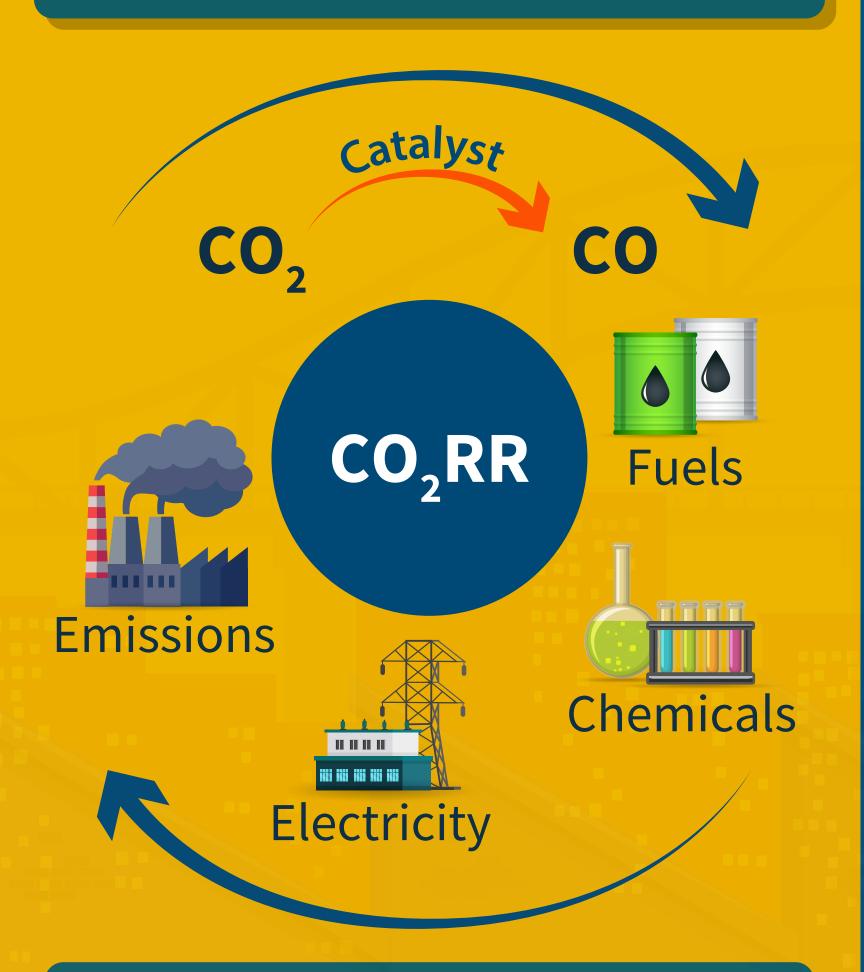
Novel Quasi-Square-Shaped Cadmium Hydroxide Nanocatalyst to Improve Electrochemical Reduction of Carbon Dioxide to Carbon Monoxide

Chemical Science



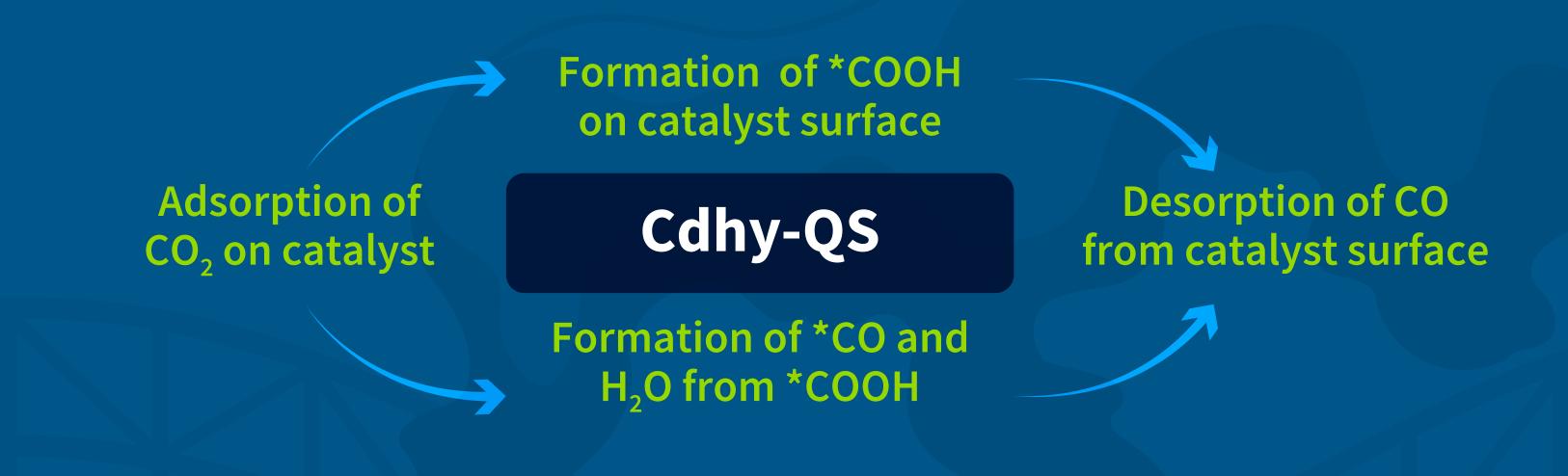
Electrochemical reduction of carbon dioxide (CO₂) using renewable electricity can reduce CO₂ emission, producing carbon monoxide (CO) for commercialisation as a feedstock for chemicals and fuels

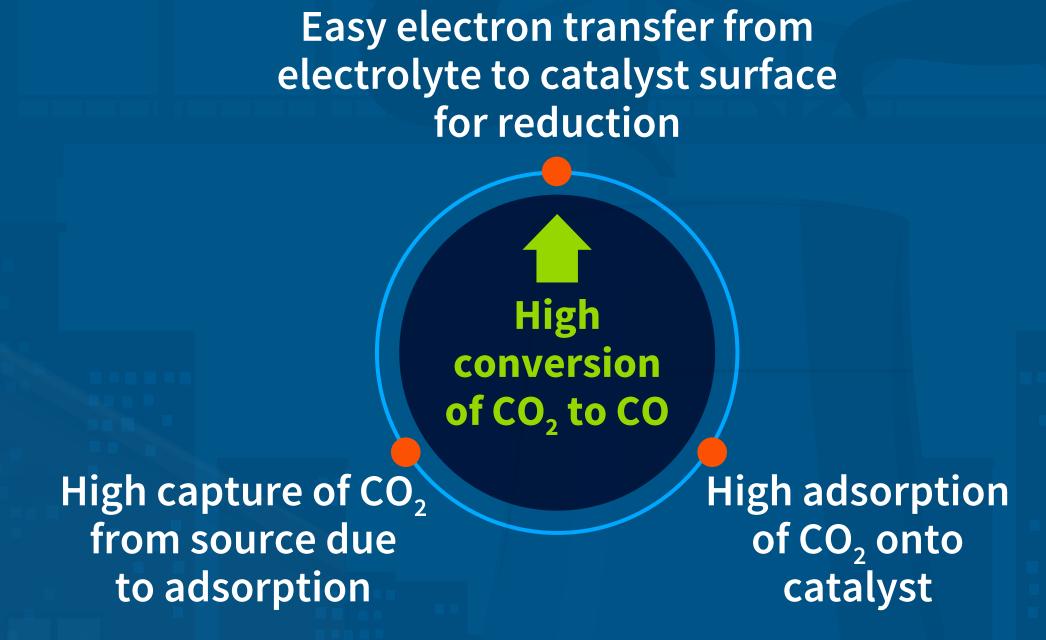


But it has disadvantages:

- Low capture rate of CO₂ at source
- Poor catalytic activity and selectivity
- Low conversion to CO

The efficient reduction of CO₂ to CO is proposed by combining a novel quasi-square-shaped cadmium hydroxide (Cdhy-QS) nanocatalyst with an adsorption-electrolysis device





- ✓ Faradaic efficiency of CO up to 94.4%
- ✓ Works for CO₂ concentrations as low as 4.9%
- Obtains CO product concentrations up to 57.0%

The combined action of Cdhy-QS nanocatalyst and an adsorption-electrolysis device system can generate high concentrations of CO at low CO, concentrations

