

Supporting Information

Porous Carbon Confined Co_xS_y Nanoparticles Derived from ZIF-67 for Boosting Lithium-Ion Storage

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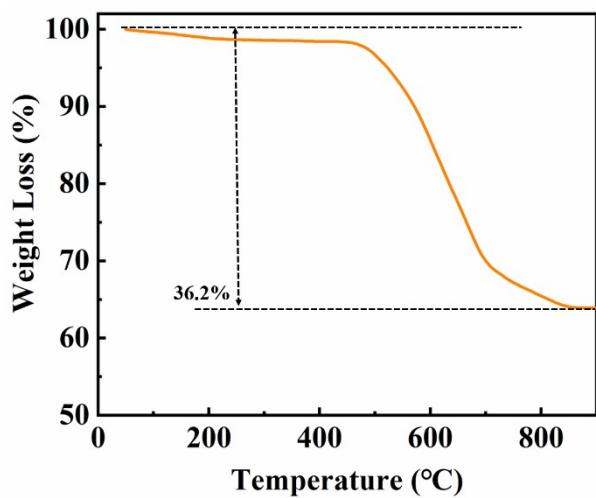


Fig. S1. TGA curve of Co_xS_y powder under nitrogen atmosphere.

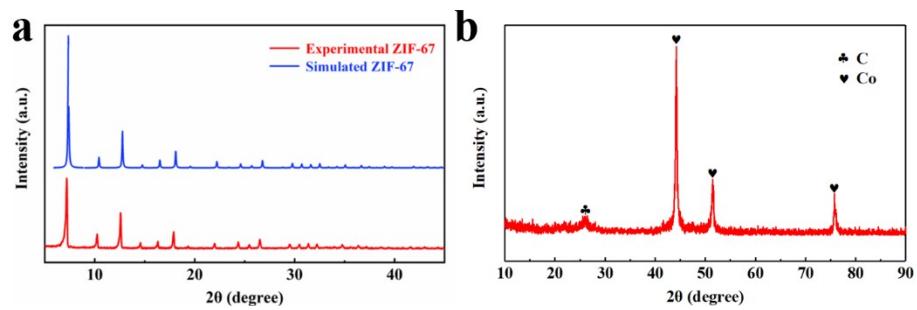


Fig. S2. XRD patterns of ZIF-67 (a) and ZIF-67(600) (b).

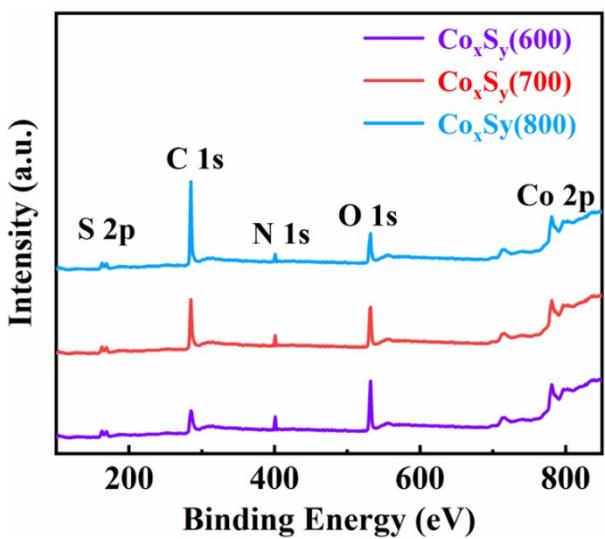


Fig. S3. XPS survey spectrum of $\text{Co}_x\text{S}_y(600)$, $\text{Co}_x\text{S}_y(700)$, and $\text{Co}_x\text{S}_y(800)$.

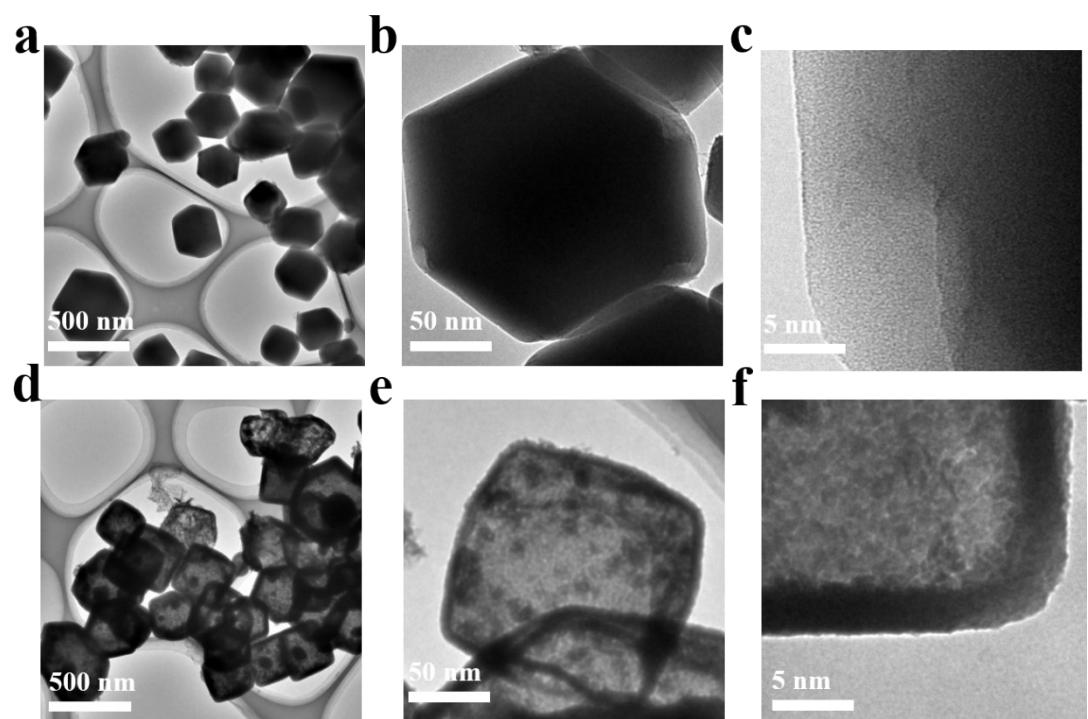


Fig. S4. TEM images of ZIF-67 (a-c) and hollow Co_xS_y (d-f).

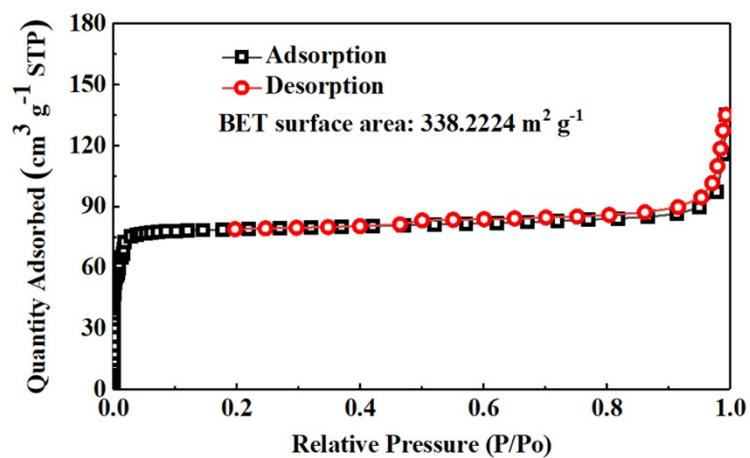


Fig. S5. N_2 adsorption-desorption isotherms of $\text{Co}_x\text{S}_y(700)$.

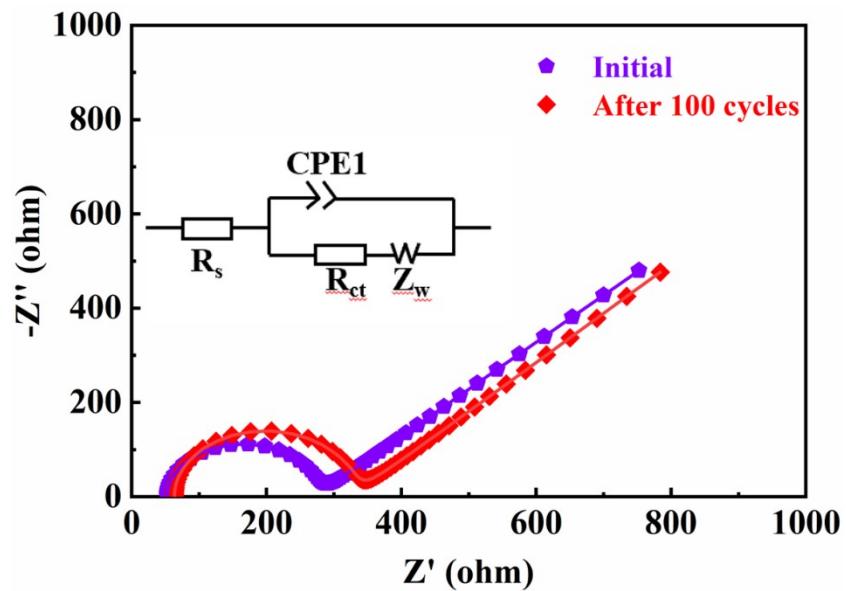


Fig. S6. Electrochemical impedance spectroscopy (EIS) of $\text{Co}_x\text{S}_y(700)$ at 1 A g^{-1} before and after 100 cycles.

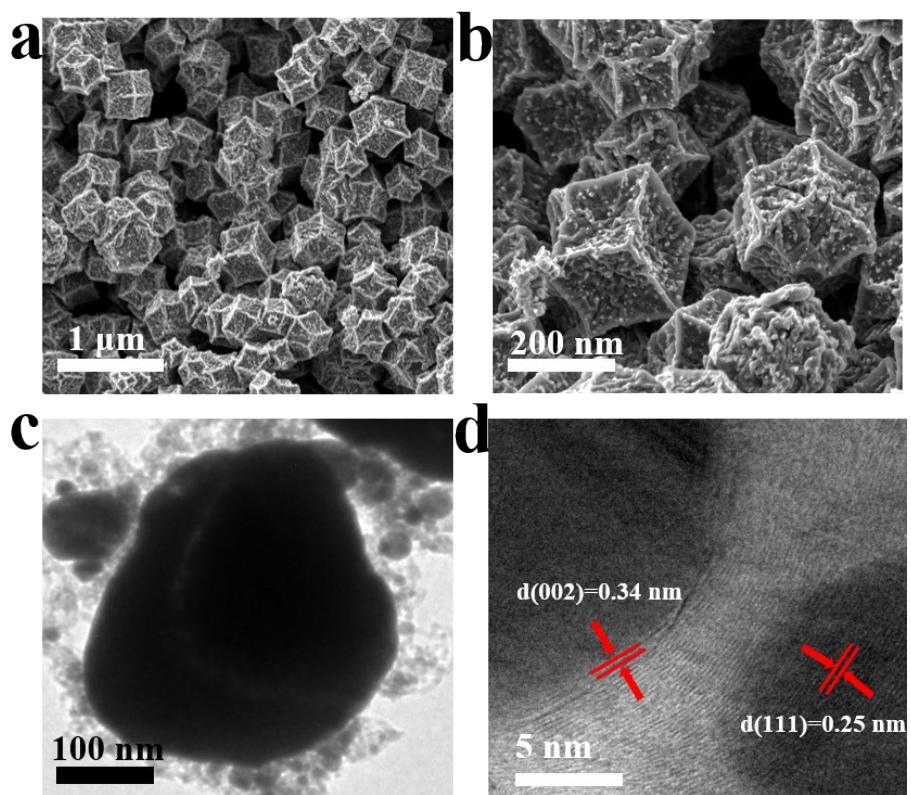


Fig. S7. (a, b) SEM images and (c, d) TEM images of ZIF-67(600).

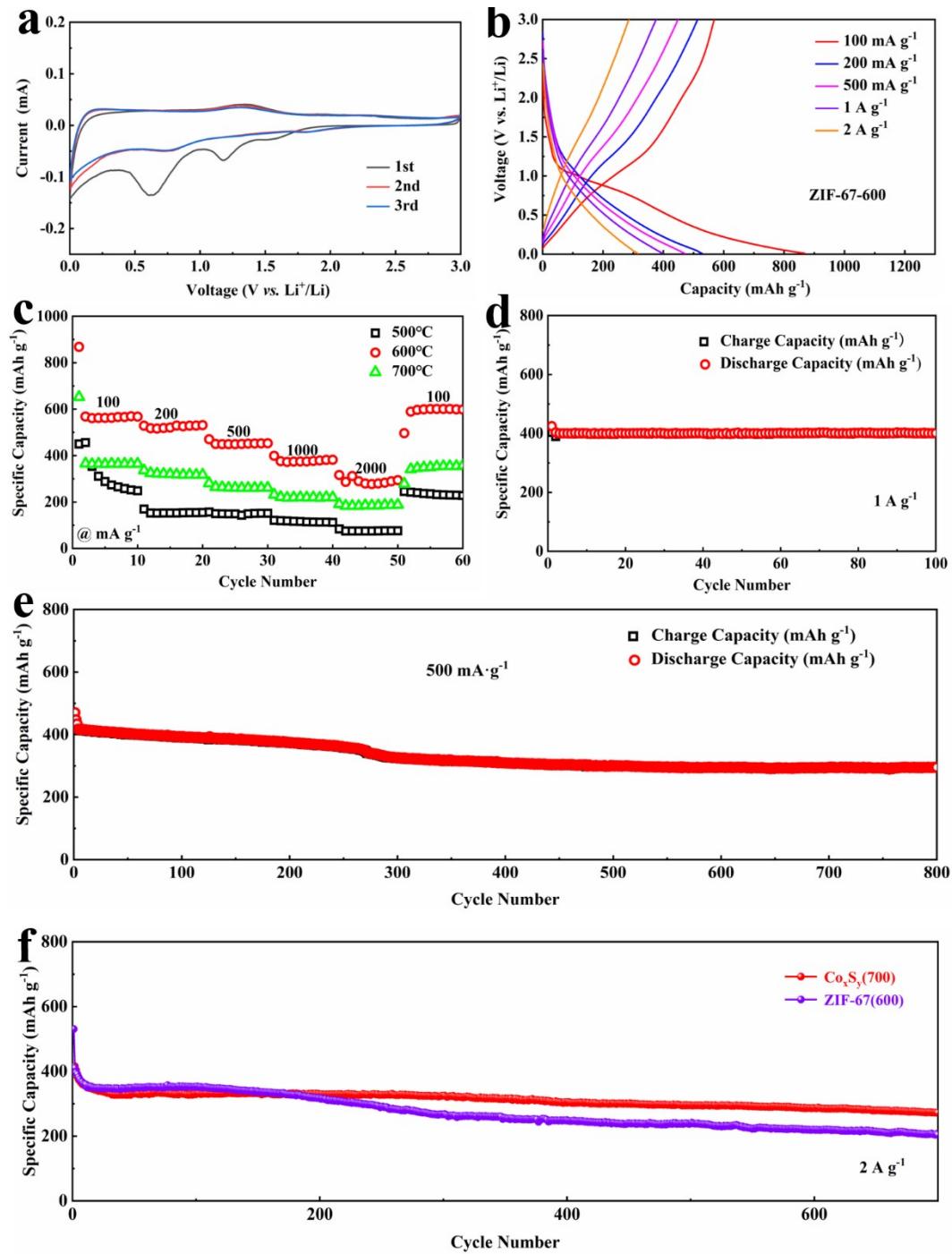


Fig. S8. CV curve (a) and galvanostatic charge-discharge curves at different rates (b) of ZIF-67(600).

Rate performance (c) of ZIF-67(*m*); cycling stability at 1 A g⁻¹ (d) and 500 mA g⁻¹ (e) of ZIF-67(600).

(f) The cycling performance comparison of ZIF-67(600) and Co_xS_y(700) at 2 A g⁻¹ for 700 cycles.

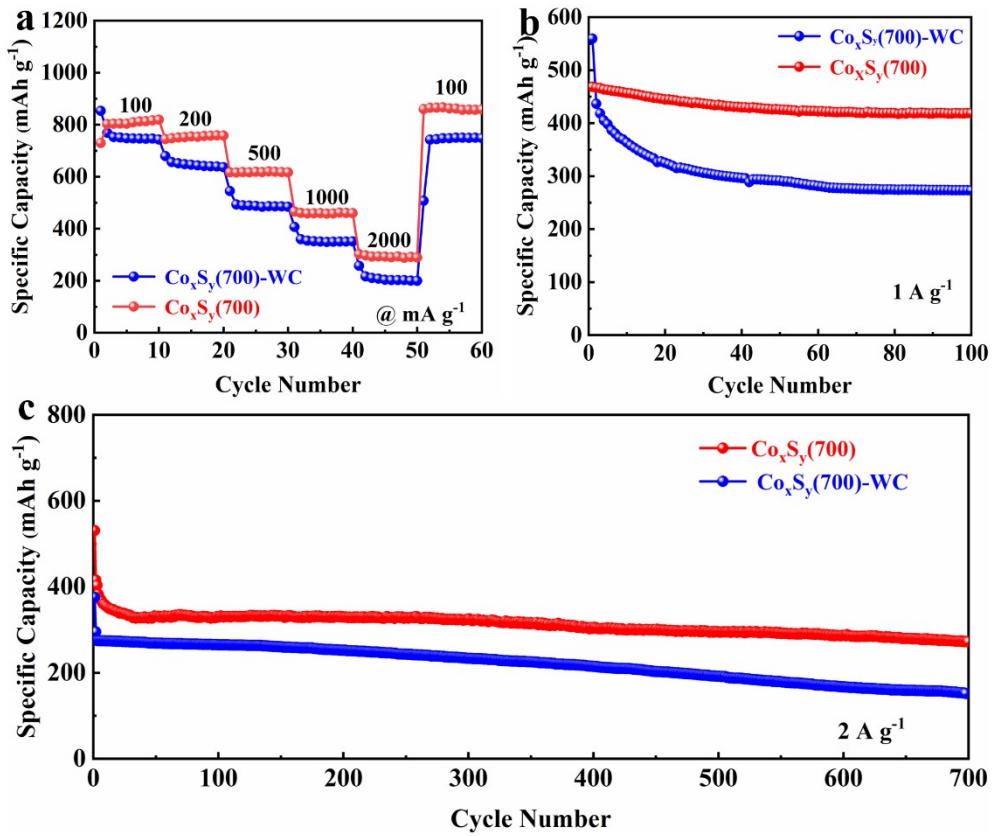


Fig. S9. The comparison of rate (a), cycle (b), and long cycle (c) for Co_xS_y(700) and Co_xS_y(700)-WC.