

1 **Electrocatalytic hydrogenation of furfural paired with photoelectrochemical oxidation**
2 **of water and furfural in the batch and flow cells**

3 **Ram Ji Dixit^a, Aditya Singh^a, Vijay Ramani^{b*}, Suddhasatwa Basu^{a,c*}**

4 ^aDepartment of Chemical Engineering, Indian Institute of Technology Delhi, Hauz Khas,
5 New Delhi 110016, India

6 ^bDepartment of Energy, Environmental & Chemical Engineering, Washington University in
7 St. Louis, St. Louis 63130, Missouri, United States

8 ^cCSIR-Institute of Minerals and Materials Technology, Bhubaneswar 751013, India

9 *corresponding author: sbasu@iitd.ac.in, ramani@wustl.edu

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

1

Supplementary information

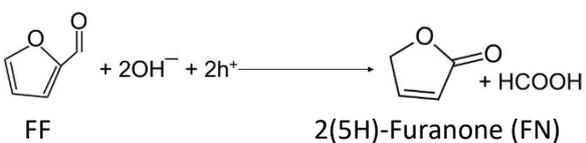
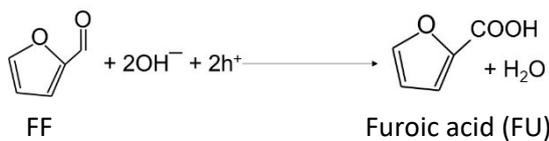


2 Cathode: Furfural (FF)

Furfuryl alcohol (FA)

3

4 Photoanode:



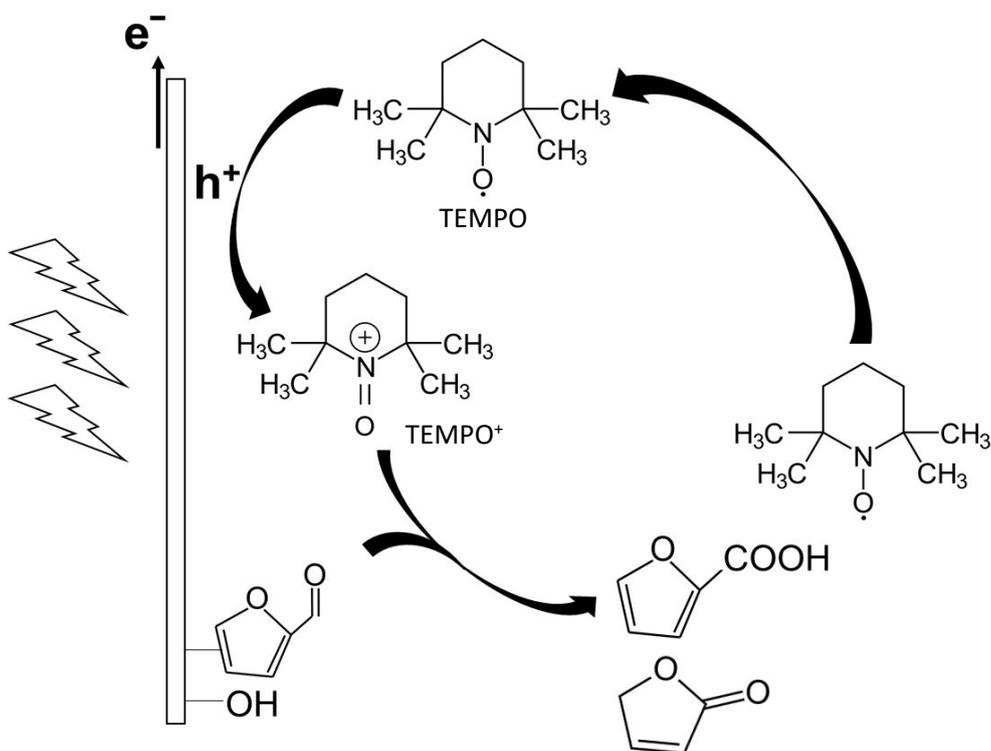
5

6

7

8 **Scheme S1.** Half-cell reactions at the Cu-Ni/NF cathode and TiO₂ NTs photoanode.

9

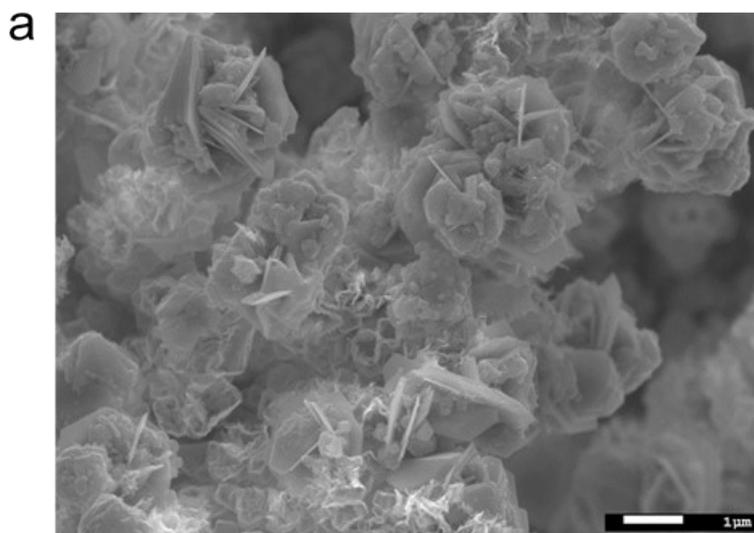


10

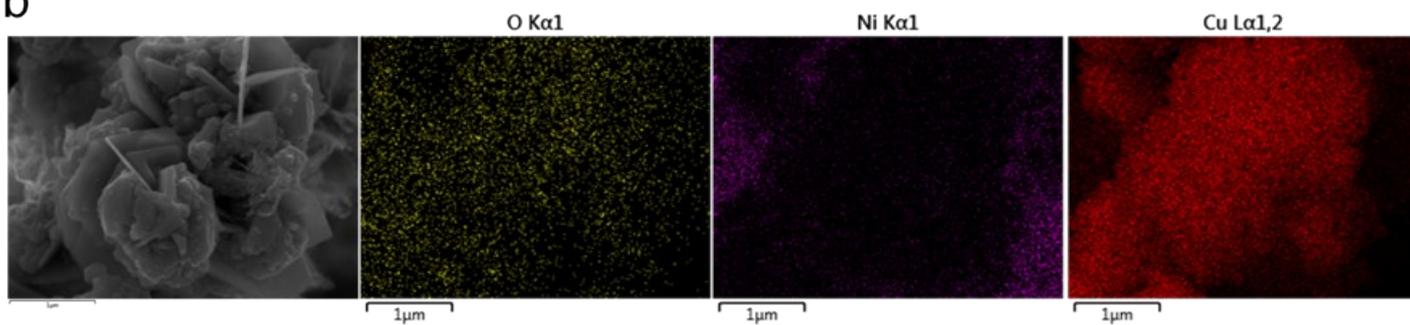
1 **Scheme S2.** TEMPO mediated FF oxidation at the TiO₂ NTs photoanode surface.

2

3



b



4

5

6 **Fig. S1. a.** SEM; **b.** EDX analysis of Cu-Ni/NF.

7

8

9

10

11

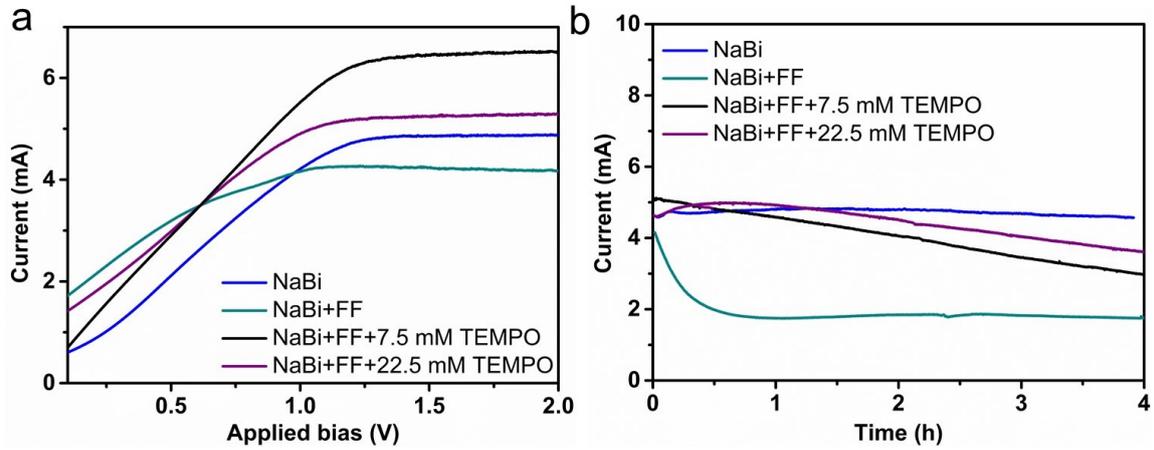
12

13

14

15

1
2
3

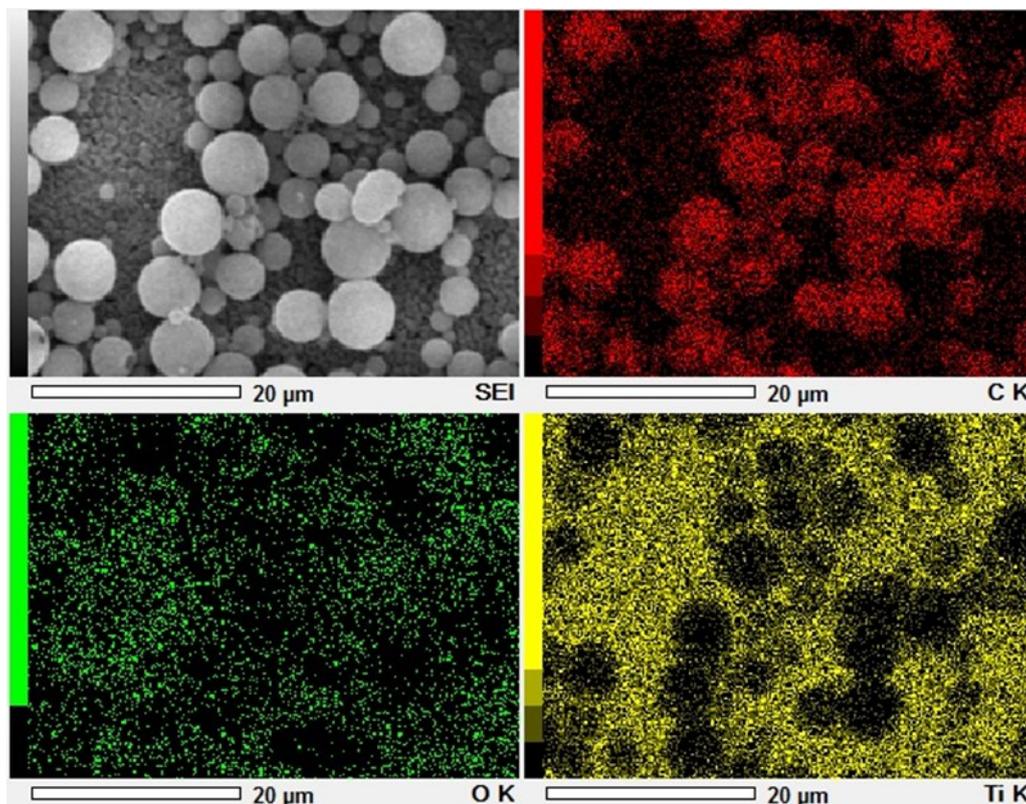


4
5

6 **Fig. S2. Photoelectrochemical cell (PEC)** | a. Linear sweep voltammetry using analytes at the
7 scan rate of 20 mV sec⁻¹; b. Chronoamperometry. Analytes: NaBi, NaBi+FF, and
8 NaBi+FF+7.5 mM TEMPO, NaBi+FF+22.5 mM TEMPO. Catholyte: NaBi+FF. Reaction
9 condition: applied bias of 1 V for 4 h for chronoamperometry. PEC: Cu-Ni/NF cathode with
10 TiO₂ NTs photoanode.

11
12
13
14
15
16
17
18
19

1



2

3 **Fig. S3.** SEM and EDX of photoanode after paired reaction (furfural electrocatalytic
4 hydrogenation paired with photoelectrochemical oxidation) using NaBi+FF anolyte at 1 V for
5 4 h in PEC. Catholyte: NaBi+FF, PEC: Cu-Ni/NF cathode and TiO₂ NTs photoanode.

6

7

8

9

10

11

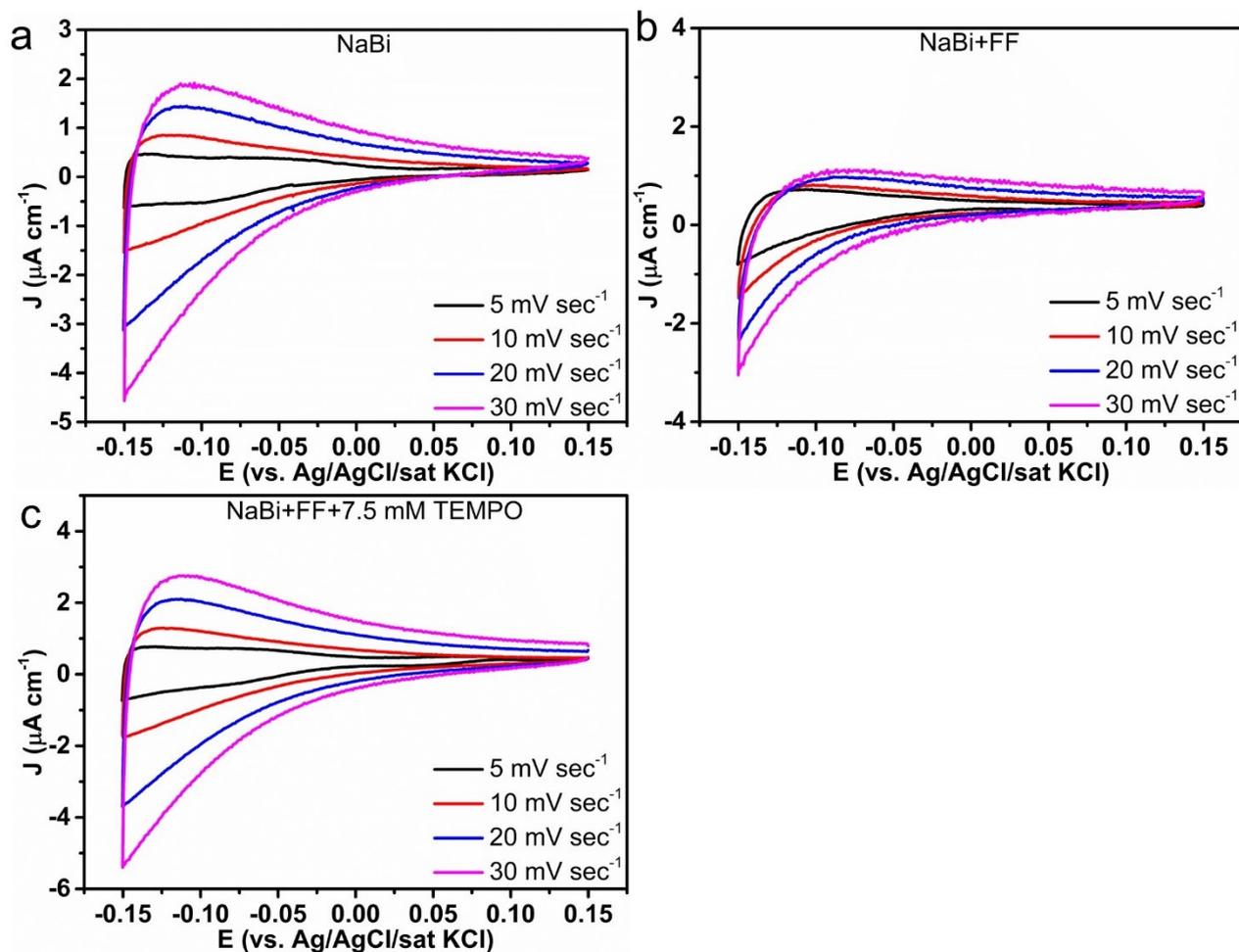
12

13

14

15

16



1

2 **Fig. S4.** CV in a 3-electrode undivided PEC under dark condition at the scan rate of 5, 10, 20,
 3 30 mV sec^{-1} using **a.** NaBi; **b.** NaBi+FF; **c.** NaBi+FF+7.5 mM TEMPO electrolytes.

4

5

6

7

8

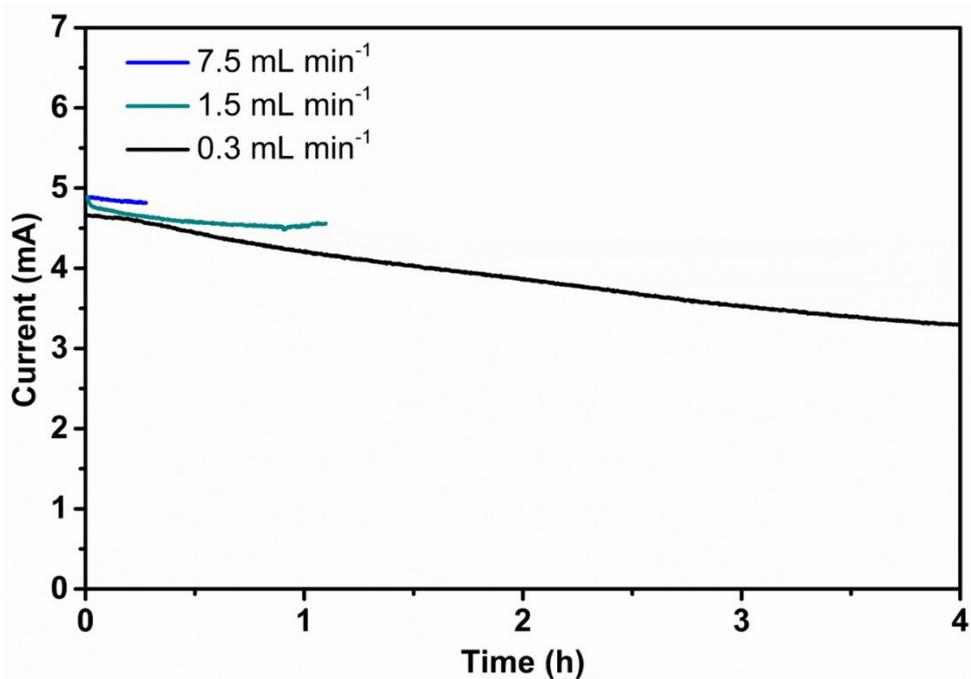
9

10

11

12

13



1

2

3 **Fig. S5.** Chronoamperometry in a flow PEC at the flow rate of 0.3, 1.5, 7.5 mL min⁻¹.

4 Catholyte: NaBi+FF, anolyte: NaBi+FF+7.5 mM TEMPO. Reaction condition: applied bias of

5 1 V. PEC: Cu-Ni/NF cathode and TiO₂ NTs photoanode.

6

7

8

9

10

11

12

13

14

15