

Photoresponsive, Switchable, Pressure-Sensitive Adhesives: Influence of UV Intensity and Hydrocarbon Chain Length of Low Molecular Weight Azobenzene Compounds

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- Synthetic route for the azobenzene-containing acrylic monomer and low molecular weight

azobenzene compounds

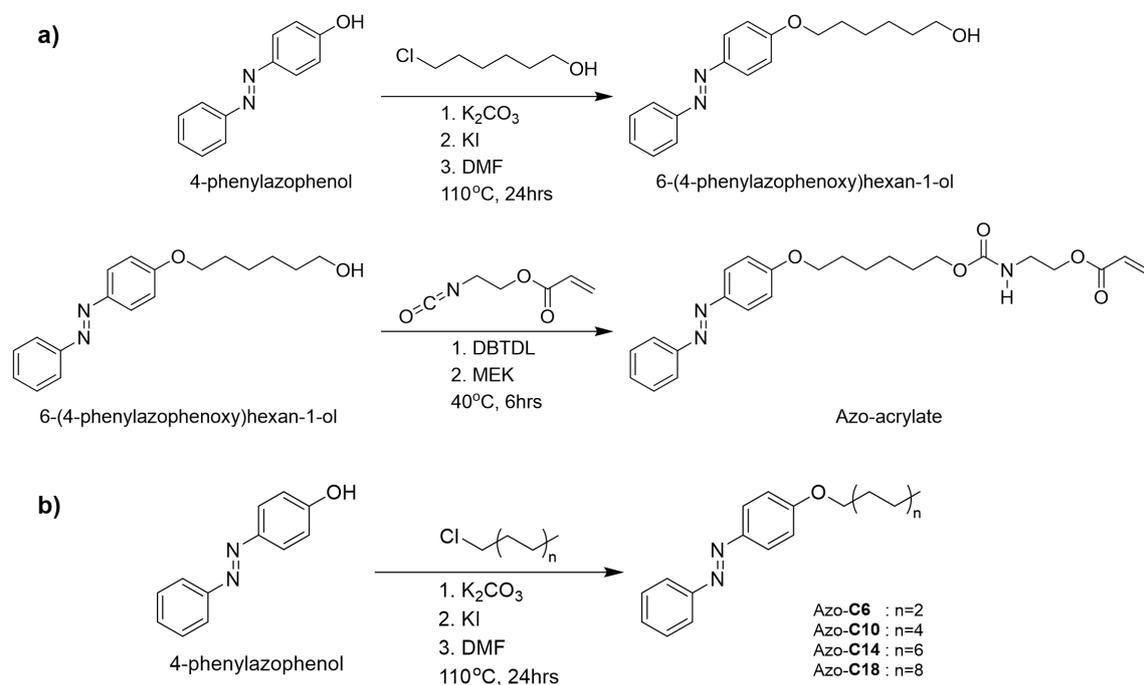


Figure S1. Synthetic route for a) the azobenzene-containing acrylic monomer (azo-acrylate) and b) the low molecular weight azobenzene compounds (azo-compounds).

- NMR spectra

6-(4-(phenyldiazenyl)phenoxy)hexan-1-ol

^1H NMR (400 MHz, DMSO- d_6 , δ): 7.83 (m, 4H, Ar H), 7.51 (m, 3H, Ar H), 7.08 (m, 2H, Ar H), 4.03 (m, 2H, OCH $_2$), 3.37 (m, 2H, CH $_2$ OH), 1.71 (m, 2H, OCH $_2$ CH $_2$), 1.38 (m, 6H, OCH $_2$ CH $_2$ CH $_2$ CH $_2$ CH $_2$)

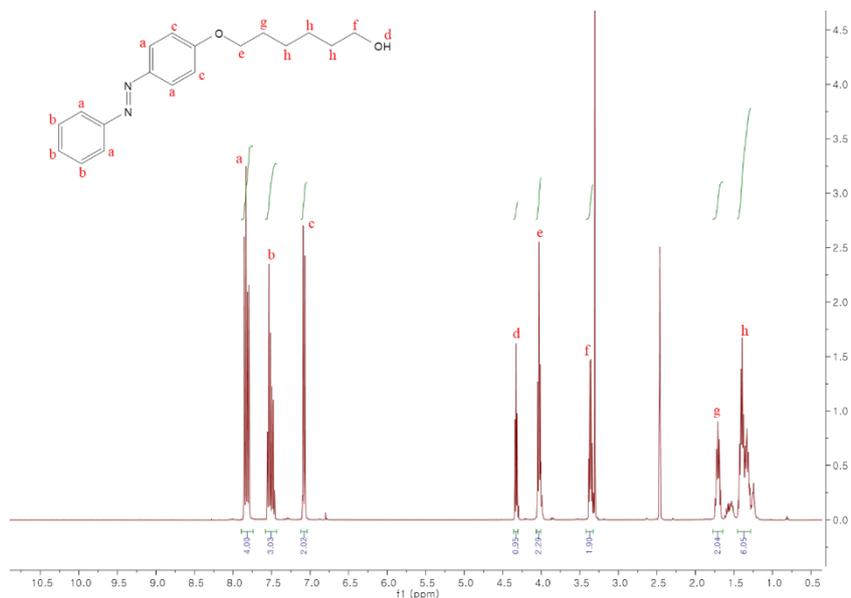


Figure S2. ^1H -NMR spectrum of 6-(4-(phenyldiazenyl)phenoxy)hexan-1-ol

2-(((6-(4-(phenyldiazenyl)phenoxy)hexyl)oxy)carbonyl)amino)ethyl acrylate (azo-acrylate)

^1H NMR (400 MHz, DMSO- d_6 , δ): 7.83 (m, 4H, Ar H), 7.51 (m, 3H, Ar H), 7.23 (m, 1H, NH), 7.08 (m, 2H, Ar H), 6.31 (m, 1H, C=CH), 6.12 (m, 1H, C=CH), 5.91 (m, 1H, C=CH), 4.04 (m, 4H, Ar-OCH $_2$, NHCH $_2$ CH $_2$), 3.90 (t, 2H, CH $_2$ OOCNH), 3.21 (m, 2H, NHCH $_2$), 1.71 (m, 2H, Ar-OCH $_2$ CH $_2$), 1.45 (m, 6H, Ar-OCH $_2$ CH $_2$ CH $_2$ CH $_2$ CH $_2$)

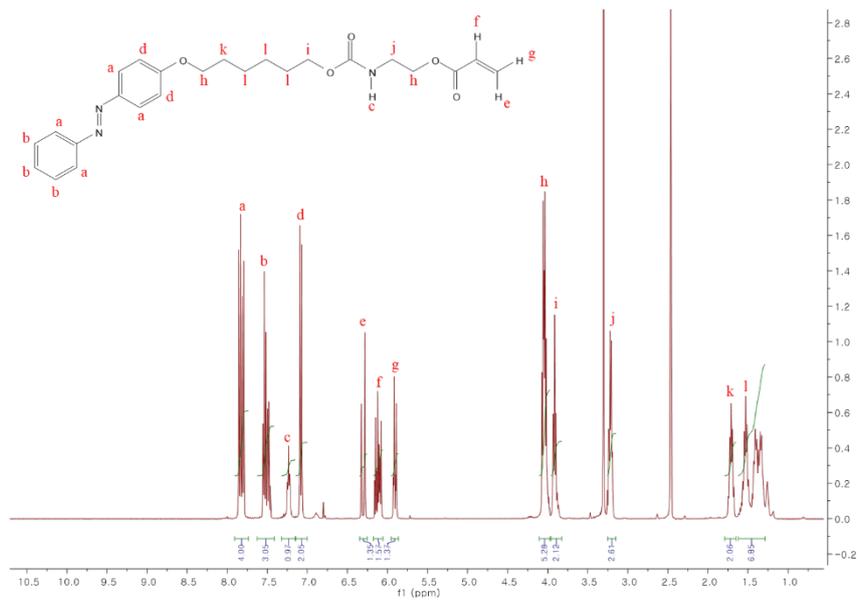


Figure S3. $^1\text{H-NMR}$ spectrum of azo-acrylate

1-(4-(hexyloxy)phenyl)-2-phenyldiazene (Azo-C6)

$^1\text{H NMR}$ (400 MHz, DMSO-d_6 , δ): 7.82 (m, 4H, Ar H), 7.52 (m, 3H, Ar H), 7.08 (m, 2H, Ar H), 4.02 (t, 2H, OCH_2), 1.70 (m, 2H, OCH_2CH_2), 1.40 (m, 2H, $\text{OCH}_2\text{CH}_2\text{CH}_2$), 1.26 (m, 4H, $\text{OCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2$), 0.84 (m, 3H, $\text{OCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$)

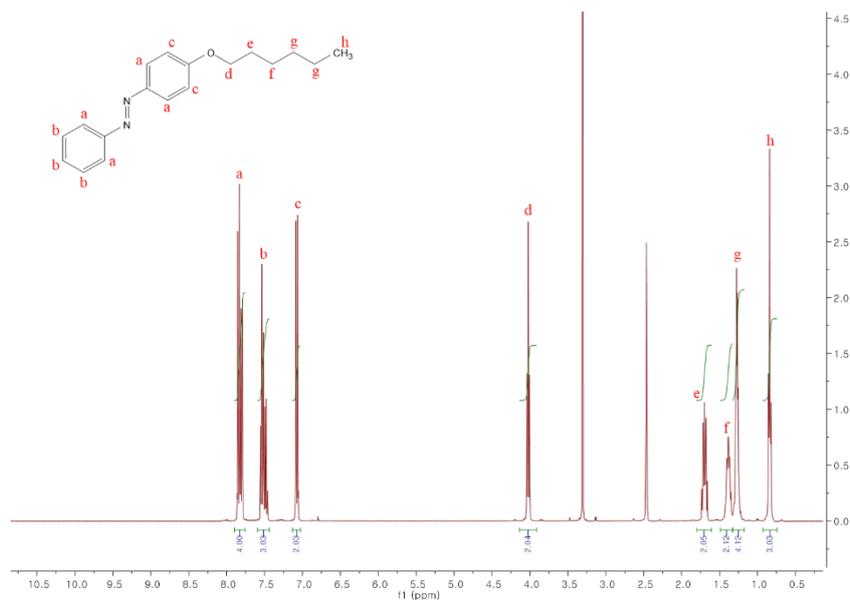


Figure S4. $^1\text{H-NMR}$ spectrum of Azo-C6

1-(4-(decyloxy)phenyl)-2-phenyldiazene (Azo-C10)

^1H NMR (400 MHz, DMSO- d_6 , δ): 7.89 (m, 4H, Ar H), 7.46 (m, 3H, Ar H), 7.00 (m, 2H, Ar H), 4.03 (t, 2H, OCH₂), 1.81 (m, 2H, OCH₂CH₂), 1.47 (m, 2H, OCH₂CH₂CH₂), 1.30 (m, 12H, OCH₂CH₂CH₂(CH₂)₆), 0.88 (m, 3H, OCH₂CH₂CH₂(CH₂)₆CH₃)

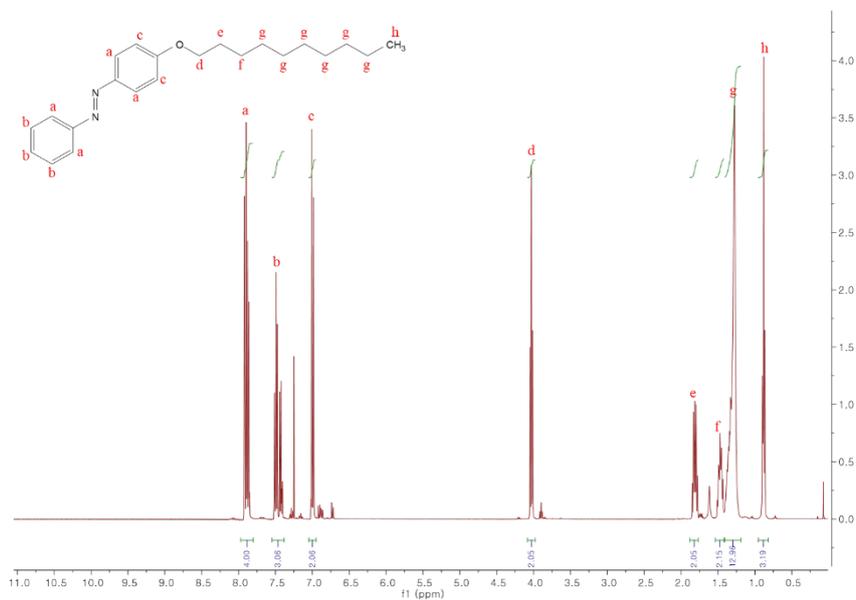


Figure S5. ^1H -NMR spectrum of Azo-C10

1-(4-(tetradecyloxy)phenyl)-2-phenyldiazene (Azo-C14)

^1H NMR (400 MHz, Chloroform- d , δ): 7.89 (m, 4H, Ar H), 7.46 (m, 3H, Ar H), 7.00 (m, 2H, Ar H), 4.03 (t, 2H, OCH₂), 1.81 (m, 2H, OCH₂CH₂), 1.47 (m, 2H, OCH₂CH₂CH₂), 1.30 (m, 20H, OCH₂CH₂CH₂(CH₂)₁₀), 0.88 (m, 3H, OCH₂CH₂CH₂(CH₂)₁₀CH₃)

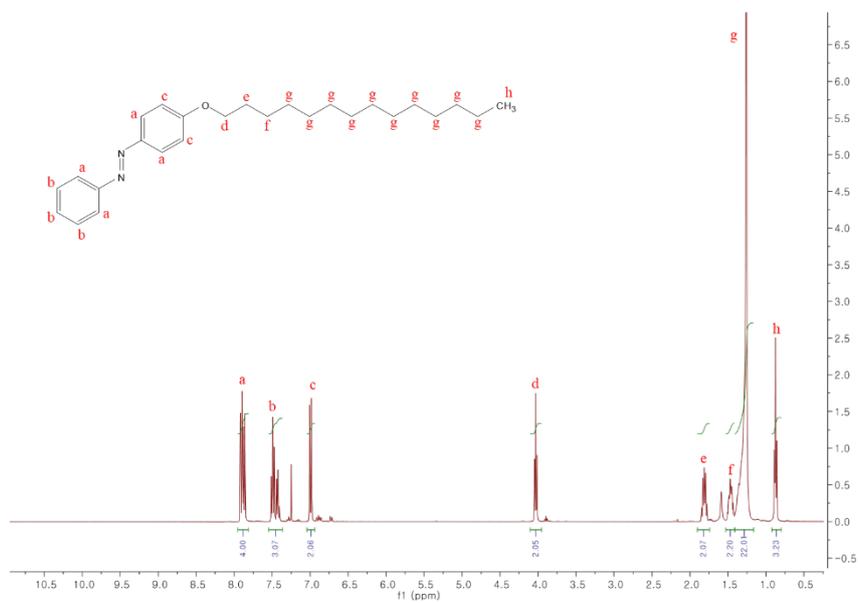


Figure S6. $^1\text{H-NMR}$ spectrum of Azo-C14

1-(4-(octadecyloxy)phenyl)-2-phenyldiazene (Azo-C18)

$^1\text{H NMR}$ (400 MHz, Chloroform- d , δ): 7.89 (m, 4H, Ar H), 7.46 (m, 3H, Ar H), 7.00 (m, 2H, Ar H), 4.03 (t, 2H, OCH_2), 1.82 (m, 2H, OCH_2CH_2), 1.47 (m, 2H, $\text{OCH}_2\text{CH}_2\text{CH}_2$), 1.30 (m, 28H, $\text{OCH}_2\text{CH}_2\text{CH}_2(\text{CH}_2)_{14}$), 0.88 (m, 3H, $\text{OCH}_2\text{CH}_2\text{CH}_2(\text{CH}_2)_{14}\text{CH}_3$)

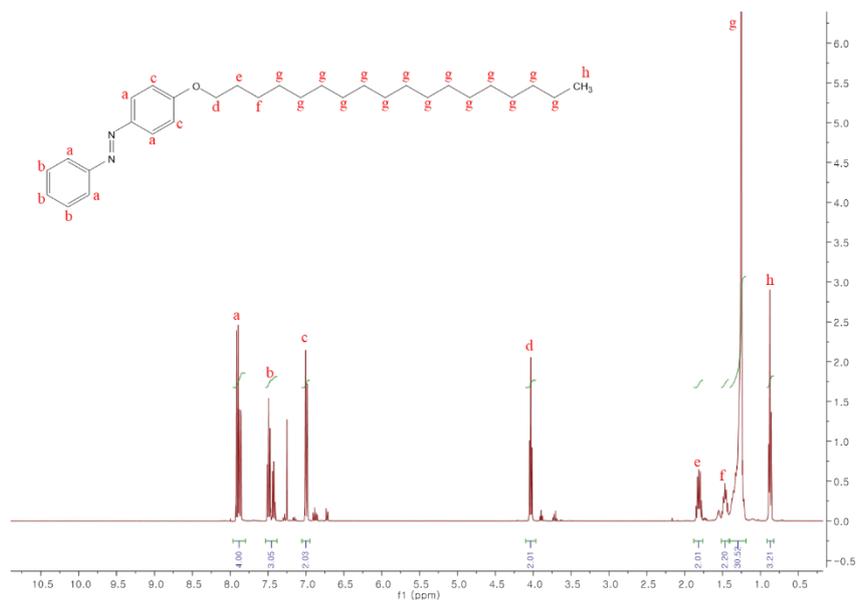


Figure S7. $^1\text{H-NMR}$ spectrum of Azo-C18

Azo-polymer (^1H NMR, 400 MHz, DMSO- d_6)

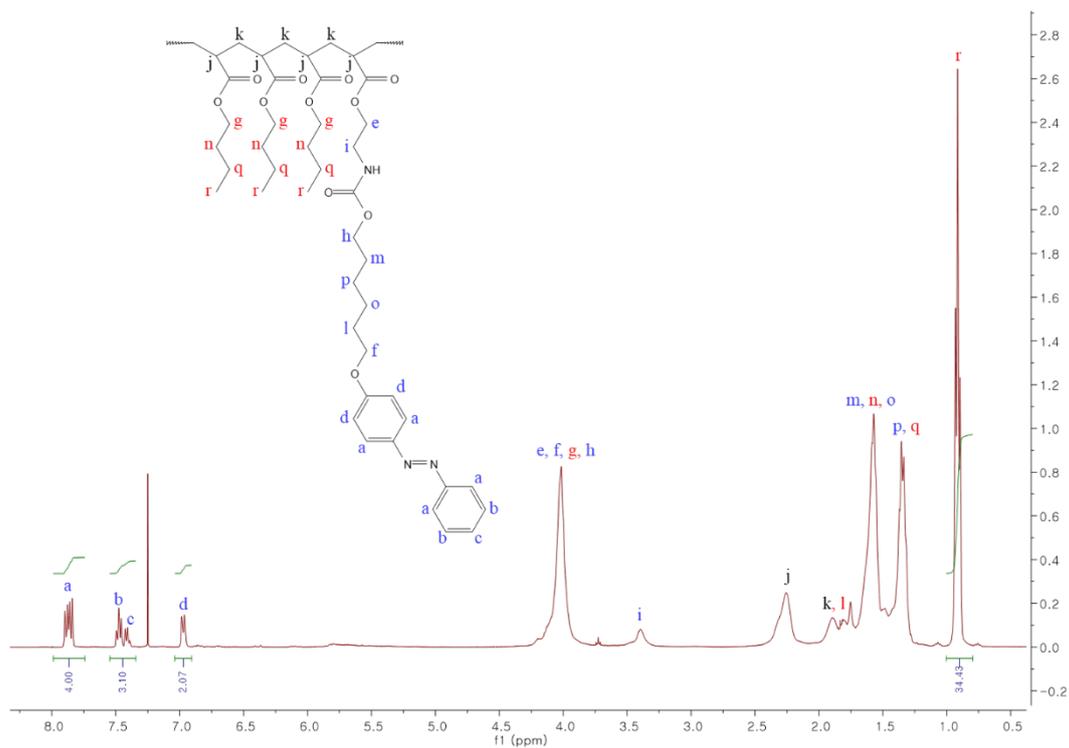


Figure S8. ^1H -NMR spectrum of azo-polymer

- Probe tack

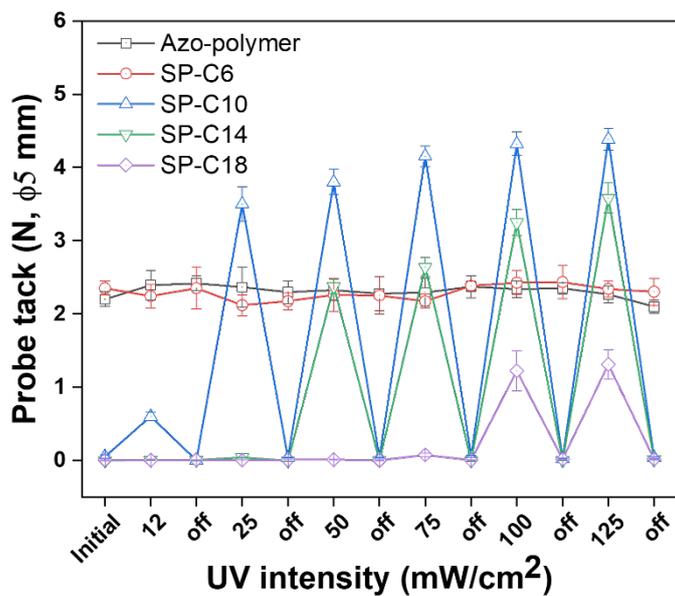


Figure S9. Probe tack forces of the switchable PSAs according to UV intensity, including switching off steps.

- Water contact angle

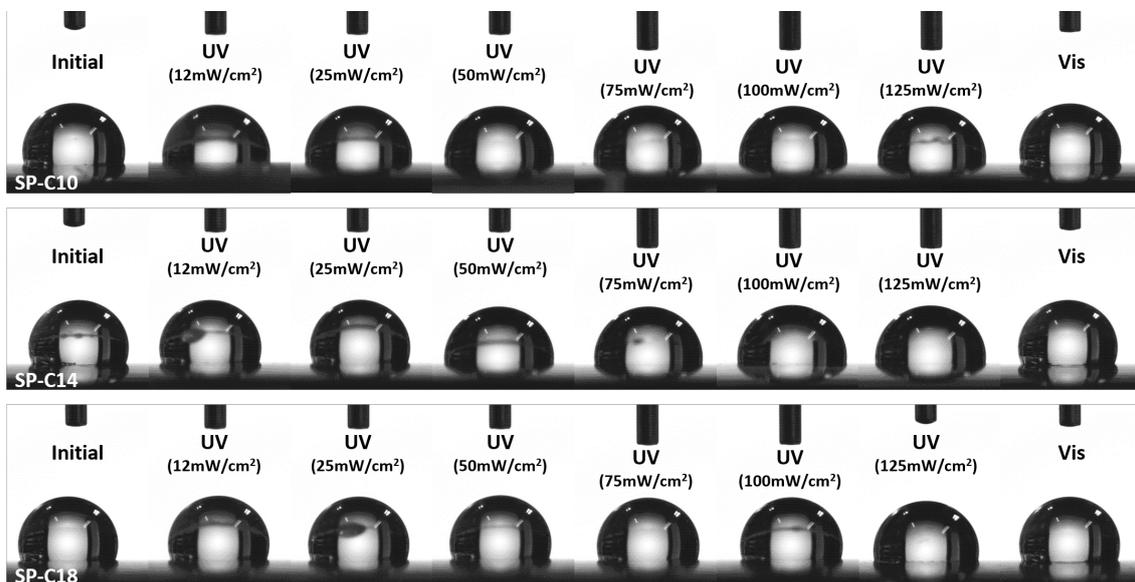


Figure S10. Images of the water droplet on the switchable PSAs according to UV intensity.