

# **Completely Green Synthesis of Rose-Shaped Au Nanostructures and Their Catalytic Applications**

Jae Hwan Jeong,<sup>a</sup> Astrini Pradyasti,<sup>a</sup> Hyeonbo Shim,<sup>a</sup> Hee-Chul Woo,<sup>b</sup> Mun Ho Kim<sup>a,\*</sup>

<sup>a</sup>Department of Polymer Engineering, Pukyong National University, 45 Yongso-ro, Nam-gu,  
Busan 48513, Republic of Korea

<sup>b</sup>Department of Chemical Engineering, Pukyong National University, 45 Yongso-ro, Nam-gu,  
Busan 48513, Republic of Korea

\* Corresponding author: M. H. Kim (munho@pknu.ac.kr)

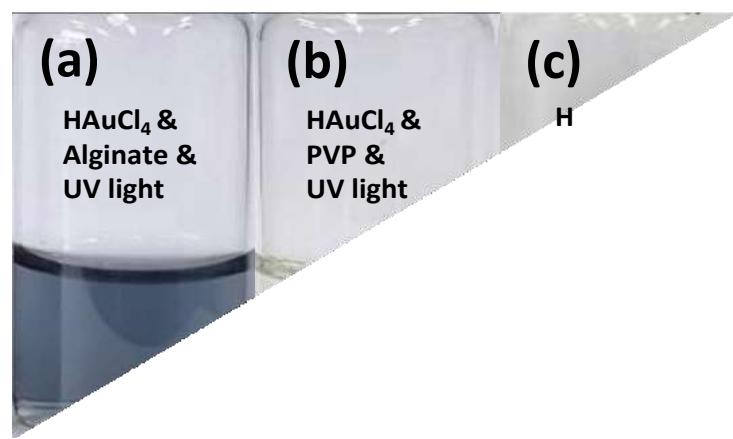
(Tel.; +82-51-629-6459, Fax; +82-51-629-6429)

## Supporting Tables

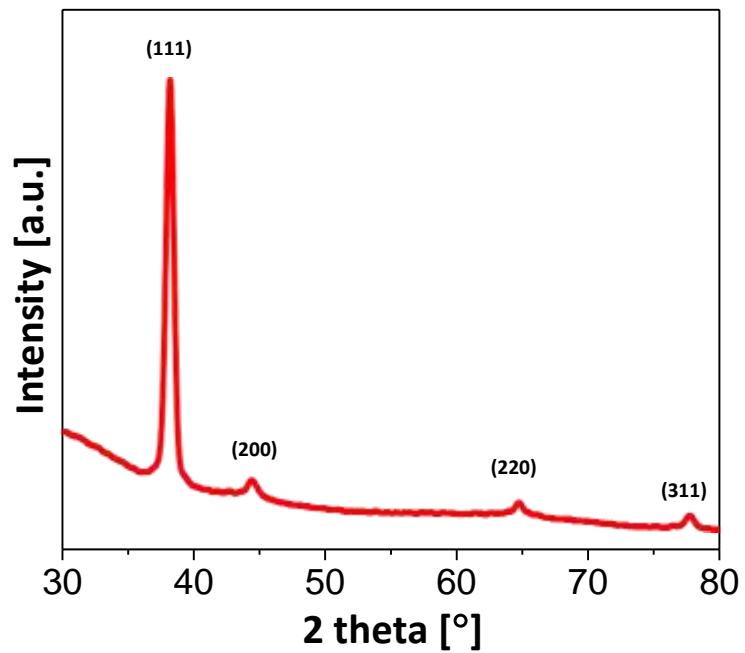
**Table S1.** Values of the activity parameter,  $\kappa$ , for various heterogeneous catalysts; the values of  $\kappa$  were obtained by dividing the reaction rate constant by the total weight of used catalyst.

Catalysts	Total weight of used catalyst [mg]	Reaction rate constant (k) [ $10^{-3} \text{ s}^{-1}$ ]	k [ $\text{s}^{-1}\text{g}^{-1}$ ]	Ref.
Smooth-edged Au nanostructures	0.28	1.7	<b>6.07</b>	<sup>1</sup>
CuNPs	1	7.11	<b>7.11</b>	<sup>2</sup>
PtAu core-shell NPs	0.5	5.92	<b>11.8</b>	<sup>3</sup>
Jagged-edged Au nanostructures	0.16	2.18	<b>13.6</b>	<sup>1</sup>
Multiply-stacked Au nanostructures	0.26	4.8	<b>18.5</b>	<sup>1</sup>
Coffee arabica seed extract stabilized Au nanoparticles	2.76	66.3	<b>24</b>	<sup>4</sup>
PtPd bimetallic nanoparticles	0.08	2.31	<b>28.9</b>	<sup>5</sup>
CuO nanosheets	0.1	4.58	<b>45.8</b>	<sup>6</sup>
Citrate stabilized Au nanoparticles	2.76	140	<b>50.8</b>	<sup>4</sup>
Avocado seed extract stabilized Au nanoparticles	0.03	1.55	<b>51.7</b>	<sup>7</sup>
Porous and Solid Au nanoparticles	0.05	4.6	<b>92</b>	<sup>8</sup>
Poly(diallyldimethylammonium chloride)-stabilized Pt nanoparticles	0.286	30	<b>105</b>	<sup>9</sup>
NiAu core-shell nanoparticles	0.06	6.4	<b>107</b>	<sup>10</sup>
Porous AuPt microparticles	0.5	55	<b>110</b>	<sup>11</sup>
PtAu alloy nanocubes	0.04	5.91	<b>147.8</b>	<sup>12</sup>
AgNi Alloy	0.2	31.1	<b>156</b>	<sup>13</sup>
Hollow porous Cu particles	0.05	9.3	<b>186</b>	<sup>14</sup>
Poly(diallyldimethylammonium chloride)-stabilized Pd nanoparticles	0.292	57	<b>195</b>	<sup>9</sup>
Porous and hollow Au nanoparticles	0.05	12.1	<b>242</b>	<sup>8</sup>
Rose-shaped Au nanostructures	0.06	17.4	<b>290</b>	This work

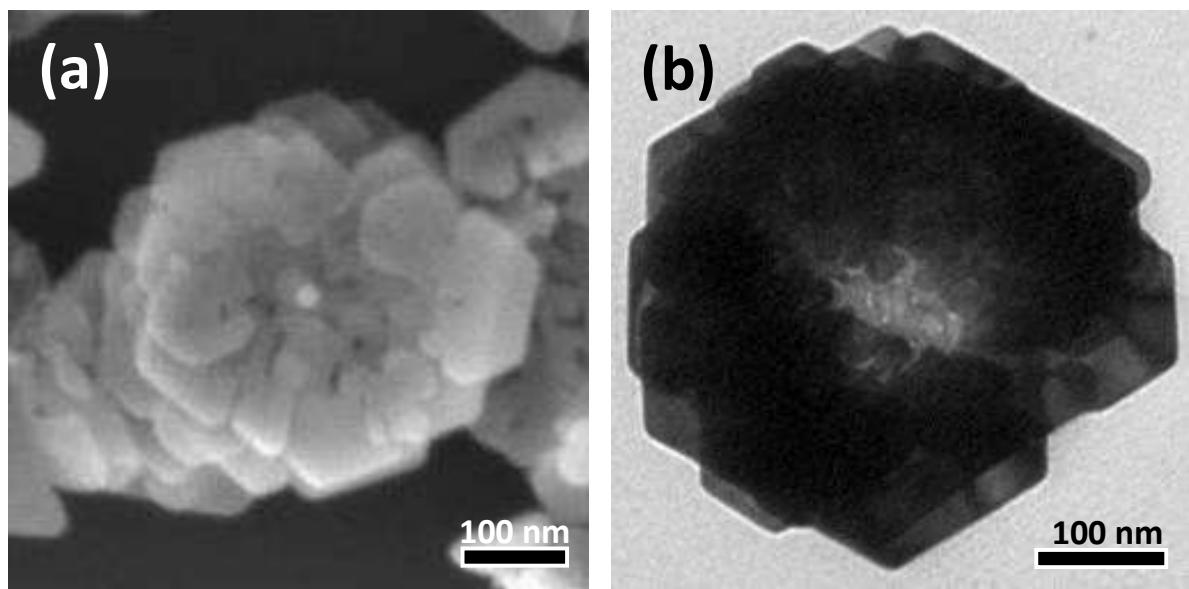
## Supporting Figures



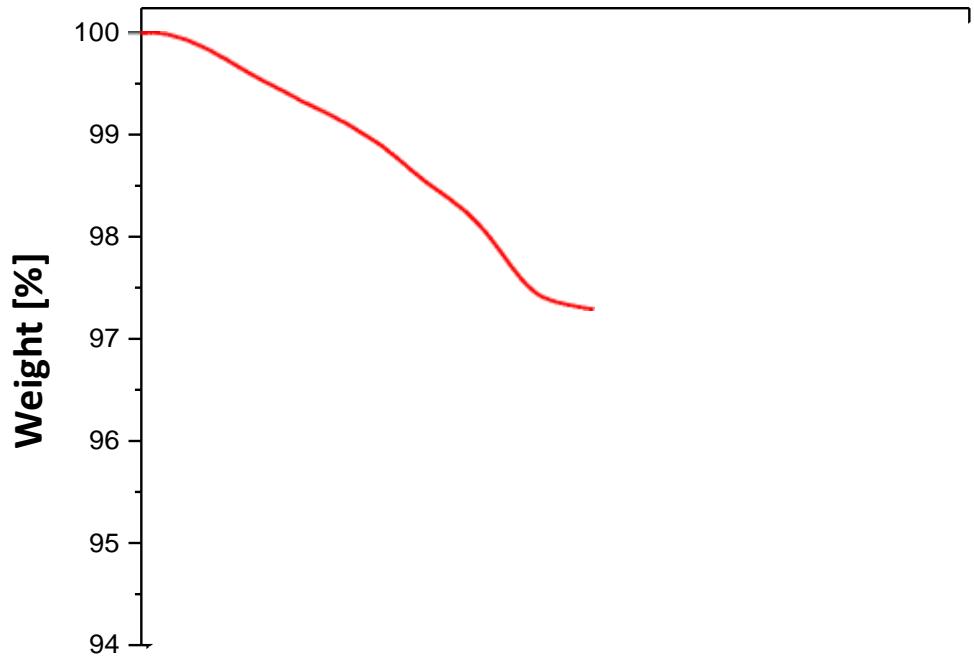
**Fig. S1.** Photographs of the reaction mixtures without (a) PVP or (b) Na-alginate and aged under UV irradiation for 3 h at room temperature. (c) Photograph of the reaction mixture including Na-alginate and PVP and aged in a dark room for 3 h at room temperature.



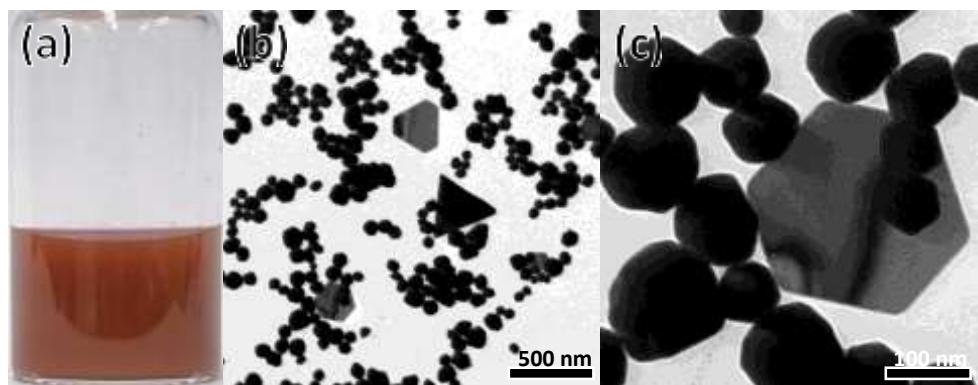
**Fig. S2.** XRD pattern of the rose-shaped Au nanostructures shown in Fig. 1.



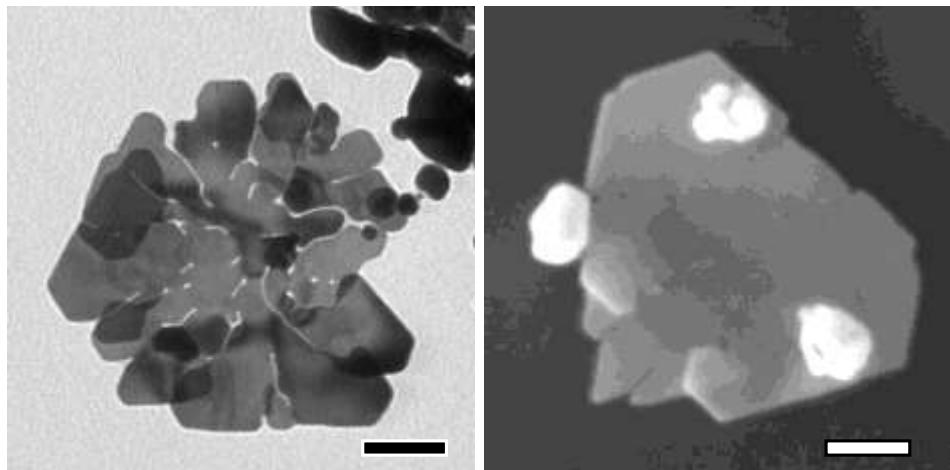
**Fig. S3.** (a) SEM and (b) TEM images of the rose-shaped Au nanostructures after aging at room temperature over 3 months.



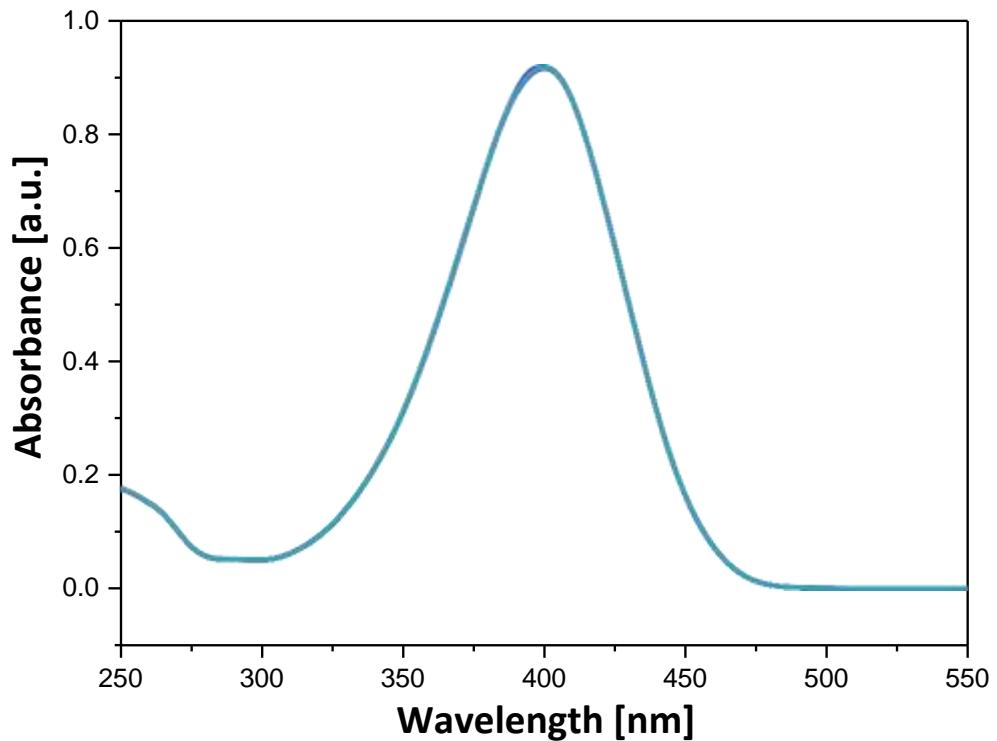
**Fig. S4.** TGA curve of the rose-shaped Au nanostructures after the complete drying of water.



**Fig. S5.** (a) Photograph of the reaction mixture containing HAuCl<sub>4</sub> and PVP but excluding Na-alginate, aged under UV irradiation for 3 h at room temperature. In the reaction, the PVP-to-HAuCl<sub>4</sub> weight ratio was approximately 18. (b)–(c) TEM images of the Au nanostructures shown in (a).



**Fig. S6.** (a) TEM and (b) SEM images of Au nanostructures grown for 1.5 h under UV irradiation at room temperature. The scale bar indicates 100 nm.



**Fig. S7.** UV–Vis absorption spectra of the solution containing 4-NP and NaBH<sub>4</sub>, as recorded after the reaction proceeded for 210 s.

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