

Supporting Information

Method for fabricating highly crystalline polyvinylidene fluoride for piezoelectric energy-harvesting and vibration sensor applications

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Experimental

Synthesis of $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ nanocubes (BNT NCs)

The hydrothermal synthesis was employed to prepare BNT cubes, and its schematically represented in Fig.S1. Initially, the 1:2 molar ratio of Bi source ($\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$) to Ti source was ($\text{Ti}(\text{OC}_4\text{H}_9)_4$) was added to the distilled water and stirred for 10 minutes. Then, the two solutions were mixed together with the addition of 16 M NaOH, which acts as a Na source and controlling agent in the shape and size of the nanomaterials. After 1 hr stirring, the precursor solutions were loaded into the autoclave, and the reaction was performed at 200°C for 80 hrs. Finally, the collected particles were washed multiple times and dried at 100°C for about 5 hrs and later used in characterizations and film fabrications.

Measurement systems

The functional properties of the prepared PVFB films were analyzed by FTIR analysis by Compact FT-IR Spectrometer ALPHA II by Bruker in ATR mode between the range of 700 -1800 cm^{-1} . The structural properties of the synthesized $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ cubes were studied using an X-ray diffractometer (Empyrean by Malvern Panalytical) and Raman spectroscopy (LabRAM) with a laser wavelength of 514 nm and power of 10 mW with an ND filter of 10 %. The Piezo Force Microscopy (PFM) analyses were performed from a Scanning Probe Microscope (SPM) by Multimode 8 (Bruker). The prepared film's morphological analysis was analyzed using field-emission scanning electron microscopy (MIRA3 by TESCAN). The Keithley electrometer (6514) and current preamplifier (Stanford systems) were used to measure the electrical output responses of PVFB PENGs upon mechanical force. The constant mechanical force is generated by the linear motor (Lin Mot,

HF01-37 model). The bath sonicator cleaning unit Elmasonic P 30 H is used in the energy conversion and demonstrating the sensor applications.

Supporting Information

Figure S1 The schematic representation of $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ preparation through the hydrothermal reaction. **Figure S2** The structural analysis of the prepared $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ through XRD and Raman analysis. **Figure S3** The functional analysis by FTIR and the morphological analysis of the prepared PVFB film at $\sim 30\%$ RH. **Figure S4** The morphological images of the prepared PVFB films with various loading of BNT at $\sim 60\%$ RH. **Figure S5** The PFM phase and amplitude images of the 0 & 30 % PVFB film at 0 V bias. **Figure S6, S7** The repeatability and location varied PFM phase and amplitude loops of the 30 % PVFB film. **Figure S8** The electrical voltage and current outputs of various PVFB PENGs and switching polarity test output. **Figure S9** The cross-sectional SEM image of the 30 % PVFB film attached with the Al foil as electrodes with PDMS as supporting layer. **Figure S10** The 30 % PVFB PENG electrical current output under varying accelerations, commercial capacitor charging and discharging analysis. **Figure S11** The 30 % PVFB PENG electrical outputs under varying sonication power @ 37 kHz and three active modes of the bath sonicator. **Figure S12** The 30 % PVFB PENG electrical outputs under 100 W power @ 80 kHz and the schematic of device attachment positions to identify the high active region. **Figure S13** The electrical output of 30 % PVFB PENG under finger, hand and foot-tapping biomechanoid force conditions.

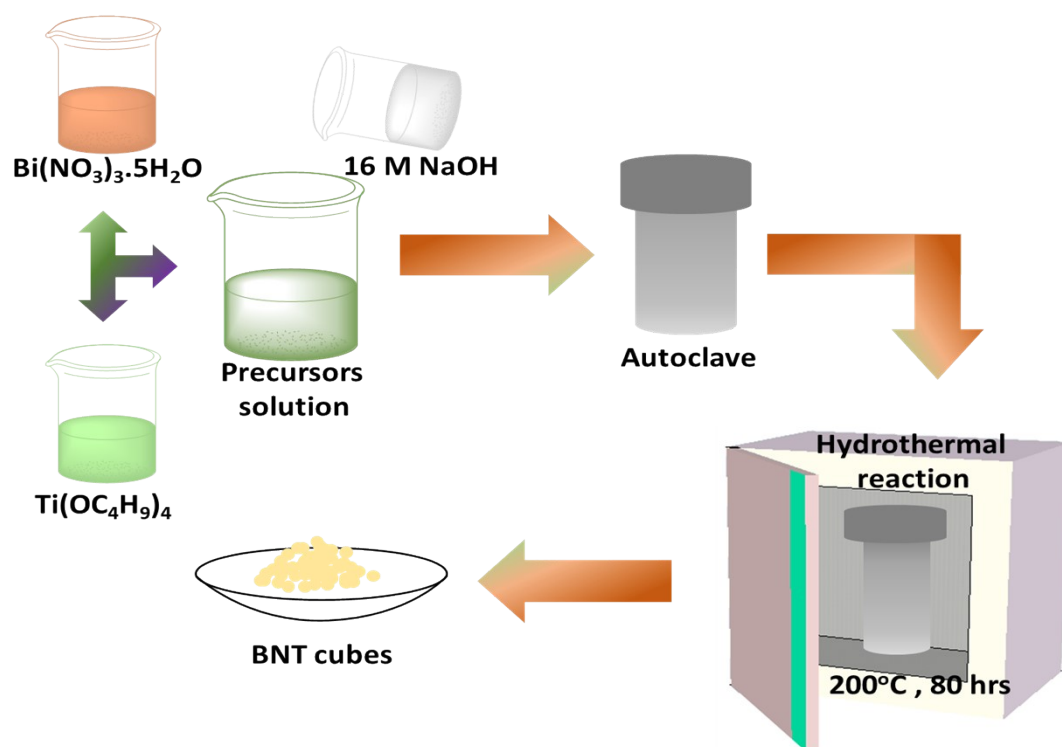


Figure S1 The schematic representation of $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ cubes preparation through the hydrothermal reaction.

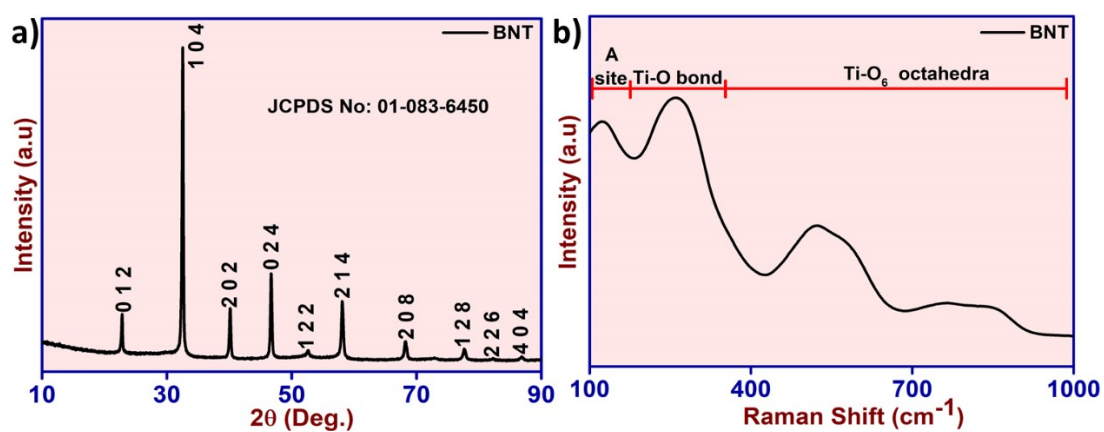


Figure S2 (a) The XRD pattern of the $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ cubes. (b) The Raman spectrum of the $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$ cubes.

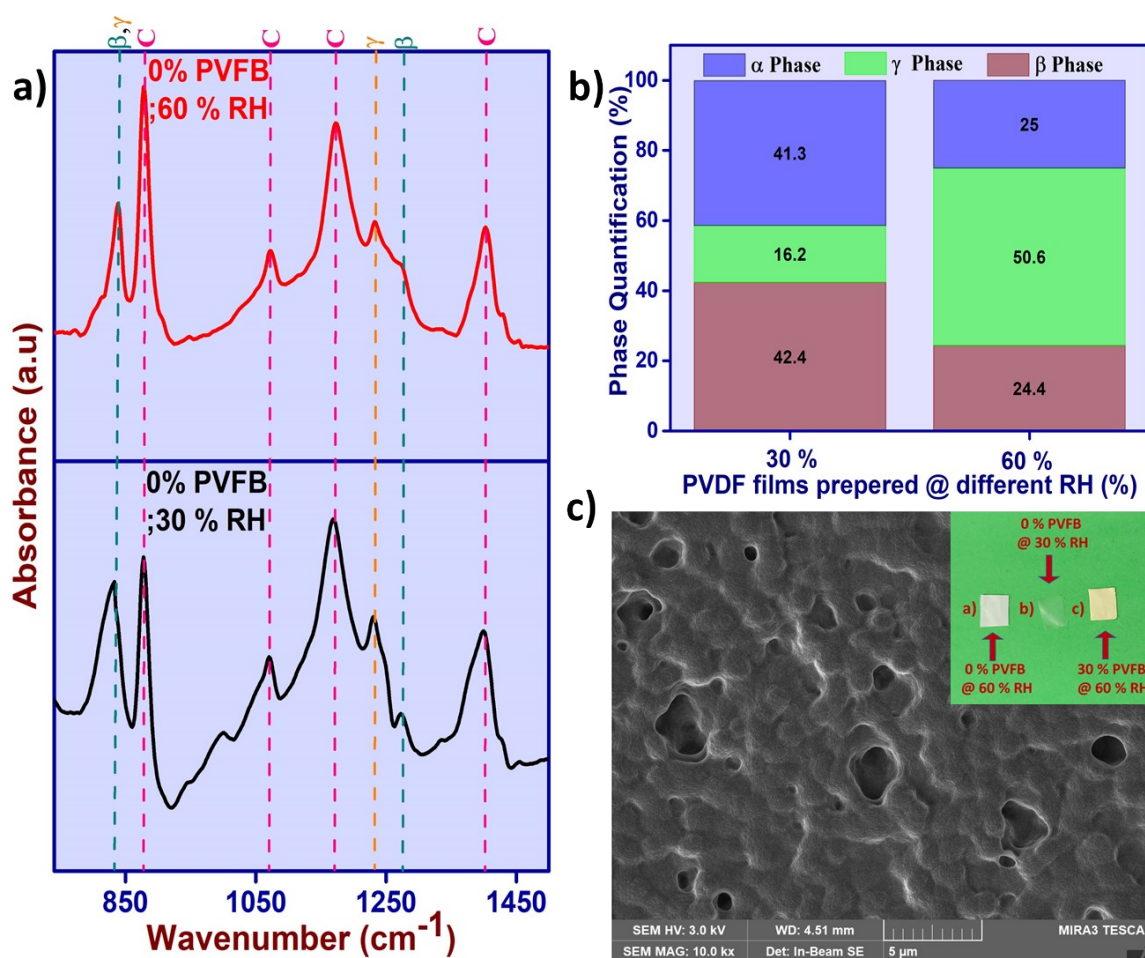


Figure S3 (a) The FTIR spectrum of 0 % PVFB film at 30 and 60 % relative humidity (RH). **(b)** The crystalline phase quantification of 0 % PVFB film at 30 and 60 % relative humidity (RH). **(c)** The morphological images of the 0 % PVFB films were prepared at 30 % relative humidity (RH). (Inset image: The digital images of PVFB films at different conditions)

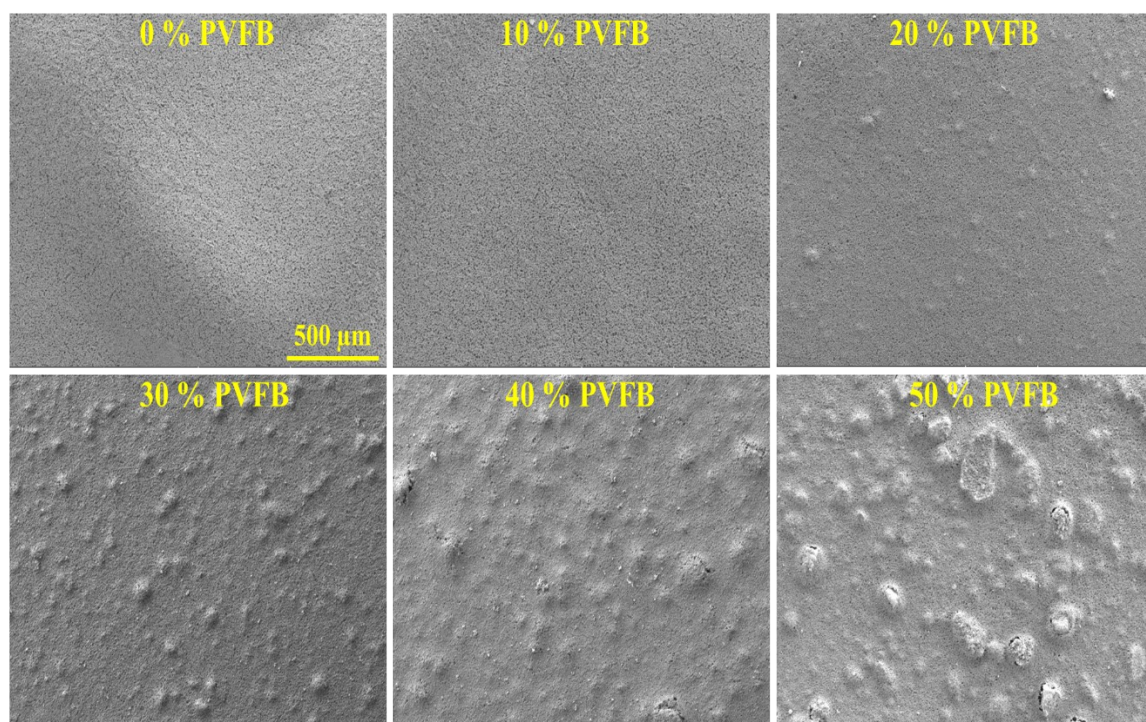


Figure S4 The morphological images of the various % of BNT cubes loaded PVFB films.

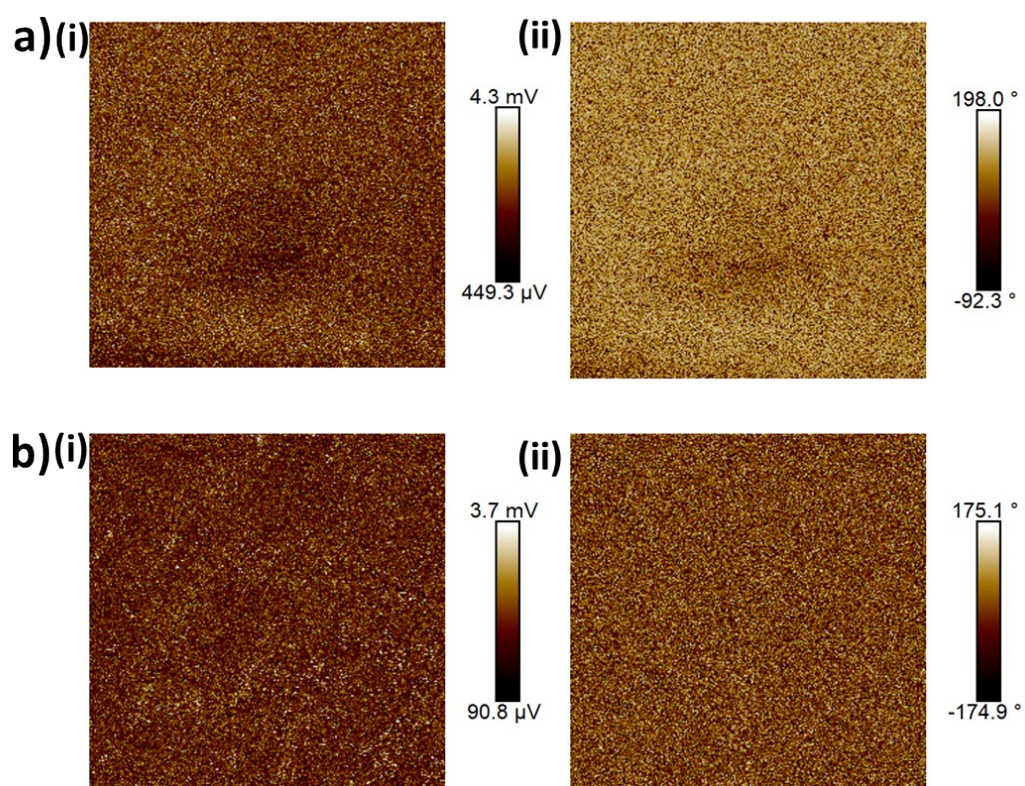


Figure S5 (a) The PFM amplitudes and phase images of the 0 % PVFB under 0 V bias. **(b)** The PFM amplitudes and phase images of the 30 % PVFB under 0 V bias.

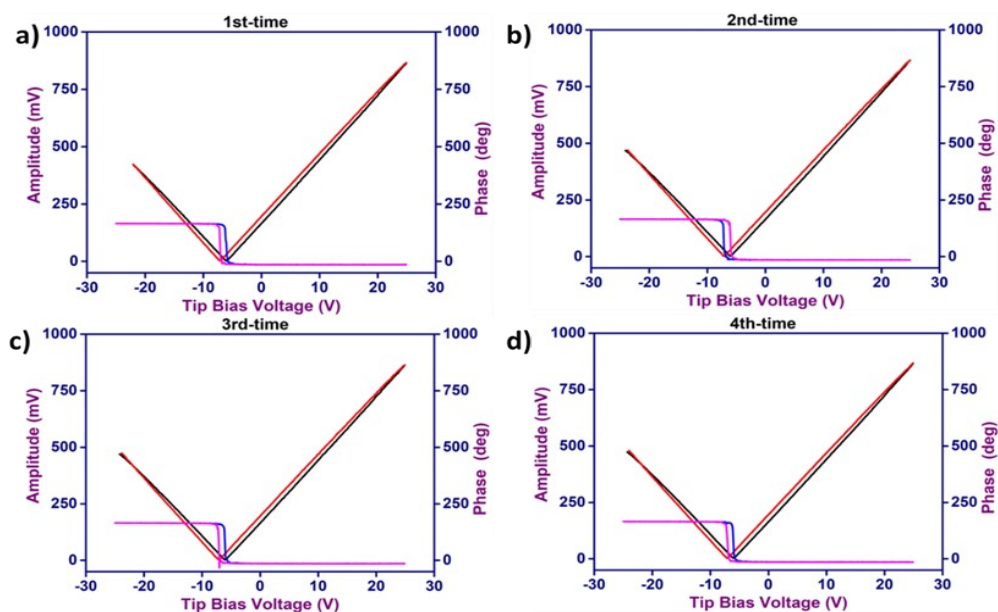


Figure S6 The repeatability test of 30 % PVFB films amplitude and phase loop.

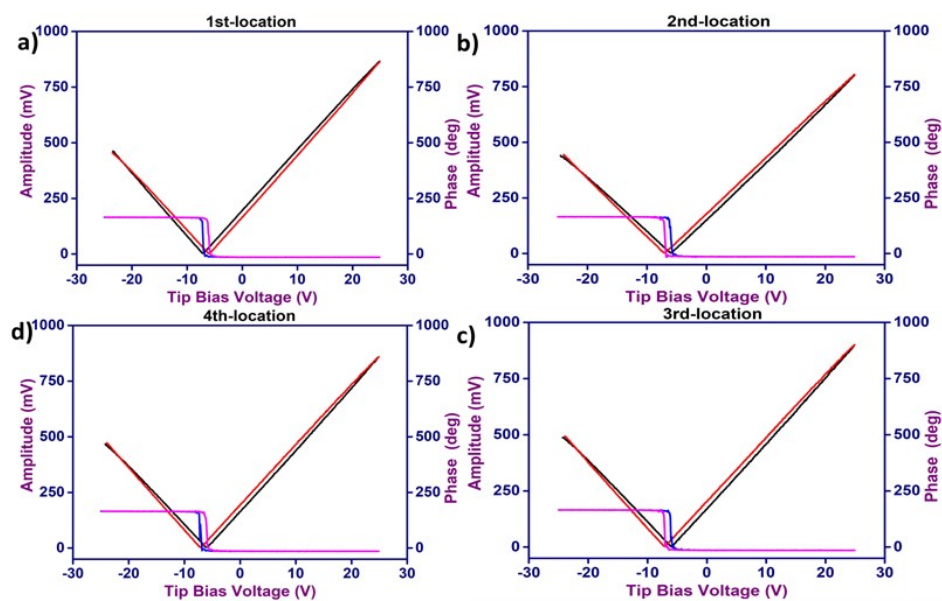


Figure S7 The reproducibility of 30 % PVFB films amplitude and phase loop at different locations.

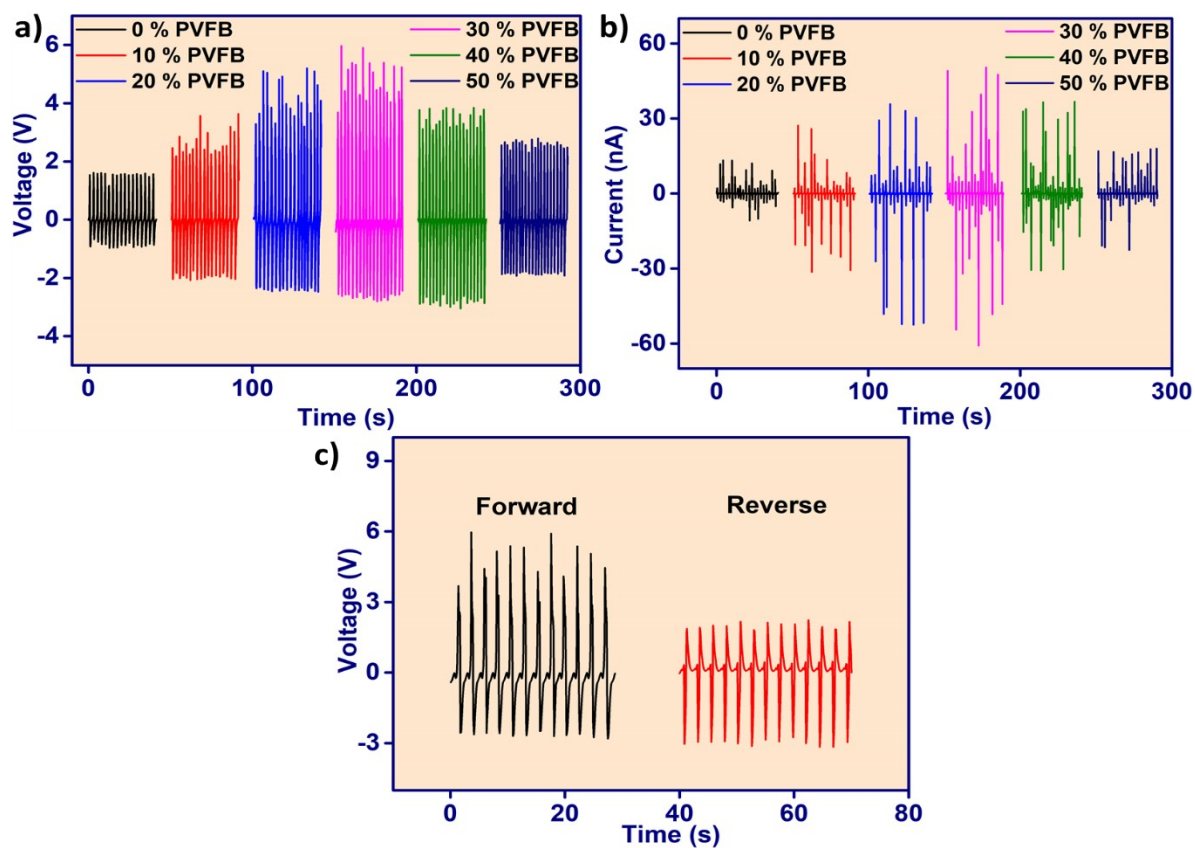


Figure S8 (a & b) The voltage and current response of unpoled PVFB PENGs under compressive force. **(c)** The switching polarity test of the 30 % PVFB PENG.

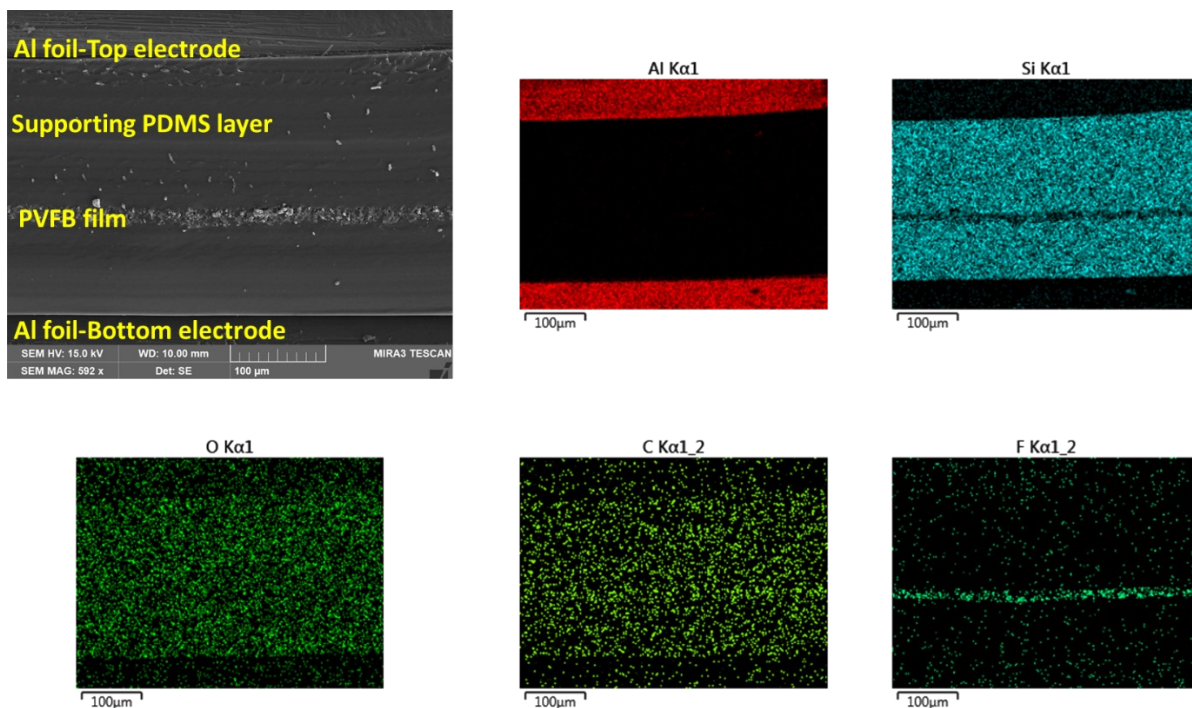


Figure S9 The cross-sectional SEM image of the 30 % PVFB film attached with the Al foil as electrodes with PDMS as supporting layer.

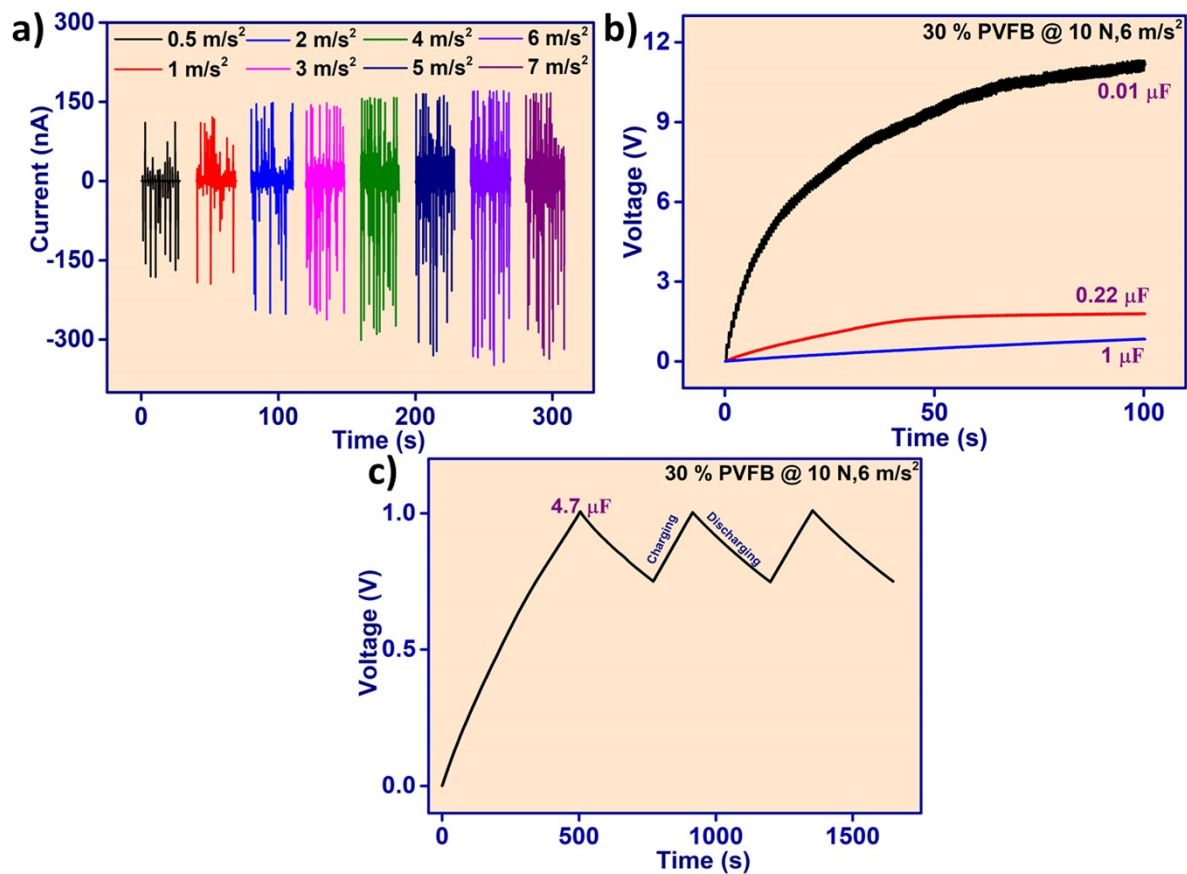


Figure S10 (a) The current response of 30 % poled PVFB PENG under various accelerations.
 (b) The charging plot of 0.01, 0.22 and 1 μF capacitor using 30 % PVFB PENG up to 100 s.
 (c) The charging-discharging analysis of 4.7 μF using 30 % PVFB PENG.

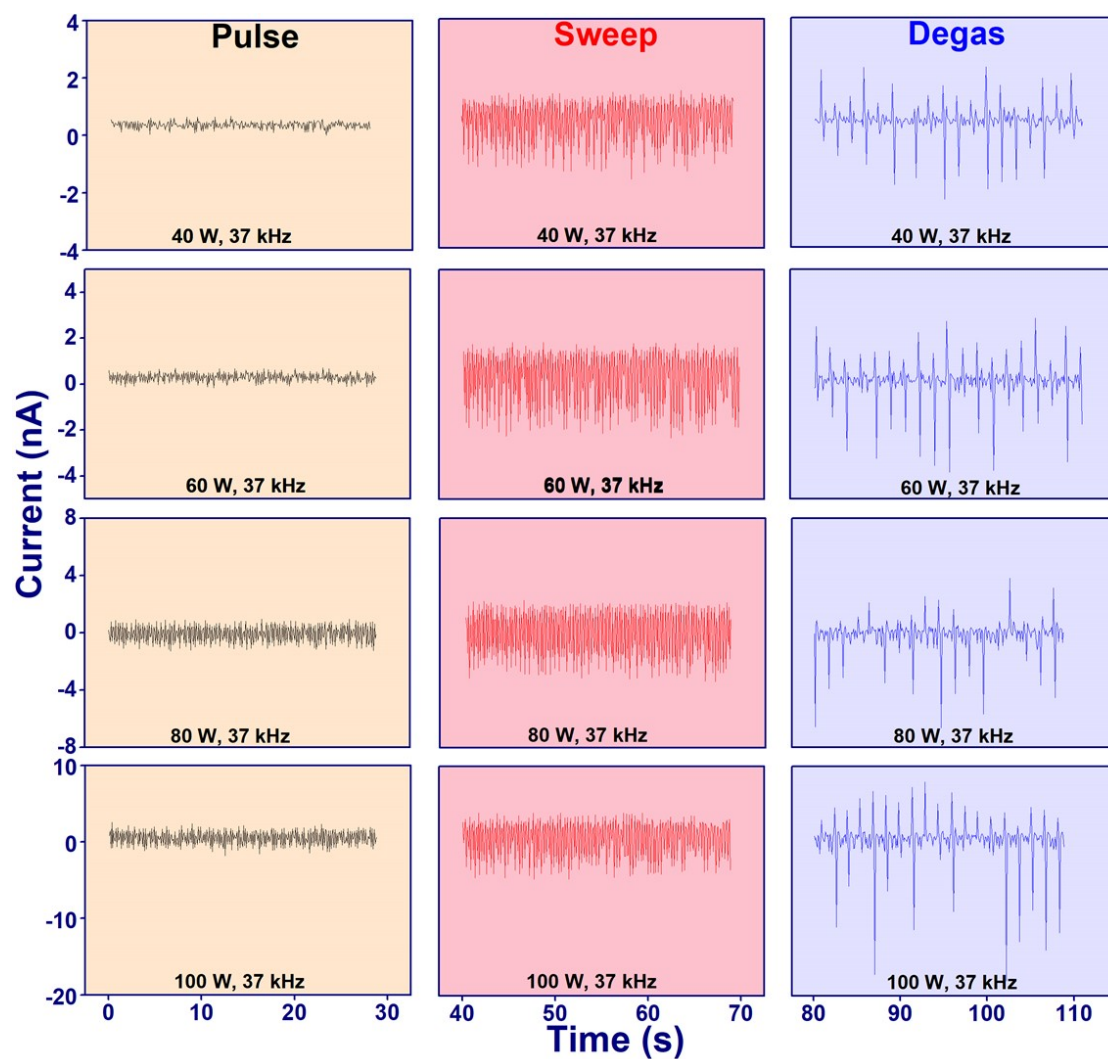


Figure S11 The current response of 30 % PVFB PENG under the varying power and sonicator operation modes.

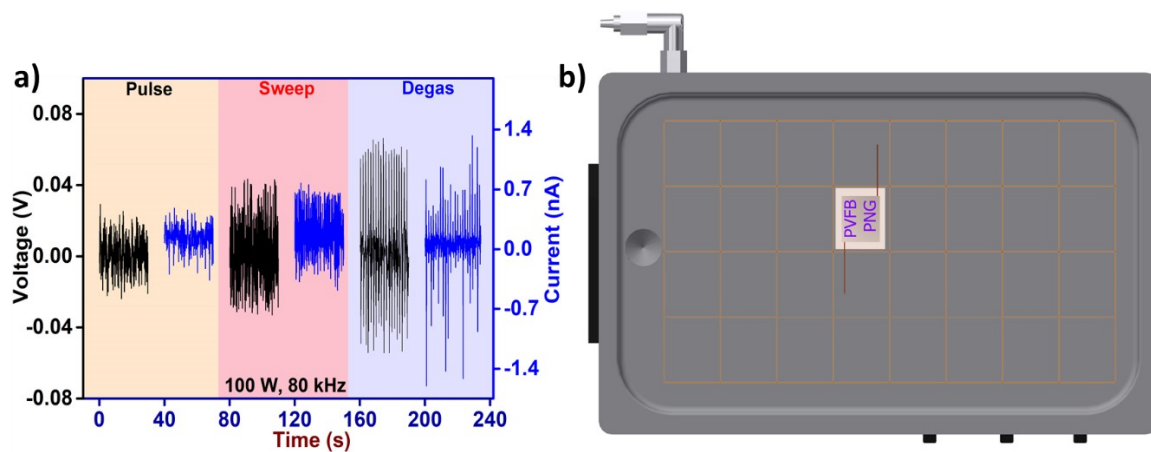


Figure S12 (a) The voltage and current response of 30 % PVFB PENG under 100 W power at 80 kHz with different sonicator operation modes. **(b)** The schematic diagram of 30 % PVFB PENG attachment on the surface of the bath sonicator as a sensing unit to identify the high active region.

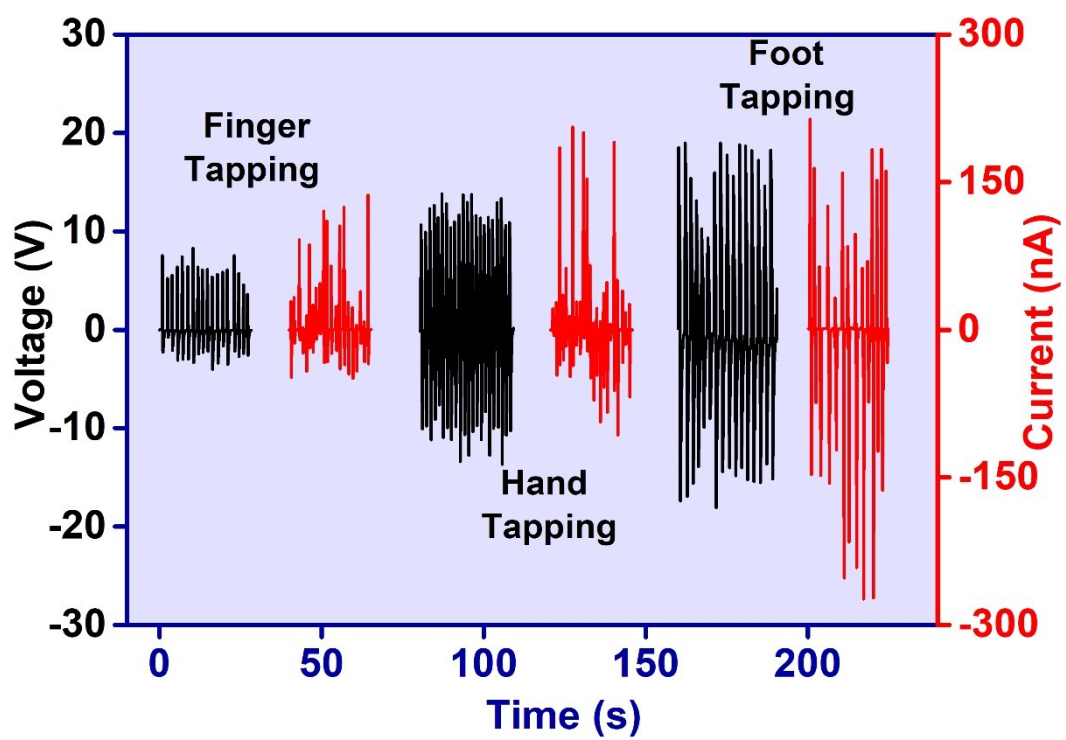


Figure S13 The electrical output of 30 % PVFB PENG under finger, hand and foot-tapping biomechanoid force conditions.

Polymer	Added Fillers	Preparation method	Solvent	Crystalline phase ~ (%)
PVDF-HFB [1]	-	Solution casting, Thermal induced phase separation (TIPS)	Dibutyl phthalate/ Dioctyl phthalate	27.70
PVDF-PEO [2]	-	Solution casting	DMF	43
PVDF [3]	BaTiO ₃	Tape casting	DMF	49.8
PVDF [4]	ZnO	Tape casting	DMAc	37.26
PVDF [5]	-	Solution casting	DMAc	46
PVDF [6]	NiO	Solution casting	DMF	72
PVDF [7]	-	Thermal melting	-	39.18
PVDF [8]	BaTiO ₃	Melt blending	-	52
PVDF [9]	A proportion of n-propanol, n-hexanol and glycerol mixture	Non-solvent induced phase separation (NIPS)	DMAc	60
		Melt spinning	-	48.7
		Solution casting	DMAc	70
PVDF [10]	Poly(ethyl methacrylate), PEMA	Melting blending	-	28
PVDF [11]	-	Melt compressed	-	39
		Solution casting	DMF	50
		Electrospun	DMF	47
PVDF [11]	1-allyl-3-methylimidazolium chloride	Melt compressed	-	44
		Solution casting	DMF	54

		Electrospun	DMF	52
PVDF [12]	-	Twin screw extruder @ 240°C	-	58
	Polyamide 6			32
PVDF [13]	-	Solution Casting -under magnetic field	DMF	49
PVDF [14]	-	Melt-Hot press	DMF	78
		Electrospinning	DMAc	56
PVDF-TrFE [15]	-	Solution casting	MEK	56
	LiNbO ₃			67
PVDF [16]	-	Solution casting, TIPS	Methyl salicylate	47.7
PVDF-HFP Tri-layer [17]	Ni-ZnO	Solution casting	DMF/ Acetone	51.1
PVDF [18]	-	Solution casting, TIPS	DMC/ DMSO	15
	Protic ionic liquids			60
PVDF [19]		Solution casting	DMSO	50.9
PVDF [20]		Electrospinning	DMF/ Acetone	36.9
	Graphene			45.5
PVDF [21]	Poly(methyl methacrylate), PMMA	Spin coating	DMAc	45
PVDF [22]	-	Solution casting	DMAc	90
PVDF [23]	Polyethylene glycol in Ionic liquid	Solution casting	DMF	35
PVDF [24]	PMMA	Electrospinning	DMF/THF	50

PVDF [25]	Few layered graphene	Solution casting	DMF	53
	Graphene oxide			51
PVDF (Present work)	-	Tape casting, under 30 % RH	DMF	58.6
		Tape casting, under 60 % RH		75
	BNT			92

***DMAc: Dimethylacetamide, DMF: Dimethylformamide, THF: Tetrahydrofuran, MEK: Methyl Ethyl Ketone, DMC: Dimethyl carbonate, DMSO: Dimethyl sulfoxide.**

Table S1 The comparison table of the attained crystalline phase of PVFB with the recently reported articles on PVDF under various preparation techniques and added filler materials.

S.No	Material	$\sim d_{33}^*$ (pm/V)	References
1	PVDF/BCZT	5	[26]
2	PVDF	49.6	[27]
3	PVDF	58	[28]
4	PVDF	30	[29]
5	PVDF	2	[30]
6	PVDF-TrFE	20	[31]
7	PVDF	43	[32]
8	PVDF	30	[33]
9	PVDF/CNT	35	
10	PVDF/Ag-CNTs	54	
11	PVDF	30	[34]
12	PVDF	20	[35]
13	PVDF/GO	110	
14	PVDF	28	[36]
15	PVDF	34	[37]
16	PVDF/BaTiO ₃	48	[38]
17	PVDF	17.5	[39]
18	PVDF	49.6	[40]
19	PVDF-TrFE/GO	29.8	[41]
20	PVDF-TrFE	18	[42]
21	PVDF/Fluro doped Graphene	63	[43]
22	PVDF	64	[44]
23	PVDF/rGO	87	
24	PVDF	5	[45]
25	PVDF	144	[46]
26	PVDF	122	Present work
	PVDF/BNT	242	

***BCZT: (BaCa)(ZrTi)O₃, CNT: Carbon nanotubes, GO: Graphene oxide, BNT: Na_{0.5}Bi_{0.5}TiO₃.**

Table S2 The Comparison of piezoelectric coefficient values of as-prepared PVFB films with the recently reported PVDF and its copolymer coefficients.

Polymer	Added Fillers	Electrodes	Area (cm ²)	Applied force	Voltage ~ (V)	Current ~ (μA)
PVDF-TrFE [31]	-	Copper, Silver	-	-	9.5	-
PVDF [47]	rGO-Ag	Aluminum	13	BMF	18	1.05
PVDF [48]	-	-	5	-	10	-
	hBN				68	0.1
PVDF [49]	BaTiO ₃	ITO	1	1000 N	150	1.5
PVDF-TrFE [50]	BaTiO ₃	Aluminum	0.5	-	9	-
P(VDF-TrFE) [51]	BCZT	Gold		BMF	18	-
PVDF-TrFE [52]	-	Gold	-	-	2	0.9
	GO				4	1.8
PVDF [53]	BNNT	Silver, Gold	1	40 N	22	0.6
PVDF-TrFE [54]	KNN	ITO	-	-	0.8	0.08
PVDF [55]	BaTiO ₃ , Graphene	Aluminum	6.25	-	11	-
PVDF-TrFE [56]	BaTiO ₃	Gold	-	-	13.2	0.3
PVDF-HFP [57]	ZnO	Aluminum	-	-	2.8	-
PVDF [58]		Aluminum	-	-	12	0.4
PVDF [59]	-	Aluminum	-	-	3.8	
	Graphene				7.9	
PVDF-HFP [60]	BaTiO ₃ , hBN	-	3.14	2.5 N	2.4	-
PVDF [61]	Graphene	-	12	BMF	11	0.006

PVDF [62]	ZnO, KNN	Aluminum	4	BMF	25	1.81
PVDF [63]	Platinum	Copper, Nickel	14	217 N	30	-
PVDF [64]	-	Aluminum	-	-	16	0.8
PVDF [65]	BaTiO ₃	Aluminum	-	BMF	2	0.3
PVDF Present work	BNT	Aluminum	4	10 N	29.5	0.5

* **BMF: Biomechanical force, N: Newton, KNN: $K_{0.5}Na_{0.5}NbO_3$, hBN: hexagonal Boron nitride, BNNT: Boron nitride nanotubes.**

Table S3 The Energy harvesting performance comparison of PVFB PENGs with other recently reported PVDF and its copolymer-based piezoelectric nanogenerators.

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