## Thermal annealing effects on hydrothermally synthesized unsupported MoS<sub>2</sub> for enhanced deoxygenation of Propylguaiacol and Kraft lignin

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Aluminium, Al	18	mg/kg	Manganese, Mn	58	mg/kg
Antimony,Sb	0.06	mg/kg	Molybdenum, Mo	0.7	mg/kg
Arsenic, As	0.08	mg/kg	Sodium, Na	9300	mg/kg
Barium, Ba	2	mg/kg	Neodymium, Nd	0.005	mg/kg
Beryllium, Be	0.013	mg/kg	Niobium, Nb	<0.005	mg/kg
Lead, Pb	0.04	mg/kg	Nickel, Ni	0.4	mg/kg
Boron, B	22	mg/kg	Osmium, Os	<0.005	mg/kg
Bromine, Br	1.2	mg/kg	Palladium, Pd	<0.005	mg/kg
Cerium, Ce	0.012	mg/kg	Platinum, Pt	<0.005	mg/kg
Cesium, Cs	0.13	mg/kg	Praseodymium, Pr	<0.005	mg/kg
Dysprosium, Dy	<0.005	mg/kg	Rhenium, Re	< 0.005	mg/kg
Erbium, Er	<0.005	mg/kg	Rhodium, Rh	<0.005	mg/kg
Europium, Eu	<0.005	mg/kg	Rubidium, Rb	4.2	mg/kg
Phosphorus, P	12	mg/kg	Ruthenium, Ru	< 0.005	mg/kg
Gadolinium, Gd	<0.005	mg/kg	Samarium, Sm	< 0.005	mg/kg
Gallium, Ga	0.15	mg/kg	Selenium, Se	0.09	mg/kg
Germanium, Ge	0.03	mg/kg	Silver, Ag	0.03	mg/kg
Gold, Au	<0.005	mg/kg	Scandium, Sc	< 0.005	mg/kg
Hafnium, Hf	<0.005	mg/kg	Strontium, Sr	1	mg/kg
Holmium, Ho	<0.005	mg/kg	Sulphur, S	21000	mg/kg
tridium, Ir	<0.005	mg/kg	Tantalum, Ta	< 0.005	mg/kg
lodine, l	0.5	mg/kg	Tellurium, Te	< 0.005	mg/kg
Iron, Fe	30	mg/kg	Thallium, Tl	0.03	mg/kg
Cadmium, Cd	0.17	mg/kg	Tin, Sn	0.1	mg/kg
Calcium, Ca	200	mg/kg	Terbium, Tb	<0.005	mg/kg
Potassium, K	1100	mg/kg	Titanium, Ti	1.4	mg/kg
Silicon, Si	6000	mg/kg	Thorium, Th	< 0.005	mg/kg
Cobalt, Co	0.08	mg/kg	Thulium, Tm	<0.005	mg/kg
Copper, Cu	0.9	mg/kg	Uranium, U	< 0.005	mg/kg
Chromium, Cr	0.3	mg/kg	Vanadium, V	19	mg/kg
Mercury, Hg	< 0.005	mg/kg	Bismuth, Bi	<0.005	mg/kg
Lanthanum, La	0.007	mg/kg	Tungsten, W	0.3	mg/kg
Lithium, Li	0.07	mg/kg	Ytterbium, Yb	<0.005	mg/kg
Lutetium, Lu	<0.005	mg/kg	Yttrium, Y	0.005	mg/kg
Magnesium, Mg	21	mg/kg	Zinc, Zn	13	mg/kg
			Zirconium, Zr	0.007	mg/kg

Figure S1 ICP analysis of impurities in kraft lignin.

Table S1 Elemental analysis for kraft lignin.

Elemental	Wt%
analysis (%)	
С	62.1
Н	5.85
Ν	0.35
S	2.18
О	29.5*

\*by difference and neglecting other impurities



Figure S2 Nitrogen adsorption-desorption isotherms for studied catalysts.



Figure S3 Additional HRTEM images of a-c) MoS<sub>2</sub>-12, d-f) MoS<sub>2</sub>-12a, g-i) MoS<sub>2</sub>-24, and j-l) MoS<sub>2</sub>-24a.



*Figure S4a)* Distribution of the number of stacks and b) MoS<sub>2</sub> slab length for different unsupported MoS<sub>2</sub>.



Figure S5 Comparison of PG conversion and product selectivity for HDO of PG over  $MoS_2$ -24 following an annealing treatment under different atmosphere (air or  $N_2$ ) at 400 °C for 2 h. Reaction conditions: 50 bar total  $H_2$  pressure, 300 °C, and 1000 rpm.



Figure S6 Reaction product distribution for HDO of PG over a) Bulk  $MoS_2$  b) 13.2 wt%  $MoS_2$  supported on alumina at 50 bar total  $H_2$  pressure, 300 °C and 1000 rpm.



Figure S7 GC spectrum of the lignin fraction obtained from the hydrotreatment of kraft lignin over commercial MoS<sub>2</sub> (blue line) and MoS<sub>2</sub>-12a (black line). Reaction conditions: 3:1 lignin to catalyst ratio, 340 °C, 40 bar initial H<sub>2</sub> pressure, and 1000 rpm. The major compounds were labeled in the spectrum as (1) Methylcyclopentane, (2) Cyclohexane, (3) Methylcyclohexane, (4) Ethylcyclopentane, (5) Toluene, (6) Ethylcyclohexane, (7) 1,3-dimethylbenzene, (8) Propylcyclohexane, (9) Propylbenzene, (10) Guaiacol, (11) Creosol, (12) 4-ethyl-2methoxyphenol and (13) Propylguaiacol.

Table S2 Products identified from GC-MS spectra and product yields for hydrotreatment of Kraft lignin over bulk  $MoS_2$  and  $MoS_2$ -12a. Reaction conditions: 3:1 lignin to catalyst mass ratio, 340 °C, 40 bar initial  $H_2$  pressure, and 1000 rpm.

Retention time (min)	Compound identified	Compound chemical formula	FID peak area (Bulk MoS <sub>2</sub> )	Bulk MoS <sub>2</sub> Product yield (area %)	FID peak area (MoS <sub>2</sub> -12a)	MoS <sub>2</sub> -12a Product yield (area %)
7.135	Methylcyclopentane	C <sub>6</sub> H <sub>12</sub>	-	-	$2.10 \times 10^{6}$	2.53
8.089	Cyclohexane	C <sub>6</sub> H <sub>10</sub>	$6.19 \times 10^{5}$	1.9	$1.56 \times 10^{7}$	18.83

8.148	1,3-	$C_{7}H_{14}$	-	_	$9.45 \times 10^{5}$	1.14
	dimethylcyclopentane					
9.073	Methylcyclohexane	$C_7H_{14}$	-	-	$1.90 \times 10^{7}$	13.20
9.276	Ethylcyclopentane	$C_{7}H_{14}$	-	-	$1.34 \times 10^{6}$	1.61
10.044	Toluene	C <sub>7</sub> H <sub>8</sub>	$3.98 \times 10^{5}$	1.22	$3.47 \times 10^{6}$	4.18
10.126	Cyclopentene	C <sub>7</sub> H <sub>12</sub>	$3.27 \times 10^{5}$	1.00	-	-
10.392	1,3-	C <sub>8</sub> H <sub>16</sub>	-	-	$5.77 \times 10^{5}$	0.69
	dimethylcyclohexane					
10.536	Pentylcyclopentane	$C_8H_{16}$	$7.76 \times 10^{5}$	2.38	-	-
10.587	1-ethyl-3-	$C_8H_{16}$	-	-	$9.70 \times 10^{5}$	1.17
	methylcyclopentane					
10.906	1,2-	$C_8H_{16}$	-	-	$5.65 \times 10^{5}$	0.68
	dimethylcyclohexane					
11.650	Propylcyclopentane	$C_8H_{16}$	-	-	$1.44 \times 10^{6}$	1.74
11.757	Ethylcyclohexane	$C_8H_{16}$	-	-	$1.06 \times 10^{7}$	12.78
12.343	1,3-dimethylbenzene	$C_8H_{10}$	-	-	$3.68 \times 10^{6}$	4.44
12.547	Xylene	$C_8H_{10}$	-	-	$7.12 \times 10^{5}$	0.86
12.920	1-methyl-2-	$C_{9}H_{18}$	-	-	$1.39 \times 10^{6}$	1.69
	propylcyclopentane					
12.946	Octylcyclopentane	$C_{11}H_{22}$	$9.91 \times 10^{5}$	3.04	-	-
13.480	1-ethyl-4-	$C_{9}H_{18}$	-	-	$6.73 \times 10^{5}$	0.82
	methylcyclohexane					
13.782	(1-	$C_{10}H_{20}$	-	-	$1.11 \times 10^{6}$	1.34
	methylethyl)cyclohexa					
	ne					
13.984	Propylcyclohexane	$C_{9}H_{18}$	-	-	$1.42 \times 10^{7}$	17.16
14.433	Propylbenzene	C9H12	-	-	$4.24 \times 10^{6}$	5.12
14.573	1-ethyl-3-	$C_{9}H_{12}$	-	-	$4.48 \times 10^{5}$	0.54
	methylbenzene		<i>.</i>			
14.748	Phenol	$C_6H_6O$	$1.87 \times 10^{6}$	5.71	-	-
15.077	1-methyl-2-	$C_{10}H_{20}$	-	-	$4.39 \times 10^{5}$	0.53
	propylcyclohexane					
15.132	(2-methylpropyl)-	$C_{10}H_{20}$	-	-	$2.31 \times 10^{5}$	0.28
	cyclohexane					
15.424	1-Methyl-4-(1-	$C_{10}H_{18}$	-	-	$2.22 \times 10^{3}$	0.27
	methylethyl)-					
	cyclohexane					
15.924	(1-methylpropyl)-	$C_{10}H_{20}$	-	-	$3.29 \times 10^{3}$	0.40
16.024	Cyclohexane	C II			<b>5</b> ( <b>7</b> 105	0.60
16.034	n-butylbenzene	$C_{10}H_{14}$	-	-	$5.67 \times 10^{5}$	0.68
16.136	Cyclopropylbenzene	$C_9H_{10}$	-	-	$1.08 \times 10^{5}$	1.31
16.321	I-methyl-3-	$C_{10}H_{14}$	-	-	$9.61 \times 10^{5}$	1.16
16 4 4 1	propylbenzene	C II			2.00 × 1.05	0.47
16.441	n-butylbenzene	$C_{10}H_{14}$	-	-	$3.89 \times 10^{5}$	0.4/
16.588	3-methylphenol	$C_7H_8O$	$2.04 \times 10^{\circ}$	6.25	-	-
16.944	Guaiacol	$C_7H_8O_2$	$5.13 \times 10^{\circ}$	15./1	-	-
17.134	(2-	$C_{11}H_{22}$	-	-	$3.31 \times 10^{5}$	0.40
	wietnyibutyi)cyclohex					
17.922	ane Deutsclasse 1 also asso	C II			5 45× 105	0.((
17.832	2.4. directionexane	$C_{11}H_{22}$	-	-	3.45× 10°	0.00
17.010	2,4-uineinyipnenoi	$C_8\Pi_{10}U$	2.09 × 10°	0.88	-	-
17.919	(3-memyi-2-butenyi)- benzene	$C_{11}\Pi_{14}$	-	-	3.30× 10°	0.41

18.151	3-ethylphenol	$C_8H_{10}O$	$1.43 \times 10^{6}$	4.37	-	-
18.152	2,5-Dimethylphenyl	$C_{10}H_{14}O$	-	-	$1.03 \times 10^{6}$	1.24
	methyl carbinol					
18.395	1,2,3,4-tetrahydro-	$C_{10}H_{12}$	-	-	$8.78 \times 10^{5}$	1.06
	Naphthalene					
18.658	Creosol	$C_8H_{10}O_2$	$7.09 \times 10^{6}$	21.71	-	-
19.247	3,4-dimethoxyltoluene	$C_9H_{12}O_2$	$4.63 \times 10^{5}$	1.42	-	-
19.249	1,2,3,4-tetrahydro-2-	$C_{11}H_{14}$	-	-	$5.24 \times 10^{5}$	0.63
	methyl- Naphthalene					
19.563	3-propylphenol	$C_9H_{12}O$	$1.21 \times 10^{6}$	3.69	-	-
19.927	4-ethyl-2-	$C_9H_{12}O_2$	$4.73 \times 10^{6}$	14.50	-	-
	methoxyphenol					
20.460	4-ethyl-1,2-	$C_{10}H_{14}O_2$	$3.93 \times 10^{5}$	1.20	-	-
	dimethoxybenzene					
20.634	2,5-diol-p-cymene	$C_{10}H_{14}O_2$	$2.71 \times 10^{5}$	0.83	-	-
20.884	4-(2-propenyl)-phenol	$C_9H_{10}O$	$6.56 \times 10^{5}$	2.01	-	-
21.171	2-methoxy-4-	$C_{10}H_{14}O_2$	$2.76 \times 10^{6}$	8.28	-	-
	propylphenol					
21.502	4-(1-methylethyl)-	$C_{10}H_{12}O$	$1.27 \times 10^{6}$	3.89	-	-
	benzaldehyde					