

Efficient utilization of crude bio-oil: synthesizing of nitrogen-doped hierarchically porous carbons as electrocatalysts for oxygen reduction reaction

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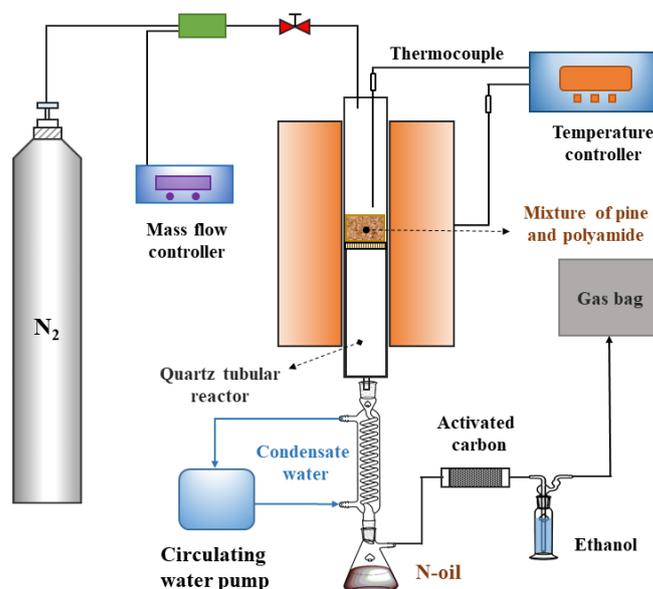


Fig. S1 Schematic diagram of the fixed bed for N-oil production

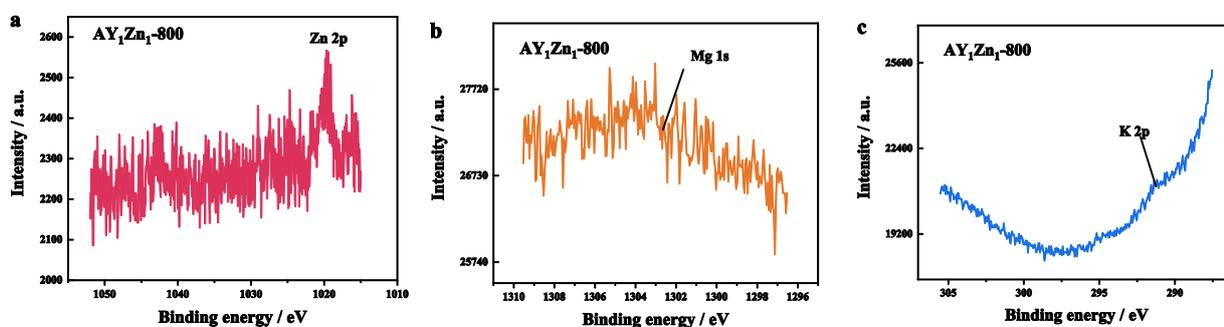


Fig. S2 (a) High-resolution Zn2p spectrum (b) High-resolution Mg1s spectrum (c) High-resolution K2p spectrum of AY₁Zn₁-800. The content of Zn, Mg, K were 0.33, 0.69 and 0.24 at%, respectively.

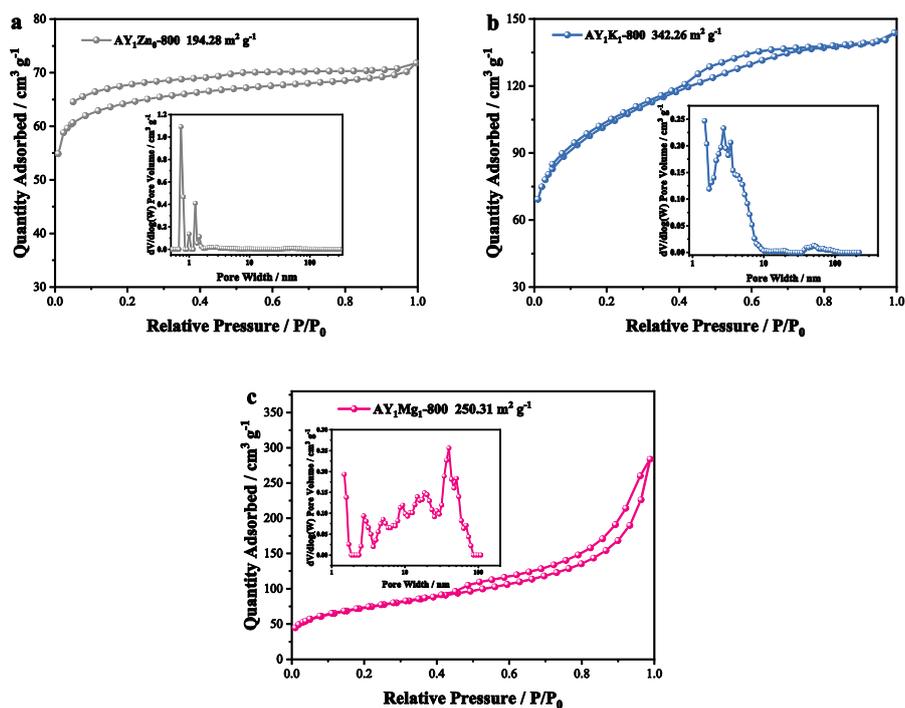


Fig. S3 N₂ adsorption-desorption isotherms of (a) AY₁Zn₀-800; (b) AY₁K₁-800 and (c) AY₁Mg₁-800. The insets are corresponding pore size distribution curves.

Table S1 Parameters from N₂ adsorption-desorption analysis

| Sample | S _{BET} m ² g ⁻¹ | S _{micro} ^[a] m ² g ⁻¹ | S _{meso} m ² g ⁻¹ | V _{total} ^[b] cm ³ g ⁻¹ | V _{micro} ^[a] cm ³ g ⁻¹ | V _{meso} cm ³ g ⁻¹ | V _{meso} / V _{micro} | V _{meso} /V _{total} % | D _{average} ^[c] nm | Porosity ^[d] % |
|--------------------------------------|--|---|---|--|--|--|---|--|---|------------------------------|
| AY ₁ Zn ₀ -800 | 194.28 | 155.93 | 38.35 | 0.110 | 0.082 | 0.028 | 0.34 | 25.45 | 2.84 | 66.92 |
| AY ₁ K ₁ -800 | 343.26 | 108.20 | 235.06 | 0.222 | 0.055 | 0.167 | 3.05 | 75.30 | 3.17 | 78.46 |
| AY ₁ Mg ₁ -800 | 250.31 | 41.43 | 208.88 | 0.439 | 0.021 | 0.418 | 19.90 | 95.22 | 8.42 | 88.28 |

[a] Surface area and pore volume of micropores determined by t-plot method; [b] Total pore volume of pores at P/P₀=0.99; [c] BJH desorption average pore diameter; [d] Porosity=V_{total}/(V_{total}+ Mass of tested sample/ Density of carbon), here the density of carbon is 1.99 g cm⁻³.

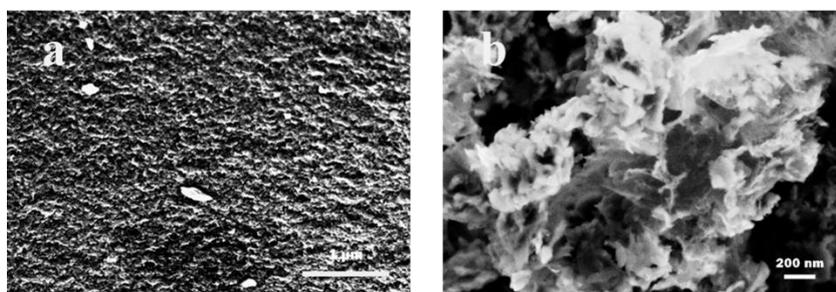


Fig. S4 The SEM images of (a) AY₁K₁-800 and (b) AY₁Mg₁-800.

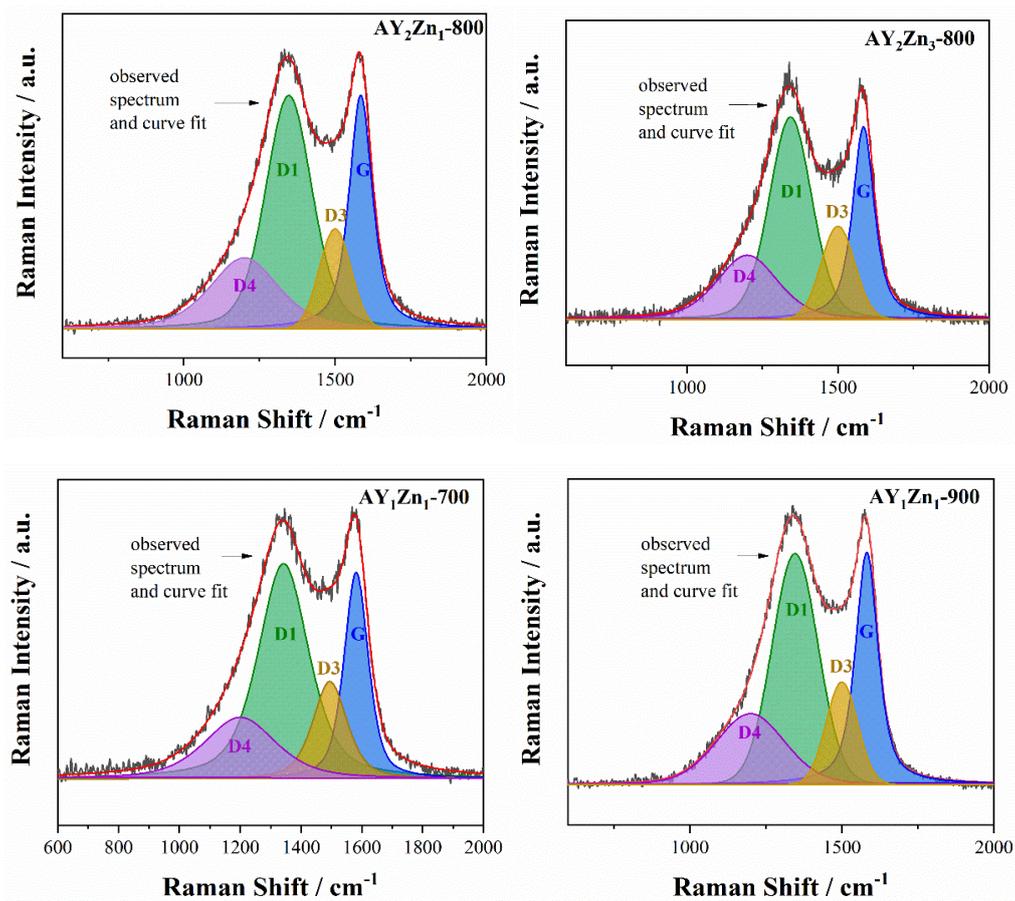


Fig. S5 Fitted Raman spectra of other samples

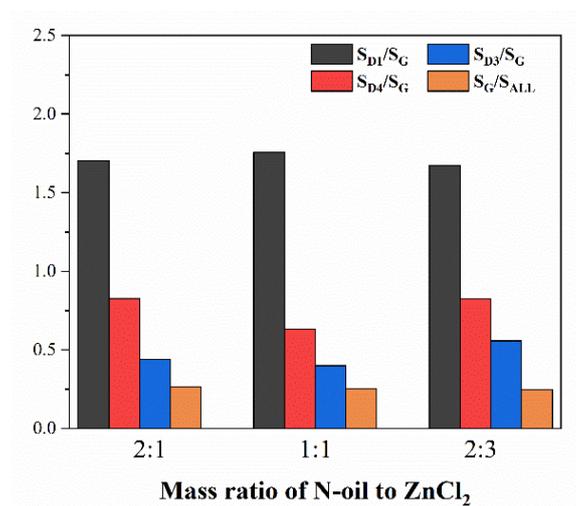


Fig. S6 The values of S_{D1}/S_G, S_{D3}/S_G, S_{D4}/S_G and S_G/S_{ALL} of samples prepared at different mass ratio of N-oil/ZnCl₂.

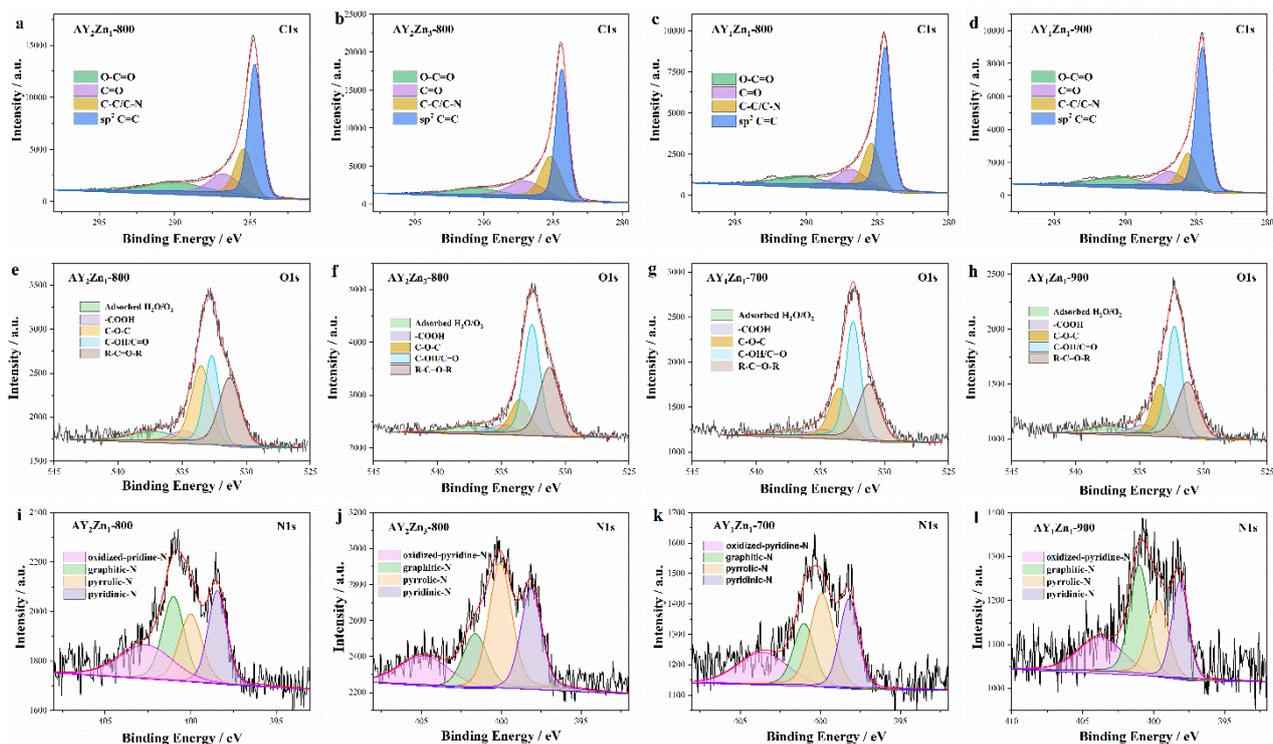


Fig. S7 (a-d) High-resolution C1s spectra of other samples; (e-h) High-resolution O1s spectra of other samples; (i-l) High-resolution N1s spectra of other samples.

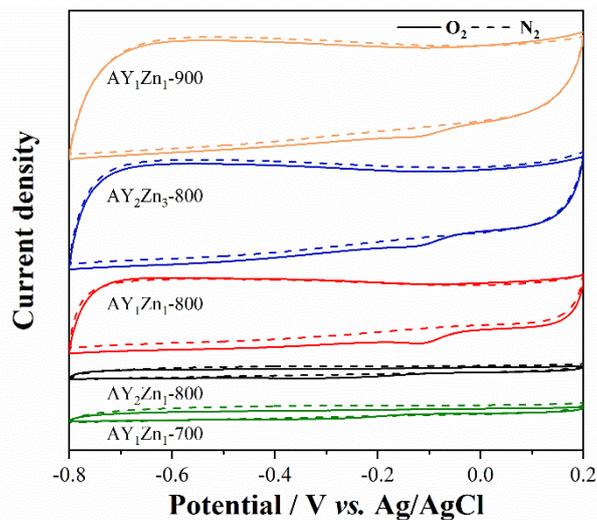


Fig. S8 CV curves of AY_xZn_T in O_2 -saturated and N_2 -saturated 0.1 M KOH electrolyte at a scan rate of 10 mV s^{-1} .

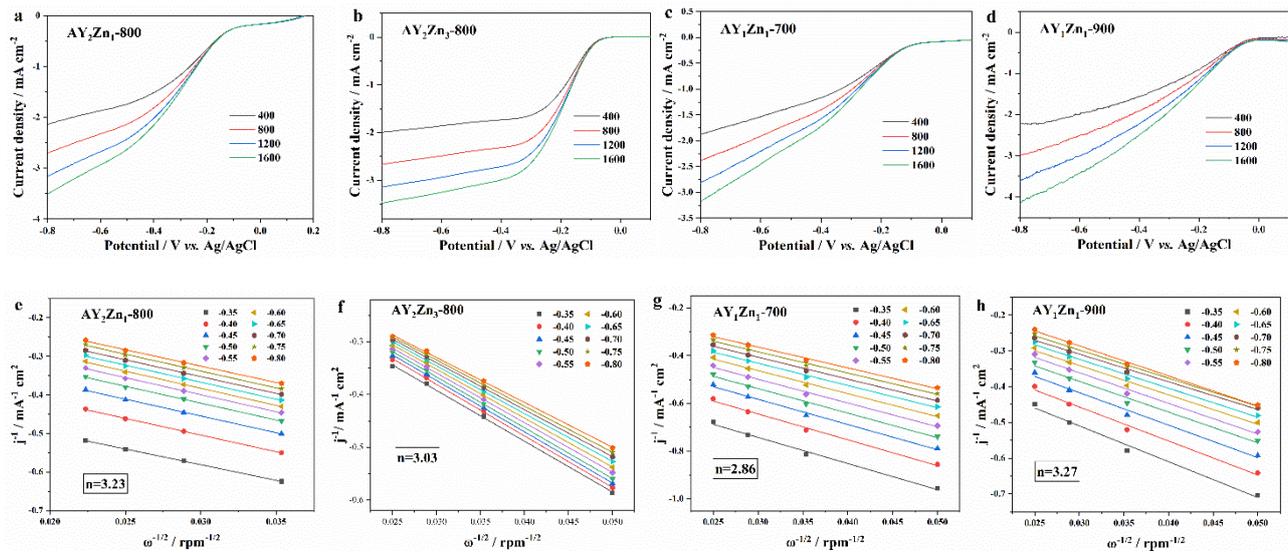


Fig. S9 (a-d) LSV curves for other samples in O₂-saturated 0.1 M KOH at various rotating speeds; (e-h) the corresponding K-L plots for ORR in O₂-saturated 0.1 M KOH.

Table S2 The LSV parameters for the carbons and commercial Pt/C catalyst tested by RDE

| Sample | E_{onset} V vs. Ag/AgCl | $E_{1/2}$ V vs. Ag/AgCl | j_{lim} mA cm ⁻² | $n^{[a]}$ |
|--------------------------------------|------------------------------|----------------------------|----------------------------------|-----------|
| AY ₁ Zn ₁ -800 | 0.044 | -0.174 | 5.91 | 3.75 |
| AY ₂ Zn ₁ -800 | -0.054 | -0.354 | 3.49 | 3.23 |
| AY ₂ Zn ₁ -800 | 0 | -0.214 | 3.47 | 3.03 |
| AY ₁ Zn ₁ -700 | -0.059 | -0.38 | 3.17 | 2.86 |
| AY ₁ Zn ₁ -900 | 0.016 | -0.317 | 3.99 | 3.27 |
| Pt/C (20 wt%) | 0.048 | -0.152 | 6.26 | ~4 |

[a] The electron transfer number calculated based on K-L plot method