Supplementary Information

Electrochemically Driven Stereoselective Approach to *syn*-1,2-Diol Derivatives from Vinylarenes and DMF

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Section 1. General Information

All reactions were performed in oven-dried two-neck glass tubes unless otherwise noted. The tubes were fitted with a rubber septum and a threaded Teflon cap with airtight, electrical feed-throughs. The reactions were conducted under a nitrogen atmosphere. Flash chromatography was performed using silica gel 60 (230-400 mesh) from SiliCycle. Commercial reagents were purchased from Sigma Aldrich, Alfa Aesar, Acros, and TCI and used as received. (8*R*,9*S*,13*S*,14*S*)-13-methyl-3-vinyl-6,7,8,9,11,12,13,14,15,16-decahydro-17*H*-

cyclopenta[a]phenanthren-17-one,¹ ethyl 2-benzamido-3-(4-vinylphenyl)propanoate², Ncinnamyl-4-methylbenzenesulfonamide³ (starting materials for 16, 17, and 23 respectivly) and 29⁴ were synthesized by the previously reported procedures. Proton nuclear magnetic resonance (¹H NMR) spectra was recorded on 300 MHz or 600 MHz, carbon nuclear magnetic resonance (13C NMR) spectra was recorded on 75 MHz or 100 MHz and fluorine nuclear magnetic resonance (19F NMR) was recorded on 282 MHz. Chemical shifts for protons are reported in parts per million downfield from tetramethylsilane and are referenced to residual protium in the NMR solvent (CHCl₃ = δ 7.26). Chemical shifts for carbon are reported in parts per million downfield from tetramethylsilane and are referenced to the carbon resonances of the solvent (CDCl₃ = δ 77.0). Data are represented as follows: chemical shift, multiplicity (br. s = broad, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constants in Hertz (Hz), integration. Infrared (IR) spectra of the newly synthesized compounds were obtained using a Bruker Alpha FT-IR spectrometer. Cyclic voltammetry data were measured with a Biologics SP-50 potentiostat. High resolution mass spectra was obtained from Korea Basic Science Institute (Daegu) by electron ionization (EI) and fast atom bombardment (FAB) method or from KAIST Research Analysis Center by using electrospray ionization (ESI) method.

Electrolysis experiments were performed using a Biologics SP-50 potentiostat/galvanostat or a DC power supply. Carbon Felt was purchased from Fuel Cell Store. The carbon was cut into 1 x 0.5 x 0.6 cm³ pieces before use, and was connected to electrical feed-through on the Teflon cap of the electrochemical cell via a piece of graphite (2B pencil lead, 2 mm in diameter). The platinum plate was cut into 1 x 0.5 x 0.02 cm³ and was connected to electrical feed-through on the Teflon cap of the electrochemical cell via a piece of graphite (2B pencil lead, 2 mm in diameter). Saturated calomel electrode (SCE) reference electrodes were obtained from CH Instruments.

Abbreviations: ^{*t*}Bu—*tert*-butyl, Me—methyl, Ac—acetyl, Ph—phenyl, Bz—benzoyl, DCM dichloromethane, MeCN—acetonitrile, THF—tetrahydrofuran, Tf—trifluoromethansulfonyl, TFA—Trifluoroacetic acid, TBA—tetrabutylammonium, DMF—*N*,*N*-dimethylformamide.

Section 2. General Procedures for Electrochemically Driven syn-Dioxygenation of Olefins



An oven-dried, 10 mL two-neck glass tube was equipped with a magnetic stir bar, a rubber septum, a threaded Teflon cap fitted with electrical feed-throughs, a carbon felt anode (1.0 *0.5 cm²) (connected to the electrical feedthrough via a 9 cm in length, 2 mm in diameter graphite rod), and a platinum plate cathode (0.5 * 1.0 cm²). To this reaction vessel, TBABF4 (131.6 mg, 0.4 mmol) was added. The cell was sealed and flushed with nitrogen gas for 5 minutes, followed by the sequential addition via syringe of olefin substrate (0.2 mmol, 1.0 equiv) in 3 mL of DMF and trifluoroacetic acid (1.0 mmol, 77 μ L) and water (0.6 mmol, 11 μ L). A nitrogen-filled balloon was adapted through the septum to sustain a nitrogen atmosphere. Electrolysis was initiated at a constant voltage of 2.5 mA at room temperature (22 °C) for 12 h. The mixture was then diluted with ethyl acetate (60 mL) and then washed with water, brine, dried over anhydrous Na₂SO₄, and concentrated under reduced pressure. The residue was subjected to flash column chromatography on silica gel (eluted with hexanes/ethyl acetate) to yield the desired product.



Figure S1. Setup for Electrochemical syn-Dioxygenation of Olefins

Unsuccessful Substrates

Unactivated Simple Olefins



decomposed decomposed S.M. recovered 99% Scheme S1. Unsuccessful Substrate Scopes: Unactivated Simple Olefins



Unsuccessful Vinyl Heteroarene Scopes



Scheme S2. Unsuccessful Vinyl Heteroarene Scopes

Section 3. Quantum Chemical Simulations

Oxidation potential of **C** (Figure 2)

All DFT calculations were carried out using density functional theory⁵ with Gaussian 09 prog ram.⁶ Geometry optimizations to the stationary points were performed with B3LYP^{7–11} levels of theory and the 6-31G** basis set.¹²⁻¹⁴ The electronic energies of optimized sturctures were ree valuated by single point calculations of M06 functionals¹⁵ with 6-311++G** basis set.¹⁶ Thermo dynamic parameters including Gibbs free energies at 298.15 K were obtained by frequency cal culations. Vibrational frequencies were carried out at the same level of theory as the geometry optimizations. Then, solvation correction energies were calculated by using the optimized gas-phase geometries at the same level of theory as the single point calculations. Solvation of DM F (ϵ = 37.51) was considered by single point calculation with integral equation formalism PCM model.

We performed quantum chemical simulations to predict electrochemical property of the proposed intermediate **C** (Figure 2). Oxidation potential (*E*) could be estimated from the Gibbs free energy change during the reduction:

$$E(V \text{ vs. SCE}) = -\frac{\Delta G}{nF} - E(\text{SCE})$$

where ΔG is the Gibbs free energy difference between cation **C** and dication **D**, where **D** is for med by an electrochemical oxidation of **C**. n is the number of electrons involved (n=1 in this st udy), and F is Faraday constant of 23.06 kcal/mol. The calculated oxidation potential was refer enced to the absolute reduction potential of NHE— 4.43 V.^{17,18} For the calculations of oxidation potential, electronic energies were further correcte d by single point calculations of M06-D3 and M06-2X functionals.¹⁹ We chose an oxidation pot ential calculated from M06 functional, because the results from three different functions showe d less than 0.02 V of deviations.

Table S1. Cartesian Coodinates of DFT-optimized Stucture of C



C12NH17O

С	2.870356000000	-0.938036000000	-1.398648000000
С	1.799839000000	-0.115625000000	-1.089933000000
С	1.657646000000	0.438056000000	0.218521000000
С	2.651915000000	0.106265000000	1.188369000000
С	3.718034000000	-0.715579000000	0.866908000000
С	3.835820000000	-1.243149000000	-0.427268000000
Н	2.965335000000	-1.345072000000	-2.400259000000
Н	1.067980000000	0.109581000000	-1.859633000000
Н	2.566222000000	0.521035000000	2.188666000000
Н	4.466411000000	-0.948581000000	1.617373000000
Н	4.674303000000	-1.884241000000	-0.678901000000
С	0.590548000000	1.292824000000	0.595838000000
Н	0.560868000000	1.640125000000	1.626264000000
С	-1.051995000000	3.139420000000	0.097560000000
Н	-0.285352000000	3.912917000000	-0.008650000000
н	-1 892589000000	3 400972000000	-0 548428000000

Н	-1.386927000000	3.130501000000	1.138362000000
0	-1.748881000000	0.846408000000	-0.363860000000
С	-1.734873000000	-0.312577000000	0.180090000000
Н	-0.841782000000	-0.654866000000	0.703282000000
Ν	-2.77001000000	-1.114032000000	0.131867000000
С	-4.021812000000	-0.742930000000	-0.545179000000
Н	-3.927733000000	0.252875000000	-0.971673000000
Н	-4.228730000000	-1.468159000000	-1.336237000000
Н	-4.839124000000	-0.760867000000	0.180323000000
С	-2.726112000000	-2.439956000000	0.763965000000
Н	-1.754848000000	-2.600709000000	1.232777000000
Н	-3.509556000000	-2.507260000000	1.523003000000
Н	-2.893493000000	-3.209845000000	0.006518000000
С	-0.453272000000	1.807508000000	-0.296121000000
Н	-0.176429000000	1.781232000000	-1.349580000000

Table S2. Cartesian Coodinates of DFT-optimized Stucture of D



C12NH17O

С	-3.920388000000	1.121018000000	0.683402000000
С	-2.579546000000	1.011267000000	0.409372000000
С	-2.035122000000	-0.252658000000	-0.033573000000
С	-2.924767000000	-1.387292000000	-0.164452000000
С	-4.266012000000	-1.257683000000	0.116344000000
С	-4.761936000000	-0.008589000000	0.537456000000
Н	-4.341072000000	2.063437000000	1.016849000000
Н	-1.933070000000	1.873954000000	0.531324000000
Н	-2.519321000000	-2.339609000000	-0.493489000000
Н	-4.937412000000	-2.103522000000	0.016781000000
Н	-5.820597000000	0.092270000000	0.762292000000
С	-0.708970000000	-0.442054000000	-0.351866000000
Н	-0.397316000000	-1.438313000000	-0.665188000000
С	0.397567000000	1.308355000000	-1.757527000000
Н	-0.527179000000	1.873385000000	-1.895558000000
Н	1.234523000000	2.012010000000	-1.787801000000
Н	0.502354000000	0.595591000000	-2.578610000000
0	1.628680000000	-0.175025000000	-0.234789000000
С	2.671001000000	0.422781000000	0.293043000000
Н	2.565211000000	1.427840000000	0.699196000000
Ν	3.824517000000	-0.159874000000	0.355178000000
С	4.077138000000	-1.515298000000	-0.178392000000
Н	3.178705000000	-1.911490000000	-0.644017000000
Н	4.883697000000	-1.450652000000	-0.912217000000
Н	4.390386000000	-2.159406000000	0.646543000000
С	4.977772000000	0.520986000000	0.987903000000
Н	4.685095000000	1.507949000000	1.346349000000
Н	5.327286000000	-0.087081000000	1.824937000000
Н	5.775328000000	0.617881000000	0.248290000000
С	0.378293000000	0.586677000000	-0.400192000000
Н	0.302137000000	1.300010000000	0.427278000000

Table S3. Vibrational Frequencies of Optimized Geometries

The vibrational frequencies of optimized geometries are given in below (units: cm-1).

С		
C 16.5092 89.4316 141.3342 231.5333 283.3787 406.0083 459.0018 614.7477 689.2711 835.3893 926.1160 996.5558 1027.7776 1101.8325 1150.7645 1191.2381 1263.0436 1358.0273 1428.1401 1469.5943 1493.0423 1505.5942 1536.5815 1730.5587 3074.7412 3146.5509 3165.5927 3190 5200	$\begin{array}{c} 29.4367\\ 100.5585\\ 160.5697\\ 253.7423\\ 371.9902\\ 409.6778\\ 482.6692\\ 621.1488\\ 746.0455\\ 837.1480\\ 942.5387\\ 1011.0963\\ 1044.7560\\ 1125.7914\\ 1177.4477\\ 1209.3991\\ 1270.0378\\ 1369.6581\\ 1456.8214\\ 1476.5402\\ 1502.5380\\ 1508.1434\\ 1589.5363\\ 3063.7211\\ 3135.1215\\ 3146.6543\\ 3174.3395\\ 3192\ 4315 \end{array}$	51.5236 131.5983 190.8940 265.9705 401.0506 414.5405 534.9034 680.7811 790.6258 875.6887 983.5881 1022.4235 1070.5122 1135.5814 1179.6152 1247.9095 1337.8350 1378.3943 1458.2410 1485.1481 1503.3739 1526.7218 3068.2909 3145.8162 3175.5499 32055361
3206.7727	3212.8751	3223.0329
D 28.0620 79.1857 160.5034 228.5289 316.9544 409.4189 474.2384	31.6703 93.5247 166.5480 259.0968 341.0859 431.5030 517 3516	56.3303 102.5728 180.7970 275.9886 377.8783 433.5552 558.6833
605.1076 797.6734 839.5481 948.7817 1000.1036 1028.8098 1099.6098 1153.0059 1207.1140	636.9188 825.6086 885.9098 984.5515 1017.7494 1051.8083 1117.4051 1169.8935 1222.2174	648.8749 827.5683 927.4206 998.6774 1024.2644 1055.4946 1119.9743 1176.2748 1244.9268
1272.1769 1361.6264 1428.8875 1469.9673 1488.7754 1500.7443 1576.7384 1761.3301	1297.6814 1381.2718 1446.8583 1475.8629 1491.1160 1502.5885 1602.4902 3067.0071	1332.9551 1420.9901 1457.0437 1481.2393 1495.4720 1524.2877 1663.8219 3078.1595
3082.3241 3162.2019 3177.7064 3209.8027 3217.7599	3084.2074 3165.4571 3183.4308 3212.5289 3232.5649	3150.9894 3167.8684 3188.1334 3217.5708 3235.0889

Section 4. Procedures of Mechanistic Studies

- 1. Deuterium Labeling Experiments
- 1.1. Reaction under DMF- d_7 as a solvent

An identical procedure was followed as described in **Section 2** with the exception that the reaction was carried out in 3 mL of DMF- d_7 as a solvent. When the reaction was finished, the mixture was then diluted with ethyl acetate (60 mL) and then washed with water, brine, dried over anhydrous Na₂SO₄, and concentrated under reduced pressure. The ¹H NMR yield and the ratio of proton/deuterium (H/D) of desired product was determined by integration using an internal standard (1,2-dimethoxyethane).



Figure S2. 1H NMR of Deuterium Scrambling Experiment Using DMF-d7

1.2. Reaction under formic acid- d_2 as an acid additive

An identical procedure was followed as described in **Section 2** with the exception that the reaction was carried out with 5.0 equiv of formic acid- d_2 (1.0 mmol, 38 µL) instead of trifluoroacetic acid. When the reaction was finished, the mixture was then diluted with ethyl acetate (60 mL) and then washed with water, brine, dried over anhydrous Na₂SO₄, and concentrated under reduced pressure. The ¹H NMR yield and the ratio of proton/deuterium (H/D) of desired product was determined by integration using an internal standard (1,2-dimethoxyethane).



Figure S3. 1H NMR of Deuterium Scrambling Experiment Using Formic acid-d₂

2. ¹⁸O-Oxygen labeling experiments

	CF₃COOŀ	<i>Undivided Cell</i> I (10 equiv), H ₂ ¹⁸ O (10 equiv)	\land	
36	<pre>✓ C(+)/Pt(-), U_{cell} = 2.5 V TBABF₄ (0.13 M), DMF (3 mL), N₂ 22 °C, 12 h</pre>		26 (syn- or anti-)	
calculated of I	ESI-MS	m/z [M+Na] ⁺	syn- 26	anti -26
unlabeled		229.0471	4.0%	3.2%
¹⁸ O-labeled		231.0514	26.7%	26.9%
doubly ¹⁸ O-la	beled	233.0556	69.3%	69.9%

abundance was calculated by integration of peak area

An identical procedure was followed as described in **Section 2** with the exception that the reaction was carried out with 10.0 equiv of $H_2^{18}O$ (97 % ¹⁸O-enriched, 2 mmol, 36 µL) instead of water. When the reaction was finished, the mixture was then diluted with ethyl acetate (60 mL) and then washed with water, brine, dried over anhydrous Na₂SO₄, and concentrated under reduced pressure. The residue was subjected to flash column chromatography on silica gel (eluted with hexanes/ethyl acetate) to separate each isomer (*syn*-**26** (40%) and *anti*-**26** (9%)). High-resolution mass spectrum (ESI) showed the following isotope pattern:











A solution of the syn- or anti-26 obtained in the ¹⁸O-labeling experiment (5.2 mg, 0.035 mmol, in 1 mL of THF) was stirred with NaOH (5.6 mg, 0.14 mmol, 4.0 equiv.) and H₂O (2.5 µL, 0.14 mmol, 4.0 equiv.) at room temperature for 6 hours. The resulting mixture was filtered through Na₂SO₄ and concentrated on a rotary evaporator. The desired product (syn-37 or anti-37) was obtained by column chromatography in quantative yields. High-resolution mass spectrum (ESI) of each isomers of 37 showed the following isotope pattern:







3. Voltammetric studies

General information: Cyclic voltammetry (CV) was conducted in a 10 mL glass vial fitted with a glassy carbon working electrode (3 mm in diameter), SCE reference electrode, and a platinum wire counter electrode. The solution of interest was sparged with nitrogen for 3-5 minutes before data collection.



Figure S8. Cyclic voltammogram of **1**, DMF, TFA and their mixtures in MeCN. Conditions: TBABF₄ (0.10 M), **1** (5 mM), DMF (5 mM), TFA (5 mM). Scan rate: 100 mV/s.

4. Evaluation of Diastereoselectivity by Control Experiments with Different Acids





1-[4-(tert-Butyl)phenyl]ethane-1,2-diyl diformate (2). Purified using silica gel chromatography to give 45.0 mg (90% yield) of **2** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.14 (s, 1H), 8.07 (s, 1H), 7.41 (d, J = 8.7 Hz, 2H), 7.32 (d, J = 8.2 Hz, 2H), 6.16 (dd, J = 7.3, 4.9 Hz, 1H), 4.49 – 4.37 (m, 2H), 1.31 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 160.5, 160.1, 152.3, 132.4, 126.7, 125.9, 72.7, 65.3, 34.8, 31.4; IR (Film): 2959, 2906, 2859, 1722, 1512, 1463, 1365, 1269, 1151, 1111 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₄H₁₈O₄⁺]: 250.1200, found 250.1205.



1-(*p***-Tolyl)ethane-1,2-diyl diformate (3)**. Purified using silica gel chromatography to give 20.8 mg (50% yield) of **3** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.16 (s, 1H), 8.09 (s, 1H), 7.31 (d, *J* = 8.2 Hz, 2H), 7.22 (d, *J* = 7.8 Hz, 2H), 6.16 (dd, *J* = 7.6, 4.6 Hz, 1H), 4.63 – 4.32 (m, 2H), 2.38 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.3, 159.9, 139.0, 132.4, 129.5, 126.7, 72.7, 65.1, 21.2. IR (Film): 2952, 2919, 2850, 1721, 1516, 1454, 1377, 1313, 1285, 1153, 1037, cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₁H₁₂O₄⁺]: 208.0736, found 208.0734.



1-Phenylethane-1,2-diyl diformate (4). Purified using silica gel chromatography to give 28.0 mg (72% yield) of **4** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.17 (s, 1H), 8.08 (s, 1H), 7.42 (d, *J* = 3.2 Hz, 5H), 6.19 (t, *J* = 6.5 Hz, 1H), 4.53 – 4.40 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.3, 159.8, 135.3, 129.0, 128.8, 126.7, 72.7, 65.1; IR (Film): 3055, 2984, 2937, 1726, 1606, 1496, 1454, 1374, 1265, 1156, 1027 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₀H₁₀O₄⁺]: 194.0579, found 194.0581.



4-[1,2-Bis(formyloxy)ethyl]phenyl acetate (5). Purified using silica gel chromatography to give 25.2 mg (50% yield) of **5** as a white solid; ¹H NMR (300 MHz, CDCl₃) δ 8.15 (s, 1H), 8.08 (s, 1H), 7.44 (d, *J* = 8.6 Hz, 2H), 7.15 (d, *J* = 8.6 Hz, 2H), 6.20 (t, *J* = 6.0 Hz, 1H), 4.45 (d, *J* = 6.0 Hz, 2H), 2.33 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 169.2, 160.3, 159.7, 151.1, 132.9, 128.0, 122.0, 72.1, 64.9, 21.1; IR (Film): 2925, 2852, 1753, 1719, 1608, 1509, 1428, 1370, 1182, 1157, 1042 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₂H₁₂O₆⁺]: 252.0634, found 252.0634.



Methyl 4-(1,2-bis(formyloxy)ethyl)benzoate (6). Purified using silica gel chromatography to give 24.2 mg (48% yield) of **6** as a white solid; ¹H NMR (300 MHz, CDCl₃) δ 8.18 (s, 1H), 8.14 – 8.01 (m, 3H), 7.50 (d, J = 8.3 Hz, 2H), 6.23 (dd, J = 7.1, 4.5 Hz, 1H), 4.56 – 4.42 (m, 2H), 3.95 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.4, 160.2, 159.6, 140.2, 130.8, 130.1, 126.7, 72.2, 64.8, 52.3; IR (Film): 2953, 2923, 2851, 1718, 1613, 1436, 1280, 1151, 1112, 1019 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₂H₁₂O₆⁺]: 252.0634, found 252.0634.



1-(4-Fluorophenyl)ethane-1,2-diyl diformate (7). Purified using silica gel chromatography to give 31.8 mg (75% yield) of **7** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.15 (s, 1H), 8.08 (s, 1H), 7.41 (dd, *J* = 8.6, 5.3 Hz, 2H), 7.10 (t, *J* = 8.6 Hz, 2H), 6.16 (t, *J* = 6.0 Hz, 1H), 4.65 – 4.32 (m, 2H); δ ¹³C NMR (75 MHz, CDCl₃) δ 163.0 (d, *J* = 246.8 Hz), 160.3, 159.8, 131.3 (d, *J* = 3.5 Hz), 128.7 (d, *J* = 8.4 Hz), 115.9 (d, *J* = 21.7 Hz), 72.1, 64.9; ¹⁹F NMR (282 MHz, CDCl₃) δ -112.14 (dtt, *J* = 10.7, 6.5, 2.6 Hz); IR (Film): 3032, 2923, 2850, 1718, 1495, 1454, 1374, 1333, 1266, 1155, 1026 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₀H₉FO₄⁺]: 212.0485, found 212.0487.



1-(4-Chlorophenyl)ethane-1,2-diyl diformate (8). Purified using silica gel chromatography to give 27.4 mg (60% yield) of **8** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.12 (s, 1H), 8.05 (s, 1H), 7.41 – 7.29 (m, 4H), 6.12 (t, *J* = 5.9 Hz, 1H), 4.42 (d, *J* = 6.0 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.2, 159.7, 135.1, 133.9, 129.1, 128.2, 72.0, 64.8; IR (Film): 2922, 2851, 1720, 1600, 1493, 1376, 1345, 1149, 1092, 1014 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₀H₉ClO₄⁺]: 228.0189, found 228.0188.

1-(4-Bromophenyl)ethane-1,2-diyl diformate (9). Purified using silica gel chromatography to give 60% yield of **9**. The yield was determined by ¹H NMR using 1,2-dimethoxyethane as an internal standard for this case because of an inseparable impurity; ¹H NMR (300 MHz, CDCl₃) δ 8.15 (s, 1H), 8.07 (s, 1H), 7.55 (d, J = 8.5 Hz, 2H), 7.30 (d, J = 8.0 Hz, 2H), 6.13 (t, J = 5.9 Hz, 1H), 4.52 – 4.29 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.2, 159.7, 134.4, 132.0, 128.4, 123.1, 72.0, 64.7; IR (Film): 2924, 2850, 1720, 1593, 1489, 1407, 1376, 1345, 1152, 1073 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₀H₉BrO₄⁺]: 271.9684, found 271.9687.



1-(o-Tolyl)ethane-1,2-diyl diformate (10). Purified using silica gel chromatography to give 29.1 mg (70% yield) of **10** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.16 (s, 1H), 8.10 (s, 1H), 7.44 – 7.38 (m, 1H), 7.30 – 7.19 (m, 3H), 6.41 (dd, *J* = 7.3, 4.8 Hz, 1H), 4.60 – 4.14 (m, 2H), 2.49 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.4, 159.9, 135.5, 133.7, 130.7, 128.8, 126.4, 126.2, 69.8, 64.5, 19.1; IR (Film): 2923, 2852, 1719, 1491, 1462, 1376, 1358, 1287, 1264, 1148 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₁H₁₂O₄⁺]: 208.0736, found 208.0737.



1-(Naphthalen-2-yl)ethane-1,2-diyl diformate (11). Purified using silica gel chromatography to give 31.7 mg (65% yield) of **11** as a white solid; ¹H NMR (300 MHz, CDCl₃) δ 8.20 (s, 1H), 8.08 (s, 1H), 8.00 – 7.73 (m, 4H), 7.61 – 7.39 (m, 3H), 6.34 (dd, *J* = 7.2, 4.8 Hz, 1H), 4.64 – 4.44 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.3, 159.9, 133.4, 133.0, 132.7, 128.8, 128.1, 127.7, 126.7, 126.6, 126.4, 123.9, 72.9, 65.1; IR (Film): 3055, 2927, 2851, 1717, 1495, 1447, 1376, 1286, 1152, 1070 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₄H₁₂O₄⁺]: 244.0736, found 244.0734.



2-Phenylpropane-1,2-diyl diformate (12). Purified using silica gel chromatography to give 27.0 mg (65% yield) of **12** as a white solid; ¹H NMR (300 MHz, CDCl₃) δ 8.10 (s, 2H), 7.54 – 7.33 (m, 5H), 4.70 – 4.26 (m, 2H), 1.99 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.2, 159.5, 139.9, 128.7, 128.3, 125.1, 82.3, 69.0, 22.1; IR (Film): 2979, 2922, 2851, 1717, 1495, 1447, 1376, 1286, 1152, 1070 cm⁻¹; HRMS (EI) exact mass calculated for [M-H⁺, C₁₁H₁₂O₄⁺]: 207.0652, found 207.0659.



1-(Dibenzo[b,d]furan-2-yl)ethane-1,2-diyl diformate (13). Purified using silica gel chromatography to give 31.2 mg (55% yield) of **13** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.21 (s, 1H), 8.12 (s, 1H), 8.07 – 7.92 (m, 2H), 7.66 – 7.58 (m, 2H), 7.52 (ddd, J = 8.3, 4.4, 2.4 Hz, 2H), 7.40 (td, J = 7.5, 1.1 Hz, 1H), 6.36 (dd, J = 8.1, 4.1 Hz, 1H), 4.68 – 4.45 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.3, 159.9, 156.6, 156.3, 130.0, 127.7, 125.8, 124.7, 123.6, 123.0, 120.8, 119.4, 112.1, 111.8, 72.8, 65.2; IR (Film): 2939, 1719, 1603, 1589, 1481, 1450, 1377, 1339, 1322, 1245, 1153 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₆H₁₂O₅⁺]: 284.0685, found 284.0687.



1-(9-Tosyl-9*H***-carbazol-3-yl)ethane-1,2-diyl diformate (14).** Purified using silica gel chromatography to give 48.1 mg (55% yield) of **14** as a pale yellow oil; ¹H NMR (300 MHz, CDCl₃) δ 8.33 (dd, J = 8.5, 5.0 Hz, 2H), 8.18 (s, 1H), 8.08 (s, 1H), 7.97 – 7.87 (m, 2H), 7.70 (d, J = 8.4 Hz, 2H), 7.59 – 7.47 (m, 2H), 7.38 (td, J = 7.5, 1.0 Hz, 1H), 7.11 (d, J = 8.1 Hz, 2H), 6.31 (dd, J = 7.7, 4.4 Hz, 1H), 4.63 – 4.41 (m, 2H), 2.26 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.4, 159.9, 145.2, 138.7, 138.6, 134.9, 131.1, 129.8, 127.9, 126.7, 126.5, 125.9, 125.7, 124.1, 120.2, 118.6, 115.4, 115.1, 72.7, 65.2, 21.6; IR (Film): 2946, 1724, 1598, 1485, 1446, 1369, 1231, 1172, 1153, 1090 cm⁻¹; HRMS (ESI) exact mass calculated for [M+Na⁺, C23H19NNaO6S⁺]: 460.0825, found 460.0855.



1-[4-(*tert***-butyl)phenyl]ethane-1,2-diyl diacetate (15).** Purified using silica gel chromatography to give 17.2 mg (31% yield) of **15** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 7.38 (d, *J* = 8.5 Hz, 2H), 7.29 (d, *J* = 8.4 Hz, 2H), 6.01 (dd, *J* = 7.8, 4.2 Hz, 1H), 4.41 – 4.19 (m, 2H), 2.11 (s, 3H), 2.06 (s, 3H), 1.31 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 170.7, 170.1, 151.6, 133.4, 126.5, 125.5, 73.1, 66.1, 34.6, 31.2, 21.1, 20.8; IR (Film): 2959, 2868, 1741, 1512, 1462, 1435, 1366, 1219, 1107, 1042, 1018 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₆H₂₂O₄⁺]: 278.1518, found 278.1518.



1-((8R,9S,13S,14S)-13-methyl-17-oxo-7,8,9,11,12,13,14,15,16,17-decahydro-6H-

cyclopenta[a]phenanthren-3-yl)ethane-1,2-diyl diformate (16). Isolated as a 1:1 mixture of diastereomers. Purified using silica gel chromatography to give 46.6 mg (63% yield) of **16** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.16 (s, 1H), 8.09 (s, 1H), 7.34 (d, *J* = 8.1 Hz, 1H), 7.19 (d, *J* = 8.0 Hz, 1H), 7.14 (s, 1H), 6.13 (dd, *J* = 7.5, 4.7 Hz, 1H), 4.54 – 4.37 (m, 2H), 2.97 – 2.93 (m, 2H), 2.62 – 2.41 (m, 2H), 2.32 (td, *J* = 10.4, 4.0 Hz, 1H), 2.25 – 1.93 (m, 4H), 1.80 – 1.31 (m, 6H), 0.93 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 220.7, 160.4, 159.9, 140.8, 137.1, 132.8, 127.4 (2C), 125.9, 124.1 (2C), 72.6 (2C), 65.1, 50.4, 47.9, 44.3, 37.9, 35.8, 31.5, 29.3 (2C), 26.3, 25.6, 21.5, 13.8; IR (Film): x cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₂₂H₂₆O₅⁺]: 370.1780, found 370.1780.



Ethyl 2-benzamido-3-{4-[1,2-bis(formyloxy)ethyl]phenyl}propanoate (17). Purified using silica gel chromatography to give 53.7 mg (65% yield) of **17** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.12 (s, 1H), 8.04 (s, 1H), 7.73 (d, *J* = 8.3 Hz, 2H), 7.51 (t, *J* = 6.6 Hz, 1H), 7.46 – 7.37 (m, 2H), 7.31 (d, *J* = 8.3 Hz, 2H), 7.18 (d, *J* = 8.2 Hz, 2H), 6.64 (d, *J* = 7.6 Hz, 1H), 6.14 (t, *J* = 6.0 Hz, 1H), 5.12 – 4.99 (m, 1H), 4.41 (d, *J* = 6.0 Hz, 2H), 4.21 (q, *J* = 7.1 Hz, 2H), 3.38 – 3.15 (m, 2H), 1.26 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.4, 166.8, 160.3, 159.8, 137.1, 134.1, 133.8, 131.8, 129.9, 128.6, 127.0, 126.9, 72.4, 65.0, 61.7, 53.4, 37.6, 14.1; IR (Film): 3341, 2979, 2933, 1723, 1646, 1525, 1487, 1446, 1374, 1153 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₂₂H₂₃NO₇⁺]: 413.1475, found 413.1476.



1-Phenoxyethane-1,2-diyl diformate (18). Purified using silica gel chromatography to give 15.5 mg (37% yield) of **18** as a colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 8.10 (d, *J* = 4.0 Hz, 2H), 7.38 – 7.29 (m, 2H), 7.10 (t, *J* = 7.3 Hz, 1H), 7.01 (d, *J* = 7.6 Hz, 3H), 6.78 (t, *J* = 5.0 Hz, 1H), 4.50 (ddd, *J* = 58.8, 11.5, 5.0 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.0, 159.3, 155.3, 129.8, 123.8, 116.9, 91.4, 62.5; IR (Film): 2920, 2850, 1723, 1495, 1449, 1377, 1151, 1096, 1062, 1030 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₀H₁₀O₅⁺]: 210.0528, found 210.0528.



1-Phenylpropane-1,2-diyl diformate (19). ¹H NMR analysis [integration of formyl resonances at 8.11 (major) and 8.17 (minor) ppm] or the unpurified reaction indicated a 10:1 d.r.; **18** was obtained as a major diastereomer from the reaction using *trans-***34** (*trans-*β-methylstyrene) (Figure 4C) as a starting material. Purified using silica gel chromatography to give 26.6 mg (64% yield) of **18** as a colorless oil. Relative configuration was determined after hydrolysis of **18** using the same method provided in page S10 and compared with the spectral data known in the precedent literature²⁰; ¹H NMR (300 MHz, CDCl₃) δ 8.11 (s, 1H), 8.06 (s, 1H), 7.42 – 7.33 (m, 5H), 5.88 (d, *J* = 7.5 Hz, 1H), 5.43 (p, *J* = 6.6 Hz, 1H), 1.15 (d, *J* = 6.6 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.2, 159.8, 135.8, 129.0, 128.8, 127.4, 76.8, 71.1, 16.6.



1-Phenylpropane-1,2-diyl diformate (19'). ¹H NMR analysis [integration of formyl resonances at 8.17 (major) and 8.11 (minor) ppm] or the unpurified reaction indicated a 3:1 d.r.; **18'** was obtained as a major diastereomer from the reaction using *cis*-**34** (*cis*-β-methylstyrene) (Figure 4C) as a starting material. Purified using silica gel chromatography to give 20.4 mg (49% yield) of **18'** as a colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 8.20 (s, 1H), 8.04 (s, 1H), 7.41 – 7.37 (m, 5H), 6.08 (d, *J* = 5.2 Hz, 1H), 5.54 – 5.28 (m, 1H), 1.27 (d, *J* = 6.5 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.3, 159.9, 135.6, 128.8, 128.7, 127.2, 75.8, 71.4, 14.8; Properties for mixture of diastereomers: IR (Film): 2987, 2934, 1718, 1495, 1454, 1381, 1263, 1155, 1079, 1051 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₁H₁₂O₄⁺]: 208.0736, found 208.0736.



1,2-Diphenylethane-1,2-diyl diformate (20). Purified using silica gel chromatography to give 9.2 mg (17% yield) of **20** was obtained as a sole diastereomer as a white solid; ¹H NMR (300 MHz, CDCl₃) δ 8.13 (s, 2H), 7.25 – 7.10 (m, 10H), 6.19 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 159.7, 135.1, 128.8, 128.4, 127.6, 76.6; IR (Film): 3034, 2932, 1720, 1494, 1455, 1329, 1282, 1254, 1145, 1076 cm⁻¹; HRMS (ESI) exact mass calculated for [M+Na⁺, C₁₆H₁₄O₄Na⁺]: 293.0784, found 293.0785.



3-Chloro-1-phenylpropane-1,2-diyl diformate (21). ¹H NMR analysis [integration of formyl resonances at 8.09 (major) and 8.03 (minor) ppm] or the unpurified reaction indicated a 10:1 d.r. Purified using silica gel chromatography to afford the title as a colorless oil (24.2 mg, 50% yield).

Major diastereomer: ¹H NMR (300 MHz, CDCl₃) δ 8.13 (s, 1H), 8.10 (s, 1H), 7.46 – 7.35 (m, 8H), 6.18 (d, *J* = 7.4 Hz, 1H), 5.56 (dt, *J* = 8.3, 4.6 Hz, 1H), 3.64 (dd, *J* = 12.3, 4.0 Hz, 1H), 3.34 (dd, *J* = 12.3, 4.8 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 159.7, 159.5, 134.8, 129.5, 129.1, 127.4, 73.7, 73.0, 42.3; Properties for mixture of diastereomers: IR (Film): 2925, 2852, 1722, 1587, 1495, 1455, 1432, 1374, 1330, 1143, 1079 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₁H₁₁ClO₄⁺]: 242.0346, found 242.0348.



2,3-bis(formyloxy)-3-phenylpropyl acetate (22). ¹H NMR analysis [integration of benzyl resonances at 6.10 (major) and 5.90 (minor) ppm] or the unpurified reaction indicated a 10:1 d.r. Purified using silica gel chromatography to afford the title as a colorless oil (31.9 mg, 56% yield).

Major diastereomer: ¹H NMR (600 MHz, cdcl₃) δ 8.09 (s, 2H), 7.43 – 7.32 (m, 5H), 6.10 (d, J = 7.7 Hz, 1H), 5.65 – 5.53 (m, 1H), 4.27 (dd, J = 12.4, 3.3 Hz, 1H), 3.83 (dd, J = 12.3, 5.6 Hz, 1H), 2.06 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 170.3, 159.8, 159.5, 134.8, 129.4, 129.0, 127.2, 73.4, 71.6, 61.9, 20.6; Properties for mixture of diastereomers: IR (Film): 2926, 2853, 1722, 1666, 1496, 1454, 1370, 1229, 1151, 1047 cm⁻¹; HRMS (ESI) exact mass calculated for [M+Na⁺, C₁₃H₁₄O₆Na⁺]: 289.0683, found 289.0732.



2,3-Bis(formyloxy)-3-phenylpropyl benzoate (23). ¹H NMR analysis [integration of formyl resonances at 8.13 (major) and 8.17 (minor) ppm] or the unpurified reaction indicated a 16:1 d.r. Purified using silica gel chromatography to afford the title as a colorless oil (38.7 mg, 59% yield);

Major diastereomer: ¹H NMR (300 MHz, CDCl₃) δ 8.12 (d, J = 4.4 Hz, 2H), 8.06 – 7.96 (m, 2H), 7.63 – 7.54 (m, 1H), 7.51 – 7.33 (m, 7H), 6.20 (d, J = 8.0 Hz, 1H), 5.82 – 5.69 (m, 1H), 4.46 (dd, J = 12.3, 3.3 Hz, 1H), 4.12 (dd, J = 12.3, 5.6 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 165.8, 159.8, 159.5, 134.8, 133.4, 129.7, 129.4, 129.2, 129.0, 128.5, 127.3, 73.6, 71.7, 62.5; Properties for mixture of diastereomers: IR (Film): 2935, 1720, 1601, 1584, 1494, 1452, 1315, 1217, 1150, 1115, 1071 cm⁻¹; HRMS (ESI) exact mass calculated for [M+Na⁺, C₁₈H₁₆O₆Na⁺]: 351.0839, found 351.0895.



3-[(4-Methylphenyl)sulfonamide]-1-phenylpropane-1,2-diyl diformate (24). ¹H NMR analysis [integration of aromatic resonances at 7.66 (major) and 7.81 (minor) ppm] or the unpurified reaction indicated a 7:1 d.r. Purified using silica gel chromatography to afford the title as a colorless oil (32.4 mg , 43% yield).

Major diastereomer: ¹H NMR (300 MHz, CDCl₃) δ 8.06 (s, 1H), 7.93 (s, 1H), 7.66 (d, *J* = 8.4 Hz, 2H), 7.37 – 7.19 (m, 8H), 6.02 (d, *J* = 6.9 Hz, 1H), 5.36 (q, *J* = 5.8 Hz, 1H), 5.19 (t, *J* = 6.6 Hz, 1H), 3.05 (t, *J* = 5.5 Hz, 2H), 2.42 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.8, 159.6, 143.8, 136.3, 134.7, 129.8, 129.2, 128.9, 127.2, 127.1, 73.5, 72.6, 42.7, 21.5; Properties for mixture of diastereomers: IR (Film): 3036, 2928, 1724, 1597, 1494, 1432, 1376, 1330, 1267, 1155, 1093 cm⁻¹ HRMS (ESI) exact mass calculated for [M+Na⁺, C₁₈H₁₆O₆Na⁺]: 400.0825, found 400.0860.



5-Oxo-10,11-dihydro-5H-dibenzo[*a,d*][7]annulene-10,11-diyl diformate (25). ¹H NMR analysis [integration of benzyl resonances at 6.57 (major) and 6.47 (minor) ppm] or the unpurified reaction indicated a 8:1 d.r. Purified using silica gel chromatography to afford the title as a white solid (35.5 mg, 60% yield).

Major diastereomer: ¹H NMR (300 MHz, CDCl₃) δ 8.16 (s, 2H), 8.01 (dd, *J* = 8.1, 1.6 Hz, 2H), 7.63 – 7.55 (m, 2H), 7.50 (t, *J* = 6.9 Hz, 4H), 6.57 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 193.5, 159.8, 137.5, 132.8 (2C), 130.4, 129.3, 128.5, 73.9; Properties for mixture of diastereomers: IR (Film): 2929, 1723, 1655, 1597, 1451, 1344, 1294, 1240, 1143, 1076 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₇H₁₂O₅⁺]: 296.0685, found 296.0683.



2,3-Dihydro-1*H***-indene-1,2-diyl diformate (26).** ¹H NMR analysis [integration of benzyl resonances at 6.36 (major) and 6.43 (minor) ppm] or the unpurified reaction indicated a 4:1 d.r. Purified using silica gel chromatography to afford the title as a white solid (22.4 mg, 49% yield).

Major diastereomer: ¹H NMR (300 MHz, CDCl₃) δ 8.14 (s, 1H), 8.09 (s, 1H), 7.48 – 7.39 (m, 1H), 7.43 – 7.31 (m, 1H), 7.35 – 7.26 (m, 2H), 6.36 (d, *J* = 5.2 Hz, 1H), 5.66 (q, *J* = 5.5 Hz, 1H), 3.38 – 3.08 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.2 (2C), 130.1, 127.7, 126.0, 125.1, 74.6, 72.8, 35.8; Properties for mixture of diastereomers: IR (Film): 2924, 2850, 1720, 1593, 1489, 1407, 1376, 1345, 1152, 1073 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₁H₁₀O₄⁺]: 206.0579, found 206.0582.



2,3-Bis(formyloxy)-3-phenylpropyl cinnamate (27). Purified using silica gel chromatography to afford the title as a white solid (62.3 mg, 88% yield); ¹H NMR (300 MHz, CDCl₃) δ 8.14 (d, *J* = 4.3 Hz, 2H), 7.72 (d, *J* = 16.0 Hz, 1H), 7.61 – 7.52 (m, 2H), 7.49 – 7.37 (m, 8H), 6.47 (d, *J* = 16.0 Hz, 1H), 6.21 (d, *J* = 7.9 Hz, 1H), 5.77 – 5.65 (m, 1H), 4.39 (dd, *J* = 12.3, 3.3 Hz, 1H), 4.04 (dd, *J* = 12.3, 5.5 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 166.1, 159.8, 159.5, 146.0, 134.9, 134.0, 130.6, 129.4, 129.0, 128.9, 128.2, 127.3, 116.8, 73.5, 71.8, 62.0; IR (Film): 2942, 1718, 1635, 1578, 1495, 1451, 1310, 1270, 1202, 1148 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₂₀H₁₈O₆⁺]: 354.1103, found 354.1102.



2-Methyl-1-phenylpropane-1,2-diyl diformate (28). Purified using silica gel chromatography to afford the title as a white solid (19.1 mg, 43% yield); ¹H NMR (300 MHz, CDCl₃) δ 8.16 (d, *J* = 1.0 Hz, 1H), 8.01 (s, 1H), 7.45 – 7.30 (m, 5H), 6.08 (s, 1H), 1.54 (d, *J* = 2.7 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 160.07, 159.70, 135.51, 128.68, 128.25, 128.17, 83.34, 78.94, 22.97, 22.09; IR (Film): 2988, 2925, 2853, 1718, 1495, 1454, 1386, 1370, 1285, 1142 cm⁻¹; HRMS (ESI) exact mass calculated for [M+Na⁺, C₁₂H₁₄O₄Na⁺]: 245.0784, found 245.0824.



1,2-Diphenylpropane-1,2-diyl diformate (29). ¹H NMR analysis [integration of formyl resonances at 8.08 (major) and 8.02 (minor) ppm] or the unpurified reaction indicated a 10:1 d.r. Purified using silica gel chromatography to afford the title as a colorless oil (27.8 mg, 49% yield).

Major diastereomer: ¹H NMR (600 MHz, CDCl₃) δ 8.14 (s, 1H), 8.08 (s, 1H), 7.31 – 7.26 (m, 4H), 7.20 – 7.11 (m, 4H), 6.91 (d, *J* = 7.6 Hz, 2H), 6.15 (s, 1H), 1.95 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.5 (2C), 139.4, 134.3, 128.5, 128.3 (2C), 128.2, 127.7, 126.2, 85.4, 79.9, 19.0; Properties for mixture of diastereomers: IR (Film): 2929, 2853, 1720, 1606, 1512, 1452, 1422, 1376, 1347, 1153 cm⁻¹; HRMS (ESI) exact mass calculated for [M+Na⁺, C₁₇H₁₆O₄Na⁺]: 307.0941, found 307.0979.



2,5-Diphenylpent-2-ene-1,5-diyl diformate (31). Purified using silica gel chromatography to afford **30** as a mixture of diastereomers (white solid, 30.4 mg, 49% yield).

Major diastereomer: ¹H NMR (300 MHz, CDCl₃) δ 8.15 (s, 1H), 8.02 (s, 1H), 7.45 – 7.36 (m, 4H), 7.40 – 7.29 (m, 6H), 6.04 (t, *J* = 6.7 Hz, 1H), 5.97 (t, *J* = 7.5 Hz, 1H), 5.07 - 4.98 (m, 2H), 3.22 – 2.81 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 160.8, 160.2, 140.0, 139.0, 137.1, 128.8, 128.7, 128.5 (2C), 127.7, 126.5, 126.3, 75.0, 60.3, 35.6; Properties for mixture of diastereomers: IR (Film): 3032, 2922, 2848, 1718, 1494, 1454, 1374, 1333, 1158, 1026 cm⁻¹; HRMS (EI) exact mass calculated for [M⁺, C₁₉H₁₈O₄⁺]: 310.1205, found 310.1207.



(*E*)-1-(Formyloxy)-4-phenylbut-3-en-2-yl 2,2,2-trifluoroacetate (33). Purified using silica gel chromatography to afford **32** as a colorless oil (25.9 mg, 45% yield); ¹H NMR (300 MHz, CDCl₃) δ 8.02 (s, 1H), 7.42 – 7.19 (m, 5H), 6.77 (d, *J* = 15.8 Hz, 1H), 6.07 (dd, *J* = 15.9, 7.9 Hz, 1H), 5.77 (td, *J* = 7.8, 3.5 Hz, 1H), 4.45 (ddd, *J* = 12.2, 3.5, 0.9 Hz, 1H), 4.31 (ddd, *J* = 12.2, 7.7, 0.7 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 160.1, 156.6 (q, *J* = 32.3 Hz), 137.5, 134.9, 129.1, 128.8, 127.0, 119.6, 114.4 (q, *J* = 285.7 Hz), 76.5, 63.5; ¹⁹F NMR (282 MHz, CDCl₃) δ -75.0; IR (Film): 2921, 2851, 1786, 1728, 1494, 1451, 1378, 1222, 1148, 1026 cm⁻¹; HRMS (EI) exact mass calculated for [C₁₃H₁₁F₃O₄+]: 288.0609, found 288.0609.



(*E*)-4-Phenylbut-3-ene-1,2-diyl diformate (34). Colorless oil; ¹H NMR (300 MHz, CDCl₃) $\delta 8.14$ (s, 1H), 8.09 (s, 1H), 7.44 – 7.28 (m, 5H), 6.77 (d, *J* = 16.2 Hz, 1H), 6.14 (dd, *J* = 16.0, 7.3 Hz, 1H), 5.83 (td, *J* = 7.2, 3.7 Hz, 1H), 4.45 (dd, *J* = 11.9, 3.7 Hz, 1H), 4.34 (dd, *J* = 11.9, 7.1 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 160.4, 159.9, 135.5, 135.4, 128.7, 128.6, 126.8, 121.8, 71.6, 64.1; IR (Film): 2924, 2849, 1787, 1721, 1493, 1450, 1376, 1352, 1154, 1029 cm⁻¹; HRMS (EI) exact mass calculated for [C₁₂H₁₂O₄+]: 220.0736, found 220.0736.



2,3-Dihydro-1H-indene-1,2-diol (*syn-***37**). Relative configuration of **36** was determined by comparing with the spectral data known in the precedent literature.²⁰ White Solid. ¹H NMR (300 MHz, CDCl₃) δ 7.50 – 7.40 (m, 1H), 7.34 – 7.14 (m, 4H), 4.97 (d, *J* = 4.9 Hz, 1H), 4.46 (q, *J* = 5.5 Hz, 1H), 3.11 (dd, *J* = 16.3, 5.7 Hz, 1H), 2.94 (dd, *J* = 16.2, 3.8 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 141.9, 140.1, 128.8, 127.1, 125.3, 125.0, 75.9, 73.4, 38.5; HRMS (EI) exact mass calculated for [C₉H₁₀O₂⁺]: 150.0681, found 150.0679.



2,3-Dihydro-1H-indene-1,2-diol (*anti-***37**). White Solid. ¹H NMR (300 MHz, MeOD) δ 7.39 – 7.32 (m, 1H), 7.28 – 7.15 (m, 4H), 4.88 (d, *J* = 4.4 Hz, 1H), 4.26 (td, *J* = 6.9, 5.3 Hz, 1H), 3.24 (dd, *J* = 15.7, 7.1 Hz, 1H), 2.74 (dd, *J* = 15.7, 6.7 Hz, 1H); ¹³C NMR (75 MHz, MeOD) δ 142.5, 139.3, 127.9, 126.5, 124.4, 124.0, 81.1, 80.2, 37.6; HRMS (EI) exact mass calculated for [C₉H₁₀O₂+]: 150.0681, found 150.0680.















































References

- [1] D. S. Rao, T. R. Reddy, A. Gurawa, M. Kumar and S. Kashyap, Org. Lett., 2019, 21, 9990-9994.
- [2] Y. Arai, R. Tomita, G. Ando, T. Koike and M. Akita, *Chem.–Eur. J.*, 2016, 22, 1262-1265.
- [3] Y. Liang and X. Zhao, ACS Catal., 2019, 9, 6896-6902.
- [4] S. Bazzi, E. Schulz and M. Mellah, Org. Lett., 2019, 21, 10033-10037.
- [5] R. G. Parr and Y. Weitao, Density-Functional Theory of Atoms and Molecules, Oxford University Press, 1994.
- [6] Gaussian 09, Revision D01 (Gaussian Inc., Wallingford, CT, 2009).
- [7] J. C. Slater, Quantum Theory of Molecules and Solids, Vol.4: The Self-Consistent Field for Molecules and Solids, McGraw-Hill, 1974.
- [8) S. H. Vosko, L. Wilk and M. Nusair, Can. J. Phys., 1980, 58, 1200.
- [9] A. D. Becke, *Phys. Rev. A*, 1988, **38**, 3098-3100.
- [10] C. Lee, W. Yang and R. G. Parr, Phys. Rev. B, 1988, 37, 785-789.
- [11] A. D. Becke, J. Chem. Phys., 1993, 98, 5648-5652.
- [12] R. Ditchfield, W. J. Hehre and J. A. Pople, J. Chem. Phys., 1971, 54, 724-728.
- [13] P. C. Hariharan and J. A. Pople, Theor. Chim. Acta., 1973, 28, 213-222.
- [14] W. J. Hehre, R. Ditchfield and J. A. Pople, J. Chem. Phys., 1972, 56, 2257-2261.
- [15] Y. Zhao and D. G. Truhlar, Theor. Chem. Acc., 2008, 120, 215-241.
- [16] T. Clark, J. Chandrasekhar, G. W. Spitznagel and P. V. R. Schleyer, J. Comput. Chem., 1983, 4, 294-301.
- [17] D. G. Truhlar, C. J. Cramer, A. Lewis and J. A. Bumpus, J. Chem. Educ., 2004, 81, 596.
- [18] H. Reiss and A. Heller, J. Phys. Chem., 1985, 89, 4207-4213.
- [19] Y. Zhao and D. G. Truhlar, Theor. Chem. Acc., 2008, 120, 215-241.
- [20] J. C. Griffith, K. M. Jones, S. Picon, M. J. Rawling, B. M. Kariuki, M. Campbell and N. C. O. Tomkinson, J. Am. Chem. Soc. 2010, 132, 14409-14411.