Supplementary Information

Tuning the Synergistic Effects of MoS₂ and Spinel NiFe₂O₄ Nanostructures for High Performance Energy Storage and Conversion Applications

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Figures



Figure S1: (a) EDS spectrum and (b) Elemental mapping image of the MN5



Figure S2: (a) Low (b) High resolution TEM images and (c) SAED pattern of NFO nanoparticles (d) Low (e) High resolution TEM images and (f) SAED pattern of MS nanoflowers



Figure S3: (a) Typical nitrogen adsorption-desorption isotherms (b) pore size distributions obtained using BJH method



Figure S4: CV curves of (a) NFO (b) MS (c) MN1 (d) MN5 and (e) MN10



Figure S5: GCD curves of (a) NFO (b) MS (c) MN1 (d) MN5 and (e) MN10

Tables

Table ST1: Comparison of the specific capacitance, energy density, power density values ofour $MoS_2/NiFe_2O_4$ nanocomposites-based electrodes with the recent literature values

Material		Elect	rolyte	Capa	acitance	Energy Density		Power Density		Ref.	
MoS ₂ nanoflowers		PVA- Na ₂ SO ₄		90 F/g at 1 A/g		16.4 Wh/kg at 0.2 A/g		1.6 kW/kg at 2 A/g		1	
MoS ₂ nanostructures		1 M KOH		244 F/g at 1 A/g		12.2 Wh/kg at 1 A/g		-		2	
MoS ₂ nanosheets		1 M Li ₂ SO ₄		47.3 F/g at 1 mA		6.56 Wh/kg at 0.5 A/g		250 W/kg at 0.5 A/g		3	
MoS ₂ /corncob-derived activated carbon		1 M Na ₂ SO ₄		38.3 F/g at 1 A/g		7.6 Wh/kg		608 W/kg		4	
MoS ₂ -Ti ₃ C ₂ T _x		1 M F	H ₂ SO ₄ 115.2		F/g at 0.5 A/g	5.1 Wh/kg		298 W/kg		5	
CNT@MoS ₂ /PDDA/PMo ₁₂		1 M F	H_2SO_4	110 F/§	g at 0.5 A/g	15.27	Wh/kg	4782 W/kg		6	
MoS ₂ nanosheets		0.5 TEA	M 14.75		F/g at 0.5 A/g	18.43 Wh/kg at 0.75 A/g		7500 W/kg		7	
AC/MoS ₂		1 M N	a ₂ SO ₄ 1		79 F/g	21 W	/h/kg	225 W/kg		8	
MoS ₂ nanoworms		1 M N	a_2SO_4	138 F/	/g at 1 A/g	12.26	Wh/kg	7.98 kW/kg		9	
	1T MoS ₂		K ₂ SO ₄		80 F/g at 5	5 mV/s 4.19		Wh/kg 225		W/kg	10
	MoS2/PANI/rGO 1 Hollow carbon-MoS2- carbon nanoplates 1 MoS2-NH2/PANI 1 MoS2 /NiFe2O4 (MN5) 1		1 M I	H ₂ SO ₄ 97.8 F/g		t 2 A/g		-		-	11
			1 M Li ₂ SO ₄		248 F/g at 0.1 A/g		78 Wh/kg		3806 W/kg		12
			1 M I	H_2SO_4	58.6 F/g at 0.5 A/g		3.5 Wh/kg		14 kW/kg		13
			1 М Л	Ja SO	246.68 F/g at 0.		at 0.5		1914.02 W/Kg at 0.5 A/g		Thi
				$ a_2SO_4 $ A/g		-	9.47 V at 4	Wh/Kg A/g	179 W/K A	72.10 g at 4 ./g	Woi

Table ST2: $MoS_2/NiFe_2O_4$ nanocomposites show the competitive electrocatalyticperformances for HER compared with the current noble-metal-free catalysts

Material	Electrode	Electrolyte	Overpotential (mV) @ 10 mA/cm ²	Tafel Slope mV/dec	Ref.
SrTiO ₃ @MoS ₂	Nickel Foam	1 М КОН	165	81.41	14
Ni-Fe _x P	Nickel Foam	1 M KOH	119	80	15
MoS ₂ -ZnO-Ni	Nickel Foam	1 M KOH	129	78	16
Cu@Ni-P@a-MoS ₂	Nickel Foil	0.5 M H ₂ SO ₄	186	60.5	17
Co ₃ S ₄ @ MoS ₂ -Ni ₃ S ₂	Nickel Foam	1 M KOH	136	69	18
ZnNi–P	Nickel Foam	1 M KOH	175	129	19
NiS-Ni ₂ P ₂ S ₆	Nickel Foam	1 M KOH	140	72.8	20
MoS ₂ /NiFe ₂ O ₄ (MN5)	Nickel Foam	0.5 M KOH	125	92.3	This Work

Table ST3: Comparison of OER properties of $MoS_2/NiFe_2O_4$ catalysts with relevant catalytic materials

Material	Material Electrode		Overpotential	Tafel Slope mV/dec	Ref.
VOx/NiS	Nickel Foam	1 M KOH	330 mV at 50 mA/cm ²	121	21
Fe-Co ₉ S ₈	Nickel Foam	1 M KOH	270 mV at 10 mA/cm ²	70	22
CuCo-Ni ₃ S ₂	Nickel Foam	1 M KOH	400 mV at 100 mA/cm ²	94.9	23
NiCo ₂ O ₄	Nickel Foam	1 M KOH	271 mV at 10 mA/cm ²	172	24
MoS ₂ /NiCo ₂ O ₄	Nickel Foam	1 M KOH	322 mV at 50 mA/cm ²	113	25
Ni ₃ S ₂ NWs	Nickel Foam	1 М КОН	317 mV at 10 mA/cm ²	53.3	26
MoS ₂ /NiFe ₂ O ₄ (MN5)	Nickel Foam	0.5 M KOH	300 mV at 50 mA/cm²	96.4	This Work

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