

Electronic Supplementary Material ESI for Sustainable Energy & Fuels

Metal-Organic Framework Mediated Nickel Doped Copper Ferrite for

Superior Lithium Storage

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Supporting Information

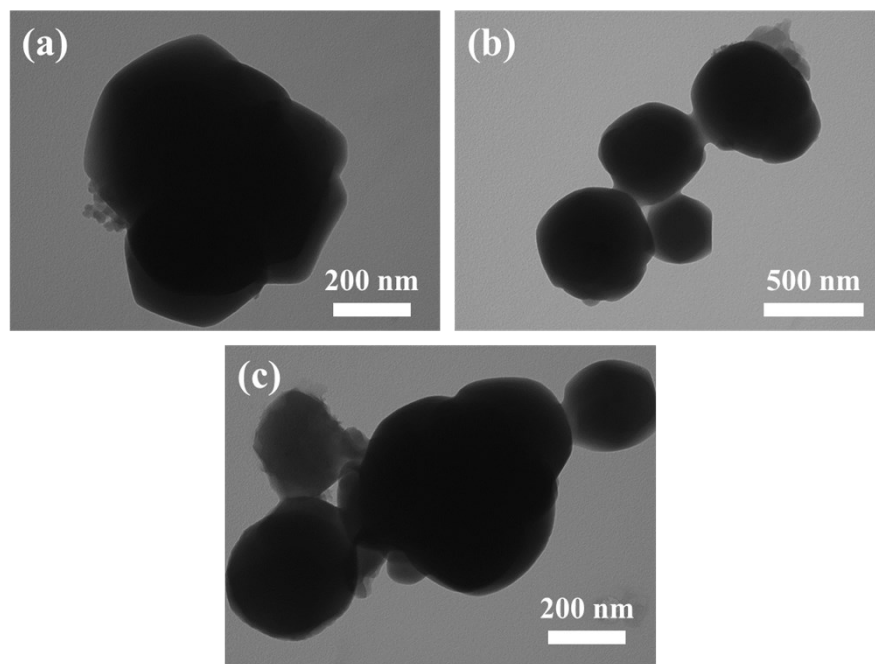


Figure S1: TEM images of $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@ZIF$ after ZIF coating.

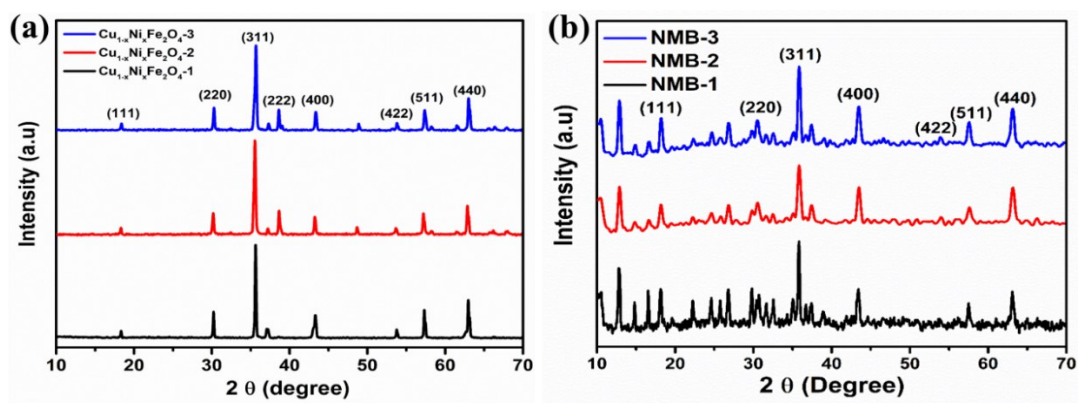


Figure S2: XRD patterns of (a) $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4$ particles, (b) $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@ZIF$ before post-calcination.

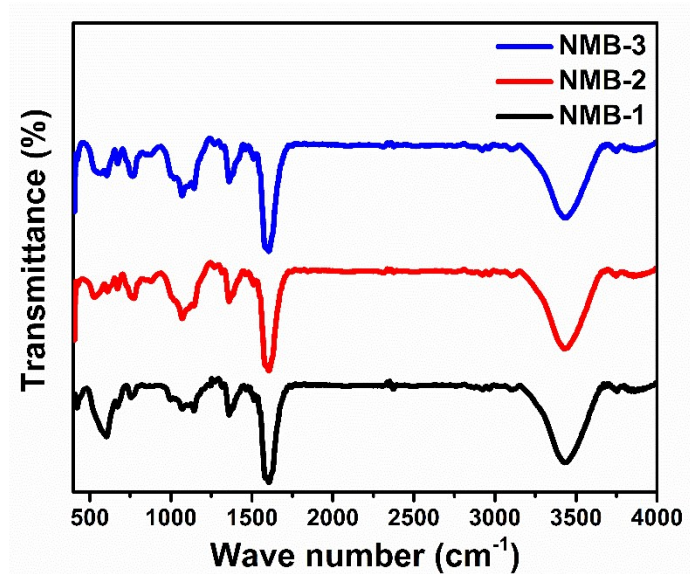


Figure S3: FTIR spectra of $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@\text{ZIF}$ before post-calcination.

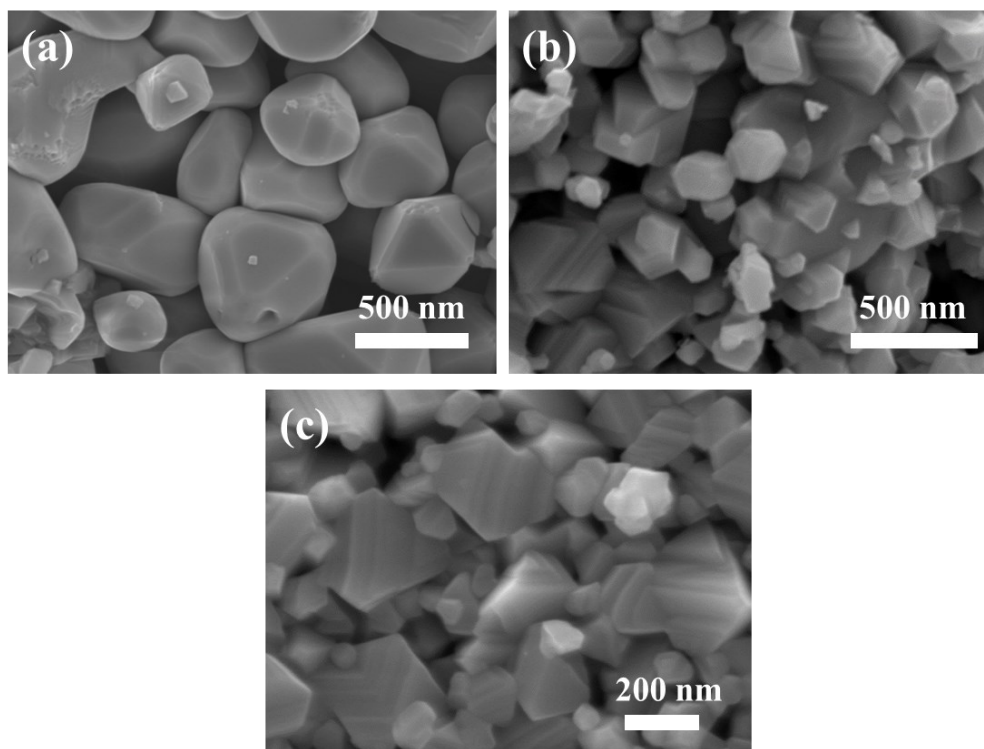


Figure S4: SEM images of as-synthesized $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4$ particles after pre-calcination.

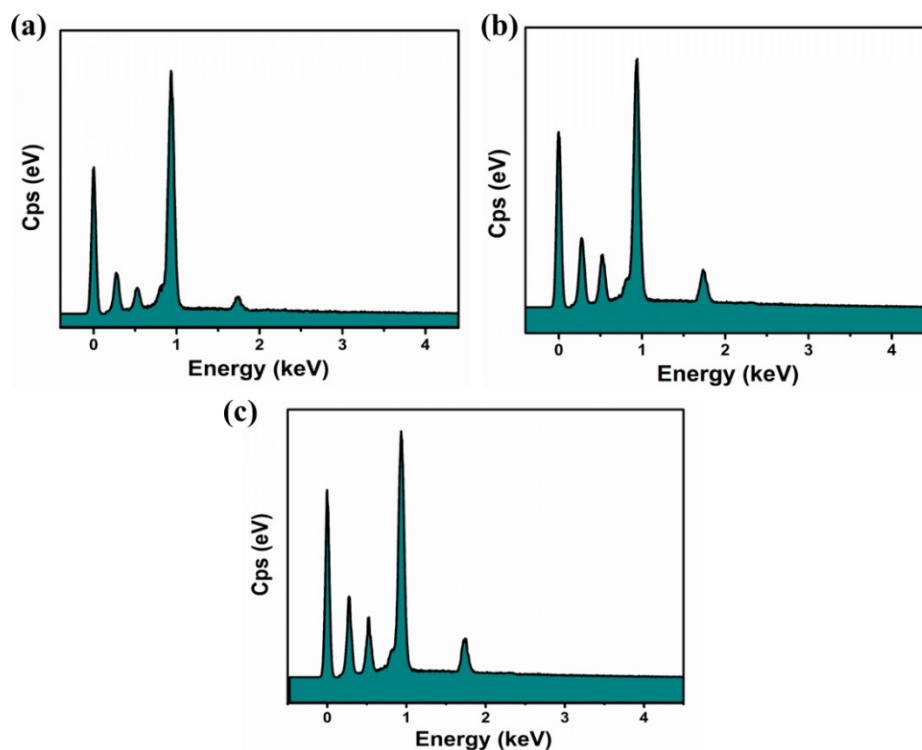


Figure S5: EDX analysis of $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@\text{C}$ after post-calcination.

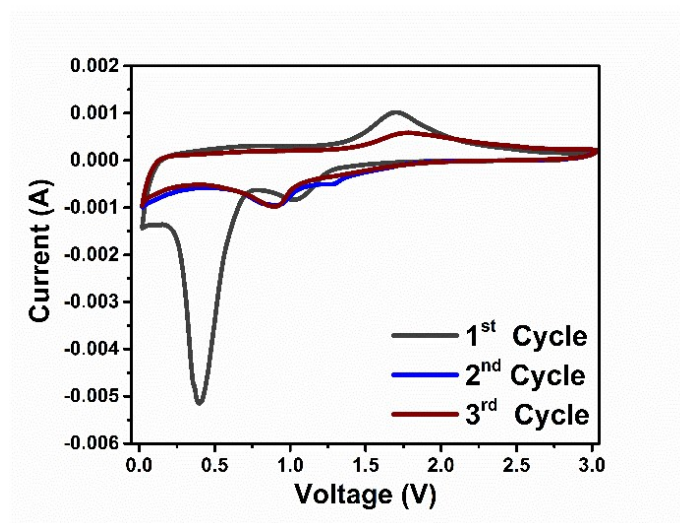


Figure S6: Cyclic voltammetry (CV) of $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@\text{C}$.

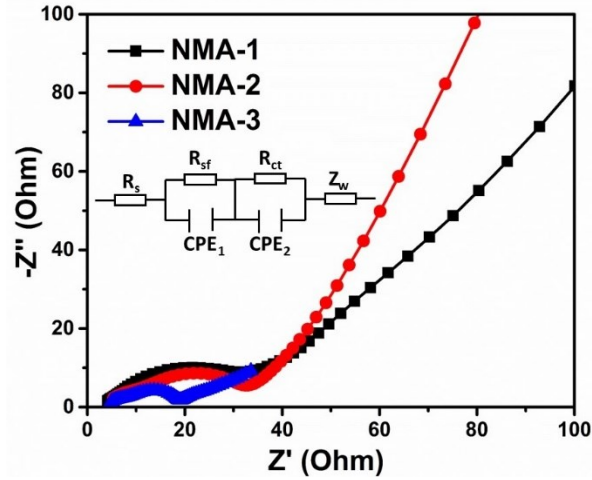


Figure S7: EIS spectra of $\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@\text{C}$ after 50 cycles.

Table R1. Comparison of various anode materials in terms of their electrochemical performances.

Anode	Cycling Stability				Columbic efficiency	Ref
	Initial Discharge capacity [mAh g^{-1}]	Capacity after n^{th} cycles	No of Cycles	Current rate [mA g^{-1}]		
$\text{CuFe}_2\text{O}_4@\text{rGO}$	1078	587	100	100	~66%	1
$\text{CoFe}_2\text{O}_4/\text{graphene}$	1683	501	25	183	53%	2
$\text{CuFe}_2\text{O}_4\text{-graphene}$	1605	687	50	100	66%	3
$\text{CuFe}_2\text{O}_4/\text{rGO}$	1200	845	25	100	73%	4
$\text{NiFe}_2\text{O}_4/\text{rGO}$	1363	1225	100	100	79%	5
$\text{NiFe}_2\text{O}_4/\text{graphene}$	1350	812	50	100	67%	6
$\text{NiFe}_2\text{O}_4@\text{SiO}_2$	1460	690	100	100	71%	7
$\text{CoFe}_2\text{O}_4/\text{C}$	~1700	600	200	185	~70%	8
$\text{NiFe}_2\text{O}_4/\text{graphene}$	1575	407	50	100	72%	9
$\text{NiFe}_2\text{O}_4/\text{MWCNT}$	1305	871	25	100	79%	10
$\text{Cu}_{1-x}\text{Ni}_x\text{Fe}_2\text{O}_4@\text{C}$	1428	722	500	500	70%	Our Work

References:

- [1] Junyong W.; Qinglin D.; Mengjiao L.; Kai J.; Jinzhong Z.; Zhigao H.; Junhao C.; Copper ferrites@ reduced graphene oxide anode materials for advanced lithium storage applications, *Sci. Rep.*, 2017, **7**, 8903.
- [2] Songmei L.; Bo W.; Jianghua L.; Mei Y.; In situ one-step synthesis of CoFe_2O_4 /graphene nanocomposites as high-performance anode for lithium-ion batteries, *Electrochim. Acta*, 2014, **129**, 33.
- [3] Yongsheng F.; Qun C.; Mingyang H.; Yunhai W.; Xiaoqiang S.; Hui X.; Xin W.; Copper ferrite-graphene hybrid: A multifunctional heteroarchitecture for photocatalysis and energy storage, *Ind. Eng. Chem. Res.* 2012, **51**, 11700.
- [4] Sumair A S.; Iftikhar H G.; Hashim N.; Shafiqullah M.; Muhammad M.; Improved performance of CuFe_2O_4 /rGO nanohybrid as an anode material for lithium-ion batteries prepared via facile one-step method, *Current Nanosci.* 2019, **15**, 420.
- [5] Xiaobin T.; Longze Z.; Xiaoping L.; Xi P.; Jianmin Z.; Xiaochuan D.; Qiuhong L.; High index faceted nickel ferrite nanocrystals encapsulated by graphene with high performance for lithium-ion batteries, *Electrochim. Acta*, 2017, **257**, 99.
- [6] Yongsheng F.; Yunhai W.; Hui X.; Xin W.; Nickel ferrite-graphene hetero-architectures: Toward high-performance anode materials for lithium-ion batteries, *J. Power Sources*, 2012, **213**, 338.
- [7] Getong Q.; Xin W.; Jianwu W.; Jing L.; Min Z.; A Core-Shell NiFe_2O_4 @ SiO_2 Structure as a High Performance Anode Material for Lithium-Ion Batteries, *ChemElectroChem*, 2019, **6**, 911.
- [8] Yun H J.; Seung D S.; Hyun W S.; Kyung S. P.; Dong W K.; Synthesis of core/shell spinel ferrite/carbon nanoparticles with enhanced cycling stability for lithium ion battery anodes, *Nanotech.* 2012, **23**, 125402.
- [9] Xuefang C.; Ying H.; Kaichaung Z.; Xuansheng F.; S. Li, Self-assembled flower-like NiFe_2O_4 decorated on 2D graphene nanosheets composite and their excellent electrochemical performance as anode materials for LIBs, *J. Alloys Comp.* 2016, **686**, 905.
- [10] Muhammad M.; Rafi U K.; Muhammad M.; Mubasher.; Sumai A. S.; Shafiq U.; NiFe_2O_4 nanoparticles/MWCNTs nanohybrid as anode material for lithium-ion battery, *Cer. Intl.*, 2019, **45**, 8486.