

Supporting Information

The Highly Selective Synthesis of 5-Methy Vanillin from the By-product in Vanilla Industry and The Scent Influence for Vanillin

Haifang Mao^a, Hongzhao Wang^a, Ting Tang^{b*}, Qixuan Shi^b, Haiyan Yu^b,
Xiaojun Hu^{a*}, Zuobing Xiao^b, Pingyi Zhang^a, Jibo Liu^{a*}

a. School of Chemical and Environmental Engineering, Shanghai Institute of Technology, 100 Haiquan Road, 201418, Shanghai, China.

b. School of Perfume and Aroma Technology, Shanghai Institute of Technology, 100 Haiquan Road, Shanghai, 201418, China.

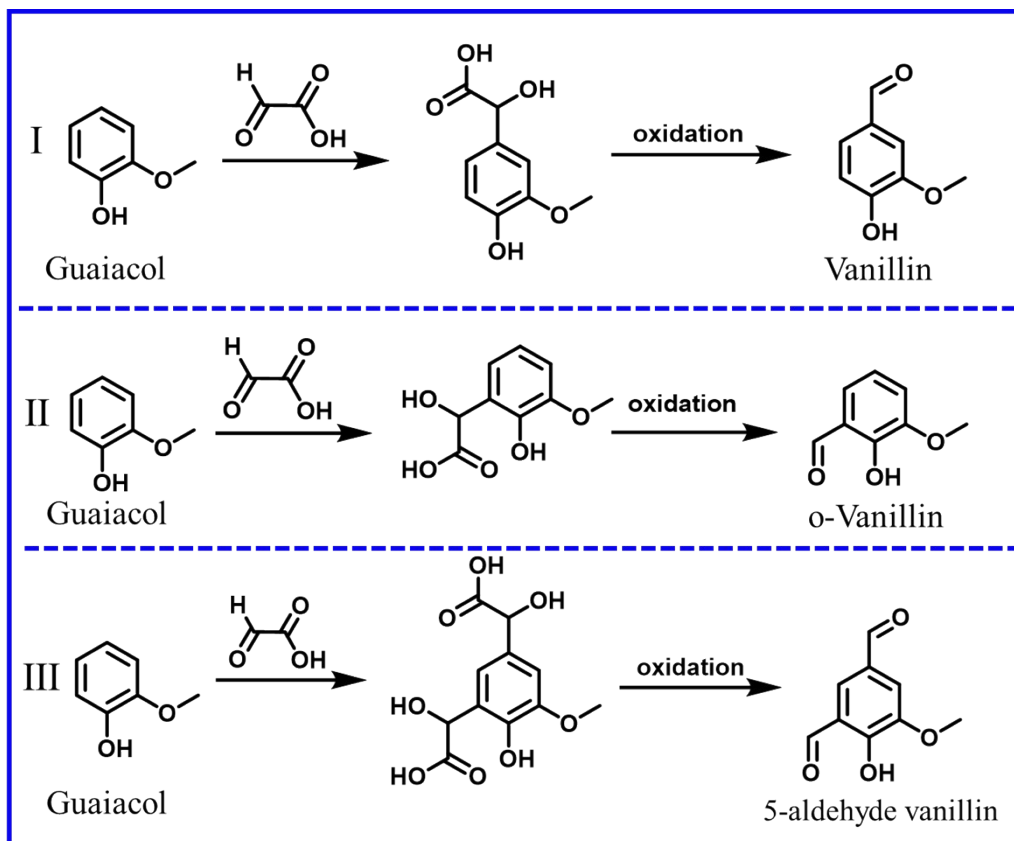
1 Preparation and characterization of Pd/CSC

1.1 Preparation of Catalyst Pd/CSC

To remove impurities and modify surface functional groups, the CSC (Coconut shell carbon) (10.0 g) was refluxed at 90 °C in HNO₃ (100 mL) for 6 hours. After that, the mixture was cooled to room temperature and then filtered and washed repeatedly using distilled water for 3 times and dried in vacuum. Finally, the obtained CSC was dried at 120 °C for 12 hours. The supported Pd catalysts were prepared by incipient-wetness impregnation of CSC with aqueous solutions of Pd(NO₃)₂ respectively. After impregnation, the catalysts were dried in air at 110 °C for 3 hours and finally reduced in H₂ at 400 °C for 3 hours.

1.2 Characterization of the Prepared Pd/CSC Composite Catalyst

X-ray data on suitable single crystals were collected at 293 K with a Focus D8 (Bruker, Germany) with Cu K α radiation ($\lambda = 0.1542$ nm). The composition and structure of these catalyst samples were investigated using a Raman spectrometer (DXR, Thermo Fisher Scientific, America). The X-ray photoelectron spectroscopy (XPS) experiments were performed on a Thermo Escalab 250Xi system using Al K α radiation ($h\nu = 1486.6$ eV). The Brunauer-Emmett-Teller (BET) specific surface areas of typical products were obtained at 77 K in a Micromeritics ASAP 2020 system. The morphology of the as-prepared catalyst was characterized by field-emission scanning electron microscopy (S-3400N, Hitachi, Japan).



Scheme S1 The preparation method of Vanillin form guaiacol.

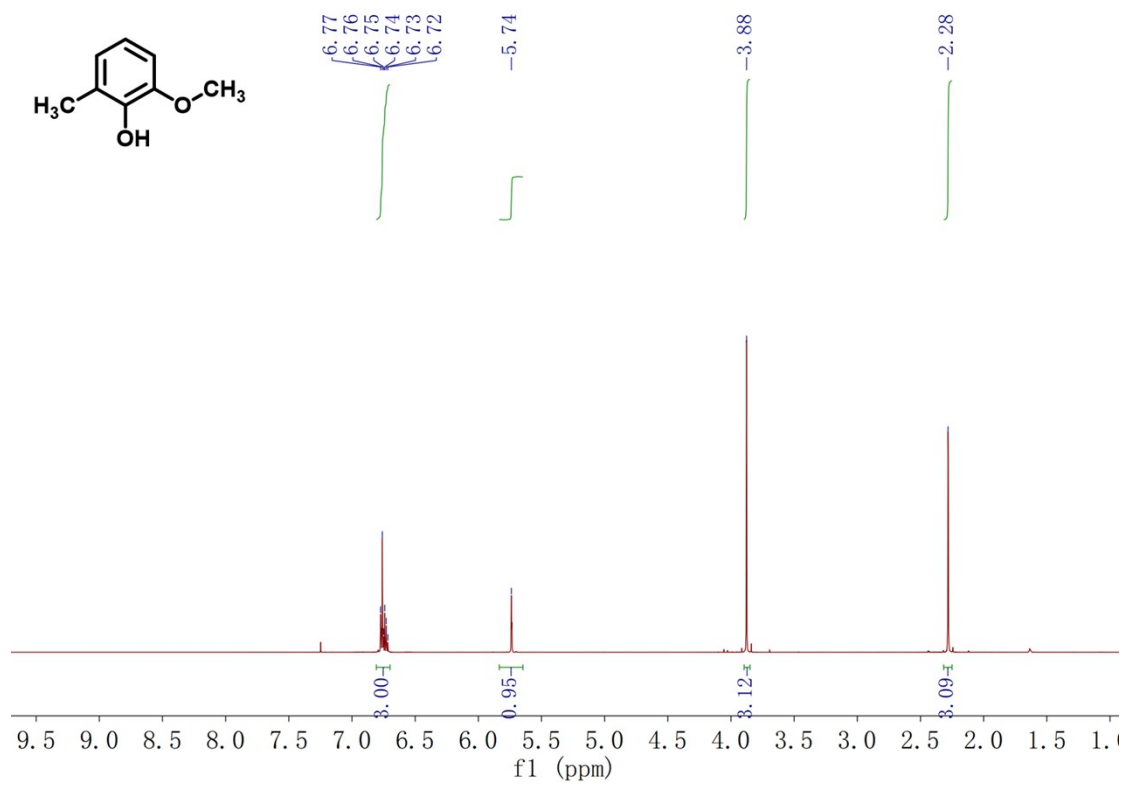


Fig.S1 The ¹H NMR of M-Guaiacol

¹H NMR (600 MHz, CDCl₃, Me₄Si) 6.77-6.73 (3H, m), 5.74 (1H, s), 3.88 (3H, s), 2.28 (3H, s).

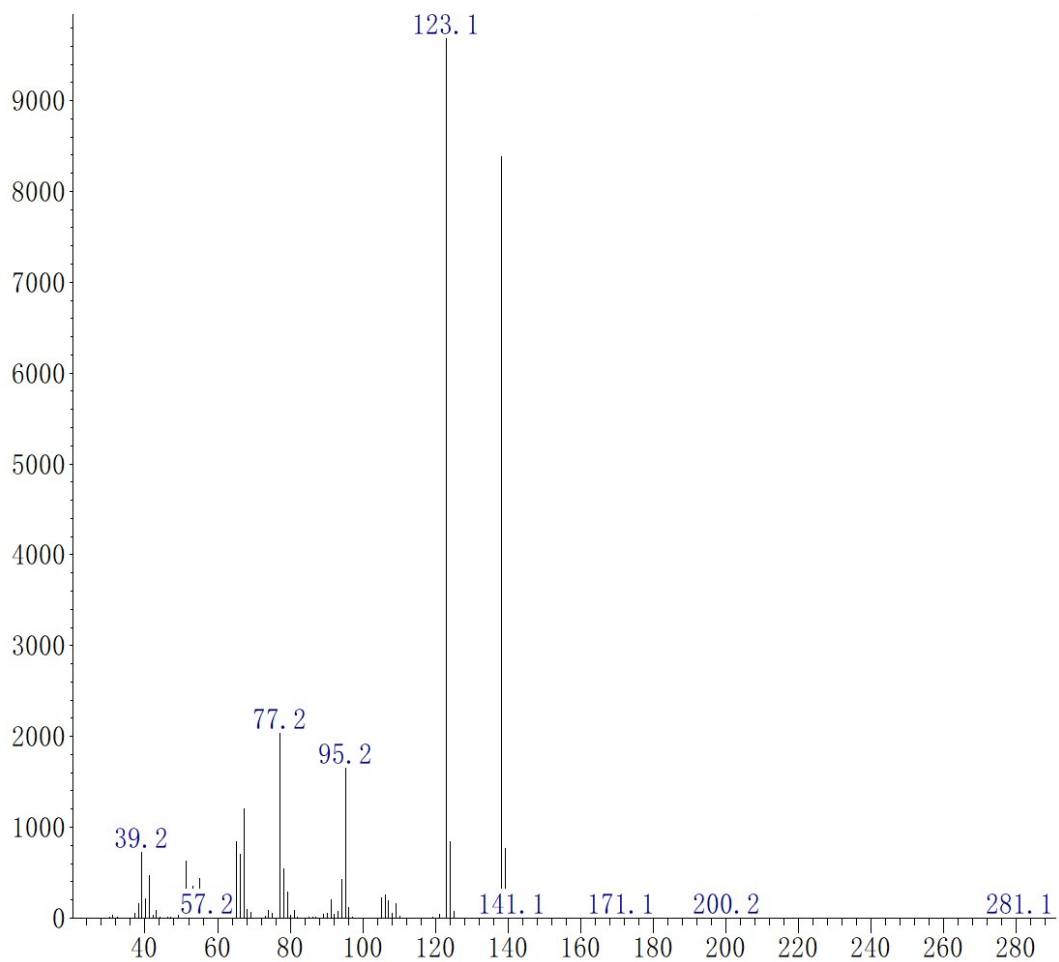


Fig.S2 Gas chromatography - mass spectrometry(GC-MS) of M-Guaiacol.

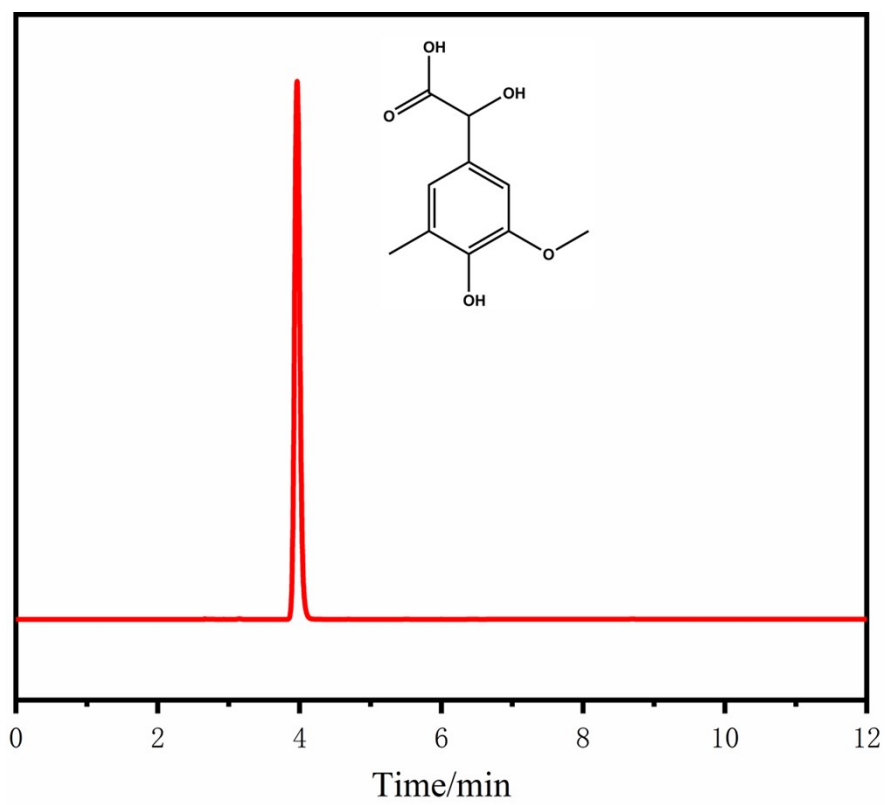


Fig.S3 Liquid phase spectrogram of 5-MVMA.

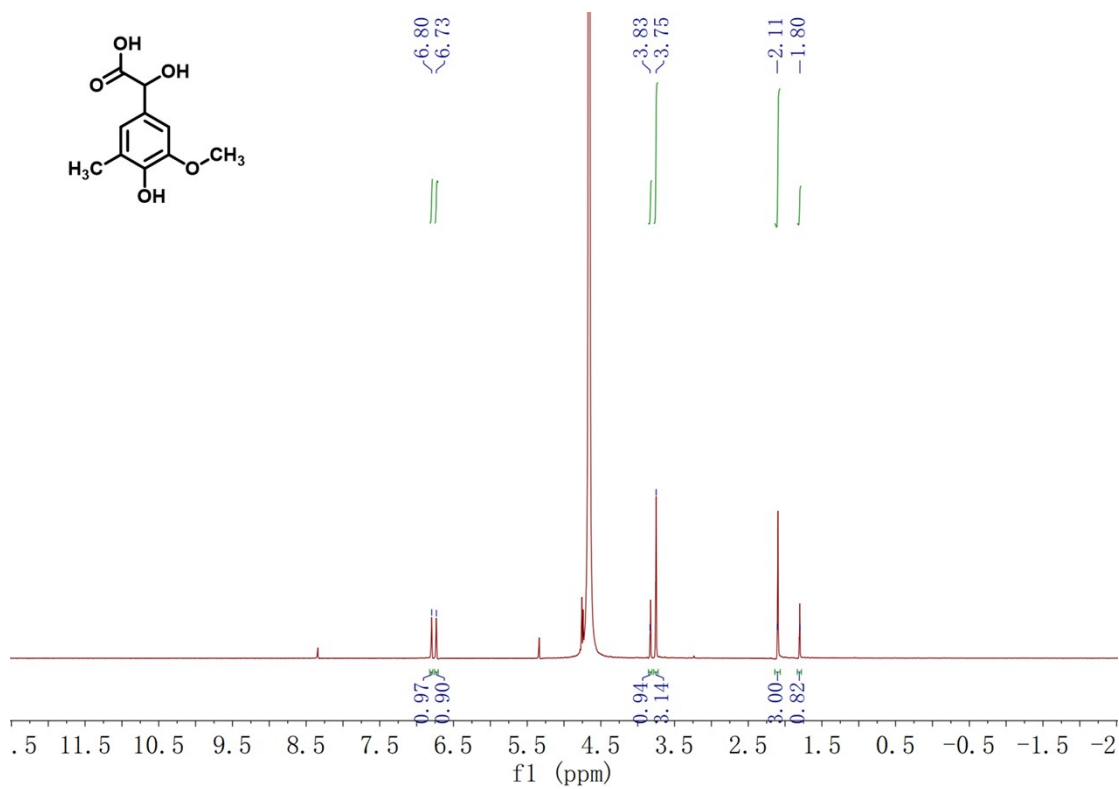


Fig.S4 The ¹H NMR of 5-MVMA

¹H NMR (600 MHz, D₂O, Me₄Si) 6.80 (1H, s), 6.73 (1H, s), 3.83 (1H, s), 3.75 (3H, s), 2.11 (3H, s), 1.80 (1H, s).

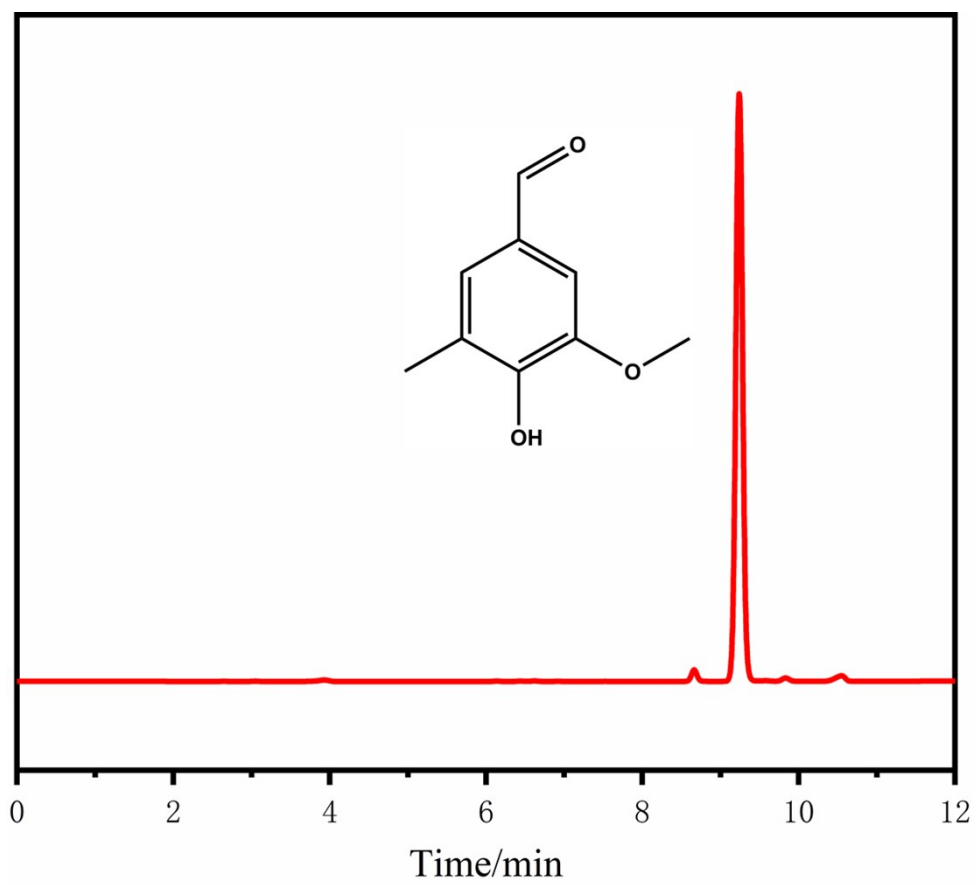


Fig.S5 Liquid phase spectrogram of M-Vanillin.

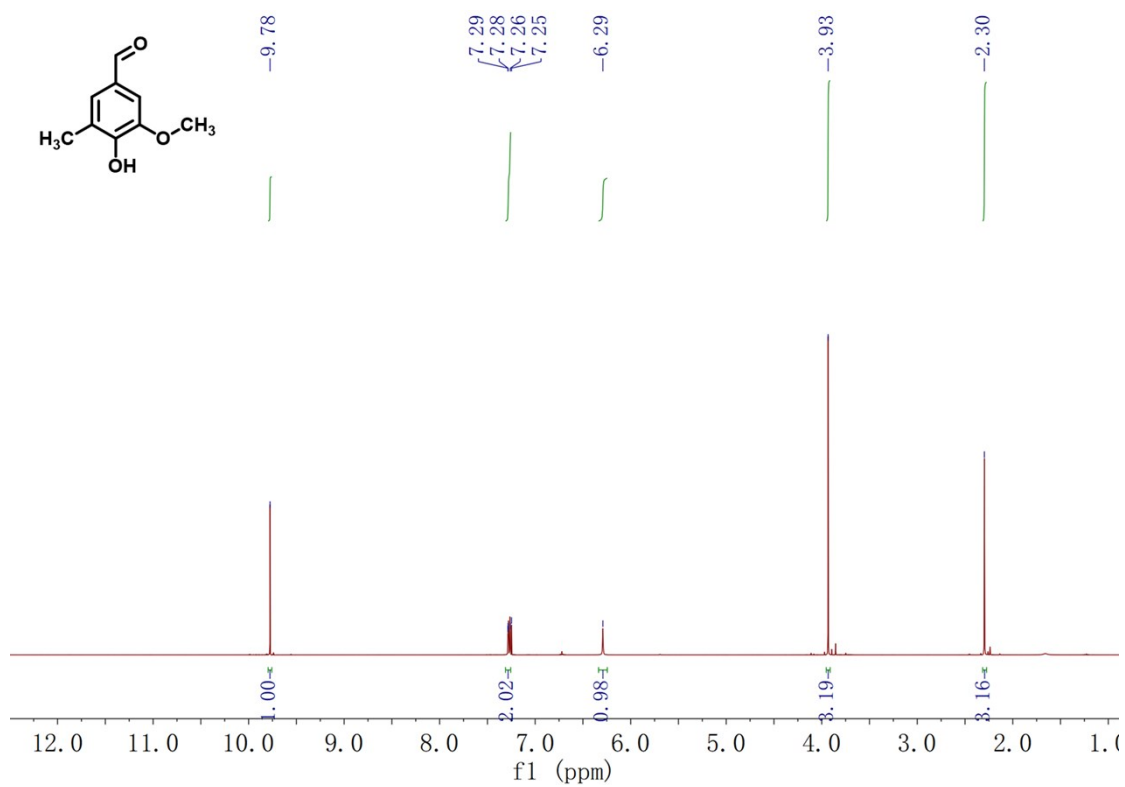


Fig.S6 The ¹H NMR of M-Vanillin

¹H NMR (600 MHz, CDCl₃, Me₄Si) 9.79 (1H, s), 7.29-7.26 (2H, m), 6.29 (1H, s), 3.93 (3H, s), 2.30 (3H, s).

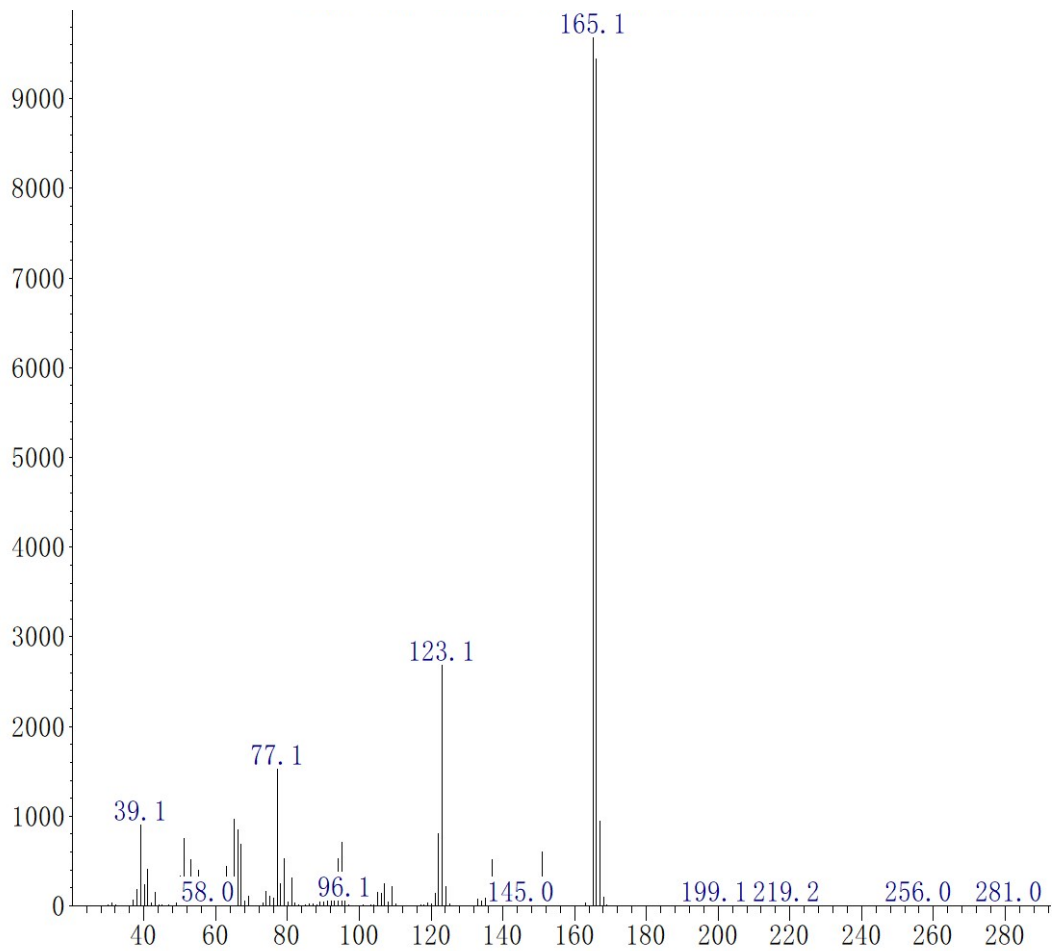


Fig.S7 Gas chromatography - mass spectrometry (GC-MS) of M-Vanillin.

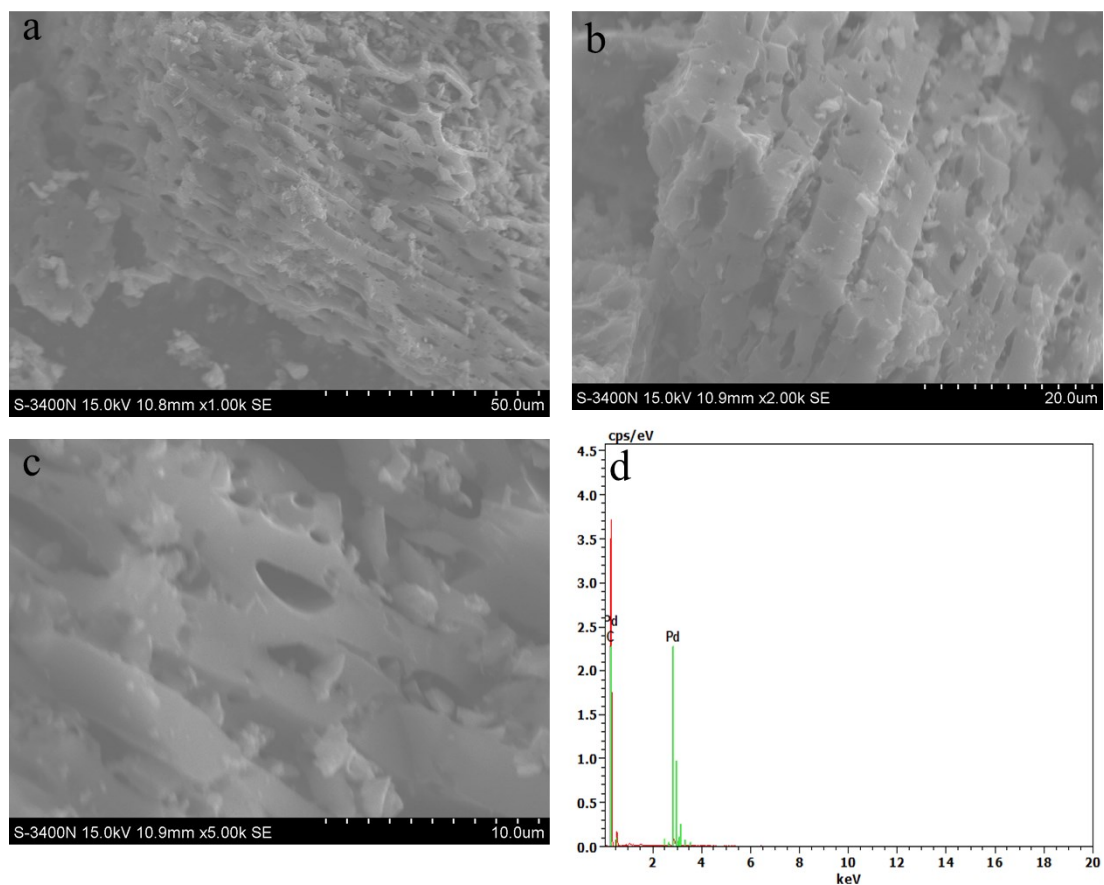


Fig.S8 (a~c) Field Emission Scanning Electron Microscope (SEM) sketch of Pd/CSC, (d) Energy dispersive spectroscopy (EDS) sketch of Pd/CSC.