## SUPPORTING INFORMATION

## Three-dimensional coral-like Zn,O-codoped Ni<sub>3</sub>S<sub>2</sub> electrocatalyst for efficient overall water splitting at large current density

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Figure S.1 (A) Raman spectra; (B) SEM image of NiZn-LDH.



Figure S.2SEM images of  $Ni_3S_2$  samples grown on foamed nickel with different sulfur source<br/>concentrations: (A) 0.0005 mol/L TAA; (B) 0.001 mol/L TAA; (C) 0.005 mol/L TAA ; (D) 0.01<br/>mol/L TAA; Polarization curves of hydrogen evolution and oxygen evolution in 1 M KOH: (E)<br/>HER polarization curves; (F) OER polarization curves.



Figure S.3 XRD patterns of Ni<sub>3</sub>S<sub>2</sub>/NF and ZO-Ni<sub>3</sub>S<sub>2</sub>/NF.



Figure S.4 (A,B) SEM images; (C) TEM images and (D) High-resolution TEM of  $Ni_3S_2/NF$ .



**Figure S.5** Cyclic voltammograms at different speeds at a potential range of -0.88-1.02 V vs. RHE: (A) ZO-Ni<sub>3</sub>S<sub>2</sub>/NF; (B) Ni<sub>3</sub>S<sub>2</sub>/NF; (C) Summary of overpotential at 100, 500, and 1000 mA cm<sup>-2</sup> for ZO-Ni<sub>3</sub>S<sub>2</sub>/NF, Ni<sub>3</sub>S<sub>2</sub>/NF-np and 25% PtC/NF;(D) Multi-step chronoamperometric curves of HER over ZO-Ni<sub>3</sub>S<sub>2</sub>/NF.



Figure S.6 Cyclic voltammograms at different speeds at a potential range of 0.05-0.15 V vs. RHE:
(A) ZO-Ni<sub>3</sub>S<sub>2</sub>/NF; (B) Ni<sub>3</sub>S<sub>2</sub>/NF; (C) Multi-step chronoamperometric curves of OER over ZO-Ni<sub>3</sub>S<sub>2</sub>/NF; (D) The polarization curve of ZO-Ni<sub>3</sub>S<sub>2</sub>/NF-np for water splitting.



Figure S.7 SEM images (A-C) and (D-F) TEM images of ZO-Ni $_3S_2$ /NF after electrochemical test

Table S.1	The ICP	test data	of zinc	content i	n ZO-Ni <sub>2</sub> S <sub>2</sub> /NF
	I HC ICI	test aata		content h	

	1	2	3		
Initial Amount	9.81 mg				
Residual Amount	4.285 mg	4.771 mg	4.468 mg		
	Average: 4.508 mg				
NiZn-LDH/NF	5.302 mg				
Residual Amount	2.4 mg	2.1 mg	1.95 mg		
ZO-Ni <sub>3</sub> S <sub>2</sub> /NF	3.152 mg				

Catalyst	Overpotential (mV, at 100 mA cm <sup>-2</sup> )	Overpotential (mV, at 500 mA cm <sup>-2</sup> )	Tafel slope (mV dec <sup>-1</sup> )	Reference
ZO-Ni <sub>3</sub> S <sub>2</sub> /NF	145	235	102	This work
Cu-Ni <sub>3</sub> S <sub>2</sub> /Co <sub>3</sub> S <sub>4</sub> /NF	150	-	50.4	Appl. Catal. B Environ. 2021, 293, 120225
Ni <sub>2(1-x)</sub> Mo <sub>2x</sub> P	162	240	46.6	Nano Energy, 2018, 53, 492-500
Ni <sub>3</sub> S <sub>2</sub> @NiV-LDH/NF	256	~580	69	Nanoscale, 2019, 11, 8855-8863
NiFe <sub>2</sub> O <sub>4</sub> /NiFe LDH	-	297	67.1	ACS Appl. Mater. Interfaces 2018, 10, 31, 26283–26292
Ni <sub>3</sub> S <sub>2</sub> @NGCLs/NF	225	-	74	Chem. Eng. J. 2020, 401, 126045
Co-Ni <sub>3</sub> S <sub>2</sub> @CNTs/GNF	350	-	138	J. Mater. Chem. A, 2018, 6, 10490-10496
NiWO <sub>4</sub> /Ni <sub>3</sub> S <sub>2</sub>	274	-	112	Appl. Catal. B Environ. 2020, 274, 119120
δ-FeOOH/Ni <sub>3</sub> S <sub>2</sub> /NF	210	-	82.6	J. Mater. Chem. A, 2020,8, 21199-21207
NS-horn/NF	350	-	139	Appl. Catal. B Environ. 2019, 257, 117911
CoFe-OOH@Co <sub>2</sub> P/NF	~230	-	76	Electrochimica Acta, 2020, 360:136994

Table S.2 Comparison of the HER catalytic performance of  $ZO-Ni_3S_2/NF$  with other recently reported hydroxide-based electrocatalysts in alkaline solution.

catalyst	Overpotential (mV, at 100 mA cm <sup>-2</sup> )	Overpotential (mV, at 500 mA cm <sup>-2</sup> )	Tafel slope (mV dec- 1)	Reference
ZO-Ni <sub>3</sub> S <sub>2</sub> /NF	360	450	74.5	This work
Ni <sub>3</sub> S <sub>2</sub> NTFs	400	-	101.2	Appl. Catal. B: Environ. 2019, 243, 693-702
Ni <sub>2(1-x)</sub> Mo <sub>2x</sub> P	340	-	-	Nano Energy, 2018, 53, 492-500
Ni <sub>3</sub> S <sub>2</sub> @NiV-LDH/NF	320	~490	57	Nanoscale, 2019, 11, 8855-8863
Ni <sub>3</sub> S <sub>2</sub>	400 @ 10 mA cm <sup>-2</sup>	-	51	ACS Appl. Mater. Interfaces. 2018, 10, 12807-12815
Ni <sub>3</sub> S <sub>2</sub> -GQDs/NF	394	-	95.5	Small 2017, 13, 1700264
Fe-Ni <sub>3</sub> S <sub>2</sub> /FeNi	495	-	54	Small 2017, 13, 1604161
Ni(OH) <sub>2</sub> /Ni <sub>3</sub> S <sub>2</sub>	497	-	94.6	J. Mater. Chem. A 2018, 6, 6938-6946
Ni <sub>2</sub> P/Ni <sub>3</sub> S <sub>2</sub> /NF	330	-	62	Nano Energy, 2018, 51, 26-36
NS-horn/NF	370	-	120	Appl. Catal. B Environ. 2019, 257, 117911
CoFe-OOH@Co <sub>2</sub> P/NF	~220	-	39	Electrochimica Acta, 2020, 360:136994

Table S.3 Comparison of the OER catalytic performance of  $ZO-Ni_3S_2/NF$  with other recently reported hydroxide-based electrocatalysts in alkaline solution.

catalyst	Water splitting potential (V, at 10 mA cm <sup>-2</sup> )	Water splitting potential (V, at 100 mA cm <sup>-2</sup> )	Reference
ZO-Ni <sub>3</sub> S <sub>2</sub> /NF	1.548	1.806	This work
$Ni_3S_2$ NTFs	1.611	-	Appl. Catal. B: Environ. 2019, 243, 693-702
Ni <sub>3</sub> S <sub>2</sub>	2	-	ACS Appl. Mater. Interfaces. 2018, 10, 12807-12815
Ni <sub>3</sub> S <sub>2</sub> -GQDs/NF	1.58	-	Small 2017, 13, 1700264
Ni(OH) <sub>2</sub> /Ni <sub>3</sub> S <sub>2</sub>	1.57	-	J. Mater. Chem. A 2018, 6, 6938-6946
NS-horn/NF	1.57	~1.95	Appl. Catal. B Environ. 2019, 257, 117911
Cu-Ni <sub>3</sub> S <sub>2</sub> /Co <sub>3</sub> S <sub>4</sub> /NF	1.49	~1.85	Appl. Catal. B: Environ. 2021, 293, 120225
Ni <sub>3</sub> S <sub>2</sub> @NGCLs/NF	1.55	-	Chem. Eng. J. 2020, 401, 126045
δ-FeOOH/Ni <sub>3</sub> S <sub>2</sub> /NF	1.525	~1.84	J. Mater. Chem. A, 2020,8, 21199-21207
CoS <sub>x</sub> /Ni <sub>3</sub> S <sub>2</sub> @NF	1.573	-	ACS Appl. Mater. Interfaces, 2018, 10, 27712–27722

Table S.4 Comparison of the water splitting performance of  $ZO-Ni_3S_2/NF$  with other recently reported hydroxide-based electrocatalysts in alkaline solution.