Supporting Information for "Effects of applied voltage on water at a gold electrode interface from *ab initio* molecular dynamics"

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1 Additional electrostatic data

The computed electrode potentials for each trajectory presented in the main text follow from the solution and metal potentials Φ_S and Φ_M , respectively. The values of these quantities at every ~48 fs for each trajectory are given in Figs. S1 and S2. As is mentioned in the main text and is visible here, Φ_M was the more volatile quantity of these two and contributed to most of the magnitude of the fluctuations in Φ_e .



Figure S1: Solution potentials Φ_S as a function of simulation time *t* for each trajectory, designated by their electrode charges σ .



Figure S2: Metal potentials Φ_M as a function of simulation time *t* for each trajectory, designated by their electrode charges σ .

We chose to measure Φ_M as the value of $\overline{\Phi}$ in the back 1 Å of the Au(111) slab because of the stability of this quantity relative to alternatives. To illustrate this, we plotted the value of Φ_M throughout the $\sigma = +0.2 \ e$ trajectory as measured three different ways in Fig. S3. The relative behaviors of Φ_M measured each way were similar for each trajectory. These values demonstrate that the method utilized for the data in Fig. S2 and in the main text avoided the very large fluctuations likely associated with the sampling of unusual, or far from equilibrium atomic structures during the dynamics. We can further rationalize this choice by the limitation of relatively short trajectories and our interest in water under bias at thermal equilibrium. The fluctuations of Φ_M even with our chosen method were large compared to those of Φ_S , underscoring the trade-off between fluctuations in potential vs. charge in fixed-charge and fixed-potential methodologies, respectively.



Figure S3: Values of Φ_M for the $\sigma = +0.2 e$ trajectory as computed as the value of $\overline{\Phi}$ at a fixed *z* near the middle of the slab (black), the value of $\overline{\Phi}$ at the average value of *z* for all Au atoms (green), and as the average value of $\overline{\Phi}$ in the back 1 Å of the Au slab, within the fixed layer (purple; same as corresponding data in Fig. S2).

The system's Fermi level $E_{\rm F}$ exhibits even larger fluctuations than $\Phi_{\rm M}$, measured any way, during the trajectories, as shown in Fig. S4. While the system's Fermi energy has been used as a reliable metal potential in static or other applications, ^{1–4} it was satisfactory in this work to utilize the more stable $\Phi_{\rm M}$ as measured at the partially-frozen back of the Au(111) slab.



Figure S4: Computed Fermi energies E_F of the system every 100 MD steps (~0.48 fs) in each trajectory. Note that the scale on which this quantity is plotted is much larger than that of Φ_M in Fig. S2.

2 Water hydrogen bonding and orientations

The structural properties of water at the Au(111) interface were further quantified by counting the average numbers of hydrogen bonds formed by water as functions of both its layer and the applied voltage vs. PZC, φ . Hydrogen bonds were defined by O atoms being less than 3.5 Å apart with an ODO angle within 30° of colinear.⁵ The average numbers of hydrogen bonds formed by water molecules in each layer were further distinguished by whether the hydrogen bond was formed with another water molecule of the same or of a neighboring layer. These values are reflected in Figs. S5-S7.



Figure S5: Average numbers of hydrogen bonds in layer 1 of water as a function of φ , delineated by intra- (cyan) and inter-layer (magenta) contributions. Layer 1 water formed fewer hydrogen bonds than water in layers 2 and 3. Water in this layer formed more hydrogen bonds at $\varphi \ge 0$ V vs. PZC and the proportion of intra-layer hydrogen bonds generally increased with φ .



Figure S6: Average numbers of hydrogen bonds in layer 2 of water as a function of φ , delineated by intra- (cyan) and inter-layer (magenta) contributions. Water in layer 2 demonstrated the most overall hydrogen bonds of the three layers studied. The overall number of hydrogen bonds in this layer was not apparently φ -dependent, although the proportion of intra-layer hydrogen bonds generally decreased with φ , in opposition to the trend in layer 1.



Figure S7: Average numbers of hydrogen bonds in layer 3 of water as a function of φ , delineated by intra- (cyan) and inter-layer (magenta) contributions. Layer 3 water formed slightly fewer hydrogen bonds on average compared to layer 2, and neither the overall number nor the proportion of intra- to inter-layer hydrogen bonds demonstrated a clear dependence on φ .

From Fig. S5 it is evident that the total average number of hydrogen bonds formed by layer 1 water is lower at negative φ than at PZC and positive φ . Additionally in layer 1, we observed an increase in the number intra-layer hydrogen bonding and a decrease in inter-layer hydrogen bonding when going from negative φ to zero and positive φ . That layer 1 water at $\varphi \ge 0$ V exhibits more overall and intra-layer hydrogen bonds than at $\varphi < 0$ is consistent with the higher densities observed at these potentials. This observation is likely also a consequence of the O atom being more likely than not oriented towards the Au surface at non-negative φ given that the O atom typically participates in two hydrogen bonds compared to one for H atoms. Therefore, when O is oriented towards Au and not towards layer 2, its ability to form *inter-layer* hydrogen bonds is suppressed. The hydrogen bonding similarities between the system at PZC and at $\varphi > 0$ in layer 1 are consistent with their similarities in density as a function of *z* as well.

Figs. S6 and S7 demonstrate that layer 2 water participates in the most hydrogen bonds overall of the 3 studied. The average overall numbers of hydrogen bonds formed per water in layers 2 and 3 did not show any clear dependence on φ , consistent with the diminished dependence of D_{\parallel} and the VDOS spectra on φ in these two layers. In addition, the proportion of intra- and inter-layer hydrogen bonds in layer 2 with respect to φ was opposite to that in layer 1. In layer 3, the hydrogen bonding data did not seem to depend on φ and the differences with φ were likely within the error of the short simulations.

The effects of φ on water's orientation persisted beyond layer 1, the data for which is given in the main text. The orientations of the water bisectors and OD bonds with respect to surface normal for all layers and φ are given in Figs. S8 and S9.



Figure S8: Histograms of the cosines of the angles formed between water's bisector and a unit vector in z for all φ and all layers (1: magenta, 2: red, 3: orange, 4: gray). The inset illustration indicates the relevant angle θ in which the yellow rectangle represents the Au slab, the O atom is red, and the D atoms are white. The effects of φ on water's orientations relative to surface-normal were strongest in layer 1, but persisted into layers 2 and 3.



Figure S9: Histograms of the cosines of the angles formed between OD bond vectors and a unit vector in z for all φ and all layers (1: magenta, 2: red, 3: orange, 4: gray). The inset illustration indicates the relevant angles θ in which the yellow rectangle represents the Au slab, the O atom is red, and the D atoms are white. The effects of φ on the orientations of water's OD bonds relative to surface-normal were strongest in layer 1, but persisted into layers 2 and 3.

3 Mean square displacements

The O atom MSDs in *x* and *y* up to about half of the total trajectory times for each φ and layers 1-3 were computed to determine the values of D_{\parallel} shown in the main text. These are depicted here in Figs. S10-S12. The best fit slopes of the *xy* MSDs were used in Einstein's relation to obtain the 2D diffusion coefficients.



Figure S10: Mean-square displacements in *x* and *y* for water O atoms in layer 1 for each φ as a function of simulation time *t* up to about half of the total BOMD trajectory time. The slopes of the best linear fits to these MSDs were related to the parallel diffusion coefficients D_{\parallel} shown in the main text. Water in this layer diffused parallel to the surface most at negative φ , consistent with their lower densities and the layer's greater distance from the surface. That the displacements of O atoms in layer 1 were smaller at positive φ is also consistent with the O atom being electrostatically attracted to the surface. Note that deviations from linearity for some trajectories are reflected in the large error bars in Fig. 7.



Figure S11: Mean-square displacements in *x* and *y* for water O atoms in layer 2 for each φ as a function of simulation time *t* up to about half of the total BOMD trajectory time. The slopes of the best linear fits to these MSDs were related to the parallel diffusion coefficients D_{\parallel} shown in the main text. Water in layer 2 diffused parallel to the surface most at small, non-zero φ , regardless of sign. At the largest in magnitude values of φ , water diffused parallel to the surface more than at PZC. Note that deviations from linearity for some trajectories are reflected in the large error bars in Fig. 7.



Figure S12: Mean-square displacements in *x* and *y* for water O atoms in layer 3 for each φ as a function of simulation time *t* up to about half of the total BOMD trajectory time. The slopes of the best linear fits to these MSDs were related to the parallel diffusion coefficients D_{\parallel} shown in the main text. Water in layer 3 diffused parallel to the surface most at small, non-zero φ , regardless of sign. At the largest in magnitude values of φ , water diffused parallel to the surface more than at PZC. Note that deviations from linearity for some trajectories are reflected in the large error bars in Fig. 7.

4 Velocity autocorrelation functions

The velocity autocorrelation functions of all water atoms for each φ and layers 1-3 were computed and shown in Figs. S13-S15. The first 1200 fs of each were Fourier transformed to obtain the corresponding VDOS spectra.



Figure S13: Velocity autocorrelation functions $\langle v(0)v(t)\rangle$ of water O and D atoms in layer 1 for each φ . The first 1200 fs were Fourier transformed to obtained the VDOS spectra shown in the main text.



Figure S14: Velocity autocorrelation functions $\langle v(0)v(t)\rangle$ of water O and D atoms in layer 2 for each φ . The first 1200 fs were Fourier transformed to obtained the VDOS spectra shown in the main text.



Figure S15: Velocity autocorrelation functions $\langle v(0)v(t)\rangle$ of water O and D atoms in layer 3 for each φ . The first 1200 fs were Fourier transformed to obtained the VDOS spectra shown in the main text.

5 Additional VDOS spectra and analysis

The VDOS spectra shown in the main text were obtained from the Fourier-transformed velocity autocorrelation functions, followed by Gaussian convolution. The resolution of the VDOS spectra that corresponds to the ~0.48 fs time step was about ~27 cm⁻¹. We further note here that the interpretations of the VDOS spectra are limited to mainly qualitative observations regarding the effects of distance from the surface and φ given the relatively short BOMD trajectories.

Figs. S16-S18 show the referenced VDOS spectra for layers 1-3 corresponding to those in the main text, with the $\varphi = 0$ V spectrum of each layer as the reference. These referenced spectra are all plotted on the same scale. The referenced spectra highlight shifts and differences in intensity at different φ for each of the first three layers of water from the electrode. These spectra are clearly most intense in layer 1, where the effects of the electrode electrification are greatest. The referenced spectra for layer 1 demonstrate that both the δ DOD and vOD peaks are blue shifted at non-zero φ . The referenced vOD signals in layers 2 and 3 for non-zero φ are slightly negative at the lower frequencies characteristic of vOD and slightly positive at the higher frequencies characteristic of vOD. This is consistent of water molecules in layers 2 and 3 being somewhat oriented according to the screened electric field (Figs. S8 and S9).



Figure S16: VDOS spectra at each φ for layer 1 water referenced to PZC (black; zero by construction). Negative intensities at lower frequencies and positive intensities at higher frequencies around both regions of interest are representative of blue shifts of both peaks at non-zero φ .



Figure S17: VDOS spectra at each φ for layer 2 water referenced to PZC (black; zero by construction). The lower intensities relative to layer 1 are indicative of a lesser effect of the electrode potential on the vibrational spectra.



Figure S18: VDOS spectra at each φ for layer 3 water referenced to PZC (black; zero by construction). The lower intensities relative to layer 1 are indicative of a lesser effect of the electrode potential on the vibrational spectra.

For the purposes of measuring the positions and integrating the intensities of the two VDOS peaks of interest despite the $\sim 27 \text{ cm}^{-1}$ resolution, we interpolated between data points for the appropriate portions of each spectra. The positions of the δ DOD peaks were sensitive to both φ and the layer. These data are shown in Fig. S19. In layer 1, the peak occurs at a higher frequency at all non-zero φ , as discussed in the main text. We observed blue shifts of this vibrational mode's frequency in layer 1 was up to about 10 cm⁻¹ at modest positive and negative φ . Layer 2 demonstrated very little effect of φ on the δ DOD peak position in layer 3 appeared to blue shift with increasing φ . Moreover, the position of the δ DOD peak in layer 1 was red-shifted relative to those of layers 2 an 3. This is characteristic of water with a weakened hydrogen bonding network,⁶ consistent with the observation of the least number of hydrogen bonds formed on average in layer 1 (Figs. S5-S7).



Figure S19: Positions of the water δ DOD peaks for each φ and layers 1-3 as determined by the maximum values of the interpolating functions in this region of the spectra.

The positions of the water *v*OD peaks for each φ and layers 1-3 were also determined via interpolation are shown in Fig. S20. We only investigated the global maxima of the interpolating functions for each spectrum despite there likely being several characteristic *v*OD peaks.^{7–9} These data also demonstrate that in layer 1 the position of the peak was blue-shifted at all non-zero φ , with shifts of the positions of maximum frequency of over 100 cm⁻¹. There was no discernible pattern in the *v*OD peak positions beyond layer 1.



Figure S20: Positions of the water *v*OD peaks for each φ and layers 1-3 as determined by the maximum values of the interpolating functions in this region of the spectra.

We also integrated the total intensities of the δ DOD peaks using the interpolated spectra. These integrated intensities are shown in Fig. S21. For layer 1, we observed a minimum integrated spectral intensity for δ DOD at the PZC, which was also observed for SEIRAS spectra of aqueous electrolytes on Au(111).⁷ We observed a maximal integrated

 δ DOD peak intensity at the most negative φ studied. There were no discernible patterns observed among the integrated intensities of the δ DOD peaks in layers 2 and 3.



Figure S21: Integrated intensities (arbitrary units) of the water δ DOD peaks for each φ and layers 1-3. The integration range was 1000-1400 cm⁻¹.

For comparison, we also computed the VDOS spectra from the velocity autocorrelation functions only in the surface-normal dimension z given the importance of surface-normal dipoles in many surface-enhanced spectroscopies. These are shown alongside their corresponding 3D VDOS spectra in Figs. S22-S24. For the most part, the z-only VDOS spectra hew closely to the full VDOS spectra, especially in the second and third layers. In layer 1, the z-only VDOS spectra demonstrate some subtle differences from the 3D spectra: The δ DOD peaks appear slightly red-shifted in the z-only spectra relative to the full spectra. In addition, the z-only spectra for layer 1 generally hew to only the second vOD peaks of the full spectra for $\varphi \neq 0$ V. That this is more evident when there is a non-zero potential is consistent with the observation of water molecules that are more oriented along the surface-normal electric fields associated with the applied biases. This difference between the full and z-only spectra furthermore suggests that there are more than one characteristic vOD mode in this heterogeneous environment. However, it is likely that all vibrational modes are strongly coupled between the (arbitrary) Cartesian dimensions given the overall similarities between the spectra.



Figure S22: Layer 1 water VDOS spectra obtained from the 3D velocity autocorrelation functions (solid; same as main text) and from the *z* velocity autocorrelation functions (dashed). The spectra for each φ are shifted arbitrarily for clarity.



Figure S23: Layer 2 water VDOS spectra obtained from the 3D velocity autocorrelation functions (solid; same as main text) and from the *z* velocity autocorrelation functions (dashed). The spectra for each φ are shifted arbitrarily for clarity.



Figure S24: Layer 3 water VDOS spectra obtained from the 3D velocity autocorrelation functions (solid; same as main text) and from the *z* velocity autocorrelation functions (dashed). The spectra for each φ are shifted arbitrarily for clarity.

The integrated intensities of the δ DOD peaks for the 3D and *z*-only VDOS spectra for layer 1 water are shown in Fig. S25. These integrals show discrepancies only at positive φ , and for the *z*-only spectra the integrated intensity is no longer minimal at PZC.



Figure S25: Integrated intensities (arbitrary units) of the water δ DOD peaks in the 3D (magenta) and *z*-only (light pink) VDOS spectra for layer 1 water.

6 Representative input files and initial Cartesian coordi-

nates

Example Quantum-ESPRESSO^{10,11} PWscf input file, including initial coordinates, for the

σ = -0.4 *e* trajectory:

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   outdir
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   forc\_conv\_thr = 1.0D-3,
   dt = 10.0,
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   tefield = .true.,
   dipfield = .true.,
   gate = .true.,
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               = 0,
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!
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   nspin
   tot_charge
              = -0.4,
```

```
= 216,
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               = 3,
   ntyp
                = 60,
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   edir = 3,
   emaxpos = 0.99,
   eopreg = 0.95,
   zgate = 0.87,
   block = .true.,
   block_1 = 0.743665,
   block_2 = 0.91,
   block_height = 0.25,
&ELECTRONS
   conv thr
              = 1.D-4,
   mixing_mode = 'plain',
   mixing_beta = 0.5D0,
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   startingpot = 'file',
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&IONS
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   pot_extrapolation = 'first_order',
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 Η
    15.999 O_ONCV_PBE-1.0.UPF
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   0.00000000
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                                                            1
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                                                       1
                                                            1
                                                                0
Au
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                                                       1
                                                            1
                                                                0
Au
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                                                                0
Au
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Н	2.071304724	9.713494133	18.160454727
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О Н Н О Н Н О Н	8.784266081 7.861763259 9.297933137 6.817099986 6.776928047 7.702839380 1.104121552 1.248163681	8.106721185 7.871519202 7.791784221 8.348986720 7.387478443 8.352776968 8.635969590 7.739032244	22.922853506 22.656446073 22.132015071 18.081828776 18.326417907 17.656090659 15.292988970 14.848938860
H	0.184883987	8.592578060	15.616902938
0	10.082773089	7.106578078	20.737625620
Н	0.763857614	7.480847628	20.760174626
Η	0.006667415	6.136304498	20.700411653
0	5.340199259	11.240343292	14.919425159
Η	5.731463722	10.363702969	14.680736780
H	4.368268790	11.058479676	14.892990557
0	7.187144861	2.890747058	23.681981996
H	8.090164596	2.938/48/66	24.UZIZZIZ65
П	0.920910203	1.939340003 10.245507241	18 053635333
н	5 535400463	9 616309354	18 075549348
H	4.068274911	9.715128541	17.606037243
0	4.643734084	3.352721579	17.488688572
H	4.650260212	2.532359508	16.956464823
Н	5.321849959	3.189297114	18.169821351
0	1.357360322	6.196853128	14.571333565
Н	0.527620207	5.693428033	14.813738514
H	1.686111298	5.775095878	13.750884539
0	8.701776102	7.718298499	15.292727958
H	1.142451933	/.53/135286	15.36/4/9610
H	9.03//41243	6.81/1694Z3 5 /27750272	13.0/282608/
U H	9.909077433	J.42//JU2/J / 508601130	24.023243730
H	9 825442120	5 739686861	23 120100759
0	2.188042221	1.037771562	17.156754963
H	2.406764988	0.773520384	16.245266186
Н	2.044255473	12.029971974	17.683522319
0	6.108185019	8.772426972	14.332849826
Н	6.512709617	7.998699955	13.882821341
H	5.206968795	8.377721544	14.512851591
0	2.40/504330	2.33/531190	20.368124086
H	1.820377219	1.626642689	20./19900//6
H	3.21136/483	1.8609/9536 2.125566974	20.111648467
U H	4.710720303 5 119816000	1 25/770380	14.020277000
H	3.763570969	1.928102807	14.656994079
0	5.928667169	0.732223829	17.371483246
H	6.091221470	12.376808279	16.426037003
Н	5.453171943	11.708238437	17.622063179
0	3.915635019	0.708999047	23.001575333
Н	2.982259975	0.458710579	23.117727809
H	4.057554913	0.651962033	22.034145573
U U	4.18/830893	4.865685626	14.830041/57
п u	4.02404U332 3.35030/101	4.1/3668/3U / /2022010/	1/ 520020267
11	J.JJUJU4IUI	4.430237134	17.3092930/

0	2.734930166	6.912155050	20.123100889
Н	2.342701406	6.707709944	19.241110277
Н	3.426032423	6.216702418	20.243083337
0	5.215148398	9.245526769	21.556350353
Н	4.671907214	9.283395995	22.369309437
Н	5.530929727	8.306644976	21.524715444
0	1.034654221	2.574185280	23.336568339
Н	1.203145716	1.776481719	22.785872268
Н	0.399288146	2.259355231	23.996447712
0	2.960577986	10.250997397	14.322452975
Н	2.097855703	9.960546212	14.703456519
H	3.338466165	9.402438925	14.032273083
0	2.318035393	0.982402524	14.210167537
H	2.700310798	0.187608399	13.784531538
Н	1.416799019	0.652175916	14.483836850
0	9.519799416	8.071971769	18.075852980
H	9.574320183	7.863576501	19.032485268
Н	10.030791243	7.336541678	17.684751977
0	6.437874476	6.879296944	22.015009570
Н	6.913604910	6.204263999	21.477493482
Н	6.090332561	6.331210333	22.761748149
0	7.536596241	10.031564962	20.211838393
Н	6.765485488	9.800110846	20.770287063
Н	7.405208096	9.484800858	19.399728936
0	6.593252890	6.157429864	14.402972366
Н	5.662517227	5.851047496	14.418310548
Н	7.083337222	5.306785394	14.368605413
0	1.962559959	3.590632904	14.008047808
Н	2.185224005	2.739006292	13.579945928
Н	1.097275312	3.358570465	14.431524385
0	4.320369268	6.147603890	17.444023372
Н	5.272127100	6.125501049	17.670962284
Н	4.142719942	5.189664553	17.317894939
0	1.335442762	3.382910157	18.114275868
Н	1.837484628	3.083071404	18.909505056
Н	1.548836692	2.613279755	17.520719692
0	7.594248263	0.772338415	14.536015698
Н	8.263070039	11.932337678	14.267258595
Н	6.742224287	0.298724870	14.444621387
0	8.945901579	3.860722780	17.442625259
Н	8.586300400	2.941156546	17.655562648
Н	9.871368663	3.798414073	17.783055100

Accompanying Environ^{12,13} input file for the $\sigma = -0.4 e$ trajectory:

```
&ENVIRON
  !
  verbose = 2
  environ_thr = 1.d-1
  environ_type = 'input'
  env_electrostatic = .true.
  env_static_permittivity = 78.3D0
  env_dielectric_regions = 1
```

```
!
 /
 &BOUNDARY
  solvent_mode = 'electronic'
  solvent_radius = 7.0D0,
 /
 &ELECTROSTATIC
  !
  pbc_correction = 'none'
  pbc_dim = 2
  pbc_axis = 3
   !
  tol = 1.d-7
  mix = 0.6
   solver = 'iterative'
   auxiliary = 'full'
   !
 /
DIELECTRIC_REGIONS {angstrom}
1.0 1.0 5.5 5.5 6.4104 6.4104 0.5 2 3
```

Initial coordinates of the $\sigma = -0.2 e$ trajectory:

Au	5.113345000	0.000000000	6.410460000
Au	7.670027730	1.476096250	6.410460000
Au	5.113345000	2.952192500	6.410460000
Au	7.670027730	4.428288750	6.410460000
Au	5.113345000	5.904385000	6.410460000
Au	7.670027730	7.380481250	6.410460000
Au	5.113345000	8.856577500	6.410460000
Au	7.670027730	10.332673750	6.410460000
Au	0.00000000	0.000000000	6.410460000
Au	2.556672500	1.476096250	6.410460000
Au	0.000000000	2.952192500	6.410460000
Au	2.556672500	4.428288750	6.410460000
Au	0.000000000	5.904385000	6.410460000
Au	2.556672500	7.380481250	6.410460000
Au	0.000000000	8.856577500	6.410460000
Au	2.556672500	10.332673750	6.410460000
Au	6.817796740	0.000000000	8.820910000
Au	9.374469240	1.476096250	8.820910000
Au	6.817796740	2.952192500	8.820910000
Au	9.374469240	4.428288750	8.820910000
Au	6.817796740	5.904385000	8.820910000
Au	9.374469240	7.380481250	8.820910000
Au	6.817796740	8.856577500	8.820910000
Au	9.374469240	10.332673750	8.820910000
Au	1.704451740	0.000000000	8.820910000
Au	4.261124240	1.476096250	8.820910000
Au	1.704451740	2.952192500	8.820910000
Au	4.261124240	4.428288750	8.820910000
Au	1.704451740	5.904385000	8.820910000
Au	4.261124240	7.380481250	8.820910000

Au	1.704451740	8.856577500	8.820910000
Au	4.261124240	10.332673750	8.820910000
Au	8.522248480	0.00000000	11.231400000
Au	5.965575980	1.476096250	11.231400000
Au	8.522248480	2,952192500	11.231400000
A11	5 965575980	4 428288750	11 231400000
Δ11	8 522248480	5 90/385000	11 231/00000
Λ11	5 965575980	7 380481250	11 231/00000
Au	9 522249490	9 956577500	11 231400000
Au	5 065575000	10 222672750	11 221400000
Au	2 100002100	10.3320/3/30	11.231400000
Au	3.400903400		11.231400000
Au	0.052224050	1.4/0090230	11.231400000
Au	3.408903480	2.952192500	11.231400000
Au	0.852224850	4.428288750	11.231400000
Au	3.408903480	5.904385000	11.231400000
Au	0.852224850	7.380481250	11.231400000
Au	3.408903480	8.856577500	11.231400000
Au	0.852224850	10.332673750	11.231400000
0	3.392863129	8.809062166	23.694277586
Η	3.430799695	7.839672957	23.699482202
Н	2.626849334	9.025043682	24.247672842
0	4.100611148	4.608363981	20.958175486
Н	3.305030210	4.076227276	21.104107817
Н	4.720366595	3.941982484	20.572508445
0	2.935248419	8.581025711	17.307817174
H	3.415418403	7.718988414	17.427422307
H	2 313723223	8 502514644	16 541552606
0	0.043767103	4.235117830	21.080796629
н	0 326780003	3 663175735	21 839836725
и П	0.116168981	3 8/6553181	20 283516859
\cap	2 053216812	0 15/050530	20.200000
U	2.00042	9.104059559	21.131402093
П	2.034195247	0.429700001	20.007023032
H	2.403133132	9.143631908 7 E4201E202	22.070556559
0	3.853/16830	7.542815393	14.4/2981998
H	3.90/300409	6./06933215	14.999745493
Н	3./8136960/	/.21/562682	13.550/91354
0	6.473315818	0.334455691	23.730797688
H	5.489615292	0.354077254	23.613836238
H	6.688796059	11.284056986	24.128080726
0	4.286365818	-0.170391849	20.270686922
Н	4.682366286	10.933117828	20.826425418
Η	4.312497523	11.178936379	19.385526174
0	7.524145072	1.010448944	21.159860074
Н	7.519535202	12.049645367	20.550862367
Н	7.217591943	0.648071430	22.019052206
0	9.650007722	10.857217229	17.519543852
Η	9.279748643	9.956753942	17.555849014
Н	10.462272859	10.747492044	18.093434563
0	6.963775444	3.437084455	14.019763645
- H	7.476723509	2.639431914	14.265397769
H	6.035444876	3,137927626	14.172662914
\bigcirc	1 329990006	6 008963153	17 708314185
Ч	1 <u>1</u> 295/0200	5 0230/77/7	17 834389378
11	1 705001007	J.UZJU4//4/ 6 10501/510	16 02E1010C0
п	1.123984391	0.199214916	10.033494068

0	6.926793040	5.628968384	18.481327681
H	7.007642900	5.21369/184	19.3/0352132
H	7.613598452	5.13590/020	17.970201902
0	1.3162164/6	0.185395956	21.802899021
H	1.952418/02	11.356222185	21.4351/28/3
H	0.4/48/414/	11.441362857	21.948/13118
0	5.622012454	2.436962203	19.9/6/65160
H	6.450571807	2.070638301	20.402117659
H	5.126813352	1.600497822	19.851215938
0	-0.008905920	11.608006841	15.053907039
Н	9.833822714	10.885075591	14.496058497
Н	10.077150175	11.283038586	15.990474517
0	5.528242653	5.116966923	24.070739598
H	6.121384421	4.326029090	23.965354383
Η	4.686313923	4.853352377	23.663944226
0	9.255197468	5.103843294	15.195698020
Η	9.079881481	4.656453434	16.082406413
Н	9.078323586	4.354410241	14.594598157
0	7.596080062	4.635257600	21.011763902
Н	8.565895657	4.447180378	20.940339315
Н	7.234783876	3.844400687	21.447387692
0	9.357856864	10.691429072	22.278343105
Η	8.817783639	10.593623132	21.460746789
Н	9.297614336	9.779554327	22.659371911
0	9.824590544	2.516453493	14.447823756
H	9.810996843	2.201957951	13.515122005
H	9 672774004	1 687898062	14 947470257
0	8 268927126	1 491772945	18 164085701
н	8 819579384	0 713024541	17 946563594
н	7 339351085	1 196165445	17 963693050
\cap	8 72/775263	9 698252297	13 752235/77
U U	7 762307046	9.090252297 9.0107706	13 6316/0507
II U	9 767109227	9.014977400 0 0/2707701	1/1 200031167
	1 76/631530	10526622272	19 670232030
U	2 00727/075	0 706710072	10 019232030
П	2.00/3/40/3	9.700710072	10.210/10033
П	1.000012004	10.293991924	19.032243139
0	8.803/4/948	8.069654454	22.951234/12
H	7.92/0493/1	7.72220(244	22.8002/0443
H	9.309255230	7.733396344	22.132968847
0	6.989953918	8.3658/3024	18.116662787
H	6.//0824868	7.441304356	18.3/8438980
H	1.955520514	8.241522690	17.894974801
0	1.05396/993	8.645/89514	15.382081884
H	1.209663668	/.800699039	14.860895591
H	0.132225488	8.526/159/8	15.681//5221
0	10.139208894	/.11685/4/0	20./82128882
H	0./84326853	/.563/13795	20.860579526
H	0.105728238	6.155229322	20.794425456
0	5.277516347	11.160116836	14.952820512
H	5.768883669	10.326974559	14.761892774
Н	4.326823362	10.891478025	14.841257633
0	7.151405578	2.944582173	23.794734224
H	8.063446340	3.011237069	24.117787306
Н	6.896383048	1.991081299	23.877506162

О Н Н О Н Н О Н Н О Н Н	4.833740704 5.574640884 4.101036981 4.640576997 4.689656317 5.234367614 1.301738483 0.495178586 1.642919204 8.660042215 7.698730112	10.182384347 9.535435308 9.699402220 3.430735589 2.580093169 3.293534498 6.244361599 5.699301083 5.924235685 7.646253398 7.693409396	18.075760863 18.091168790 17.609561812 17.449023897 16.968742864 18.216416266 14.519611271 14.746449206 13.663121791 15.385727268 15.538787617
H	8.804/30382	6.684030095	15.161060350
U U	9.992294434	J.410000337 A 571778828	24.2103/0130
п	9 731877907	5 849802218	23.305000171
0	2.184738151	1.076672189	17.143034130
H	2.484589706	0.873941766	16.236373635
Н	2.134615796	12.045708620	17.645994564
0	6.164821123	8.744008270	14.273897966
Н	6.603899431	7.903979023	14.007306064
Н	5.256284665	8.364930273	14.462768238
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H	1.//9//90/3 2.122002400	1.5/9282052	20./60/15//6
H	3.122892480	1.760744124 2.092170192	20.025151046
U H	4.700332009 5 nng3gngng	2.002140192	14.737230047
H	3 743982159	1 989939913	14 595946964
0	5.822322324	0.766471908	17.348765278
H	6.097454687	12.415667891	16.429297543
Н	5.439051150	11.689585249	17.584935128
0	3.945207262	0.769331980	23.055927912
Н	3.072211441	0.406648161	23.280526747
H	3.997057137	0.724882984	22.083005228
0	4.1/864205/	4.896436028	14.9438/1056
H U	4.529915034	4.338363411	17.000112044
	2 711528955	6 851562392	20 038307922
H	2.350208519	6.661800239	19.141663349
H	3.384199253	6.142353793	20.172114482
0	5.298105298	9.300251875	21.714208869
Η	4.683895268	9.340819350	22.472340910
Н	5.601847424	8.358082246	21.694625926
0	1.061773826	2.622755182	23.264207119
H	1.223514792	1.803345786	22.746677949
H	0.490434708	2.312810572	23.984661182
U H	2.900032302	9 8615821/2	14.330332042
H	3.294953875	9.378479541	14.001965588
0	2.254508302	0.926409089	14.251154321
Н	2.656621750	0.136629576	13.835662191
Η	1.368259697	0.563006212	14.530162786
0	9.601585439	8.042264530	18.169993824
H	9.675626063	7.845251846	19.132496679
Η	10.064099513	7.276759403	17.770901390

0	6.511785937	6.858680157	22.168856232
Н	6.915481590	6.171232744	21.580777368
Н	6.080576255	6.296940853	22.866949849
0	7.491547808	10.140862092	20.254360561
Н	6.778815182	9.909798768	20.890372087
Н	7.368491283	9.499100256	19.510204876
0	6.562176495	6.125539028	14.299793150
Н	5.645246329	5.784401882	14.390471858
Н	7.082562613	5.294134937	14.285570749
0	1.990671462	3.599797976	13.948932693
Н	2.191071726	2.709355297	13.583077951
Η	1.146612387	3.398942355	14.428755345
0	4.369673646	6.247919763	17.483694310
Η	5.297479272	6.108315818	17.778203041
H	4.060848038	5.329108943	17.358884859
0	1.328804172	3.414979642	18.159122664
Η	1.795740487	3.063743493	18.960224625
Η	1.538678698	2.672561728	17.536195169
0	7.567350980	0.752360432	14.542420580
Η	8.297254609	11.923804937	14.614202639
Η	6.768183665	0.222694492	14.341589562
0	8.880130541	3.919271359	17.517617612
Н	8.567605006	2.955303556	17.626661670
Н	9.798259779	3.880242941	17.866917282

Initial coordinates of the $\sigma = 0 e$ trajectory:

Au	5.173900970	0.285728353	6.410460002
Au	7.867060373	1.441943922	6.410460002
Au	5.365586323	3.177701420	6.410460002
Au	7.666171977	4.526538966	6.410460002
Au	5.044124624	6.001230585	6.410460002
Au	7.558271223	7.427015365	6.410460002
Au	5.159361938	9.121919196	6.410460002
Au	7.565397734	10.442693153	6.410460002
Au	-0.072087954	0.021347949	6.410460002
Au	2.429646017	1.326647058	6.410460002
Au	0.036836371	3.202833250	6.410460002
Au	2.701797208	4.395996953	6.410460002
Au	-0.107874575	6.001697787	6.410460002
Au	2.599128080	7.433233039	6.410460002
Au	-0.168108525	8.959495892	6.410460002
Au	2.403573037	10.269330105	6.410460002
Au	7.015335516	0.015058390	8.921855987
Au	9.690183836	1.453864232	9.114868978
Au	7.249581800	2.896182233	8.808896377
Au	9.634691706	4.489380036	8.947297284
Au	7.038029855	5.734501999	9.492410842
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Au	6.912689379	8.518094413	8.914351224
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