

Electronic Supplementary Information (ESI)

Honeycomb-Like Carbon with Doping of Transition-Metal and Nitrogen for Highly Efficient Zinc-Air Battery and Zinc-Ion Battery

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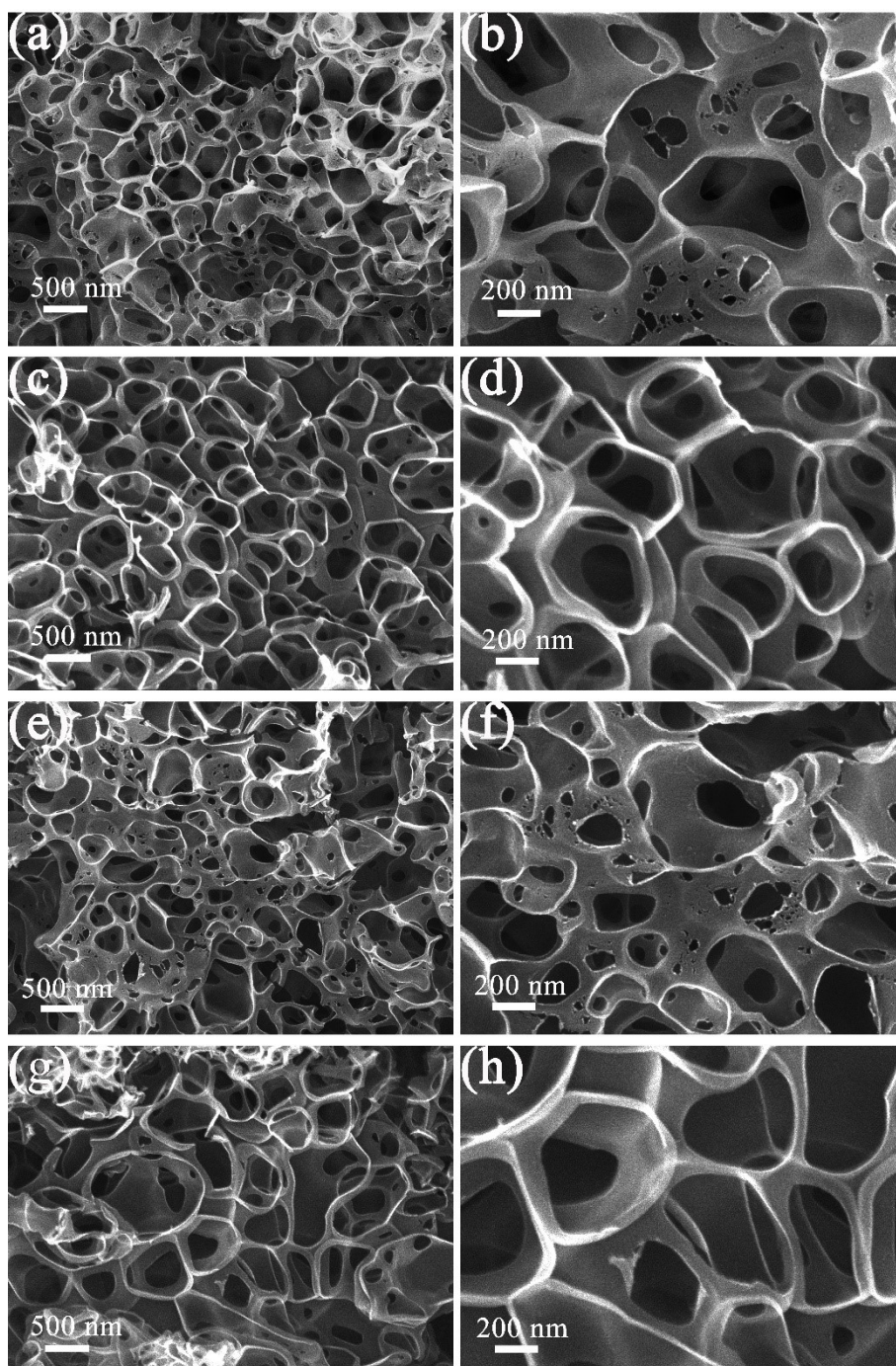


Fig. S1 SEM images of (a, b) FeZnN-C; (c, d) FeN-C; (e, f) ZnN-C; (g, h) N-C catalysts.

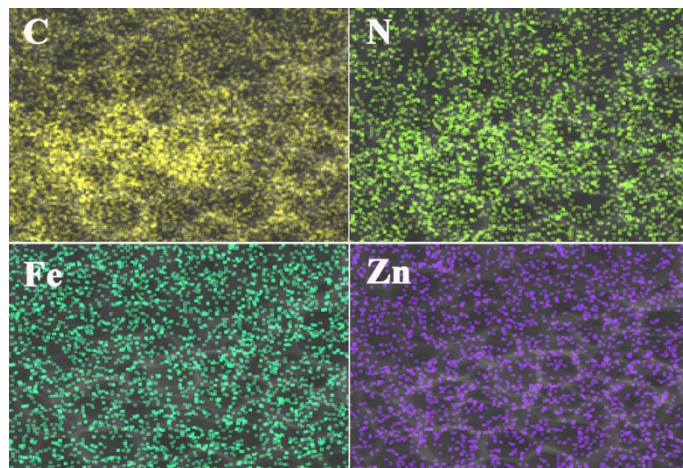


Fig. S2 The EDX mapping images of FeZnN-C for C, N, Fe and Zn elements.

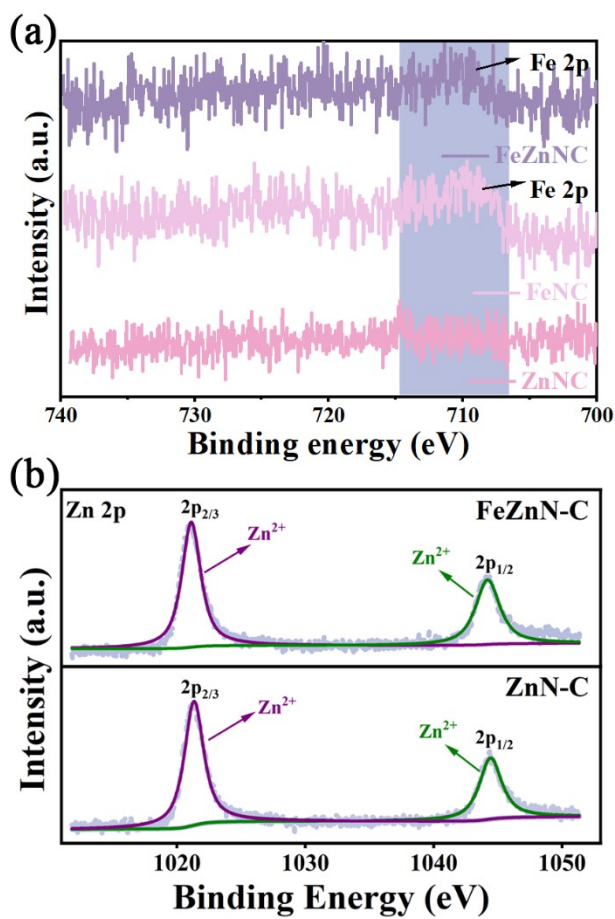


Fig. S3 (a) Fe 2p spectrum of FeN-C, ZnN-C, FeZnN-C catalysts; (b) Zn 2p spectrum of FeZnN-C and ZnN-C catalysts.

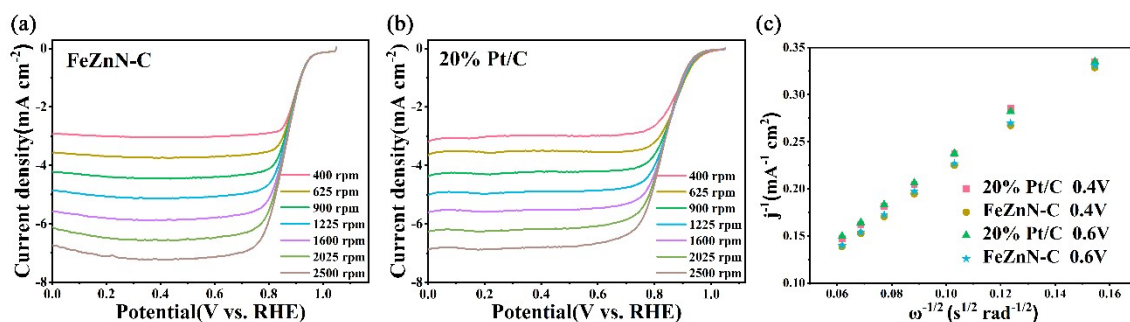


Fig. S4 The LSV curves of (a) FeZnN-C and (b) 20% Pt/C at different rotation rates; (c) the corresponding Koutecky-Levich curves at 0.4 V and 0.6 V.

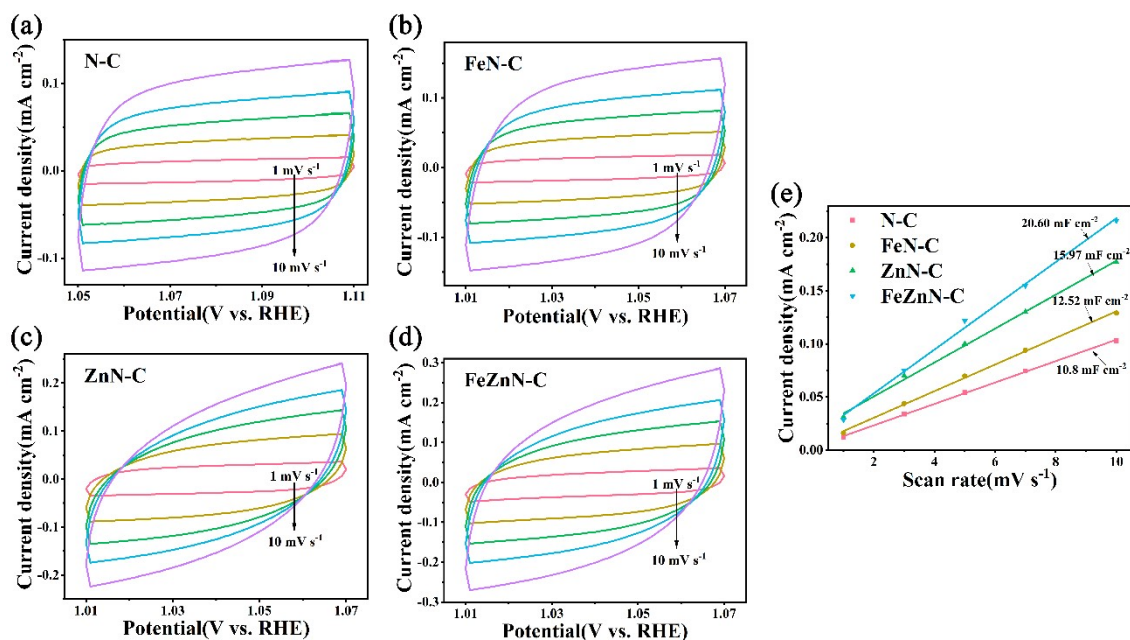


Fig. S5 CV curves of (a) N-C, (b) FeN-C, (c) ZnN-C, (d) FeZnN-C at different scan rates in Ar-saturated 1 M KOH; (e) the corresponding C_{dl} values.

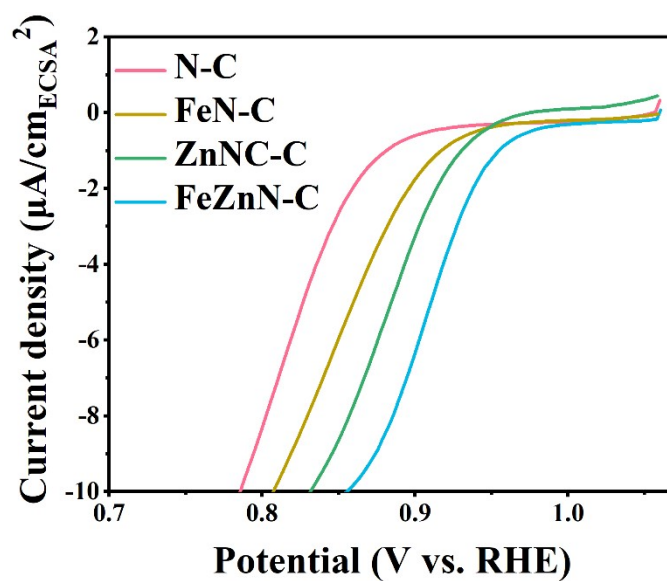


Fig. S6 The ECSA normalized LSV curves of N-C, FeN-C, ZnN-C and FeZnN-C.

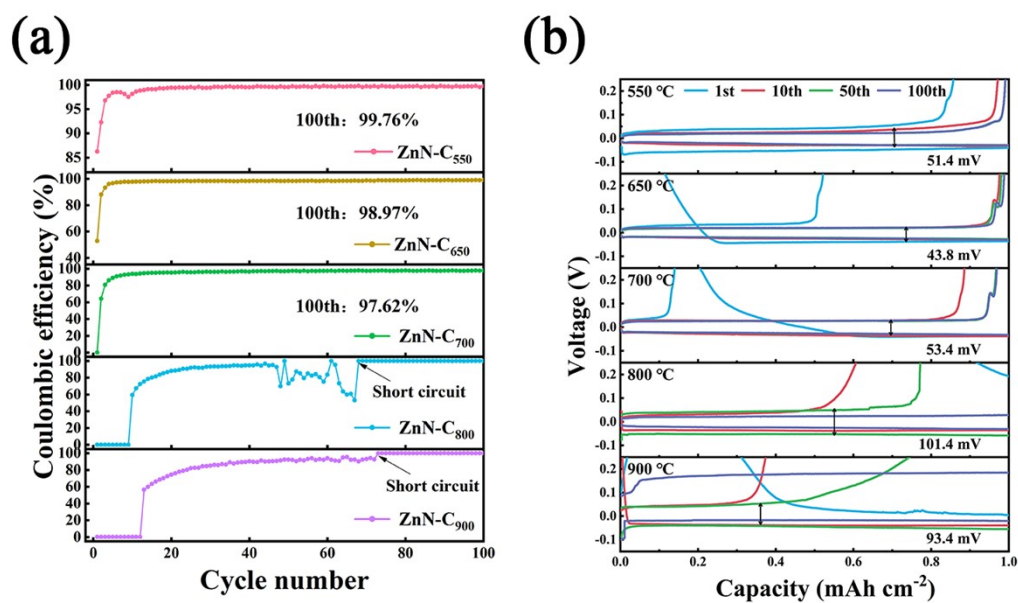


Fig. S7 (a) Coulombic efficiency of coin cell of $\text{ZnN-C}_x \parallel \text{Zn}$ plate, (b) Polarization curves of the plating/stripping of ZnN-C_x at $1 \text{ mA cm}^{-2} - 1 \text{ mAh cm}^{-2}$.

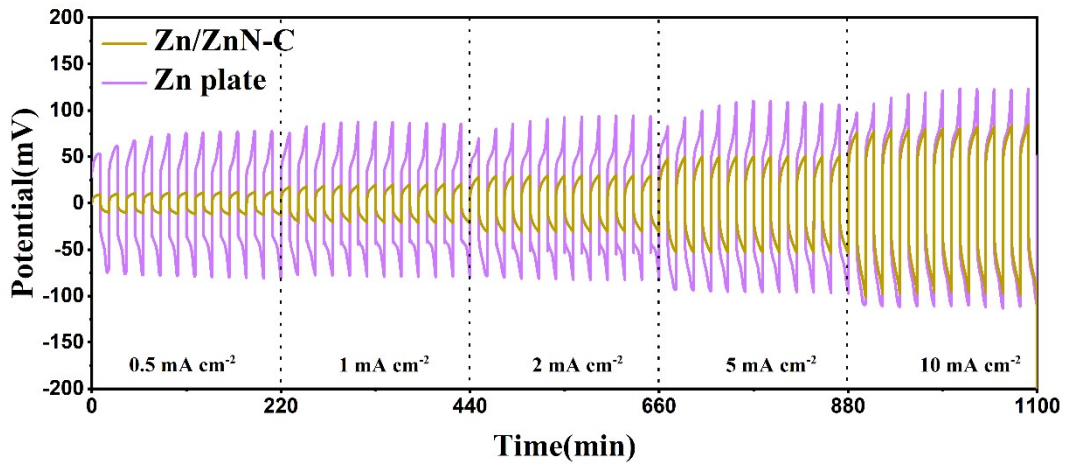


Fig. S8 Rate performance of symmetric coin cells of Zn/ZnN-C || Zn/ZnN-C and Zn plate || Zn plate.

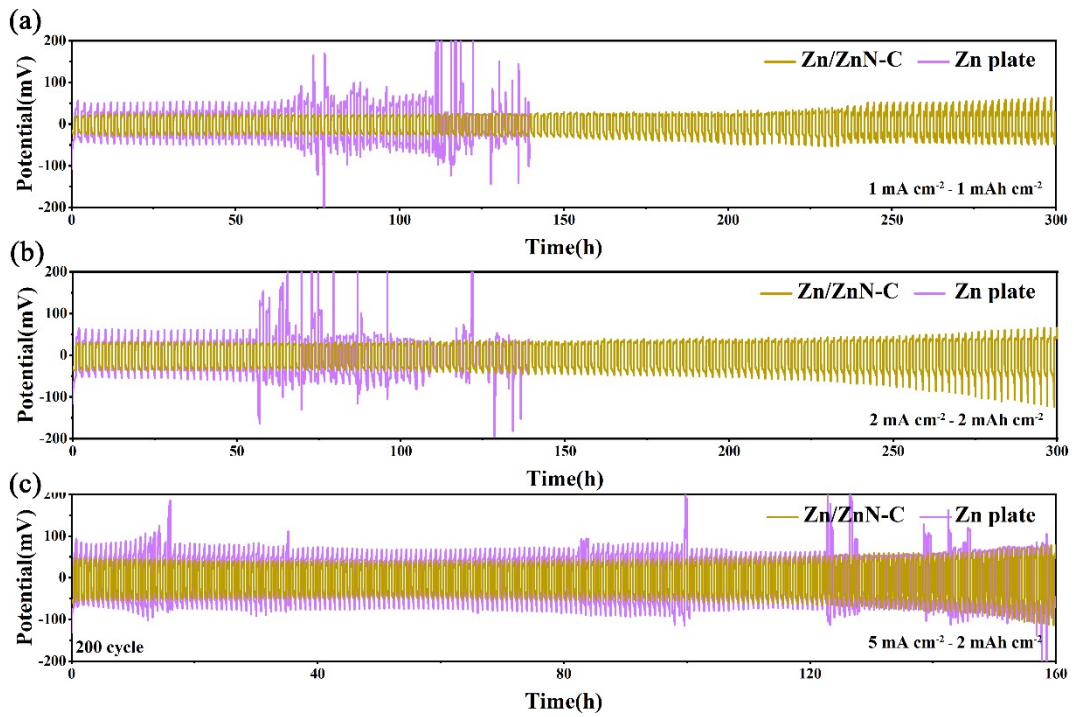


Fig. S9 Electrochemical stability (Zn plating/stripping) tests of symmetric coin cells of Zn/ZnN-C || Zn/ZnN-C and Zn plate || Zn plate at different current densities - area capacities: (a) 1 mA cm^{-2} - 1 mAh cm^{-2} ; (b) 2 mA cm^{-2} - 2 mAh cm^{-2} ; (c) 5 mA cm^{-2} - 2 mAh cm^{-2} .

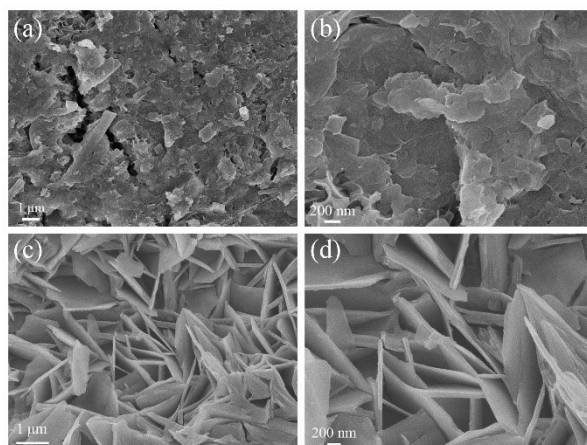


Fig. S10 The SEM images of (a-b) Zn/ZnN-C and (c-d) commercial Zn plate after plating/stripping tests.

Table S1. Atomic percentage of different elements and various N species for N-C, FeN-C, ZnN-C, FeZnN-C catalysts.

Atomic %		N-C		FeN-C		ZnN-C		FeZnN-C			
C		86.82		85.40		85.05		84.17			
N	Pyridinic N	8.95	3.019	9.58	3.459	8.10	2.544	9.85	3.158		
	Graphitic N		3.036						3.303	2.701	3.367
	Metal N		0						1.146	1.418	1.915
	Pyrrolic N		2.227						1.259	0.890	0.924
	Oxidized N		0.668						0.413	0.547	0.486
Effective N		67.70		82.50		82.30		85.70			
O		4.22		4.74		6.12		5.17			
Fe		0		0.28		0		0.21			
Zn		0		0		0.73		0.60			

Table S2. The comparison of ORR activity of different catalysts in 0.1 M KOH.

Catalysts	E_{onset} (V vs.RHE)	$E_{1/2}$ (V vs.RHE)	j_L (mA cm ⁻²)	References
FeZnN-C	1.052	0.901	-6.04	This work
ZnN-C	1.039	0.856	-5.72	This work
FeN-C	0.959	0.810	-5.16	This work
N-C	0.925	0.789	-4.51	This work
FeNi@NCNT-CP	1.0	0.85	-6.0	1
CoNi@N-DCNT	0.92	0.83	-6.0	2
Fe/Co-N/P-9	0.96	0.85	-4.5	3
SA-Fe-HPC	/	0.89	-5.4	4
NC-Co SA	1.00	0.87	/	5
FeN-HPC _{GD2}	0.98	0.888	-5.42	6
FeNC-1	/	0.90	-6.0	7
Fe-Zn-SA/NC	0.94	0.85	/	8
Fe/N-PCNs	0.96	0.86	/	9
(Zn,Co)/NSC	1.07	0.893	-5.4	10

Table S3. The performance of Zn-air batteries of nonprecious catalysts in 6 M KOH.

Catalysts	Peak power density (mW cm ⁻²)	References
FeZnN-C	257	This work
FeN-HPC _{GD2}	241	6
Ni ₃ Fe/N-S-CNTs	180	11
Fe ₃ Mn/N-C	160.8	12
BTC-Co-O-Cu-BTA MOF	200	13
GNCNTs	207	14
CoN ₄ /NG	115	15
Zn/CoN-C	230	16
Mo-N/C@MoS ₂	196.4	17

Table S4. Discharge voltages at different current densities of Zinc-air batteries using FeZnN-C and 20% Pt/C as cathodic electrocatalysts.

Current density (mA cm ⁻²)	Cell Voltage (V)	
	FeZnN-C	20% Pt/C
1	1.44	1.37
10	1.36	1.30
50	1.24	1.18
100	1.15	1.09

Table S5. The comparison of the battery performance based on different catalysts.

Electrode	Electrolyte	Current density (mA cm ⁻²)	Lifetime (h)	Voltage hysteresis (mV)	References
Zn/ZnN-C	2 M ZnSO₄	0.5	600	21	This work
CNT-Zn	2 M ZnSO ₄	2	200	27	18
3D Zn Anode	2 M ZnSO ₄	0.5	350	40	19
CC@MnO ₂ -UTF@Zn	2 M ZnSO ₄	1	150	150	20
ZnSe@Zn	2 M ZnSO ₄	1	1530	128	21
Cu foil@Zn	2M ZnSO ₄ + 0.1 M MnSO ₄	2	140	40	22
S/MX@ZnS@Zn-300	2M ZnSO ₄	1	1100	75	23
Zn micromesh	3.0 M Zn(CF ₃ SO ₃) ₂	1	500	50	24

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