Supporting Information

Atomic Origins of Strong Metal-Support Interaction in Silica Supported Catalysts

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Figure S1. **a**–**d**, STEM image (**a**) and EDX elemental mapping on pristine ETEM chip showing a thin-film of SiO_2 on SiN_x membrane (**b**–**d**). **e**, Corresponding EDX spectrum. The relative abundance of N, O, and Si were also shown.

Region No.	N (at.%)	O (at.%)	Si (at.%)
1	54.6	6.5	39.9
2	54.1	6.0	39.9
3	54.1	5.4	40.5
4	53.7	5.8	40.5
5	54.4	6.6	39.0
6	53.6	6.0	40.4
Average	54.0	6.0	40.0

Table S1. Quantitative analysis of N, O, and Si by EDX mapping on six regions of pristine ETEM chip.



Figure S2. SAED patterns of Co nanoparticle before (0 s) and after annealing (7294 s). The SAED was acquired *in-situ* at 750 $^{\circ}$ C.



Figure S3. Size distribution of Co nanoparticles on carbon substrate derived from ETEM images acquired at relative time: **a**, 0 s, **b**, 2716 s.



Figure S4. *In-situ* HAADF-STEM images of Co nanoparticles with overlayer on the edge of substrate under 3% H₂/Ar.



Figure S5. ETEM images of Co nanocrystals with overlayer obtained in different conditions at 750 °C: **a**, 3% H₂/Ar. **b**, CH₄. **c**, CH₄/H₂ (v/v=7/3), **d**, CO. The overlayer were indicated by arrows. All scale bars: 2 nm. Typical FFT patterns of fcc and hcp-Co were also shown in TEM images.



Figure S6. Statistical analysis of layer of ordered SiO_2 covered on Co nanoparticles derived from TEM images.



Figure S7. **a**–**c**, *In-situ* EELS elemental mapping of Si, O, Co on the same Co particle as shown in Figure 2a. Corresponding line scan profiles of EELS across the Co-overlayer indicated with #1–#3 in (**a**–**c**).



Figure S8. a, c, *In-situ* STEM-EELS image of a Co nanoparticles on SiO_2/SiN_x substrate annealing under H_2/CH_4 (**a**) and CO (**c**) conditions. The particle in (**a**) is the same as shown in Figure 2a. **b**, **d**, Corresponding EELS spectrum of C *K*-edge obtained at the surface of Co nanoparticle.

The inversed FFT (IFFT) formed using spots of SiO₂ in corresponding FFT matched well with the simulated TEM contrast of quartz SiO₂ [10 $\overline{1}$] (Fig. S9d–f).



Figure S9. **a**, ETEM image of SiO₂-Si-Co interface. **b**, FFT pattern as shown in Figure 3b. **c**, Simulated electron diffraction pattern of quartz SiO₂ along $[10\overline{1}]$ direction. **d**–**f**, IFFT formed using spots of SiO₂ in corresponding FFT (**d**) and simulated TEM image

(e) obtained from atomic structure model of quartz $SiO_2[10\overline{1}]$ (f).



Figure S10. a–d, HAADF-STEM images and EDX elemental mapping of Co/SiO₂ after annealing under 10% H₂/Ar at 750 $^{\circ}$ C for 6 h.



Figure S11. **a**–**d**, Single-shot ETEM images of Co nanocrystals with SiO₂ overlayer on flat facet with low index. The round surface consisting of high index crystal planes was marked by dashed line.



Figure S12. *In-situ* STEM-EELS characterization. a, b, STEM-EELS images of two Co nanoparticles on SiO_2/SiN_x in 3% H₂/Ar (50 Pa) at 750 °C. Corresponding EELS elemental mapping of N *K*-edge, Si $L_{2,3}$ -edge, O *K*-edge, Co $L_{2,3}$ -edge, and their combinatial images.



Figure S13. *In-situ* STEM-EELS characterization. **a**, SETM-EELS spectrum image showing a Co nanoparticle on SiO_2/SiN_x substrate in 3% H₂/Ar at 750 °C. **b**–**d**, EELS spectra extracted from (**a**, regions I and II), showing Si (**b**), O (**c**), and Co (**d**) signals from nanoparticle and substrate. Commercial Si, SiO₂, and CoSi₂ were used as referenced samples, CoO, Co₃O₄, Co, and Si₃N₄ from EELS database were used. All

the spectra were calibrated by using the zero-loss peak (0 eV). The EELS spectra of Co L-edge were normalized by the energy loss peak of Co L_2 -edge.



Figure S14. **a**, **b**, *In-situ* EELS Co $L_{2,3}$ spectra of Co-SiO₂ before (**a**) and after (**b**) orderto-disorder transition of SiO₂. The EELS spectra of different Co-SiO₂ nanoparticles and Co and CoSi₂ references were calibrated by the zero-loss peak (0 eV) and then normalized by the energy loss peak of Co L_2 -edge.



Figure S15. HAADF-STEM images of as-prepared Pt nanoparticles supported on SiO_2 spheres.



Figure S16. **a**, High-resolution TEM image of Pt nanoparticle supported on SiO₂. **b**, FFT derived from region marked in (**a**). **c**, Simulated FFT of Pt₃Si [61 $\overline{3}$].

Table S2. Active surface area of Pt/SiO_2 catalyst annealed at 500-900 °C, which was characterized by CO adsorption. The loading amount of Pt was controlled to 3 wt% of the support.

Catalyst reduction	Active Pt surface	
temperature (°C)	area (m ² /g-cat)	
500	0.93	
600	4.15	
750	3.91	
900	1.73	



Figure S17. Distributions of Pt catalyst size calcined at 500 (a), 600 (b), 750 (c), and 900 $^{\circ}$ C (d). The Pt catalyst was supported on SiO₂ spheres. The averaged Pt particle size was also shown in each panel.



Figure S18. a, b, TEM images of Co/SiO₂ catalyst after hydrogenation reaction. c, d, Size distribution of Pt nanoparticles before (c) and after (d) reaction. Panel (c) is the same as in Fig. S17b.