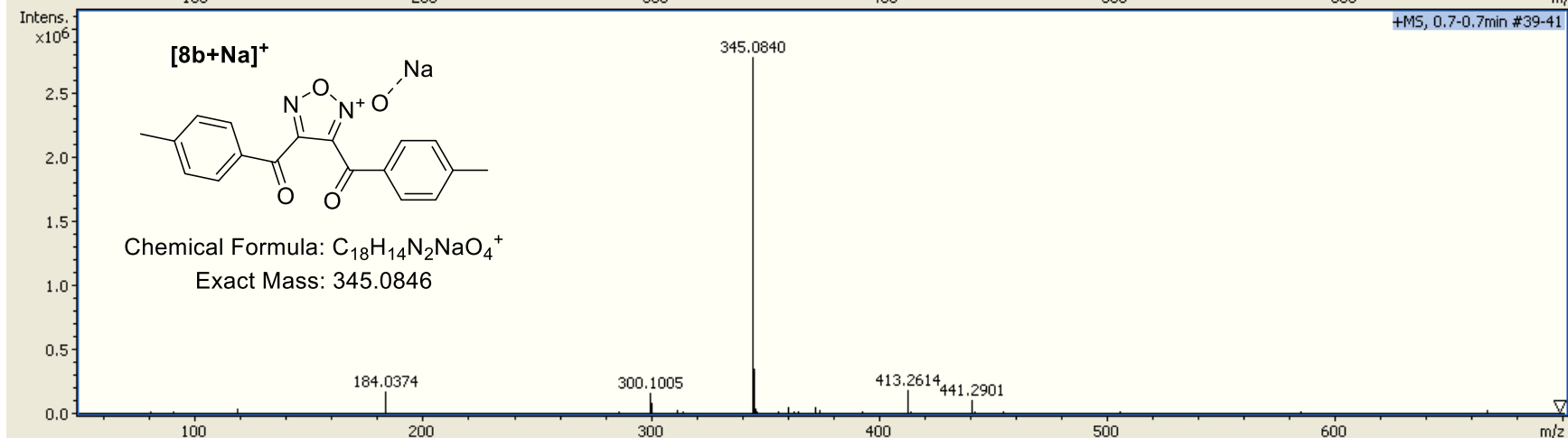
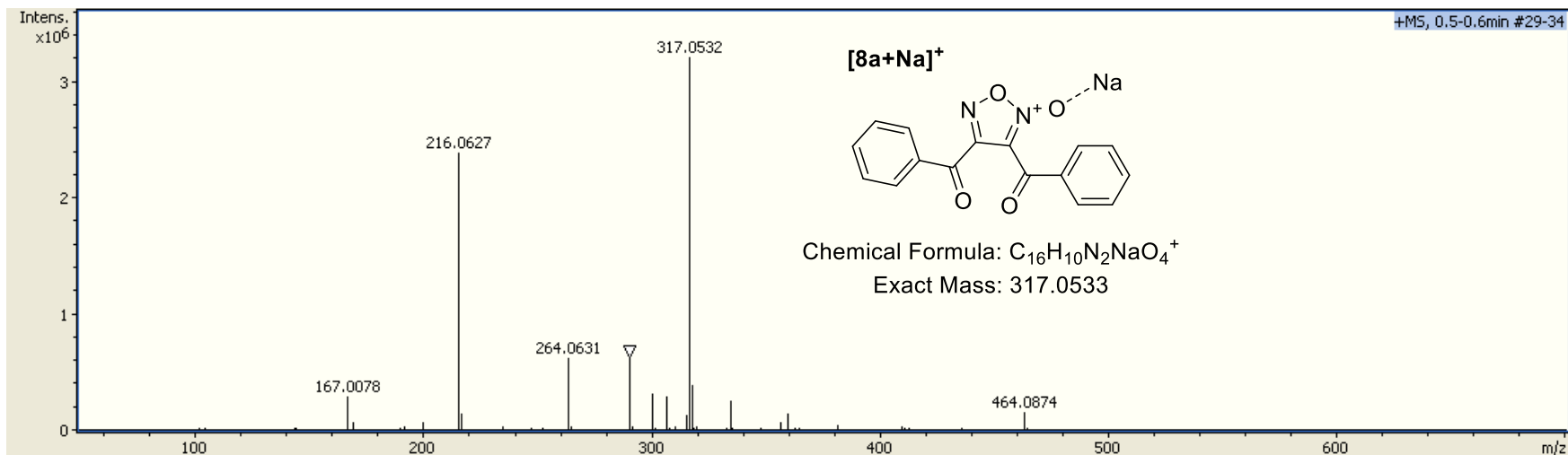


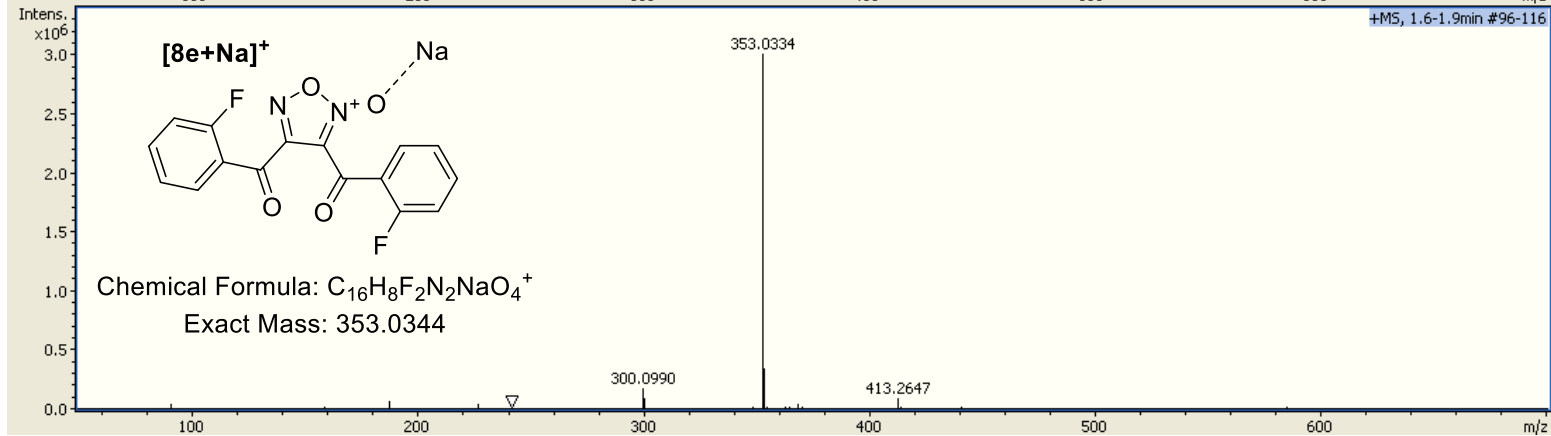
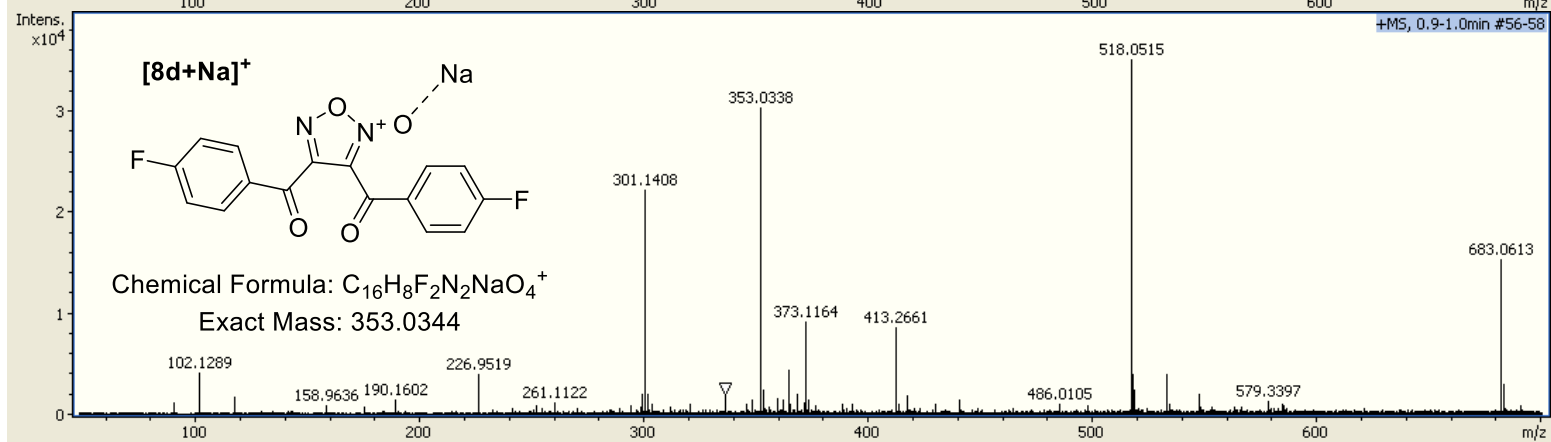
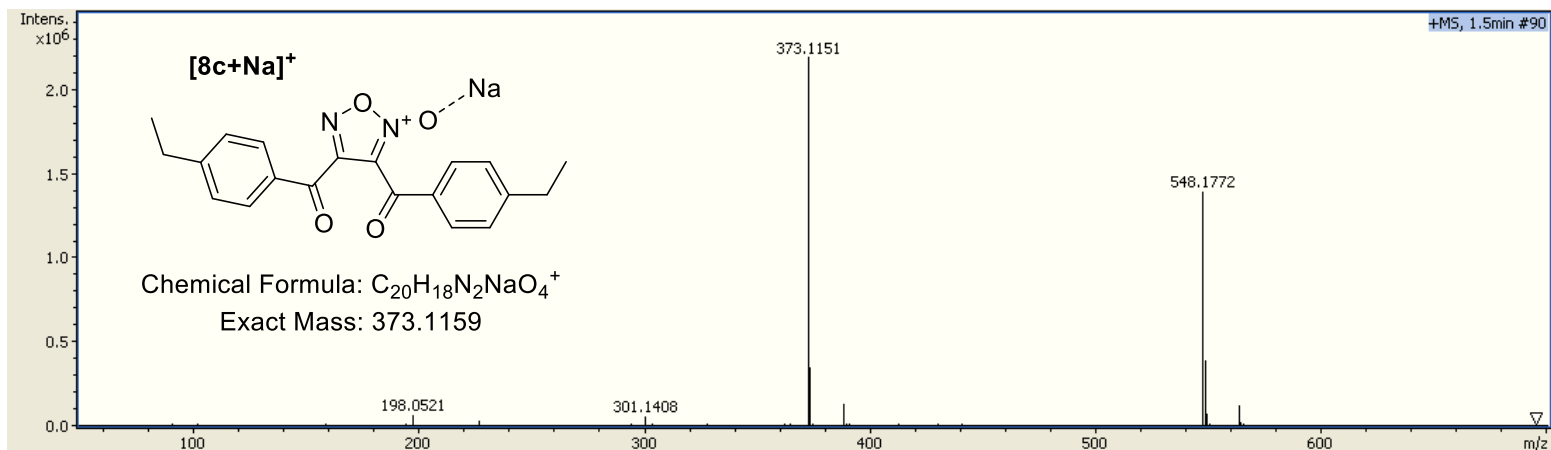
HRMS spectral charts

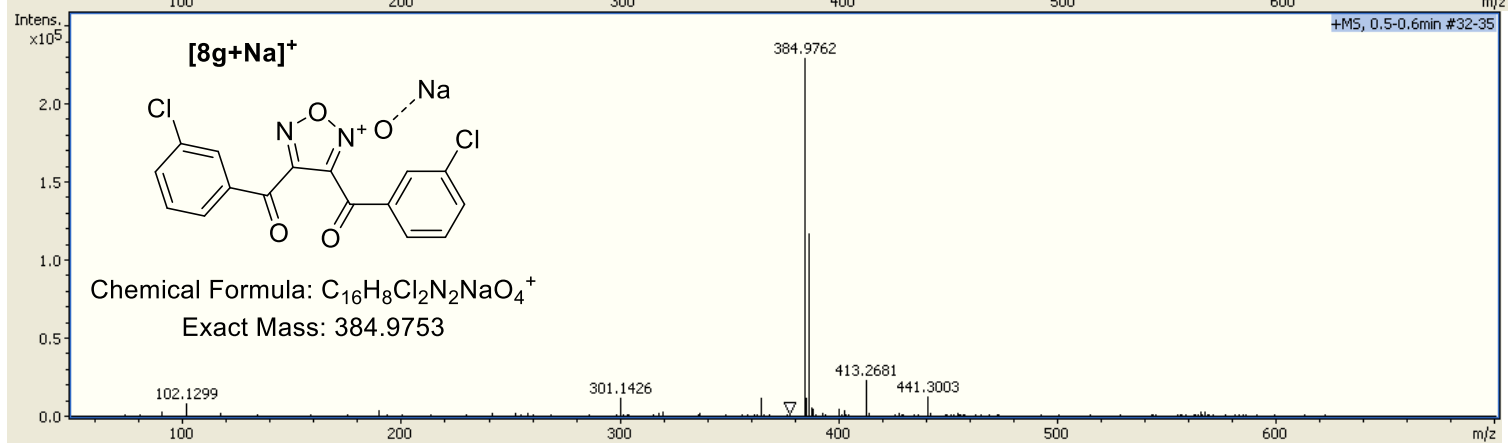
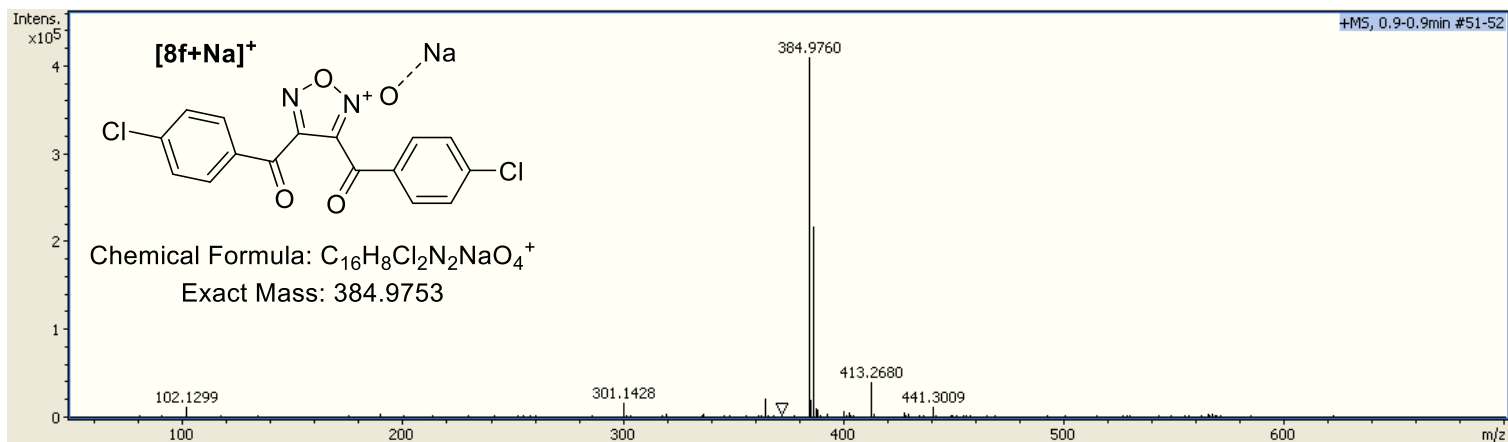
HRMS spectral charts for starting 2-nitro-4'-ethylacetophenone (1c)

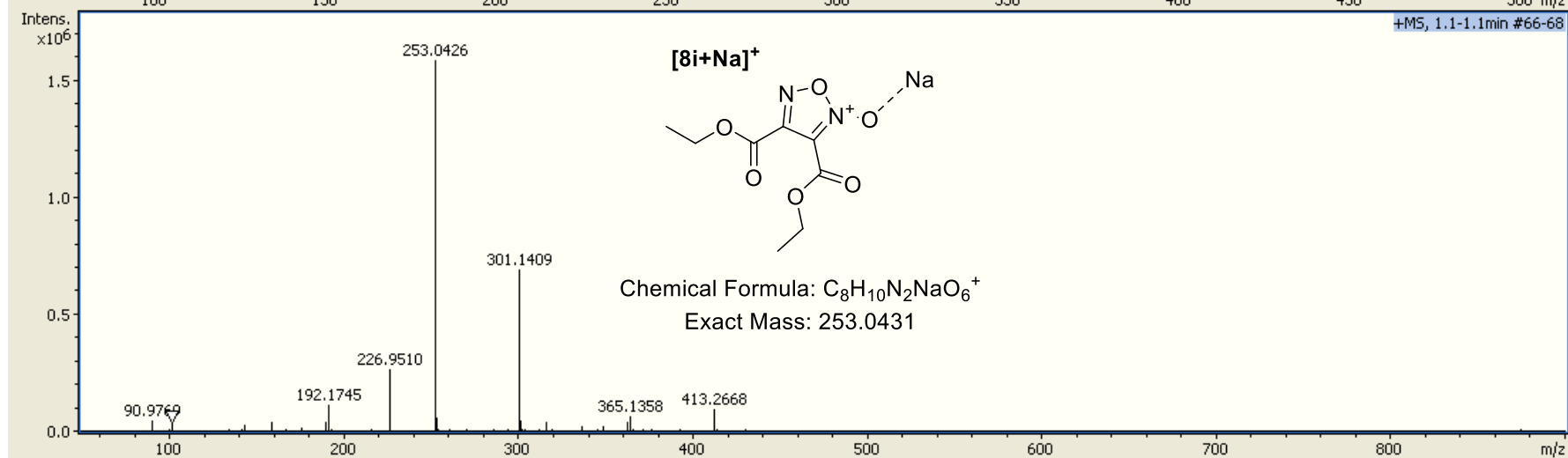
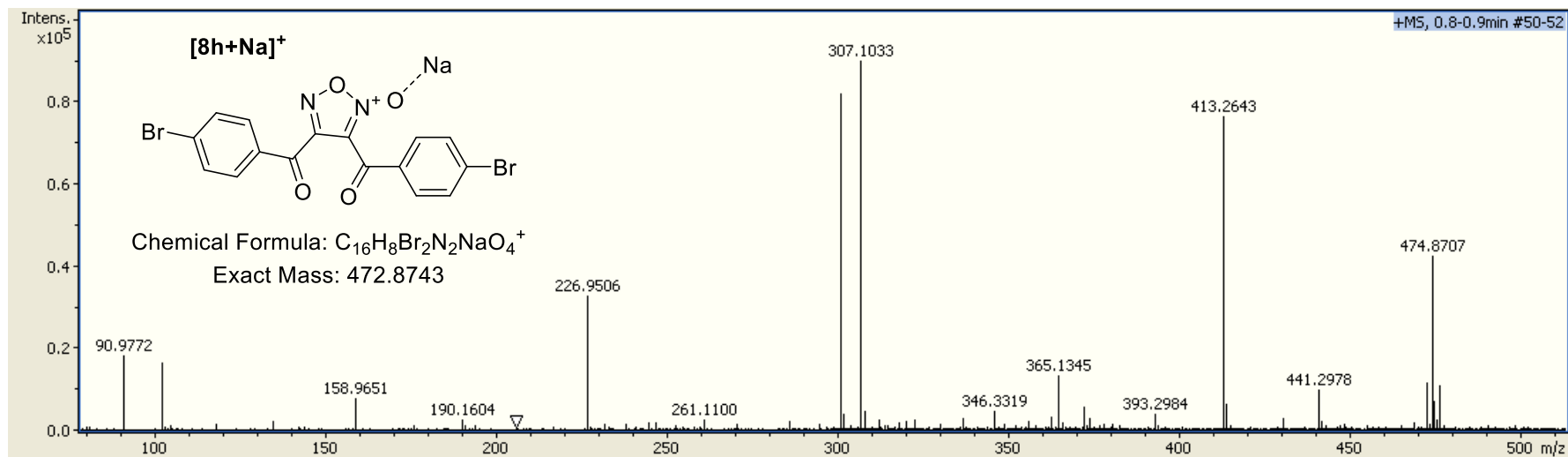


HRMS spectral charts for furoxans 8

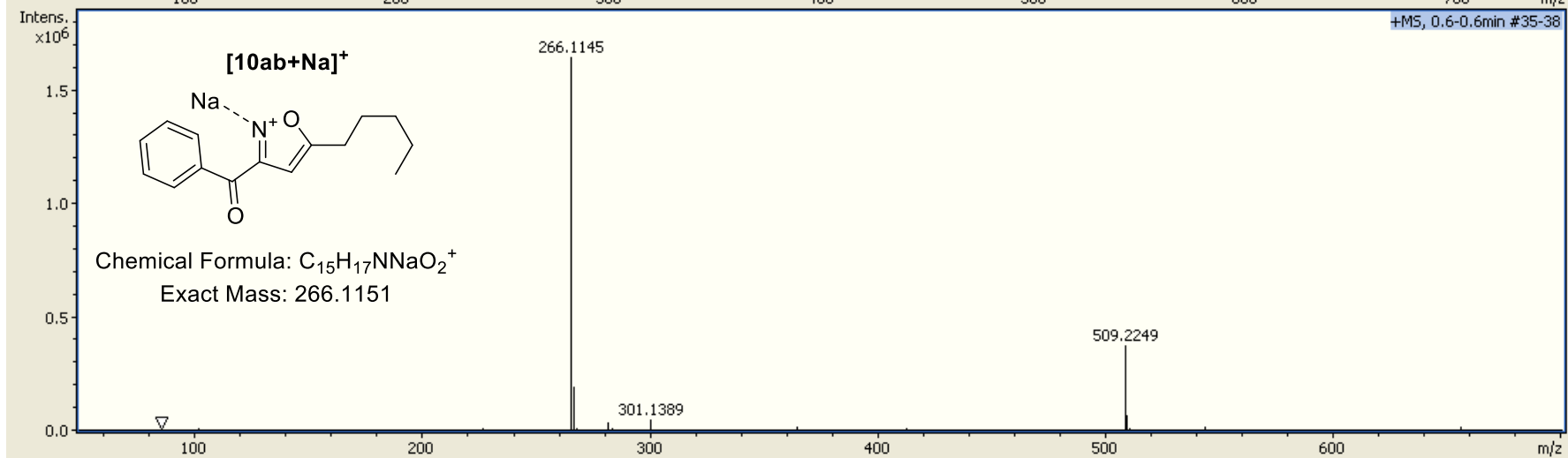
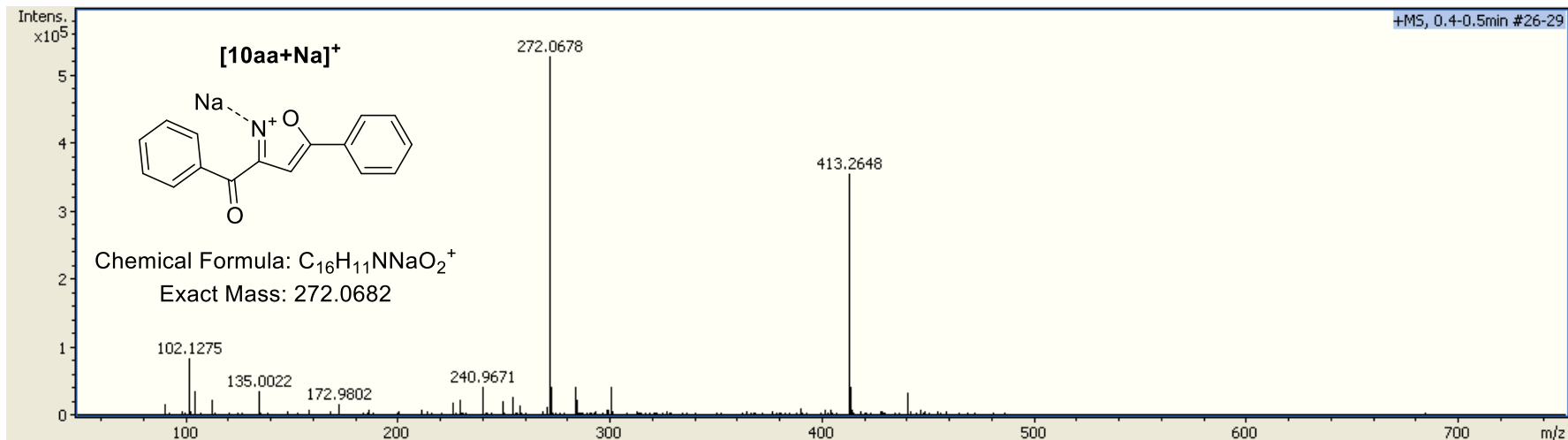


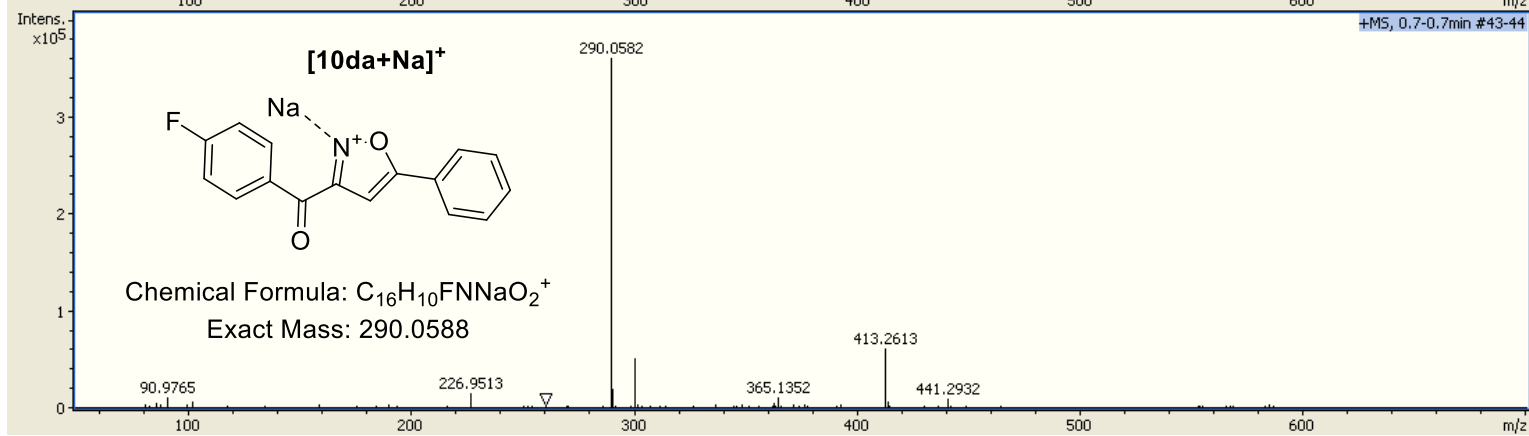
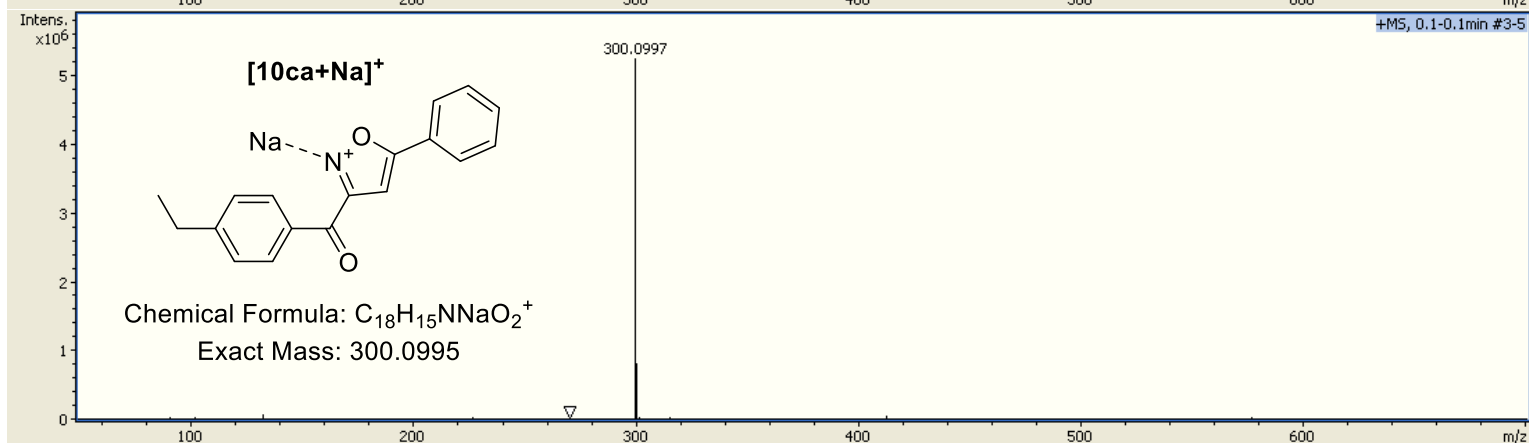
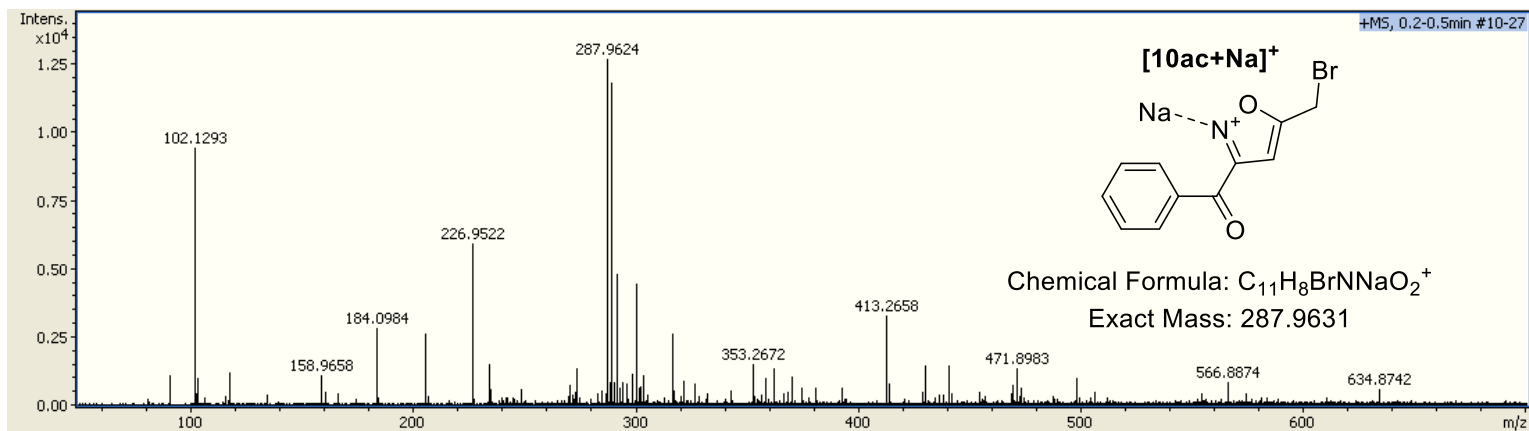


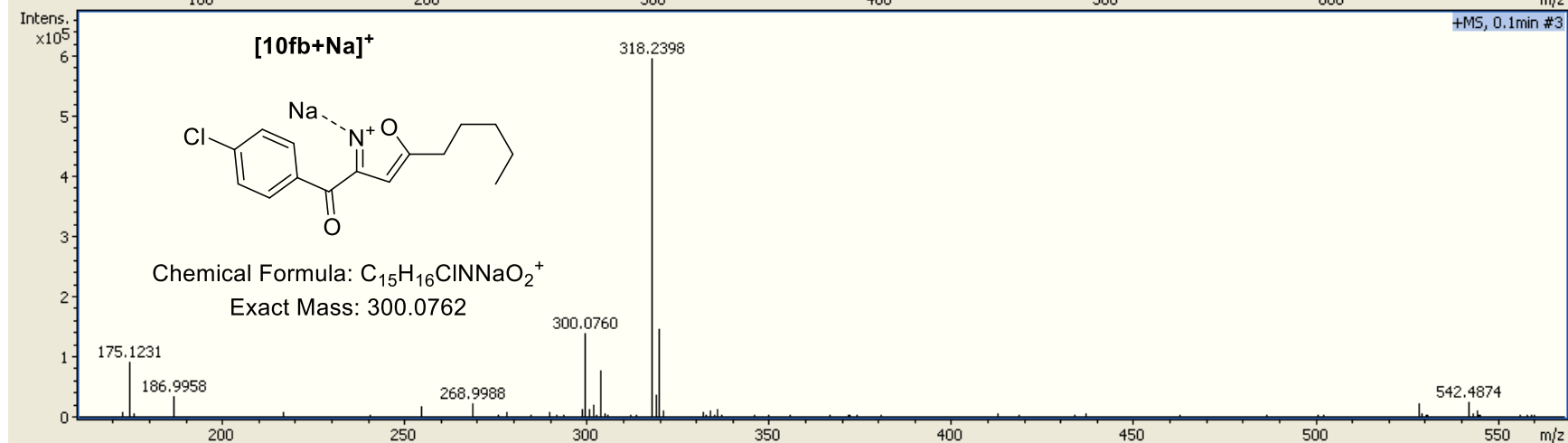
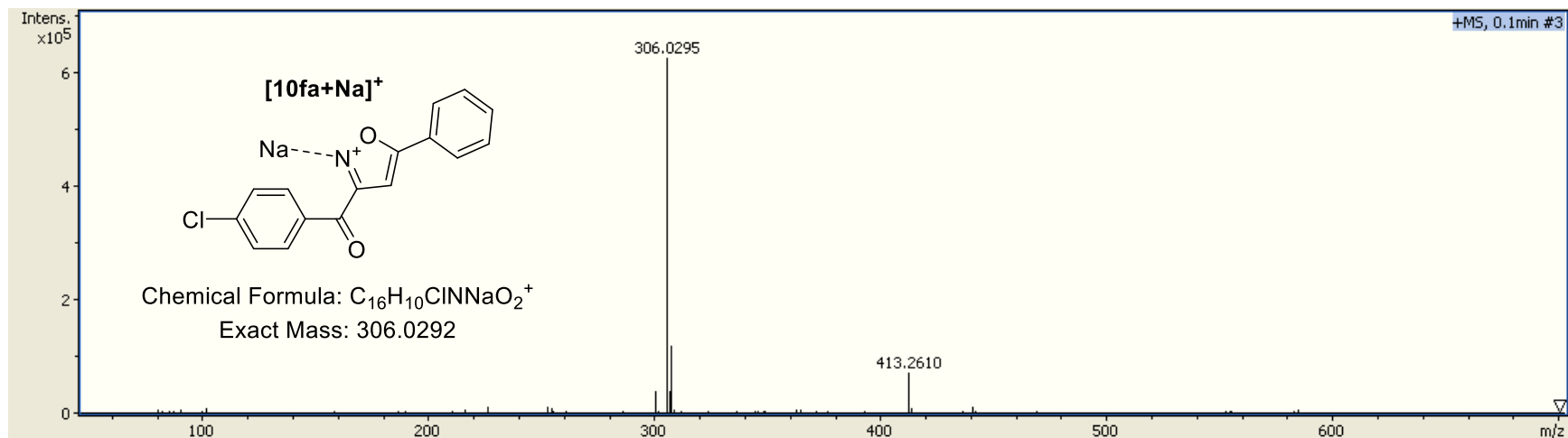


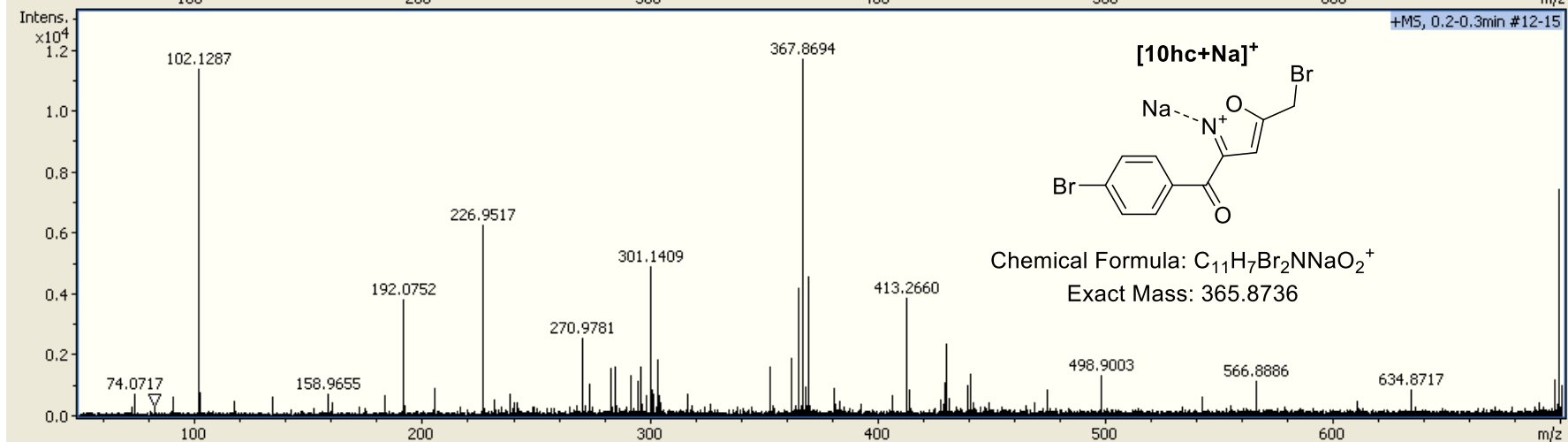
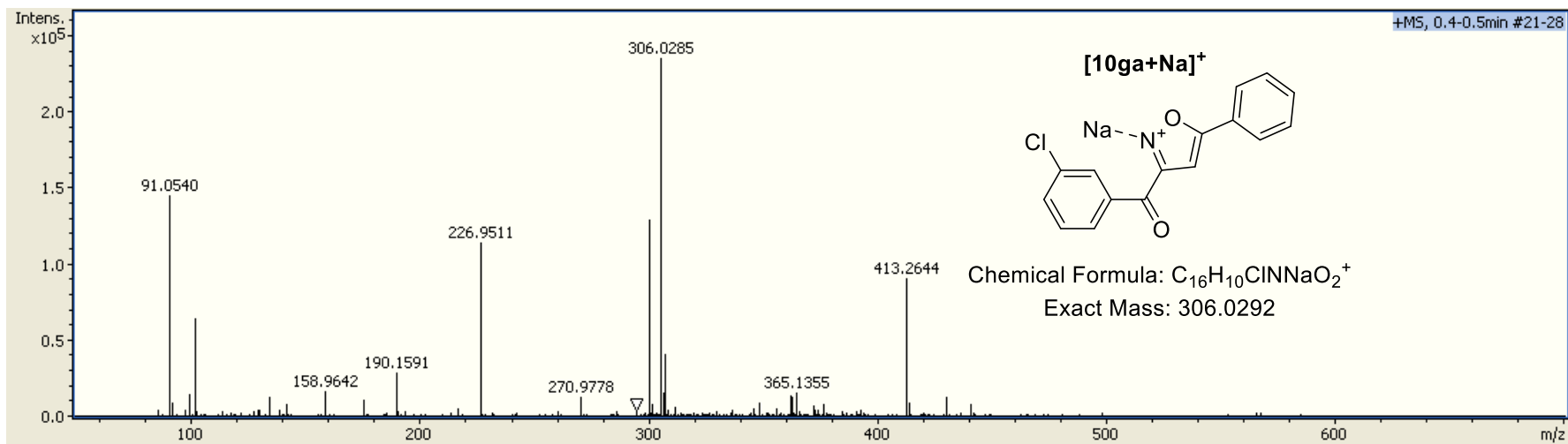


HRMS spectral charts for isoxazoles 10

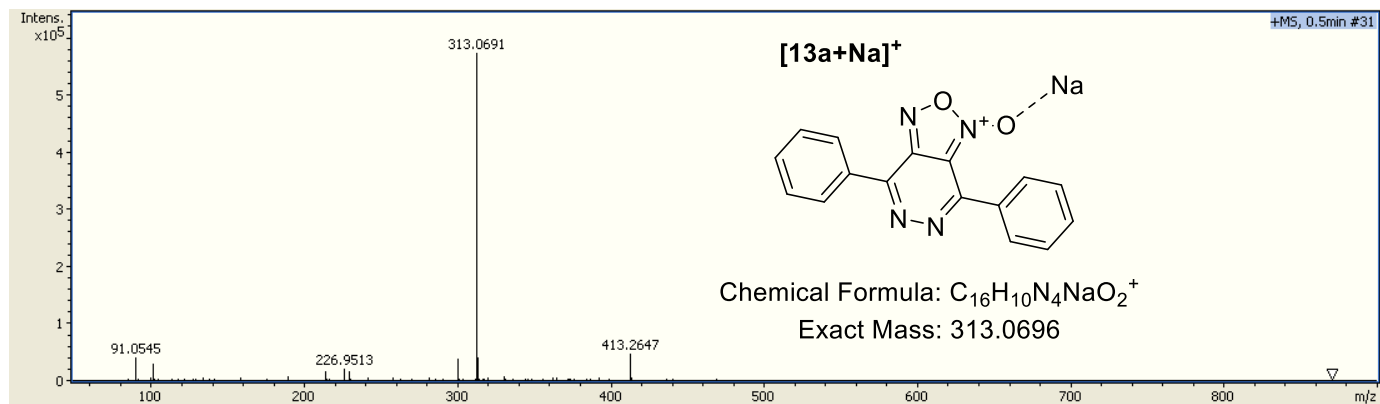




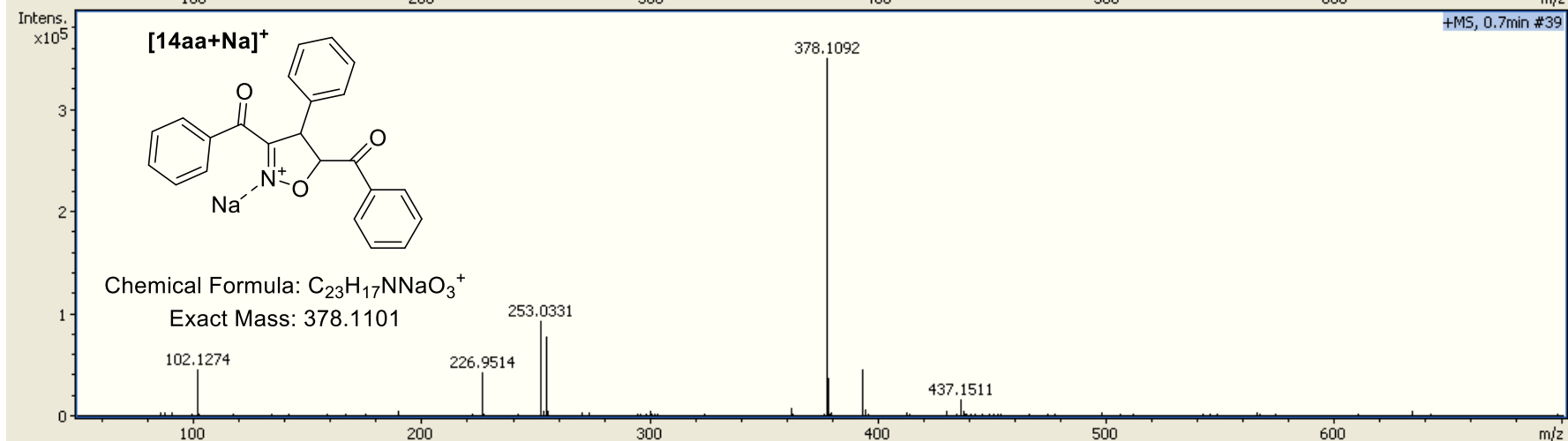
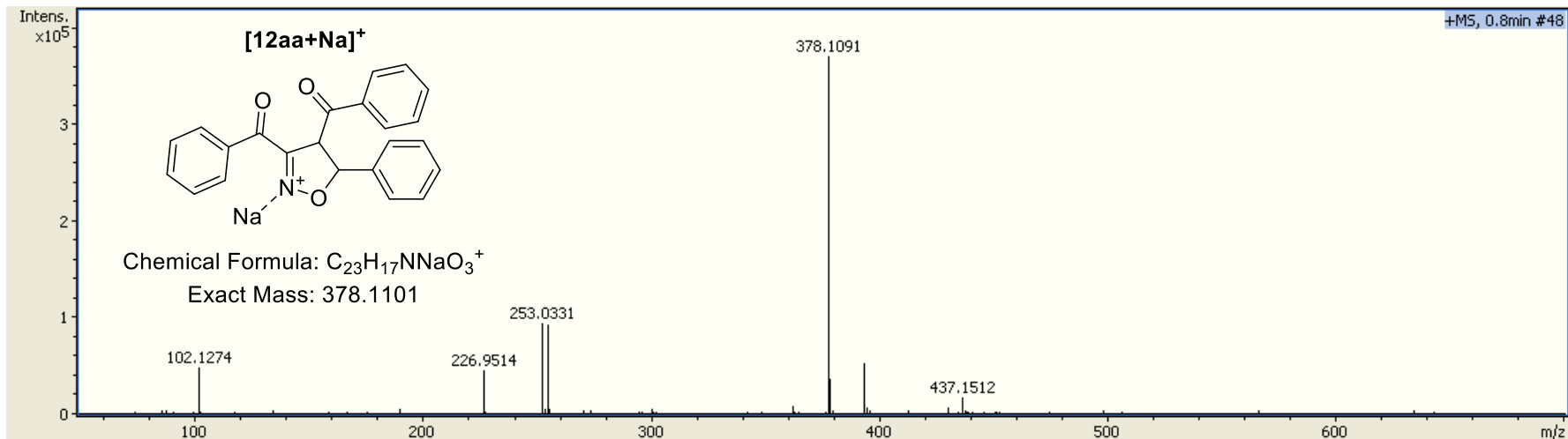


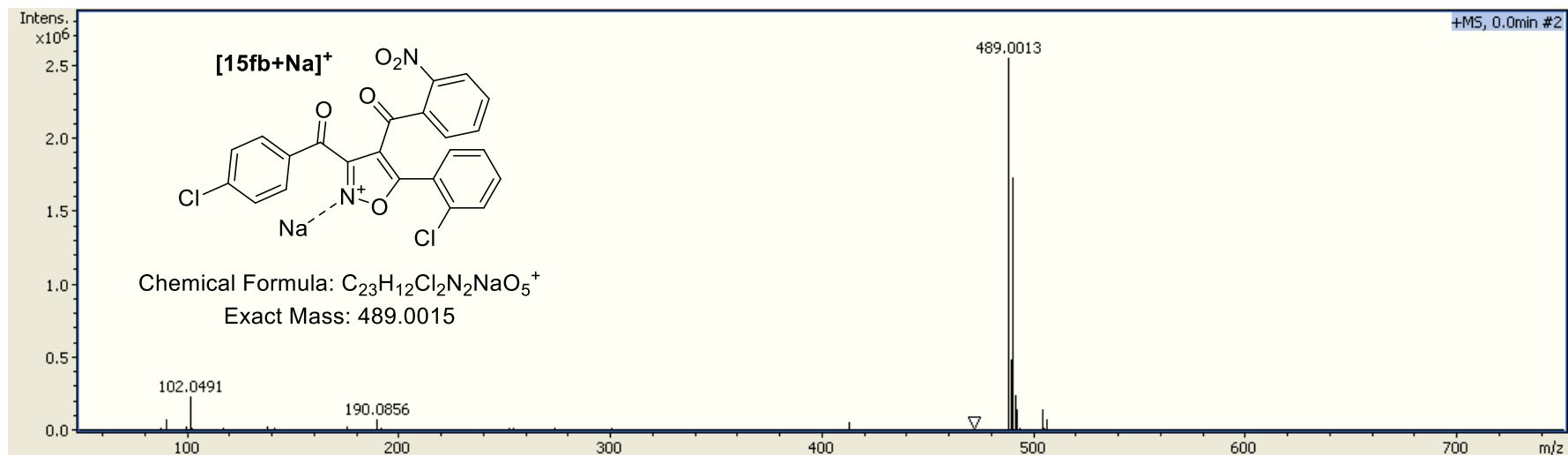


HRMS spectral charts for 4,7-diphenyl-[1,2,5]oxadiazolo[3,4-d]pyridazine 1-oxide (13a):

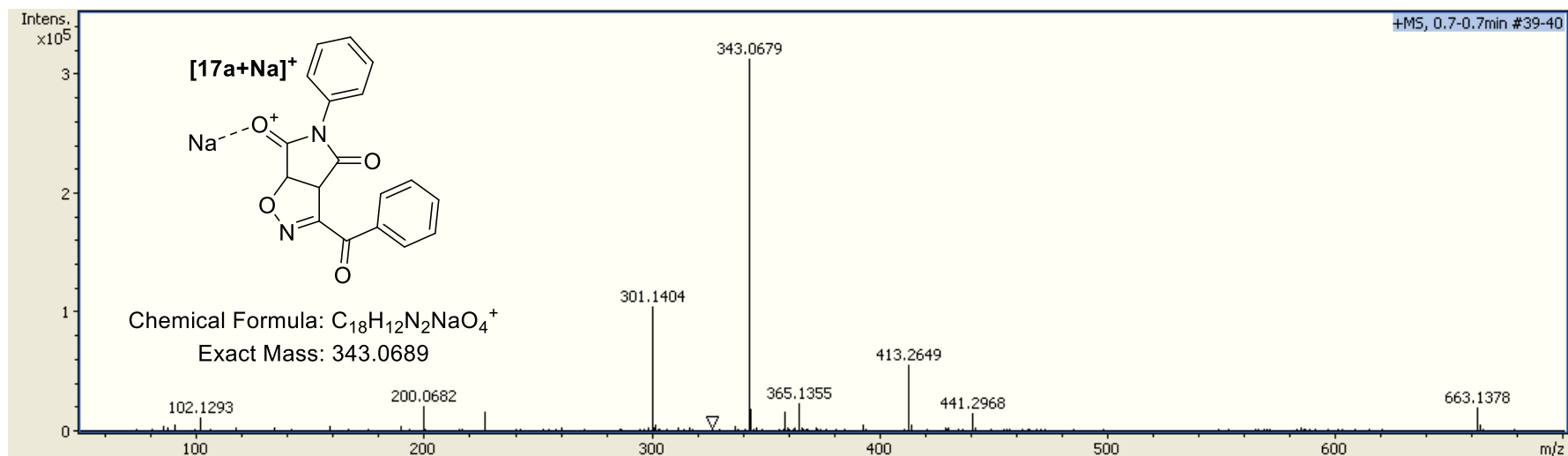


HRMS spectral charts for cycloadducts with chalcones 12aa, 14aa, 15fb:





HRMS spectral charts for 3-benzoyl-5-phenyl-3a,6a-dihydro-4H-pyrrolo[3,4-d]isoxazole-4,6(5H)-dione (17a):



X-Ray crystallography data

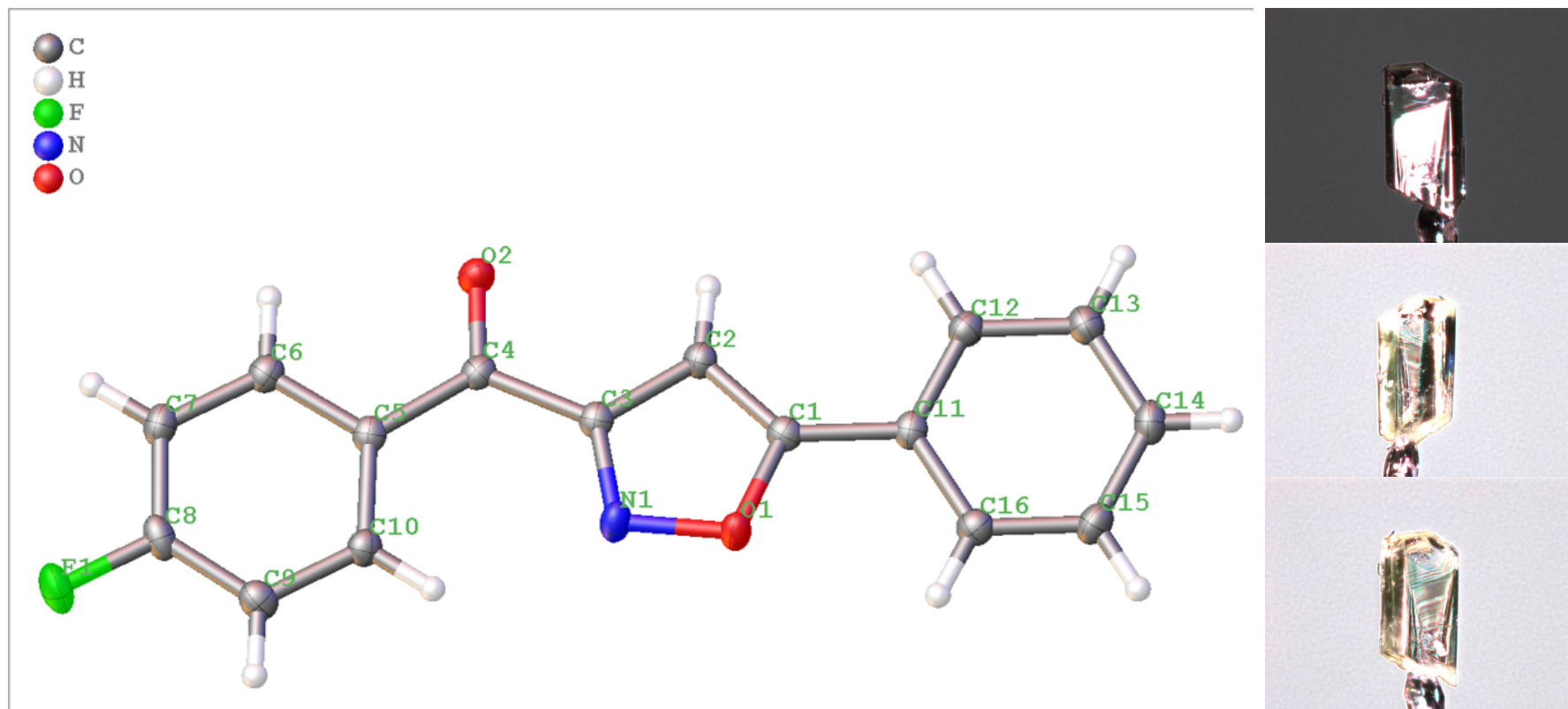


Figure S1. ORTEP drawing of the crystal structure (left) and microphotography of the single crystal of compound **10da** used for X-Ray diffraction analysis (right)

Table S1 Crystal data and structure refinement for 10da.

Identification code	ANNA_KUR182_3
Empirical formula	C ₁₆ H ₁₀ FNO ₂
Formula weight	267.25
Temperature/K	100.00(10)
Crystal system	monoclinic
Space group	P2 ₁ /n
a/Å	5.77617(12)
b/Å	28.7775(6)
c/Å	7.41435(13)
α/°	90
β/°	95.1230(19)
γ/°	90
Volume/Å ³	1227.52(4)
Z	4
ρ _{calc} /cm ³	1.446
μ/mm ⁻¹	0.888
F(000)	552.0
Crystal size/mm ³	0.576 × 0.289 × 0.124
Radiation	Cu Kα (λ = 1.54184)
2θ range for data collection/°	12.302 to 152.478
Index ranges	-7 ≤ h ≤ 7, -35 ≤ k ≤ 36, -6 ≤ l ≤ 9
Reflections collected	12715
Independent reflections	2566 [R _{int} = 0.0452, R _{sigma} = 0.0238]
Data/restraints/parameters	2566/0/169
Goodness-of-fit on F ²	1.072
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0513, wR ₂ = 0.1360
Final R indexes [all data]	R ₁ = 0.0533, wR ₂ = 0.1374
Largest diff. peak/hole / e Å ⁻³	0.38/-0.31

Table S2 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 10da. U_{eq} is defined as 1/3 of of the trace of the orthogonalised U_{IJ} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U(eq)$
F1	14576.8(19)	3189.4(4)	5379.5(14)	28.6(3)
O2	7051(2)	4471.0(4)	1342.2(16)	23.0(3)
O1	11058(2)	5764.8(4)	2646.6(16)	20.9(3)
N1	11294(3)	5288.9(5)	2987.6(19)	21.6(3)
C4	8893(3)	4587.8(5)	2164(2)	16.6(3)
C8	13225(3)	3535.0(6)	4622(2)	20.4(3)
C2	7877(3)	5435.5(5)	1330(2)	17.2(3)
C3	9396(3)	5099.7(5)	2194.9(19)	16.3(3)
C11	8482.1(17)	6329.0(2)	1154.6(13)	16.7(3)
C16	10154.3(14)	6672.9(3)	1492.5(13)	19.6(3)
C15	9661.5(16)	7128.7(3)	969.8(14)	21.5(3)
C14	7496.6(18)	7240.6(2)	109.1(14)	21.4(4)
C13	5824.4(14)	6896.6(3)	-228.9(14)	21.6(3)
C12	6317.1(15)	6440.8(3)	293.9(14)	19.3(3)
C10	12707(3)	4333.1(5)	3866(2)	18.1(3)
C1	8998(3)	5844.0(5)	1653.6(19)	16.3(3)
C9	14093(3)	3979.6(6)	4662(2)	20.7(3)
C7	11029(3)	3420.7(5)	3839(2)	20.3(3)
C5	10482(3)	4234.2(5)	3061.8(19)	16.3(3)
C6	9665(3)	3774.4(5)	3054(2)	18.8(3)

Table S3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 10da. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
F1	34.3(6)	21.2(5)	28.7(5)	1.4(4)	-5.6(4)	9.6(4)
O2	21.2(6)	19.4(6)	27.3(6)	1.4(4)	-4.4(5)	-2.7(4)
O1	22.7(6)	14.2(5)	24.0(6)	1.2(4)	-7.1(5)	-1.2(4)
N1	26.2(7)	13.7(6)	23.8(7)	0.7(5)	-4.2(5)	-1.4(5)
C4	18.7(7)	17.7(7)	13.7(7)	-0.5(5)	2.8(5)	-1.5(6)
C8	28.1(9)	19.0(8)	14.0(7)	0.2(6)	0.8(6)	8.0(6)
C2	19.7(7)	17.2(7)	14.7(7)	0.3(5)	1.2(5)	-0.1(6)
C3	19.7(7)	17.8(7)	11.4(7)	-0.8(5)	2.1(5)	0.0(6)
C11	21.9(8)	16.2(7)	12.3(7)	-0.4(5)	3.0(6)	0.1(6)
C16	20.5(8)	21.1(8)	16.8(7)	0.1(6)	0.3(6)	-1.1(6)
C15	26.8(8)	17.6(7)	20.2(8)	-1.3(6)	1.9(6)	-4.6(6)
C14	29.3(9)	15.6(7)	19.7(8)	0.8(6)	3.5(6)	2.2(6)
C13	21.9(8)	20.8(8)	22.1(8)	0.9(6)	1.6(6)	2.6(6)
C12	20.5(8)	17.8(7)	19.7(7)	-1.3(6)	2.3(6)	-1.7(6)
C10	20.9(8)	17.1(7)	16.4(7)	-2.0(5)	1.1(6)	-0.5(6)
C1	18.1(7)	19.2(8)	11.5(6)	-0.5(5)	1.1(5)	0.4(6)
C9	21.7(8)	22.8(8)	17.3(7)	-2.7(6)	-0.8(6)	2.5(6)
C7	27.5(8)	16.0(7)	17.8(7)	-0.2(6)	3.7(6)	-0.7(6)
C5	20.7(8)	17.1(7)	11.3(6)	-0.8(5)	2.6(5)	0.3(6)
C6	21.9(8)	18.5(8)	16.3(7)	-1.4(6)	2.2(6)	-1.4(6)

Table S4 Bond Lengths for 10da.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
F1	C8	1.3548(18)	C11	C16	1.3900
O2	C4	1.225(2)	C11	C12	1.3900
O1	N1	1.3970(17)	C11	C1	1.4674(17)
O1	C1	1.3615(19)	C16	C15	1.3900
N1	C3	1.315(2)	C15	C14	1.3900
C4	C3	1.501(2)	C14	C13	1.3900
C4	C5	1.487(2)	C13	C12	1.3900
C8	C9	1.374(2)	C10	C9	1.393(2)
C8	C7	1.387(2)	C10	C5	1.397(2)
C2	C3	1.419(2)	C7	C6	1.383(2)
C2	C1	1.353(2)	C5	C6	1.405(2)

Table S5 Bond Angles for 10da.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C1	O1	N1	109.17(12)	C11	C16	C15	120.0
C3	N1	O1	105.15(12)	C14	C15	C16	120.0
O2	C4	C3	115.81(14)	C15	C14	C13	120.0
O2	C4	C5	120.59(14)	C14	C13	C12	120.0
C5	C4	C3	123.60(14)	C13	C12	C11	120.0
F1	C8	C9	118.64(15)	C9	C10	C5	120.29(15)
F1	C8	C7	118.11(15)	O1	C1	C11	116.33(12)
C9	C8	C7	123.25(15)	C2	C1	O1	109.37(13)
C1	C2	C3	104.16(14)	C2	C1	C11	134.30(14)
N1	C3	C4	124.47(14)	C8	C9	C10	118.45(16)
N1	C3	C2	112.15(14)	C6	C7	C8	117.85(15)
C2	C3	C4	123.37(14)	C10	C5	C4	124.01(14)
C16	C11	C12	120.0	C10	C5	C6	119.33(14)
C16	C11	C1	120.64(9)	C6	C5	C4	116.65(14)
C12	C11	C1	119.34(9)	C7	C6	C5	120.84(15)

Table S6 Torsion Angles for 10da.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
F1	C8	C9	C10	179.37(13)	C16C15	C14	C13		0.0
F1	C8	C7	C6	-179.43(13)	C15C14	C13	C12		0.0
O2	C4	C3	N1	-179.10(15)	C14C13	C12	C11		0.0
O2	C4	C3	C2	-0.4(2)	C12C11	C16	C15		0.0
O2	C4	C5	C10	173.32(14)	C12C11	C1	O1		-173.83(9)
O2	C4	C5	C6	-5.6(2)	C12C11	C1	C2		7.2(2)
O1	N1	C3	C4	178.64(13)	C10C5	C6	C7		0.4(2)
O1	N1	C3	C2	-0.19(17)	C1	O1	N1	C3	0.17(16)
N1	O1	C1	C2	-0.09(17)	C1	C2	C3	N1	0.14(18)
N1	O1	C1	C11	-179.31(11)	C1	C2	C3	C4	-178.71(13)
C4	C5	C6	C7	179.33(13)	C1	C11	C16	C15	178.32(11)
C8	C7	C6	C5	-0.3(2)	C1	C11	C12	C13	-178.34(10)
C3	C4	C5	C10	-6.9(2)	C9	C8	C7	C6	0.4(2)
C3	C4	C5	C6	174.22(13)	C9	C10	C5	C4	-179.30(14)
C3	C2	C1	O1	-0.02(17)	C9	C10	C5	C6	-0.4(2)
C3	C2	C1	C11	179.00(15)	C7	C8	C9	C10	-0.4(2)
C11	C16	C15	C14	0.0	C5	C4	C3	N1	1.1(2)
C16	C11	C12	C13	0.0	C5	C4	C3	C2	179.77(14)
C16	C11	C1	O1	7.84(16)	C5	C10	C9	C8	0.4(2)
C16	C11	C1	C2	-171.14(15)					

Table S7 Hydrogen Atom Coordinates ($\text{\AA}\times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for 10da.

Atom	x	y	z	U(eq)
H2	6440.04	5386.07	688.25	21
H16	11602.73	6598.05	2068.4	23
H15	10780.28	7358.82	1195.86	26
H14	7166.9	7545.53	-240.69	26
H13	4375.94	6971.48	-804.73	26
H12	5198.35	6210.71	67.8	23
H10	13264.84	4636.27	3870.07	22
H9	15572.31	4043.14	5206.83	25
H7	10490.01	3116.2	3841.4	24
H6	8187.86	3706.87	2513.62	23

Experimental

Single crystals of $\text{C}_{16}\text{H}_{10}\text{FNO}_2$ **10da** were prepared by cycloaddition reaction in PPA. A suitable crystal was selected and mounted on the glass stick by acrylic glue on a SuperNova, Dual, Cu at home/near, AtlasS2 diffractometer. The crystal was kept at 100.00(10) K during data collection. Using Olex2¹, the structure was solved with the SHELXD² structure solution program using Dual Space and refined with the SHELXL³ refinement package using Least Squares minimisation.

Crystal structure determination of [10da]

Crystal Data for $\text{C}_{16}\text{H}_{10}\text{FNO}_2$ ($M=267.25$ g/mol): monoclinic, space group $\text{P}2_1/\text{n}$ (no. 14), $a = 5.77617(12)$ \AA , $b = 28.7775(6)$ \AA , $c = 7.41435(13)$ \AA , $\beta = 95.1230(19)^\circ$, $V = 1227.52(4)$ \AA^3 , $Z = 4$, $T = 100.00(10)$ K, $\mu(\text{Cu K}\alpha) = 0.888$ mm^{-1} , $D_{\text{calc}} = 1.446$ g/cm^3 , 12715 reflections measured ($12.302^\circ \leq 2\Theta \leq 152.478^\circ$), 2566 unique ($R_{\text{int}} = 0.0452$, $R_{\text{sigma}} = 0.0238$) which were used in all calculations. The final R_1 was 0.0513 ($I > 2\sigma(I)$) and wR_2 was 0.1374 (all data).

Refinement model description

Number of restraints - 0, number of constraints - unknown.

Details:

1. Fixed Uiso

At 1.2 times of:

All C(H) groups

2.a Aromatic/amide H refined with riding coordinates:

C2(H2), C16(H16), C15(H15), C14(H14), C13(H13), C12(H12), C10(H10), C9(H9),

C7(H7), C6(H6)

2.b Fitted hexagon refined as free rotating group:

C11(C16,C15,C14,C13,C12)

This report has been created with Olex2, compiled on 2020.11.12 svn.r5f609507 for OlexSys. Please [let us know](#) if there are any errors or if you would like to have additional features.

References

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2. Sheldrick, G.M. (2008). *Acta Cryst.* A64, 112-122.
3. Sheldrick, G.M. (2015). *Acta Cryst.* C71, 3-8.