Sustainable synthesis of acetals from glycerol as potential additives for biofuels under solvent-free conditions

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GENERAL TECHNIQUES

Analytical grade commercial solvents and reagents were purchased from Sigma-Aldrich, and used as received. Infrared spectra were recorded as neat using a FT-IR Varian 660 Fourier transform infrared spectrometer. Values are expressed in wavenumbers (cm⁻¹) and recorded in a range of 4000–400 cm⁻¹. NMR spectra were recorded at 25 °C in CDCl₃ on a Varian Mercury 300 spectrometer operating at 300 MHz for ¹H. All chemical shifts are reported in parts per million (ppm) and were measured relative to the solvent in which the sample was analyzed (CDCl₃ δ = 7.26). The percentage of acetals yield (%) were determined by using by $^{1}\mathrm{H}$ NMR using TMB with internal standard.

EXPERIMENTAL PROCEDURES



Fig. S1 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with benzaldehyde.



Fig. S2 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl_3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with furfural.



Fig. S3 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 2-thiophenecarboxaldehyde.



Fig. S4 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl_3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 3-methoxybenzaldehyde.



Fig. S5 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with butyraldehyde.



Fig. S6 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-cyanobenzaldehyde.



Fig. S7 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-methylbenzaldehyde.



Fig. S8 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-methoxybenzaldehyde.



Fig. S9 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 3,4,5-trimethoxybenzaldehyde.



Fig. S10 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-carboxybenzaldehyde.



Fig. S11 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-hydroxy-3-methoxybenzaldehyde.



Fig. S12 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-bromobenzaldehyde.



Fig. S13 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-chlorobenzaldehyde.



Fig. S14 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-fluorobenzaldehyde.



Fig. S15 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 2-nitrobenzaldehyde.



Fig. S16 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl_3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 3-nitrobenzaldehyde.

Fig. S17 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-nitrobenzaldehyde.

Fig. S18 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl_3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 2-hydroxybenzaldehyde.

Fig. S19 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 3-hydroxybenzaldehyde.

Fig. S20 ¹H NMR spectrum (300.069 MHz, CDCl₃, δ_{CHCl_3} 7.26, δ_{TMB} 6.09), reaction mixture of the acetalization of glycerol with 4-hydroxybenzaldehyde.