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Electronic supplementary information – Economic and life-cycle assessment of OME₃₋₅ as transport fuel: A comparison of production pathways

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This electronic supplementary information (ESI) includes the data used to conduct the environmental and economic assessment of the OME₃₋₅ fuels, namely the life-cycle inventories (Section A), monetization factors of endpoint protection areas (Section B), and production costs (Section C). Following the OME₃₋₅ infrastructure presented in Fig. 2 of the manuscript, the inventory tables S1–S12 correspond to the CO₂ procurement (SS1), H₂ procurement (SS2), OME₃₋₅ production (SS3), and fuel distribution (SS4) subsystems. This ESI also presents additional results regarding the midpoint impact categories of the main inputs in the OME infrastructure and the sensitivity of CO₂ economic allocation factors on the predicted environmental impacts.

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A. Life cycle inventories and complementary results

Table S1 LCI for H₂ from electrolysis. Values per kg H₂.¹

Input	Value	Unit
Water, deionised	18.04	kg
Electricity*	52.26	kWh
Output	Value	Unit
H ₂	1.00	kg
O ₂	8.00	kg

*In this case the electricity comes either from nuclear, solar or wind resources.

Table S2 LCI for CO₂ captured directly from air (DAC). Values per kg CO₂ captured.²

Input	Value	Unit
Air	1968.61	kg
Natural gas	0.19	m ³
Calcium carbonate, precipitated	0.02	kg
Electricity	0.37	kWh
Output	Value	Unit
CO ₂	1.00	kg

Table S3 LCI for CO₂ captured from biogas. Values per kg CO₂ captured³.

Input	Value	Unit
Biogas	1.15	m ³
Electricity	0.24	kWh
Activated carbon, granular	3.37E-4	kg
Output	Value	Unit
CO ₂ (4 bar)	1.00	kg
Methane	0.60	Nm ³
Hydrogen sulfide	1.24E-5	kg
Methane slip, non-fossil	8.23E-5	kg

Table S4 LCI for CO₂ captured from ammonia production. Values per kg CO₂ captured.³

Input	Value	Unit
Natural gas	0.49	kg
Heavy fuel oil	0.16	kg
Cooling water	0.11	kg
Electricity	0.06	kg
Water	7.32E-4	kg
Nickel	2.85E-4	kg
Solvent, organic	2.44E-5	kg
Output	Value	Unit
CO ₂ (1 bar)	1.00	kg
Ammonia, liquid	0.81	kg
Acetaldehyde	1.01E-6	kg
Acetic acid	6.80E-6	kg
Acetone	9.84E-7	kg
Ammonia	6.59E-8	kg
Arsenic	8.70E-8	kg
Benzene	7.61E-6	kg
Benzo(a)pyrene	3.75E-10	kg
Butane	1.33E-5	kg
Cadmium	2.17E-7	kg
Calcium	5.27E-7	kg
Carbon dioxide	1.19	kg
Carbon monoxide	6.83E-5	kg
Chromium	1.04E-7	kg
Chromium VI	1.06E-9	kg
Cobalt	2.17E-7	kg
Copper	3.23E-7	kg
Dinitrogen monoxide	1.24E-5	kg
Dioxins as 2,3,7,8-tetrachlorodibenzo-p-dioxin	3.54E-15	kg
Ethanol	1.98E-6	kg
Formaldehyde	4.87E-6	kg
Hydrocarbons, apliphatic alkenes	3.95E-6	kg
Hydrocarbons, apliphatic unsaturated	1.98E-7	kg
Hydrocarbons, aromatic	9.84E-7	kg
Hydrogen chloride	9.51E-6	kg
Hydrogen fluoride	9.51E-7	kg
iron	1.19E-6	kg
Lead	3.46E-7	kg
Mercury	1.56E-9	kg
Methane	9.76E-6	kg

Methanol	3.36E-6	kg
Molybdenum	1.06E-7	kg
Municipal solid waste	1.63E-4	kg
Nickel	4.28E-6	kg
Nitrogen	8.13E-5	kg
Nitrogen oxides	8.13E-4	kg
Polycyclic aromatic hydrocarbons	1.94E-7	kg
Particulates, < 2.5 μ m	2.34E-4	kg
Particulates, > 2.5 μ m	6.59E-5	kg
Particulates, > 10 μ m	3.29E-5	kg
Pentane	2.28E-5	kg
Propane	4.00E-6	kg
Propionic acid	3.80E-7	kg
Selenium	7.90E-8	kg
Sodium	4.94E-6	kg
Sulfur dioxide	8.13E-6	kg
Toluene	4.00E-6	kg
Vanadium	1.72E-5	kg
Water	0.04	kg
Water (water)	0.07	kg
Zinc	2.63E-8	kg

Table S5 LCI for CO₂ captured from a coal-based power plant. Values per kg CO₂ captured.¹

Input	Value	Unit
Coal	0.52	kg
Limestone	0.04	kg
Light fuel oil	6.20E-3	kg
Natural gas	7.80E-4	m ³
Monoethanolamine	1.55E-3	kg
Sodium hydroxide	1.20E-4	kg
Ammonia	1.15E-3	kg
Output	Value	Unit
CO ₂ (1 bar)	1.00	kg
Electricity	0.78	kWh
CO ₂ (air)	0.05	kg
Sulphur dioxide	7.00E-3	kg
Nitrogen oxide	1.60E-3	kg
Particulates, unspecified (air)	1.10E-4	kg
Ammonia	2.70E-4	kg
Monoethanolamine	9.00E-5	kg
Municipal solid waste to sanitary landfill	4.06E-3	kg

Table S6 LCI for CO₂ captured from a natural gas-based power plant. Values per kg CO₂ captured.⁴

Input	Value	Unit
Natural gas	0.62	Nm ³
Monoethanolamine	1.50E-3	kg
Sodium hydroxide	1.30E-4	kg
Activated carbon	7.50E-5	kg
Water	8.00E-4	m ³
Output	Value	Unit
CO ₂ (1 bar)	1.00	kg
Electricity	2.75	kWh
CO ₂ (air)	0.11	kg
Monoethanolamine	6.30E-5	kg
Ammonia	3.50E-5	kg
Formaldehyde	2.62E-4	g
Acetaldehyde	1.67E-4	g
Toxic solid waste to landfill	1.65E-3	kg

Table S7 LCI for methanol production from H₂ and CO₂. Values per kg methanol.¹

Input	Value	Unit
CO ₂	1.46	kg
H ₂	0.20	kg
Electricity	0.21	kWh
Cooling water	0.08	m ³
Copper oxide	2.11E-5	kg
Zinc oxide	7.93E-6	kg
Aluminium oxide	3.96E-6	kg
Output	Value	Unit
Methanol	1.00	kg
CO ₂ (air)	0.08	kg
NO ₂ (air)	1.78E-4	kg
Methanol (air)	0.01	kg
Wastewater	5.71E-4	m ³

Table S8 LCI for methanol production from natural gas via SMR.⁵

Input	Value	Unit
Natural gas	0.65	m ³
Electricity	0.07	kWh
Steam	6.93	MJ
Cooling water	8.16E-03	m ³
Water, deionised	0.85	kg
Copper oxide	9.00E-5	kg
Molybdenum	1.00E-5	kg
Zinc	3.00E-5	kg
Aluminium oxide	2.40E-04	kg
Nickel	2.00E-05	kg
Output	Value	Unit
Methanol	1.00	kg
Methane	9.80E-04	kg
CO ₂	0.08	kg
NO ₂	1.78E-04	kg
NO _X	1.50E-04	kg
Sulfur dioxide	1.38E-05	kg
AOX, Adsorbable Organic Halogen as Cl, water	1.00E-06	kg
BOD ₅ , Biological Oxygen Demand, water	1.80E-04	kg
Chloride, water	2.00E-06	kg
COD, Chemical Oxygen Demand, water	4.90E-04	kg
DOC, Dissolved Organic Carbon, water	2.40E-04	kg
Formaldehyde, water	1.00E-04	kg
Methanol, water	3.00E-05	kg
Phenol, water	1.00E-05	kg
Phosphorus, water	1.00E-05	kg
Suspended solids, water	2.00E-05	kg
Total organic carbon, water	2.40E-04	kg
Wastewater	5.67E-03	m ³

Table S9 LCI for methanol production from biomass-based syngas.⁵

Input	Value	Unit
Wood chips dry	0.72	kg
Wood chips wet	2.60	kg
Electricity	0.28	kWh
Steam	6.93	MJ
Cooling water	8.16E-03	m ³
Water, deionised	0.85	kg
Copper oxide	9.00E-5	kg
Molybdenum	1.00E-5	kg
Zinc	3.00E-5	kg
Aluminium oxide	2.40E-04	kg
Nickel	2.00E-05	kg
Output	Value	Unit
Methanol	1.00	kg
Methane	9.80E-04	kg
CO ₂ , biogenic	2.76	kg
CO, biogenic	6.51E-06	kg
Methane, biogenic	6.20E-06	kg
NO ₂	3.10E-07	kg
NO _x	2.06E-04	kg
Methanol	5.30E-04	kg
Acetaldehyde	3.10E-09	kg
Acetic acid	4.65E-07	kg
Benzene	1.24E-06	kg
Benzo(a)pyrene	3.10E-11	kg
Butane	2.17E-06	kg
Formaldehyde	3.10E-07	kg
Mercury	9.30E-11	kg
polycyclic aromatic hydrocarbons	3.10E-08	kg
Particulates, < 2.5 μm	6.20E-07	kg
Pentane	3.72E-06	kg
Propane	6.20E-07	kg
Propionic acid	6.20E-08	kg
Sulfur dioxide	1.71E-06	kg
Toluene	6.20E-07	kg
AOX, Adsorbable Organic Halogen as Cl, water	1.00E-06	kg
BOD ₅ , Biological Oxygen Demand, water	1.00E-06	kg
Chloride, water	2.00E-06	kg
COD, Chemical Oxygen Demand, water	4.90E-04	kg
DOC, Dissolved Organic Carbon, water	2.40E-04	kg
Formaldehyde, water	1.00E-04	kg
Methanol, water	3.00E-05	kg
Phenol, water	1.00E-05	kg
Phosphorus, water	1.00E-05	kg
Suspended solids, water	2.00E-05	kg
Total organic carbon, water	2.40E-04	kg
Wastewater	5.32E-03	m ³

Table S10 LCI for formaldehyde production from methanol. Values per kg formaldehyde.⁵

Input	Value	Unit
Methanol	1.20	kg
Electricity	0.15	MJ
Output	Value	Unit
Formaldehyde	1.00	kg
Steam ^a	6.60	MJ
CO ₂	0.11	kg
Formaldehyde, air	2.50E-05	kg
Methanol, air	3.10E-05	kg
NO _x	3.80E-04	kg
NMVOC	3.00E-06	kg
Particulates, < 2.5 μm and > 10 μm	5.00E-06	kg
BOD ₅ , Biological Oxygen Demand, water	8.00E-06	kg
COD, Chemical Oxygen Demand, water	8.90E-06	kg
Catalyst waste	5.00E-05	kg

^aIn our case, this steam is transferred to OME₃₋₅ process in order to partially satisfy its heat requirement.

Table S11 LCI for distribution of OME₃₋₅. Values per kg OME₃₋₅.³

Input	Value	Unit
Transport, pipeline, onshore	0.56	tkm
Transport, freight, lorry	0.06	tkm
Transport, freight train	0.04	tkm
Transport, freight, inland waterways, barge tanker	0.03	tkm
Transport, freight, light commercial vehicle	1.80E-3	tkm
Water	6.89E-4	kg
Steam	5.84E-4	MJ
Electricity	6.70E-3	kWh
Output	Value	Unit
OME ₃₋₅	1.00	kg
Water (air)	1.03E-7	m ³
Water (water)	5.86E-7	m ³
Municipal solid waste	6.27E-6	kg
Wastewater	6.89E-7	m ³

Table S12 LCI for Diesel⁶ and OME₃₋₅³ utilization phase. Values in kg per km driven.

Input	Diesel	OME₃₋₅
Fuel consumption	0.05	0.12
Output (emissions)	Diesel	OME₃₋₅
Carbon dioxide	0.17	0.20
Sulfur dioxide	1.06E-6	0.00
Cadmium	5.28E-10	0.00
Copper	8.98E-8	0.00
Chromium	2.64E-9	0.00
Nickel	3.70E-9	0.00
Zinc	5.28E-8	0.00
Lead	4.36E-15	0.00
Selenium	5.28E-10	0.00
Mercury	1.06E-12	0.00
Chromium IV	5.28E-12	0.00
Dinitrogen monoxide	2.64E-6	2.78E-6
Ammonia	8.45E-7	0.00
Polycyclic aromatic hydrocarbons	9.74E-9	1.03E-9
Carbon monoxide	6.07E-5	0.00
Nitrogen oxides	9.38E-6	6.77E-4
Particulates, < 2.5 μm	4.02E-6	0.00
Methane	1.87E-6	1.87E-6
Ethane	8.22E-8	8.22E-8
Propane	2.74E-8	2.74E-8
Butane	2.74E-8	2.74E-8
Pentane	9.96E-9	9.96E-9
Heptane	4.98E-8	4.98E-8
Cyclohexane	1.62E-7	1.62E-7
Ethene	2.73E-6	2.73E-6
Propene	8.96E-7	8.96E-7
Formaldehyde	2.99E-6	0.00
Acetaldehyde	1.61E-6	1.61E-6
Acrolein	8.91E-7	8.91E-7
Benzaldehyde	2.14E-7	2.14E-7
Acetone	7.32E-7	7.32E-7
Methyl ethyl ketone	2.99E-7	2.99E-7
Toluene	1.72E-7	1.72E-7
m-Xylene	1.52E-7	1.52E-7
o-Xylene	6.72E-8	6.72E-8
Styrene	9.21E-8	9.21E-8
Benzene	4.93E-7	4.93E-7
others NMVOC from diesel	1.47E-5	1.32E-5

Midpoint impact categories

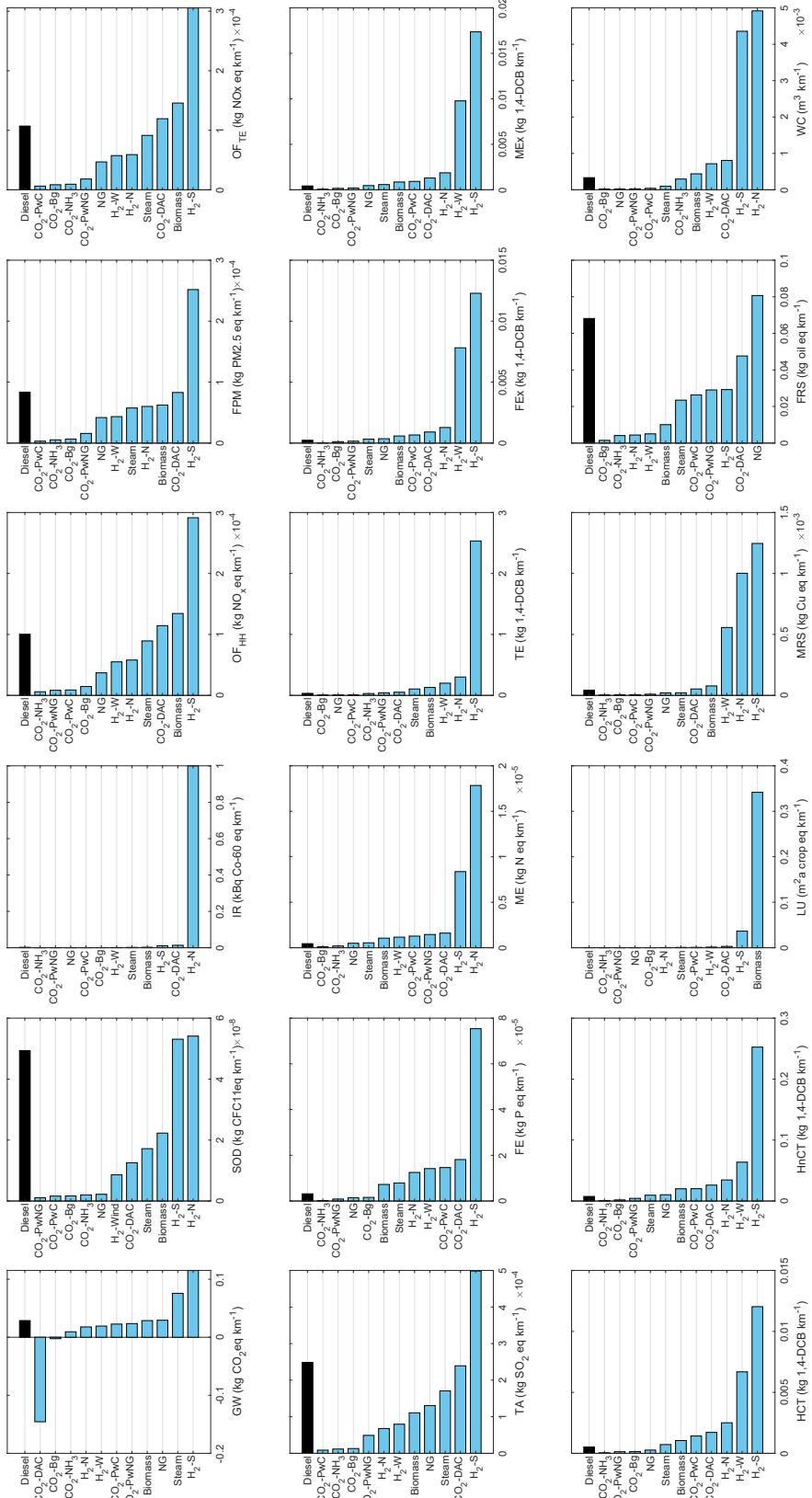


Fig. S1 Midpoint environmental impacts for the main inputs in the OME structure. GW:Global warming, SOD:Stratospheric ozone depletion, IR:Ionizing radiation, OF_{HH}:Ozone formation human health, FPM:Fine particulate matter formation, OF_{TE}:Ozone formation terrestrial ecosystem, TA:terrestrial acidification, FE:Fresh water eutrophication, ME:Marine eutrophication, TE:Terrestrial eutrophication, FEx:Fresh water ecotoxicity, MEx:Marine ecotoxicity, HCT:Human carcinogenic toxicity, HnCT:human non-carcinogenic toxicity, LU:Land use, MRS:Mineral resource scarcity, FRS:fossil resource scarcity, WC: Water consumption.

Effect of CO₂ allocation factor

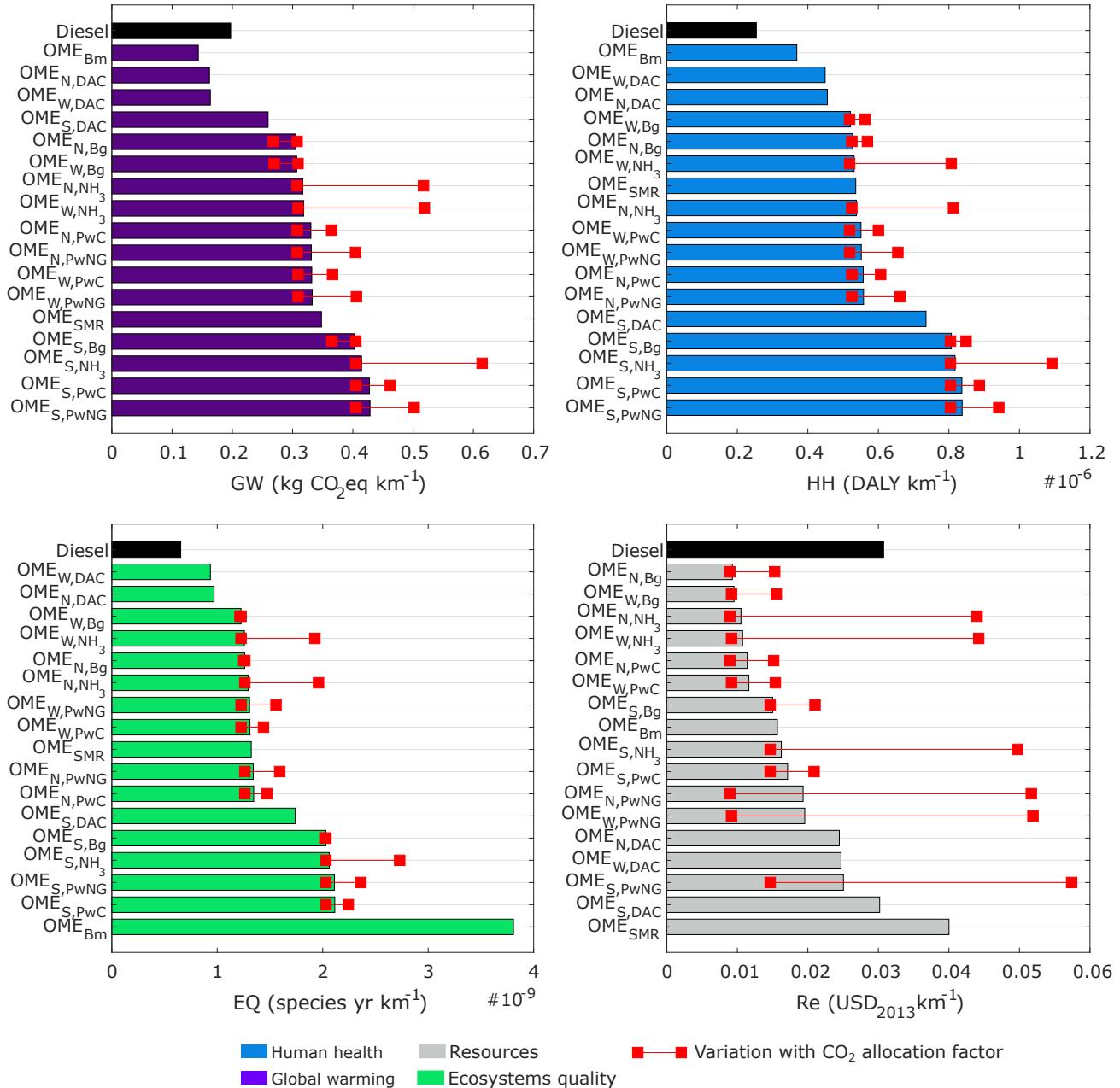


Fig. S2 Variation of GW and the endpoint protection areas (HH, EQ and Re) with the allocation factor used in the CO₂ subsystems presenting multifunctionality within the OME₃₋₅ structure.

B. Monetary valuation

Monetary valuation is employed to express the environmental endpoint impacts into monetary values, which allow us to compare those impacts against each other and with the driving cost. Thus, making possible to express the environmental indicators into a single score. Table S13 includes the conversion factors employed in our study, which are based on the values reported in the work by Weidema⁷ who compared the Eco-Indicator 99, Stepwise 2006 and ReCiPe 2008 life cycle impact assessments methods in an endpoint perspective.

Table S13 Monetary valuation factors for the three environmental endpoint impacts.⁷

Endpoint impact	Unit	Monetization factor
Human health	USD2019/DALY	116,345
Ecosystems quality	USD2019/Species·yr	14,936,175
Resources scarcity	USD2019/USD2013	1.10

C. Economic information

Table S14 Production costs employed to compute the driving cost in the OME₃₋₅ structure.

Component	Value	Unit
CO ₂ direct air capture ¹	0.16	USD/kg
CO ₂ biogas production ¹	0.02	USD/kg
CO ₂ ammonia production ⁸	0.01	USD/kg
CO ₂ coal-based power plant ¹	0.05	USD/kg
CO ₂ natural gas power plant ¹	0.08	USD/kg
H ₂ nuclear-based electrolysis ¹	4.63	USD/kg
H ₂ solar-based electrolysis ¹	8.87	USD/kg
H ₂ wind-based electrolysis ¹	5.24	USD/kg
Natural gas [?]	0.33	USD/m ³
Biomass (wood chips) [?]	0.17	USD/kg
Steam ⁹	4.77	USD/MJ
Electricity ¹	0.10	USD/kWh
Cooling water ⁹	0.02	USD/m ³
Wastewater ⁹	0.04	USD/m ³
Driving cost diesel ^{*?}	0.06	USD/km

^{*}Driving cost considers an oil price of 60 USD per bbl.

Notes and references

- 1 A. González-Garay, M. S. Frei, A. Al-Qahtani, C. Mondelli, G. Guillén-Gosálbez and J. Pérez-Ramírez, *En. Env. Sc.*, 2019, **12**, 3425–3436.
- 2 D. W. Keith, G. Holmes, D. S. Angelo and K. Heidel, *Joule*, 2018, **2**, 1573–1594.
- 3 C. Hank, L. Lazar, F. Mantei, M. Ouda, R. J. White, T. Smolinka, A. Schaadt, C. Hebling and H.-M. Henning, *Sust. En. Fuels*, 2019, **3**, 3219–3233.
- 4 B. Singh, A. H. Strømman and E. Hertwich, *Int. J. Greenhouse Gas Cont.*, 2011, **5**, 457–466.
- 5 G. Wernet, C. Bauer, B. Steubing, J. Reinhard, E. Moreno-Ruiz and B. Weidema, *Int. J. Life Cycle Assess.*, 2016, **21**, 1218–1230.
- 6 A. Simons, *Int. J. Life Cycle Assess.*, 2016, **21**, 1299–1313.
- 7 B. P. Weidema, *J. Ind. Ecol.*, 2015, **19**, 20–26.
- 8 P. Bains, P. Psarras and J. Wilcox, *Prog. En. Comb. Sc.*, 2017, **63**, 146–172.
- 9 R. Turton, R. C. Bailie, W. B. Whiting and J. A. Shaeiwitz, *Analysis, Synthesis and Design of Chemical Processes*, Pearson Education, New Jersey, 2008.