Supporting Information

One-Step Synthesis of Graphitic Carbon-Nitride doped with Black-Red Phosphorus as a Novel, Efficient and Free-Metal Bifunctional Catalyst and its Application for Electrochemical Overall Water Splitting

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Figure S1 EDS analysis for (a) BRP-g-C₃N₄ and (b) g-C₃N₄, (c-f). Elemental mapping of (c) C, (d) N, (e) P, and (f) C, N and P altogether elements in BRP-g-C₃N₄ composite



Figure S2 Raman spectra and $I_D\,/I_G$ ratio for g-C_3N_4, and BRP-g-C_3N_4 composite



Figure S3 (a) N_2 adsorption–desorption isotherm and (the insert) corresponding pore size distribution of BRP-g-C₃N₄ composite, and (b) TGA curves of RP, BRP-g-C₃N₄, and g-C₃N₄



Figure S4 XRD spectra of the (a) RP, (b) $g-C_3N_4$, and (c) BRP-g-C₃N₄



Figure S5 Cyclic voltammograms of (a) BRP-g- C_3N_4 , and (b) g- C_3N_4 modified CPE at scan rate 10 mV.s⁻¹ in 1.0 and 0.1 M KOH.



Figure S6 (a) Cyclic voltammograms and (b) electrochemical impedance spectra recorded for BRP-g C_3N_4 , RP modified and bare-CPE in 0.1 M KCl containing 5.0 mM [Fe(CN)₆]^{-3/-4}. Scan rate for CV studies; 10 mV s⁻¹. Inset: Equivalent circuit

Table S1:

Electrodes	R _{ch(ohm)}	R _{s(ohm)}	CPE-T	Chi-Squared
Bare-CPE	1827	32.39	3.0098E-7	3.8E-6
RP-CPE	1141	29.58	4.1E-5	4.0E-4
BRP-g-C ₃ N ₄ -CPE	112	42.6	2.08E-7	1.08E-4



Figure S7 Experimental setup and (the insert) hydrogen bubbles on BRP-g-C₃N₄-CPE.



Figure S8 FE-SEM images of BRP-g- C_3N_4 -CPE (a,b) before ,and (c,d) after chronoamperometry



Figure S9 (a) OER polarization curves, (b) corresponding OER Tafel plots, (c) HER polarization curves and (d) corresponding HER Tafel plots for AgFeO₂-BRP-g-C₃N₄, Fe₂O₃-BRP-g-C₃N₄, Ag₂O-BRP-g-C₃N modified CPE and bare, RuO₂ and Pt/C CPE in 1.0 M KOH with a scan rate of 5.0 mV s⁻¹.