

## **Use of a Lewis acid, a Brønsted acid, and their binary mixtures for the liquefaction of lignocellulose by Supercritical Ethanol Processing**

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**Table S1.** Identified compounds in bio-oils from the supercritical ethanol liquefaction of teak wood without and with MgClO<sub>4</sub> catalysts at 300 °C for 30 min

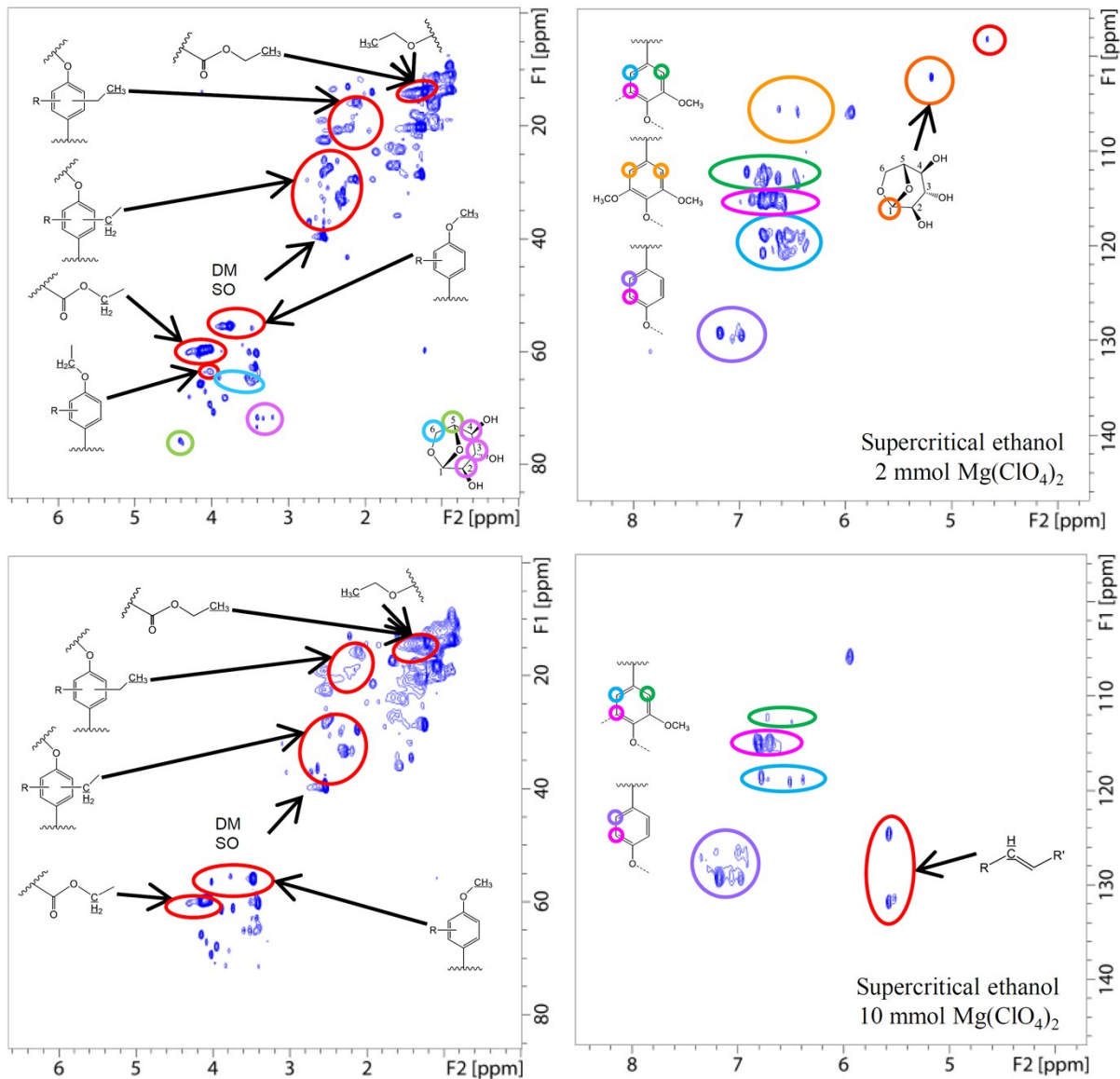
Retention Time (min)	Compounds	Area (%)			
		without catalyst	Mg(ClO <sub>4</sub> ) <sub>2</sub> (2 mmol)	Mg(ClO <sub>4</sub> ) <sub>2</sub> (5 mmol)	Mg(ClO <sub>4</sub> ) <sub>2</sub> (10 mmol)
4.28	Hydroxy-acetic acid ethyl ester	11.51	-	4.28	3.45
5.26	2-Hydroxypropanoic acid ethyl ester	3.50	26.21	17.51	24.43
5.84	2-Hydroxy-2-methylpropanoic acid ethyl ester	-	-	0.18	-
5.97	2-Cyclopenten-1-one	-	-	0.18	-
7.11	2-Furanmethanol	2.63	-	-	-
9.23	2-Methyl-2-cyclopenten-1-one	-	-	0.54	-
9.39	1,1-Dimethoxyethane	3.67	-	-	6.09
9.40	2-Hydroxybutanoic acid methyl ester	-	5.32	-	-
9.51	2-Hydroxybutanoic acid ethyl ester	-	-	4.22	-
9.81	Butyrolactone	-	-	0.25	-
9.97	Ethoxyacetic acid ethyl ester	-	-	1.79	3.90
12.16	Dihydro-5-methyl-2(3H)-furanone	-	-	1.04	2.51
12.69	5-Methyl-2-furancarboxaldehyde	-	-	0.87	-
14.75	2,3-Dimethyl-2-cyclopenten-1-one	-	-	0.93	-
15.59	3-Hexenoic acid ethyl ester	-	-	0.73	2.51
16.94	3-Methyl-1,2-cyclopentanedione	1.97	2.79	-	-
18.12	2-Hexenoic acid ethyl ester	-	-	0.74	3.42
18.73	2-Furancarboxylic acid ethyl ester	-	-	1.34	-
19.53	4-Oxo-pentanoic acid ethyl ester (Ethyl levulinate)	1.42	4.36	7.56	13.43
20.45	Butanoic acid anhydride	-	6.71	2.38	-
20.99	2-Methoxyphenol	7.24	5.07	2.49	-
21.22	4-Methylphenol	-	-	0.62	-
23.18	3-Ethyl-2-hydroxy-2-cyclopenten-1-one	2.74	2.37	0.49	-
25.41	2-Ethoxyphenol	-	-	0.48	-
27.83	Butanedioic acid diethyl ester	2.20	2.47	2.55	2.12
28.11	2-Methoxy-4-methylphenol	2.13	2.14	1.43	-
29.65	Diethyl methylsuccinate	-	-	1.24	-
33.89	4-Ethyl-2-methoxyphenol	4.88	0.95	1.64	-
34.57	Pentanedioic acid diethyl ester	1.96	-	1.29	-
38.75	2,6-Dimethoxyphenol	7.92	2.92	-	-
39.59	2-Methoxy-4-propylphenol	5.30	-	-	-
44.67	2-Methoxy-4-(1-propenyl)phenol	6.80	-	-	-
80.89	(E)-9-Octadecenoic acid	-	-	0.37	-
87.04	Diisooctyl adipate	-	1.79	-	-
93.58	Hexadecanoic acid ethyl ester	0.63	-	0.36	-
93.60	Octadecanoic acid ethyl ester	-	-	1.30	2.41
94.02	Squalene	-	5.23	-	-
95.04	1-Nonadecene	-	-	0.56	-
96.56	2,4,6-Trimethyl-tetracosanoic acid methyl ester	-	-	2.99	2.85
98.45	Cyclotetracosane	-	-	1.25	-
100.54	Docosanoic acid ethyl ester	0.93	-	-	-

**Table S2.** Identified compounds in bio-oils from the supercritical ethanol liquefaction of teak wood without and with HClO<sub>4</sub> catalysts at 300 °C for 30 min.

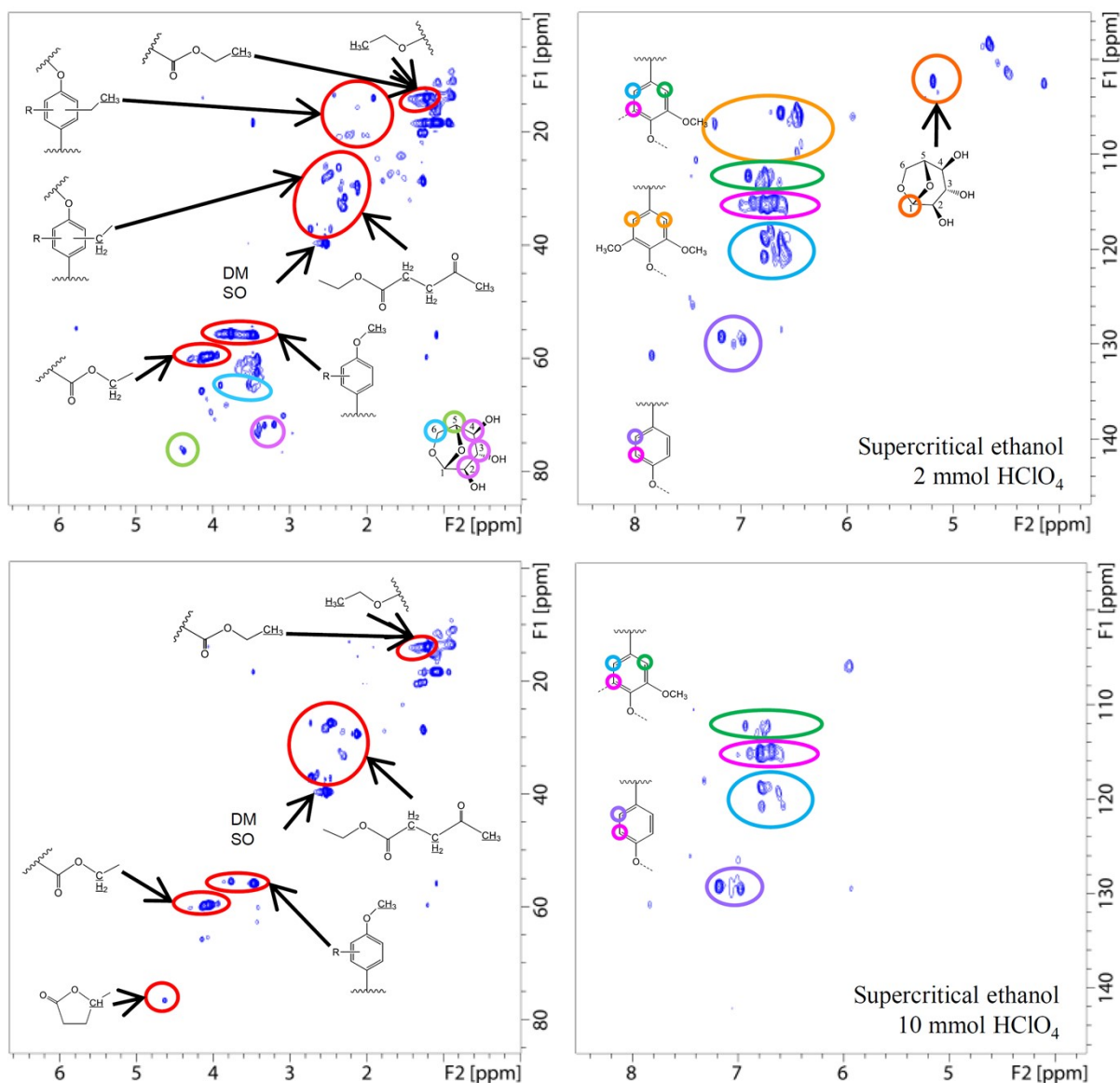
Retention Time (min)	Compounds	Area (%)			
		without catalyst	HClO <sub>4</sub> (2 mmol)	HClO <sub>4</sub> (5 mmol)	HClO <sub>4</sub> (10 mmol)
4.28	Hydroxyacetic acid ethyl ester	11.51	9.70	8.57	-
5.26	2-Hydroxypropanoic acid ethyl ester	3.50	-	5.24	1.47
5.89	Furfural	-	-	0.80	-
7.11	2-Furanmethanol	2.63	3.10	-	-
9.36	2-Hydroxybutanoic acid ethyl ester	-	1.08	-	0.41
9.39	1,1-Dimethoxyethane	3.67	-	1.37	-
12.09	Dihydro-5-methyl-2(3H)-furanone	-	-	-	1.35
12.64	5-Methyl-2-furancarboxaldehyde	-	-	1.16	-
16.94	3-Methyl-1,2-cyclopentanedione	1.97	-	-	-
17.06	2-Hydroxy-3-methyl-2-cyclopenten-1-one	-	3.56	2.34	-
18.71	2-Furancarboxylic acid ethyl ester	-	-	0.98	1.23
19.53	4-Oxo-pentanoic acid ethyl ester (Ethyl levulinate)	1.42	3.43	14.24	29.45
20.45	2-Furaldehyde diethyl acetal	-	-	3.54	1.68
20.99	2-Methoxyphenol	7.24	5.49	3.16	1.88
23.18	3-Ethyl-2-hydroxy-2-cyclopenten-1-one	2.74	2.41	1.31	-
27.83	Butanedioic acid diethyl ester	2.20	2.83	1.79	1.61
28.11	2-Methoxy-4-methylphenol	2.13	2.59	1.22	0.82
32.98	3-Methoxy-1,2-benzenediol	-	-	2.39	-
33.89	4-Ethyl-2-methoxyphenol	4.88	2.84	-	-
34.57	Pentanedioic acid diethyl ester	1.96	1.44	-	-
38.75	2,6-Dimethoxyphenol	7.92	5.99	2.61	1.20
39.59	2-Methoxy-4-propylphenol	5.30	-	-	-
44.67	2-Methoxy-4-(1-propenyl)phenol	6.80	-	-	-
44.71	3-Hydroxy-4-methoxybenzoic acid	-	-	1.26	-
44.73	4-Hydroxy-3-methoxy-benzoic acid	-	2.87	-	-
50.92	4-Hydroxy-3-methoxy-benzeneacetic acid methyl ester	-	-	-	0.90
52.87	4-Hydroxy-3-methoxy-benzoic acid ethyl ester	-	-	1.09	0.94
56.07	Ethyl homovanillate	-	-	1.35	0.92
61.55	Ethyl 3-(4-hydroxy-3-methoxyphenyl)propionate	-	-	0.87	0.47
62.89	Desaspidinol	-	2.85	-	-
62.93	1-(2,4,6-Trihydroxyphenyl)-2-pentanone	-	-	5.75	1.20
90.52	Octadecanoic acid ethyl ester	-	1.49	0.46	0.52
93.58	Hexadecanoic acid ethyl ester	0.63	-	-	-
93.59	Pentadecanoic acid ethyl ester	-	-	1.06	-
96.53	2,4,6-Trimethyl-tetracosanoic acid methyl ester	-	2.56	1.27	1.25
100.57	Docosanoic acid ethyl ester	0.93	1.99	-	0.89

**Table S3.** Identified compounds in bio-oils from the supercritical ethanol liquefaction of teak wood without and with Mg(ClO<sub>4</sub>)<sub>2</sub>/HClO<sub>4</sub> catalysts at 300 °C for 30 min.

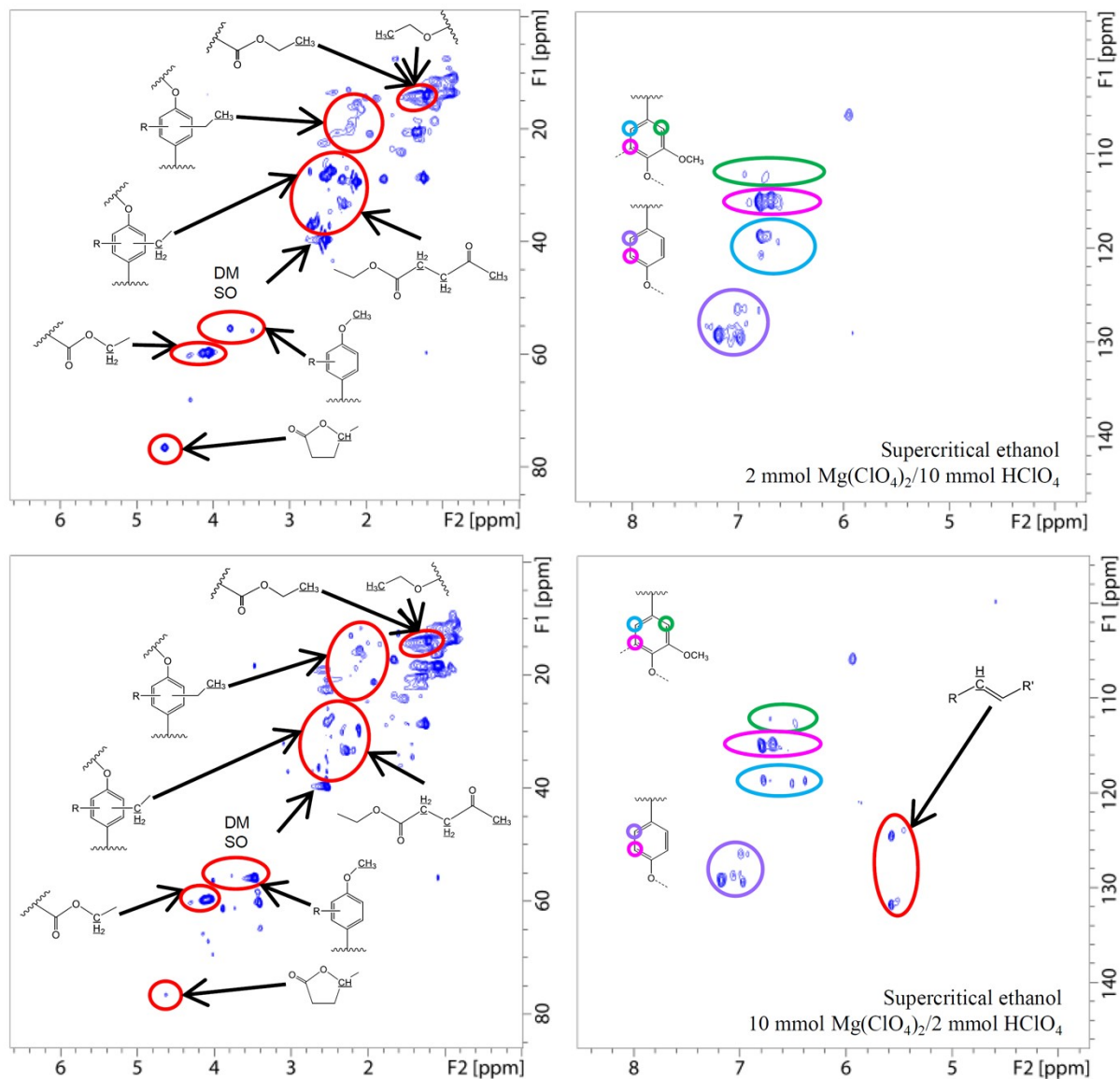
Retention Time (min)	Compounds	Area (%)			
		without catalyst	Mg(ClO <sub>4</sub> ) <sub>2</sub> /HClO <sub>4</sub> (2mmol/10mmol)	Mg(ClO <sub>4</sub> ) <sub>2</sub> /HClO <sub>4</sub> (5mmol/5mmol)	Mg(ClO <sub>4</sub> ) <sub>2</sub> /HClO <sub>4</sub> (10mmol/2mmol)
4.28	Hydroxyacetic acid ethyl ester	11.51	-	-	2.93
5.26	2-Hydroxypropanoic acid ethyl ester	3.50	3.53	12.45	15.25
7.11	2-Furanmethanol	2.63	-	-	-
9.39	1,1-Dimethoxyethane	3.67	-	-	-
9.41	2-Hydroxybutanoic acid methyl ester	-	-	-	5.12
9.42	3-Pentanol	-	-	4.57	-
9.93	Ethoxyacetic acid ethyl ester	-	-	2.16	2.56
10.70	2,5-Hexanedione	-	0.95	-	-
12.21	Dihydro-5-methyl-2(3H)-furanone	-	7.22	2.32	4.02
14.75	2,3-Dimethyl-2-cyclopenten-1-one	-	-	1.77	-
15.12	Phenol	-	2.07	0.71	-
15.60	3-Hexenoic acid ethyl ester	-	-	1.36	2.28
16.94	3-Methyl-1,2-cyclopentanedione	1.97	-	-	-
18.12	2-Hexenoic acid ethyl ester	-	-	1.55	3.17
18.73	2-Furancarboxylic acid ethyl ester	-	1.28	1.59	2.19
19.53	4-Oxo-pentanoic acid ethyl ester (Ethyl levulinate)	1.42	49.10	12.74	13.59
20.99	2-Methoxyphenol	7.24	2.57	2.33	-
22.92	2,2,5,5-Tetramethyl-3-cyclopenten-1-one	-	-	1.30	-
23.18	3-Ethyl-2-hydroxy-2-cyclopenten-1-one	2.74	-	-	-
27.83	Butanedioic acid diethyl ester	2.20	2.44	1.71	2.16
28.11	2-Methoxy-4-methylphenol	2.13	-	-	-
29.61	Diethyl methylsuccinate	-	-	1.84	2.69
33.89	4-Ethyl-2-methoxyphenol	4.88	-	-	-
34.57	Pentanedioic acid diethyl ester	1.96	-	1.49	1.93
38.75	2,6-Dimethoxyphenol	7.92	-	-	-
39.59	2-Methoxy-4-propylphenol	5.30	-	-	-
44.67	2-Methoxy-4-(1-propenyl)phenol	6.80	-	-	-
90.52	Heptadecanoic acid ethyl ester	-	0.49	0.65	-
93.58	Hexadecanoic acid ethyl ester	0.63	-	-	0.82
93.59	Octadecanoic acid ethyl ester	-	-	2.68	-
93.59	Pentadecanoic acid ethyl ester	-	1.72	-	2.78
95.02	9-Hexacosene	-	-	0.68	-
95.03	1,2-Diethyl-cyclohexadecane	-	0.47	-	-
96.53	[S-(R,S)]-2,10-Dimethylpentacosanoic acid methyl ester	-	-	4.82	-
96.54	2,4,6-Trimethyltetracosanoic acid methyl ester	-	3.82	-	-
98.10	Cyclotetracosane	-	0.45	-	-
98.42	Cyclooctacosane	-	1.01	-	-
98.43	1-Nonadecene	-	-	1.48	-
100.55	Docosanoic acid ethyl ester	0.93	-	-	3.93
100.56	2,9-Dimethyltetracosanoic acid methyl ester	-	-	3.92	-



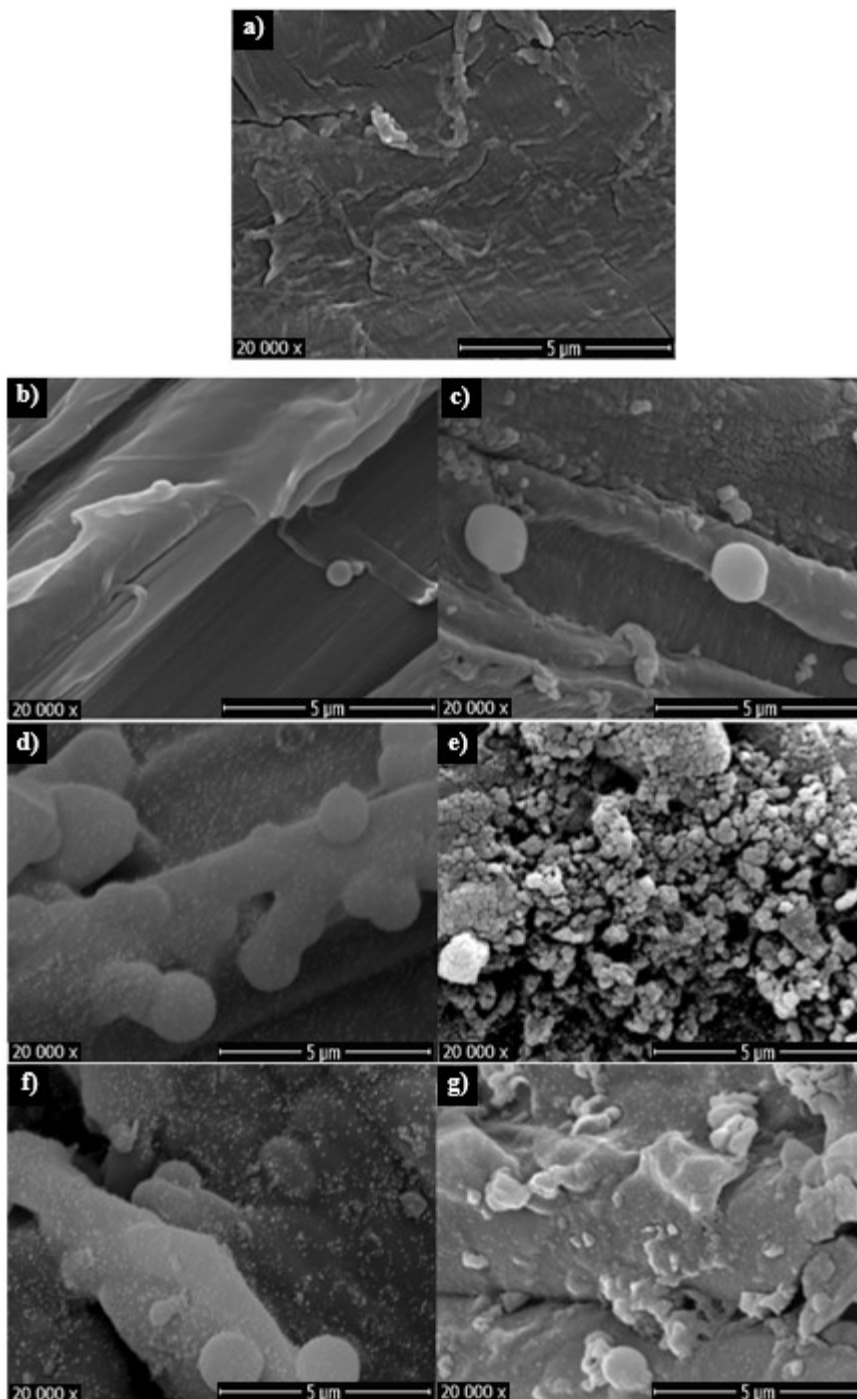
**Figure S1.** HSQC spectra of the bio-oils from the supercritical ethanol liquefaction of teak wood at 300 °C for 30 min using catalyst by 2 and 10 mmol  $\text{Mg}(\text{ClO}_4)_2$ .



**Figure S2.** HSQC spectra of the bio-oils from the supercritical ethanol liquefaction of teak wood at 300 °C for 30 min using catalyst by 2 and 10 mmol HClO<sub>4</sub>.



**Figure S3.** HSQC spectra of the bio-oils from the supercritical ethanol liquefaction of teak wood at 300 °C for 30 min using catalyst by 2 mmol/10 mmol and 10 mmol/2 mmol  $Mg(ClO_4)_2/HClO_4$ .



**Figure S4.** SEM images of teak wood and solid residues from the supercritical ethanol liquefaction of teak wood without and with catalysts at 300 °C for 30 min. a) Teak wood b) Solid residue from the non-catalytic hydrothermal liquefaction of teak wood c) Solid residue from the catalytic run with 10 mmol of  $\text{Mg}(\text{ClO}_4)_2$ , d) Solid residue from the catalytic run with 10 mmol of  $\text{HClO}_4$  e) Solid residue from the catalytic run with  $\text{Mg}(\text{ClO}_4)_2/\text{HClO}_4$  (10 mmol:2 mmol), f) Solid residue from the catalytic run with  $\text{Mg}(\text{ClO}_4)_2/\text{HClO}_4$  (2 mmol:10 mmol) g) 5 mmol  $\text{Mg}(\text{ClO}_4)_2/5$  mmol  $\text{HClO}_4$