Supplementary information for

Reinforcement of a BiVO₄ anode by an Fe₂O₃ underlayer for photoelectrochemical water splitting

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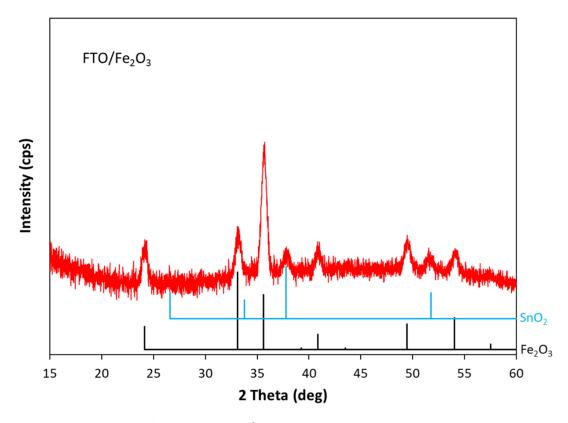


Fig. S1. XRD patterns of FTO/Fe₂O₃, reference α -hematite (JCPDS No. 01-076-8403), and reference tin oxide (JCPDS No. 00-046-1088). Data collected using a Rigaku (SmartLab) X-ray diffractometer. X-ray incident angle at 1° with Cu K α 1 (using 45 kV and 200 mA, λ = 1.540619 Å) radiation.

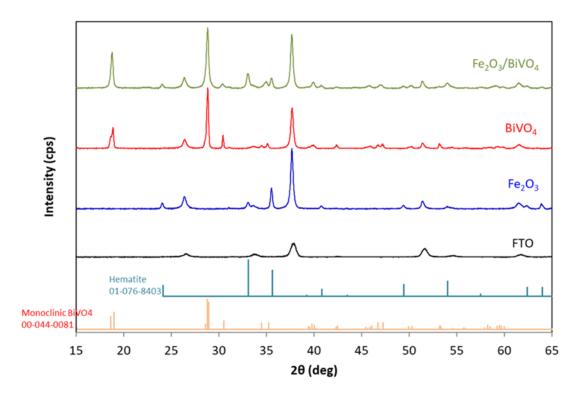


Fig. S2. XRD patterns (from top to bottom) of FTO/Fe₂O₃/BiVO₄, FTO/BiVO₄, FTO/Fe₂O₃, FTO substrate, reference α -hematite (JCPDS No. 01-076-8403), and reference scheelite-structured monoclinic BiVO₄ (JCPDS No. 00-044-0081). Data collected using a Rigaku (MiniFlex 600) X-ray diffractometer.

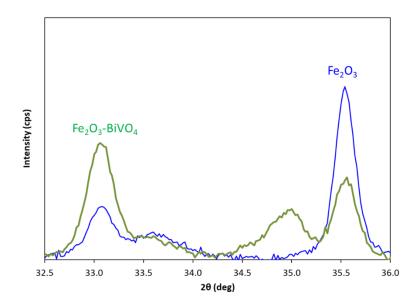
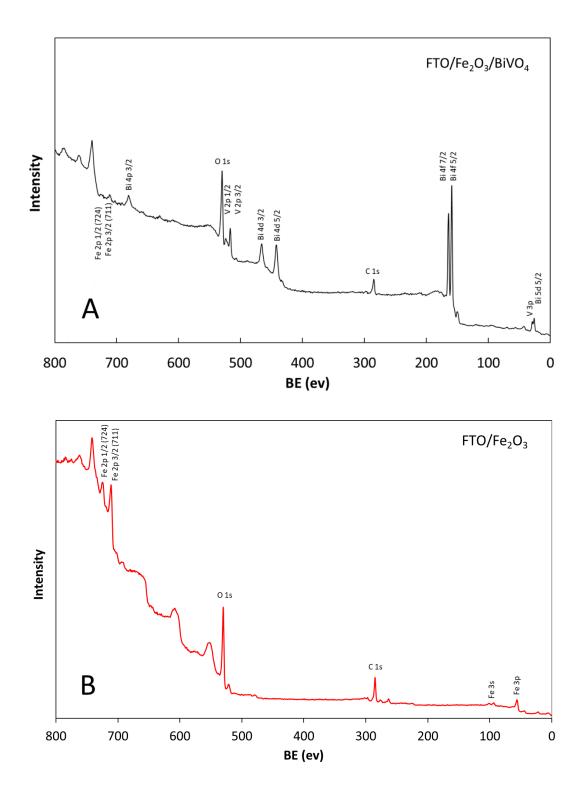


Fig. S3. Partially enlarged XRD patterns of the FTO/Fe_2O_3 film (blue line) and the $FTO/Fe_2O_3/BiVO_4$ film (green line).



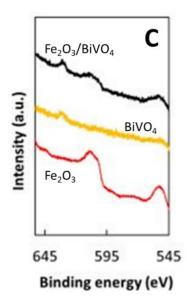
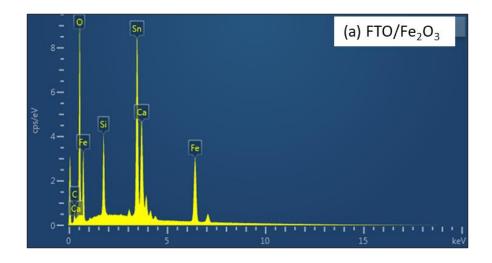
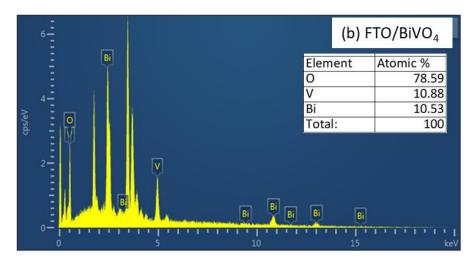


Fig. S4. X-ray photoelectron spectroscopy (XPS) spectra of (A) FTO/Fe₂O₃/BiVO₄ film and (B) FTO/Fe₂O₃ film. (C) the figure shows the XPS analysis of the LMM (Mg K α) peaks of Fe₂O₃. XPS data were collected using a JEOL (JPS-9010MC) X-ray photoelectron spectrometer under ultrahigh vacuum using a monochromatic Mg K α X-ray source, with 25 scans for each element. The adventitious carbon 1s peak was calibrated to 284.5 eV and used as an internal standard to compensate for any charging effects.





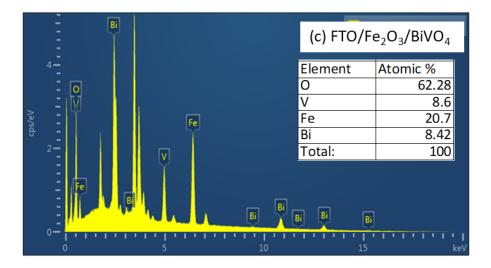


Fig. S5. The energy dispersive X-ray spectroscopy (EDS) elemental spectra of (a) FTO/Fe₂O₃ film, (b) FTO/BiVO₄ film, and (c) FTO/Fe₂O₃/BiVO₄ film.

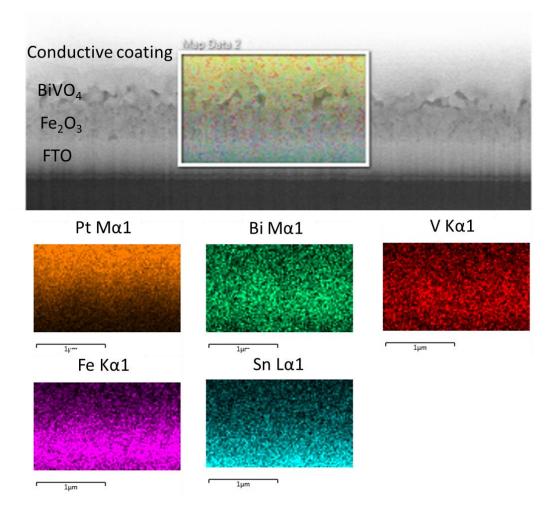


Fig. S6. EDS mapping of cross-sectional image of the FTO/Fe₂O₃/BiVO₄ film. EDS elemental analysis using an FIB-SEM. Platinum (Pt) is the conductive coating for the FIB-SEM measurement.

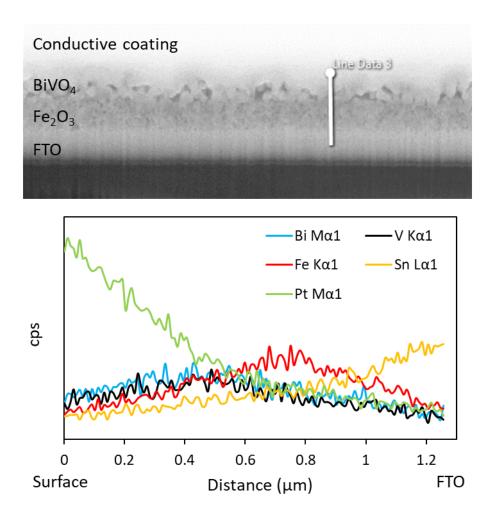


Fig. S7. EDS elemental line analysis of cross-sectional image of the FTO/Fe₂O₃/BiVO₄ film.

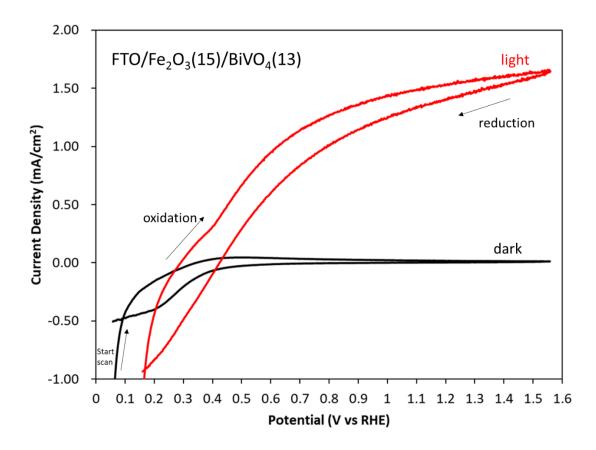


Fig. S8. CV curves of the FTO/Fe₂O₃(15)/BiVO₄(13) anode. The light CV curve is collected in $0.5 \text{ M} \text{ H}_3\text{BO}_3$ adjusted to pH 9.5 with KOH under AM1.5G with 1 sun of front illumination. The scan rate was 20 mV/s.

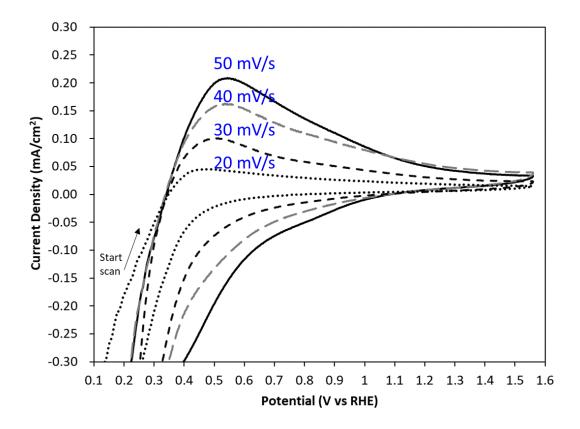


Fig. S9. Dark CV curves of the FTO/Fe₂O₃/BiVO₄ anode at different scan rates.

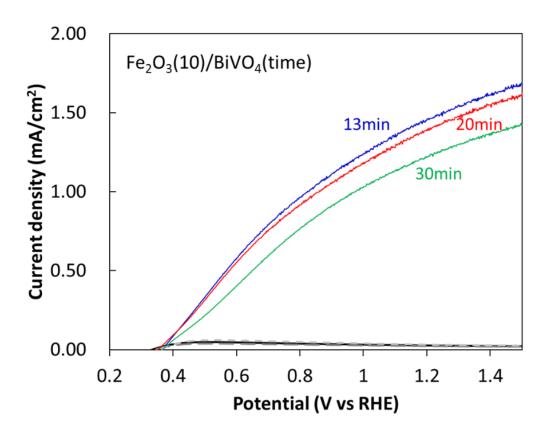


Fig. S10. PEC properties. J-V curves of the FTO/Fe₂O₃(**10**)/BiVO₄(time) anode with deposition duration of the upper layer BiVO₄ for 13, 20 and 30 min. The deposit duration of the Fe₂O₃ underlayer was **10 min**. Measurements were recorded under AM1.5G 1 sun light of front irradiation in a 0.5 M potassium borate (pH 9.5) solution.

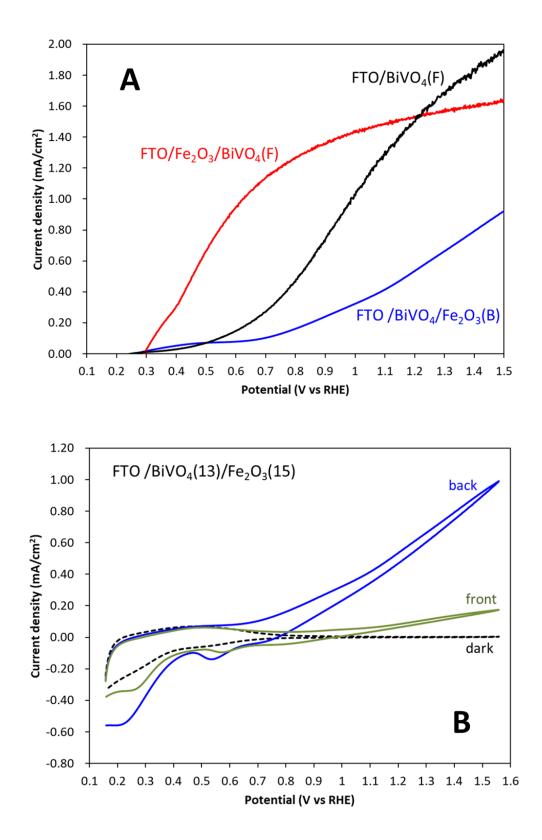


Fig. S11. (A) J-V curves of the FTO/BiVO₄(13)/Fe₂O₃(15), FTO/BiVO₄ and FTO/Fe₂O₃(15)/BiVO₄(13) anodes. (B) CV curves of the FTO/BiVO₄(13)/Fe₂O₃(15) anode.

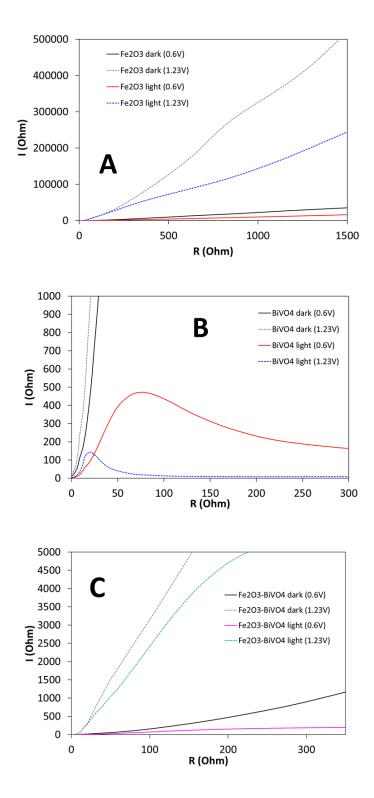


Fig. S12. Nyquist plots of (A) FTO/Fe_2O_3 , (B) $FTO/BiVO_4$ and (C) $FTO/Fe_2O_3/BiVO_4$ anodes. Data were collected in the dark and under one sun light illumination at 0.6 V and 1.23 V *vs*. RHE from 10 kHz to 10 mHz.

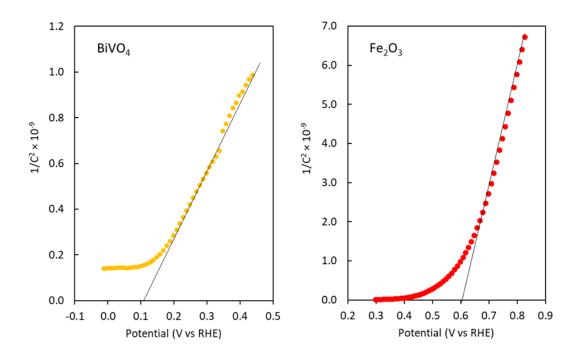


Fig. S13. Mott–Schottky plots for the $FTO/BiVO_4$ and FTO/Fe_2O_3 anodes measured in 0.5 M potassium borate (pH 9.5) solution at 1 kHz in the dark.