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# Supporting Information: Alchemical Absolute Protein-Ligand Binding Free Energies for Drug Design

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Figure S1: Scheme for non-equilibrium free energy calculations. Firstly, molecular dynamics simulations are performed to obtain equilibrium ensembles of the protein's holo and apo states. Subsequently, conformations from the equilibrium ensembles are extracted and rapid out-of-equilibrium simulations are started driving the system from holo to apo state and vice versa. Work values from these transitions are recorded and Crooks Fluctuation Theorem<sup>[5]</sup> is applied to relate the work distributions to the free energy difference between the two end-states. This protocol is particularly suited for the absolute ligand-protein binding free energy calculations. The two states, apo and holo, can be initialized independently with the crystallographically resolved structures (in case they are available) and the non-equilibrium transitions, in turn, allow connecting the two states.



Figure S2: Accuracy of relative binding free energies recalculated from absolute ones compared to experimental measurements and broken down by system. Average unsigned error (AUE) is in kJ/mol. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.



Figure S3: Calculated binding  $\Delta G$  estimates plotted against the experimental values with data separated by system. Representing the protein in decoupled ligand systems with holo (left) and apo (right) structures can lead to different accuracies and systematic shifts in the predicted  $\Delta G$  values. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.



Figure S4: The experimentally resolved apo (1wfc) and holo (3fly) structures differ in a substantial loop motion close to the ligand's binding site. The inset illustrates that in the simulations without the ligand that are started either from the 1wfc or from 3fly structure, the loop motion is sufficiently sampled.



Figure S5: Calculated binding  $\Delta G$  estimates for the p38 $\alpha$  system plotted against experimental values. Left panel: the simulations in the apo state were started from the experimental holo (3fly) structure after removing the ligand. Right panel: the structural ensemble for the apo state that was used for the calculations in the left panel was filtered by retaining only those structures that are more similar to the experimentally resolved apo (1wfc) structure in terms of the loop motion at the binding site. Bottom panel: Histogram quantifying the loop motion in terms of the distance between the tip of the loop (C $\alpha$  of Tyr 35) and the centroid of the binding pocket. Filtering captures only the extended states of the loop. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.



Figure S6: Sampling effects on the trajectories and calculated ligand binding  $\Delta G$  for the tyk2 kinase. Left panel: projection of the simulated apo tyk2 trajectories on the two principal components with the largest eigenvalues. The short (10 ns) equilibrium ensembles initialized with the 4gih<sup>[11]</sup> structure with the ligand removed are shown in blue, 1  $\mu$ s simulations in orange and 10 ns equilibrations started from the end-states of the 1  $\mu$ s runs are in red. Middle and right panels: binding  $\Delta G$  calculated by initializing apo state simulations with the 4gih structure and the final configurations from 1  $\mu$ s simulations, respectively. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.

### Long sampling of the tyk2 kinase apo state

As no crystallographic structure for the tyk2 kinase in its apo state is available, we attempted to further relax the experimentally resolved holo state 4gih<sup>[11]</sup> with the removed ligand. We performed 5 independent 1  $\mu$ s unbiased molecular dynamics simulations. The generated trajectories have explored a broad range of conformations, deviating substantially from the starting structure (Fig S11, left panel). The  $\Delta$ G estimates obtained using 5 structural apo ensembles equilibrated for 1  $\mu$ s each (right panel) highlight how the significant increase in sampling deteriorates the prediction accuracy.

#### Effects of using crystallographic water

Solvation of binding pocket in the protein-ligand system may play a significant role in determining the binding affinity. In our current protocol initial solvent positions around the protein and the ligand did not include crystallographically resolved waters. To probe the influence of crystallographic water placement on the final  $\Delta G$  estimates we have repeated the procedure by including the experimentally resolved waters (Fig. S7).

Our results show that while adding crystallographic water can improve accuracy, like in the case of tyk2, often it provides no benefit, and in the cases of galectin and pde2 is even detrimental.

For example, ligand binding free energies for the pde2 system increased by on average 8 kJ/mol when crystallographic water was used. Here, both the apo and holo states were initialized from the 6ezf crystal structure, which was previously shown to better reproduce relative binding free energies than the alternatives<sup>[12]</sup> and which we showed to better represent the apo state. However, 6ezf was resolved with a much smaller ligand than those used herein. In turn, incorporating crystallographic waters into the currently simulated protein-ligand system, results in more water molecules in the binding pocket and close to the ligands (Fig. S8A), with two waters even starting out in a sterical clash with some of the molecules of the probed set. While the clash is resolved during energy minimization, the number of waters in the binding pocket remains higher than in the simulations without crystallographic water (Fig. S8B). This increases water contacts with hydrophobic areas of the ligands, as seen from the radial distribution function of water oxygens around the ligand carbon atoms (Fig. S8C). The apo simulations meanwhile are not significantly affected, resulting in reduced binding affinity.

On the other hand, using crystallographic water in the tyk2 system increased the binding affinity of most ligands, leading to better agreement with the experimental measurement. We traced this effect to an increase in water density in areas where conserved parts of the ligands hydrogen bond with water (Fig. S9A). This results in more hydrogen bonds between the ligands and solvent when using crystallographic water (Fig. S9B), leading to more stable holo states and tighter binding.



Figure S7: Effects of crystallographic water on predicted absolute binding free energies. The left column shows the simulations started from structures where no crystallographically resolved water was used to initialize the simulations. In the column on the right, crystallographic water positions were retained for both apo and holo crystal structures. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.



Figure S8: Effects of crystallographic water on pde2-ligand systems. A Ligand 49220548 is shown in yellow, while the ligand from the 6ezf crystal structure is shown in cyan. Initial positions of some crystalographic waters are close to the studied ligands. Water molecules that directly clash with the ligand are highlighted as spheres. Crystallographic ions are also shown as larger spheres. B Distributions of the number of water oxygens within the active site (approximated here as volume within 5 Å of the starting coordinates of ligand 49220548) show an increased number of waters in holo equilibrium simulations. This holds true for all other studied pde2 ligands as well (not shown). C Radial distribution functions (RDF) of water oxygens around the carbon atoms of the ligand 49220548 show an increase of the hydrophobic contacts for the simulations initialized using crystallographic waters.



Figure S9: Effects of crystallographic water on tyk2-ligand systems. A Difference in water density around ligand ejm\_45 between simulations initialized with crystallographic water and without. The red mesh indicates an isosurface where the water density is 0.04 arb. units higher when using crystallographic water, as calculated with GROmaps<sup>[4]</sup>. Crystallographic water oxygen positions are highlighted as red as spheres. **B** Distributions of the number of ligand-water hydrogen bonds tabulated across all tyk2 ligands.



Figure S10: Distributions of uncertainty estimates for absolute and relative binding free energies. Distributions smoothed with a Gaussian kernel density estimator. Mean uncertainties and their standard errors are reported in units of kJ/mol. Computing relative binding free energies directly through alchemical transformations of the ligands results in much smaller uncertainties, as little work is dissipated. For absolute binding free energies, the uncertainties are much larger. Recalculating the relative free energies from absolute ones results in even larger uncertainties due to propagation of errors.

#### Section 1. Generation of restraints

When the ligand is decoupled from the protein (main text Fig. 7, state B), it is able to explore the whole simulation box. This leads to poor convergence of the binding free energy, as it is difficult to fully sample the movement of the ligand over a large volume. To limit the accessible volume, restraints can be applied to restrict the ligand's translational and rotational degrees of freedom<sup>[2]</sup>. In this work we followed a rigorous protein-ligand restraining scheme<sup>[3]</sup>, where three atoms of the protein and three of the ligand are connected with harmonic potentials: one for distance, two for angles, and three for dihedrals. These restraints keep the decoupled ligand close to its coupled position and orientation within the binding pocket.

As harmonic potentials translate into Gaussian probability distributions in equilibrium, the protein and ligand atoms the restraints are acting on were selected to ensure a Gaussian distribution for each of the six restrained degrees of freedom in the superpositioned ensemble (main text Fig. 7, state B). The Gaussian approximations for the distributions were fit by computing the mean and variance of the degrees of freedom to be used for restraining (after correcting for periodicity of angles and dihedrals). Atom sets where these ideal Gaussian distributions were too different from the superpositioned ensemble (p-value at or above 0.05 in a Kolmogorov-Smirnov test) were rejected.

Due to an implementation detail within GROMACS<sup>[1]</sup>, an additional condition has to be satisfied by a proper choice of atoms to restrain. As restrained angles are calculated via a cross product of two vectors  $\theta_i = \arcsin(|\hat{\mathbf{a}} \times \hat{\mathbf{b}}|)$ , crossing the periodic limits of  $0 \le \theta_i < \pi$  during a simulation is not possible leading to non-Gaussian distributions of  $\theta_i$  near those regions. To prevent this from happening in NEQ morphing simulations, atoms that would produce a restraining potential on an angle  $\theta_i$  with values  $U(\theta_i = 0) < 5kT$  or  $U(\theta_i = \pi) < 5kT$  (where *k* is the Boltzmann constant and *T* the temperature) were also rejected.

#### Section 2. Corrections for effects of superpositioning

When a ligand in the active site is coupled to the protein the potential acting between the molecules leads to correlations between their motions, resulting in reduced fluctuations in ligand coordinates. The superpositioning approach for generating the conformations of the decoupled ligand in the active site uses the coupled trajectory as a reference. This transfers some correlations to the generated conformations, limiting the variability of those poses as well (Fig. S11). The presence of correlations can be detected by evaluating the Pearson correlation coefficient between different restrained degrees of freedom after superpositioning (Fig. S12).

The six independent harmonic protein-ligand restraints are only an approximation for the potential required to produce the distribution obtained by superpositioning, which is not the same as the equilibrium distribution produced by the Boresch et al.<sup>[3]</sup> orthogonal harmonic restraints. Starting the morphing simulations that drive the decoupled state of a ligand in the protein to the coupled state using these conformations under the independent restraints would violate the assumptions of the Crooks theorem. A better approximation for the restraining potential can be obtained by treating the superpositioned distribution as a multivariate normal distribution, a correlated six-dimensional Gaussian:

$$\boldsymbol{\zeta} = [\zeta_0, \zeta_1, \zeta_2, \zeta_3, \zeta_4, \zeta_5] = [r_{\rm d}, \theta_{\rm A}, \theta_{\rm B}, \phi_{\rm A}, \phi_{\rm B}, \phi_{\rm C}] \tag{S1}$$

$$P(\boldsymbol{\zeta}) = \frac{\exp(-\frac{1}{2}(\boldsymbol{\zeta} - \boldsymbol{\mu})^{\mathrm{T}})\boldsymbol{\Sigma}^{-1}(\boldsymbol{\zeta} - \boldsymbol{\mu}))}{\sqrt{(2\pi)^{3}\det\boldsymbol{\Sigma}}} \qquad , \tag{S2}$$

where  $\mu$  is the vector of restraint equilibrium values (or means of the one dimensional distributions) and  $\Sigma$  is the covariance matrix of the restrained degrees of freedom. The restraining potential that can reproduce this can be approximated via Boltzman inversion.

$$U_{\text{correlated}}(\boldsymbol{\zeta}) = -kT \ln(P(\boldsymbol{\zeta})) \tag{S3}$$

Such a multivariate potential, however, is incompatible with the GROMACS molecular dynamics engine and implementing new potentials into the highly optimized software is beyond the scope of this study. Instead, to correct for the effect of these correlations without modifying the GROMACS source code we developed and tested two decorrelation approaches.

The first approach (Fig. S13 left), called *a priori*, reproduces the equilibrium distributions of the decoupled state expected from the uncorrelated restraints. This is achieved by independently sampling restraint coordinates for each frame from the independent Gaussian distributions. The ligand in each structure of the superpositioned ensemble is then rotated and translated to satisfy those sampled restraint coordinates. Simulations of the alchemical ligand coupling to the rest of the system are then performed starting from the generated ensemble.

The second approach (Fig. S13 right), refered to as *post hoc* and used as part of the superpositioning protocol, performs the alchemical transition simulations starting from the superpositioned configurations, but adjusts the work



Figure S11: Ensembles of ligand poses in the decoupled and coupled states and their RMSD fluctuations. Decoupled states generated with superpositioning (orange) inherit the limited range of motion and correlations with protein motion present in the coupled state (green). Adding rotational and translational noise consistent with the restraining potential (purple) to the superpositioning ensemble recovers an ensemble close to that of the explicitly simulated decoupled state (blue). Structures and RMSD data shown are limited to a single independent repeat of the 18660-1 ligand<sup>[7]</sup> with jnk1.

values associated with transitions from the decoupled ligand to its coupled state by adding the free energy difference between the two restraining potentials  $\Delta G_{MC} - \Delta G_{restr}$ . While the free energy imposed by the uncorrelated restraints can be calculated analytically<sup>[3]</sup>, that for the correlated restraints is computed via Monte Carlo integration of  $U_{correlated}(\zeta)$ . Fig. S15 shows this method to have a better agreement with the more expensive approach of explicitly simulating the decoupled state.



Figure S12: Example of correlated restraint distributions introduced by superpositioning. Restraint coordinates of frames produced by superpositioning are shown as black points while the contour lines represent the strength of the restraining potential. This data comes from the superpositioned trajectory for jnk1 in the apo conformation with the 18660-1 ligand. In this example, the dihedral restraint coordinates  $\phi_B$  and  $\phi_C$  are highly correlated.



Figure S13: Schemes for decorrelation approaches. While *a priori* decorrelation (left) reproduces the equilibrium distribution of the decoupled end state with the orthogonal restraints, *post hoc* decorrelation (right) corrects for the free energy difference between the ensembles produced by the orthogonal and correlated restraints by modifying the work values of the coupling transition.



Figure S14: Absolute binding free energies obtained with explicit equilibrium simulations of the decoupled states broken down by system. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.



Figure S15: Effects of various decorrelation methods. Explicitly equilibrating the decoupled state provides an exact reference for further comparison. The simplest protocol (top row) involves no corrections for correlation. *Post hoc* decorrelation (middle row) adjusts the coupling work values by the free energy difference between the effects of the correlated and orthogonal restraints. *A priori* decorrelation (bottom row) rotates the ligand in the superpositioned structures to orientations sampled from a distribution that satisfies the orthogonal restraints. Data is shown for the tyk2,  $p38\alpha$  and jnk1 systems. Dark and light shaded areas represent regions deviating from experiment by at most 1 and 2 kcal/mol.

	ligand	ΔG	$\Delta G (kJ/mol)$		
protein		simulation	experimental		
tyk2	jmc_30	$-29.7 \pm 2.7$	$-45.7 \pm 0.7^{[11]}$		
tyk2	jmc_28	$-29.4 \pm 2.7$	$-45.9\pm 0.7^{[11]}$		
tyk2	jmc_27	$-30.0 \pm 3.1$	$-47.2 \pm 0.7^{[11]}$		
tyk2	jmc_23	$-30.7 \pm 2.1$	$-48.9 \pm 0.7^{[11]}$		
tyk2	ejm_55	$-33.8 \pm 2.8$	$-38.5 \pm 0.7^{[10]}$		
tyk2	ejm_54	$-37.2 \pm 3.7$	$-44.1 \pm 0.7^{[10]}$		
tyk2	ejm_50	$-28.2\pm2.6$	$-37.6 \pm 0.7^{[10]}$		
tyk2	ejm_49	$-32.2\pm3.6$	$-32.4 \pm 0.7^{[10]}$		
tyk2	ejm_48	$-26.2\pm2.2$	$-37.7 \pm 0.7^{[10]}$		
tyk2	ejm 47	$-31.8 \pm 2.7$	$-40.6 \pm 0.7^{[10]}$		
tyk2	ejm 46	$-33.6 \pm 3.0$	$-47.3 \pm 0.7^{[10]}$		
tyk2	ejm 45	$-29.4 \pm 2.5$	$-40.0\pm0.7^{[10]}$		
tvk2	eim 44	$-17.4\pm2.3$	$-31.0\pm0.7^{[10]}$		
tyk2	eim 43	$-22.3 \pm 3.5$	$-34.6 \pm 0.7^{[10]}$		
tvk2	eim 42	-35.9 + 3.6	$-40.9 \pm 0.7^{[10]}$		
tyk2	eim 31	$-31.5 \pm 2.0$	$-39.9 \pm 0.7^{[10]}$		
ink1	18660-1	-405+29	$-363 \pm 03^{[14]}$		
ink1	18659-1	-420+35	$-39.4 \pm 0.3^{[14]}$		
ink1	18658-1	$-395 \pm 18$	$-40.4 \pm 0.4^{[14]}$		
jnk1	18652-1	$-40.0 \pm 22$	$-44.4 \pm 0.4^{[14]}$		
jnk1	18639-1	$-46.0 \pm 2.2$	$-40.5 \pm 0.6^{[14]}$		
jiiki inkl	18638 1	$-44.8 \pm 2.8$	$-42.0 \pm 0.4^{[14]}$		
jiiki inkl	18637 1	$-44.0 \pm 2.0$ $-42.5 \pm 2.0$	$-42.0 \pm 0.4$ $-42.4 \pm 1.5$ <sup>[14]</sup>		
jiiki inkl	18636 1	$-42.5 \pm 2.0$ $34.0 \pm 1.6$	$-42.4 \pm 1.5^{-1}$ 31.7 $\pm$ 1.7 <sup>[14]</sup>		
jiiki inkl	18635 1	$-35.0 \pm 1.0$	$-30.5 \pm 0.4$ <sup>[14]</sup>		
jiiki inkl	18634 1	$-55.0 \pm 1.0$	$-30.5 \pm 0.4^{10}$		
jiiki inki	18633 1	$-41.4 \pm 2.2$ 36.0 $\pm 1.4$	$-41.3 \pm 0.3^{11}$		
JIIK I inle1	18622 1	$-30.9 \pm 1.4$	$-38.4 \pm 1.0^{11}$		
JIIK I inle1	18621 1	$-42.4 \pm 3.1$	$-30.1 \pm 1.3^{11}$		
jiiki inlei	18620 1	$-59.5 \pm 2.0$	$-39.4 \pm 1.2^{11}$		
JIIK I inle1	18620 1	$-59.5 \pm 3.5$	$-36.1 \pm 0.2^{11}$		
JIIK I imle1	18629-1	$-36.7 \pm 2.9$	$-30.2 \pm 0.4^{11}$		
JIIK I imle1	10020-1	$-54.1 \pm 2.5$	$-30.4 \pm 0.9^{[14]}$		
JIIK I imle1	18027-1	$-40.8 \pm 3.3$	$-33.3 \pm 0.8^{14}$		
JIIK I imle1	18625 1	$-43.5 \pm 3.5$	$-37.0 \pm 0.0^{[14]}$		
JIIK I imle1	18624 1	$-29.0 \pm 2.3$	$-33.9 \pm 0.8^{11}$		
JIIK I imle1	16024-1	$-52.7 \pm 2.9$	$-33.3 \pm 0.7$		
JIIK I m28 cr	1/124-1	$-44.7 \pm 3.2$	$-40.0 \pm 1.0^{11}$		
p38a	p58a_2aa	$-44.4 \pm 2.8$	$-59.5 \pm 1.1^{[8]}$		
p38a	p38a_31mk	$-51.5 \pm 2.0$	$-50.5 \pm 0.9^{101}$		
p38α	p38a_31mn	$-40.1 \pm 2.8$	$> -45.4^{[0]}$		
p38α	p38a_3fiz	$-41.9 \pm 3.2$	$-40.5 \pm 0.9^{[8]}$		
p38α	рз8а_зпу	$-41.7 \pm 3.7$	$-48.7 \pm 0.2^{[8]}$		
$p_{38\alpha}$	p38a_3flw	$-50.0\pm2.5$	$-45.6 \pm 0.4^{[8]}$		
$p_{38\alpha}$	p38a_3flq	$-42.7 \pm 3.4$	$-42.4 \pm 1.0^{[8]}$		
$p_{38\alpha}$	p38a_3ffn	$-44.6 \pm 2.1$	$-46.4 \pm 1.0^{[8]}$		
$p_{38\alpha}$	p38a_2z	$-46.8 \pm 3.2$	$-44.1 \pm 0.8^{[8]}$		
$p_{38\alpha}$	p38a_2y	$-46.8 \pm 3.5$	$-41.6 \pm 1.0^{[8]}$		
p38α	p38a_2x	$-42.6 \pm 3.3$	$-43.6 \pm 0.7^{[8]}$		
p38α	p38a_2v	$-35.8 \pm 2.7$	$-38.2 \pm 0.7$		
p38α	p38a_2u	$-52.1 \pm 5.3$	$-52.2 \pm 0.3$		
p38α	p38a_2t	$-40.0 \pm 3.2$	$-50.5 \pm 0.4^{[8]}$		
p38α	p38a_2s	$-45.1 \pm 4.8$	$-47.3 \pm 0.4^{[8]}$		
p38α	p38a_2r	$-46.4 \pm 2.4$	$-46.0\pm0.4^{[8]}$		
p38α	p38a_2q	$-50.5 \pm 2.2$	$-52.2\pm0.5^{[8]}$		
p38α	p38a_2p	$-44.4 \pm 3.0$	$-47.7 \pm 0.2^{[8]}$		
p38α	p38a_2o	$-41.7 \pm 3.9$	$-44.7 \pm 0.5$ <sup>[8]</sup>		
p38α	p38a_2n	$-44.7 \pm 4.0$	$-50.5 \pm 0.6^{[8]}$		

p38α	p38a_2m	$-45.2 \pm 3.4$	$-46.7\pm0.3^{[8]}$
p38α	p38a_21	$-47.1 \pm 3.5$	$-52.2\pm0.3^{[8]}$
p38α	p38a_2k	$-42.6\pm1.6$	$-44.9 \pm 0.4^{[8]}$
p38α	p38a_2j	$-43.6 \pm 2.2$	$-42.9 \pm 0.7^{[8]}$
p38α	p38a_2i	$-43.1 \pm 2.7$	$-42.8 \pm 0.9^{[8]}$
p38α	p38a_2h	$-42.0 \pm 2.1$	$-40.4 \pm 0.8^{[8]}$
p38α	p38a_2gg	$-40.6 \pm 2.6$	$-45.4 \pm 0.3^{[8]}$
p38α	p38a_2g	$-43.3 \pm 2.0$	$-45.4 \pm 1.0^{[8]}$
p38α	p38a 2ff	$-45.2 \pm 2.6$	$-48.7 \pm 0.4^{[8]}$
p38α	p38a 2f	$-46.4 \pm 3.1$	$-36.1 \pm 0.9^{[8]}$
p38α	p38a 2ee	$-43.2 \pm 3.9$	$-52.2 \pm 1.5^{[8]}$
p38α	p38a 2e	$-46.5 \pm 2.5$	$-45.6 \pm 1.1^{[8]}$
p38α	p38a 2c	$-47.0 \pm 1.9$	$-44.6 \pm 1.2^{[8]}$
$p38\alpha$	p38a_2bb	$-37.5 \pm 3.8$	$-38.6 \pm 0.8^{[8]}$
nde2	43249674	$-354 \pm 41$	$-492 \pm 09^{[12]}$
nde?	50181001	$-44.0 \pm 4.3$	$-39.8 \pm 0.3^{[12]}$
nde2	50107616	-515+58	$-39.5 \pm 0.5$ $-39.5 \pm 1.8^{[12]}$
nde?	49932714	-383+44	$-41.3 \pm 0.1^{[12]}$
nde?	49932129	$-31.7 \pm 2.9$	$-40.5 \pm 0.1$
pde2 nde2	40585367	$-38.2 \pm 4.3$	$-40.2 \pm 1.1$
pdc2 nde?	49585367	$-42.8 \pm 5.4$	$-43.2 \pm 0.0$ [12]
pde2 nde2	40582300	$-41.7 \pm 4.6$	$-45.2 \pm 0.9$ -45.1 $\pm$ 1.7 <sup>[12]</sup>
puez nde?	49582590	$-41.7 \pm 4.0$ $-41.0 \pm 2.8$	$-43.1 \pm 1.7$ $-42.3 \pm 1.1$ <sup>[12]</sup>
pue2 nde2	49306113	$-41.0 \pm 2.0$	$-42.3 \pm 1.1^{-1}$
puez nde2	49390300	$-44.0 \pm 3.7$	$-43.2 \pm 0.9^{-1}$
puez mda2	49220348	$-30.1 \pm 3.2$	$-47.0 \pm 1.0^{[12]}$
puez mda2	49220392	$-47.2 \pm 0.9$	$-47.2 \pm 1.2^{11}$
pae2	49175780	$-51.0 \pm 5.8$	$-50.5 \pm 2.1^{[12]}$
pde2	491/5/89	$-32.0\pm 5.7$	$-40.3 \pm 1.3^{[12]}$
pae2	49137530	$-42.9 \pm 5.1$	$-46.8 \pm 1.7^{[12]}$
pae2	49137374	$-33.8 \pm 3.4$	$-4/.2 \pm 1.7$
pae2	49072088	$-39.5 \pm 6.2$	$-36.9 \pm 0.7$ <sup>[12]</sup>
pde2	482/1249	$-41.2 \pm 3.0$	$-45.5 \pm 0.5^{[12]}$
pae2	48108913	$-34.2 \pm 3.4$	$-43.0 \pm 1.3^{[12]}$
pde2	48022468	$-40.0 \pm 3.1$	$-50.0 \pm 1.1^{[12]}$
pde2	48009208	$-32.3 \pm 3.3$	$-46.1 \pm 1.1^{[12]}$
cdk2	32	$-29.9 \pm 5.7$	$-42.1 \pm 0.3^{[9]}$
cdk2	31	$-34.5 \pm 3.5$	$-41.2 \pm 0.3^{[9]}$
cdk2	30	$-43.9 \pm 2.0$	$-42.3 \pm 0.3^{[9]}$
cdk2	29	$-40.8 \pm 3.9$	$-42.8 \pm 1.2^{[9]}$
cdk2	28	$-36.9 \pm 2.1$	$-47.9 \pm 0.1^{[9]}$
cdk2	26	$-35.9 \pm 2.6$	$-36.5 \pm 0.1^{[9]}$
cdk2	22	$-32.6 \pm 1.9$	$-34.0\pm0.2^{[9]}$
cdk2	21	$-31.3 \pm 2.4$	$-33.9 \pm 0.3^{[9]}$
cdk2	20	$-39.7 \pm 3.0$	$-37.8 \pm 0.7^{[9]}$
cdk2	loiy	$-36.5 \pm 2.5$	$-42.6 \pm 1.7^{[9]}$
cdk2	1oiu	$-40.9 \pm 4.3$	$-39.3\pm0.5^{[9]}$
cdk2	1oi9	$-39.7 \pm 2.5$	$-42.1 \pm 0.1^{[9]}$
cdk2	1h1s	$-37.0 \pm 4.0$	$-48.6 \pm 0.5^{[9]}$
cdk2	1h1r	$-34.3 \pm 3.1$	$-29.4 \pm 1.7^{[9]}$
cdk2	lhlq	$-33.7 \pm 2.1$	$-35.4 \pm 0.2^{[9]}$
cdk2	17	$-33.0 \pm 2.8$	$-30.9 \pm 2.1^{[9]}$
cmet	CHEMBL3402761_1_21	$-50.0 \pm 5.2$	$-50.9 \pm 0.8^{[6]}$
cmet	CHEMBL3402755_4200_15	$-29.3\pm2.6$	$-30.4 \pm 0.8^{[6]}$
cmet	CHEMBL3402754_40_14	$-38.0\pm3.2$	$-41.8 \pm 0.8^{[6]}$
cmet	CHEMBL3402753_200_13	$-32.4 \pm 4.0$	$-37.9 \pm 0.8^{[6]}$
cmet	CHEMBL3402752_30000_12	$-29.6\pm3.3$	$-25.6\pm0.8^{[6]}$
cmet	CHEMBL3402751_2100_11	$-24.5\pm3.3$	$-32.1\pm0.8^{[6]}$
cmet	CHEMBL3402750_400_10	$-34.3\pm2.2$	$-36.2\pm0.8^{[6]}$
cmet	CHEMBL3402749_500_9	$-34.1 \pm 3.6$	$-35.6 \pm 0.8^{[6]}$

cmet	CHEMBL3402748_5300_8	$-20.7\pm3.2$	$-29.8 \pm 0.8^{[6]}$
cmet	CHEMBL3402747_3400_7	$-24.8 \pm 4.7$	$-30.9 \pm 0.8^{[6]}$
cmet	CHEMBL3402745_200_5	$-31.0 \pm 3.8$	$-37.9 \pm 0.8^{[6]}$
cmet	CHEMBL3402744_300_4	$-30.6 \pm 4.3$	$-36.9 \pm 0.8^{[6]}$
galectin	08_ligNH2	$-31.2 \pm 3.1$	$-27.8 \pm 0.1^{[13]}$
galectin	07_ligOH	$-36.1 \pm 4.2$	$-26.0\pm0.2^{[13]}$
galectin	06_ligPyr	$-29.6\pm3.9$	$-19.8 \pm 2.4^{[13]}$
galectin	05_ligOEt	$-28.5\pm1.5$	$-22.8 \pm 0.3^{[13]}$
galectin	04_ligNMe2	$-28.2\pm2.8$	$-24.7 \pm 0.2^{[13]}$
galectin	03_ligNHMe	$-31.5\pm2.2$	$-26.6 \pm 0.1^{[13]}$
galectin	02_ligOMe	$-30.2\pm3.3$	$-26.6 \pm 0.3^{[13]}$
galectin	01_ligF	$-29.7\pm2.9$	$-30.7 \pm 0.2^{[13]}$

Table S1: Calculated and experimental absolute free energies of binding used for Figure 1 in the main text. Holo crystallographic structures with the ligand removed were used to initialize simulations of the apo state for tyk2 and pde2, while apo crytallographic structures were used for all other proteins. Experimental free energies were obtained from published  $IC_{50}$  and  $K_d$  values at experimental temperatures. For  $IC_{50}$  derived data where no  $K_m$  was available (p38 $\alpha$ , pde2, and cmet),  $K_I \approx IC_{50}$  was used.

protein	pdb id	resolution (Å)	apo or ligand name (if in dataset)	used for modeling initial ligand poses
cdk2	1h1q	2.5	1h1q	yes
cdk2	1h27	2.2	apo	no
jnk1	2gmx	3.5	17124-1	yes
jnk1	3017	3.0	apo	no
p38α	3fly	3.0	3fly	yes
p38α	1wfc	3.0	apo	no
galectin	5e89	1.5	not in dataset	yes
galectin	3zsl	1.08	apo	no
tyk2	4gih	2.0	ejm_46	yes
cmet	4r1y	2.0	not in dataset	yes
cmet	1r1w	1.8	apo	no
pde2	6ezf	1.5	not in dataset	no
pde2	4htz	2.0	apo	no
pde2	4d08	1.9	49137530	yes
pde2	4d09	2.5	not in dataset	no

Table S2: Summary of the crystallographic structures used.

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