Electronic Supplementary Information (ESI)

A nonflammable phosphate-based localized high-concentration electrolyte for safe and high-voltage lithium metal batteries

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Fig. S1 The Photograph of flammability testing of HFE.



Fig. S2 Electrochemical performance of Li||Cu cells with different LiDFOB-DMMP/HFE electrolytes. Li plating/stripping profiles of Li||Cu cells with (a) 3.6 M LiDFOB-DMMP, (b) 1.7 M LiDFOB-DMMP/HFE, (c) 1.1 M LiDFOB-DMMP/HFE and (d) 0.8 M LiDFOB-DMMP/HFE electrolytes at a current density of 0.5 mA cm⁻² and a deposition capacity of 1 mAh cm⁻². (e) The corresponding CEs of Li||Cu cells with different electrolytes. (f) The average CEs of Li||Cu cells with different electrolytes over 30 cycles.



Fig. S3 The viscosity of 3.6 M LiDFOB-DMMP and 1.1 M LiDFOB-DMMP/HFE electrolytes.



Fig. S4 The contact angles of (a) 3.6 M LiDFOB-DMMP and (b) 1.1 M LiDFOB-DMMP/HFE electrolytes over the surface of Celgard 2325 separator.



Fig. S5 The cyclic performance of Li||Li symmetrical cells with different electrolytes at (a) 2 mA cm⁻², 1 mAh cm⁻²; (b) 3 mA cm⁻², 1 mAh cm⁻².



Fig. S6 Impedance spectra of the Li||Li cells cycled at 0.5 mA cm⁻² after 10th cycle.



Fig. S7 Morphologies of the deposited Li in 1 M LiPF₆-EC/DMC and 1.1 M LiDFOB-DMMP/HFE electrolytes. (a, b) Top and (c, d) cross-sectional SEM images of the deposited Li obtained by plating 2 mAh cm⁻² of Li on Cu substrate at a current density of 0.5 mA cm⁻².



Fig. S8 The C and F atomic contents by XPS measurement of the SEI layer formed in (a) 1 M LiPF₆-EC/DMC and (b) 1.1 M LiDFOB-DMMP/HFE electrolytes.



Fig. S9 Oxidative stabilities of 1 M LiPF₆-EC/DMC and 1.1 M LiDFOB-DMMP/HFE electrolytes in a two-electrode cell with a stainless steel as the working electrode and Li metal as the reference/counter electrode at a scan rate of 1 mV s⁻¹.



Fig. S10 The initial three charge-discharge profiles of Li||NMC622 cells with (a) 1 M LiPF₆-EC/DMC and (b) 1.1 M LiDFOB-DMMP/HFE electrolytes in the voltage range of 2.8–4.3 V at 0.1C.



Fig. S11 Electrochemical performance of Li||NMC622 full cells with different electrolytes under the charging cutoff voltage of 4.3 V. Galvanostatic charge–discharge curves with (a) 1 M LiPF₆-EC/DMC and (b) 1.1 M LiDFOB-DMMP/HFE electrolytes at 0.5 C. (c) Cycling stability and CEs at 0.5 C.



Fig. S12 The initial three charge-discharge profiles of Li||NMC622 cells with (a) 1 M LiPF₆-EC/DMC and (b) 1.1 M LiDFOB-DMMP/HFE electrolytes in the voltage range of 2.8–4.5 V at 0.1C.



Fig. S13 The representative charge–discharge profiles of Li||NMC622 cells at different C-rates with (a) 1 M LiPF₆-EC/DMC and (b) 1.1 M LiDFOB-DMMP/HFE electrolytes.



Fig. S14 Electrochemical performance of Li||NMC622 battery using 1.1 M LiDFOB-DMMP/HFE electrolyte with ultrathin Li (50 μ m), high loading cathode (18 mg cm⁻²) and medium electrolyte (50 μ L) under the voltage window of 2.8–4.5 V. (a) Representative charge–discharge profiles and (b) corresponding cycling performance. The battery is firstly activated at 0.1 C for three cycles, and then tested at 0.5 C for 100 cycles.



Fig. S15 The C and F atomic composition ratios by XPS measurement of the CEI film formed in 1 M LiPF₆-EC/DMC (Carbonate-electrolyte) and 1.1 M LiDFOB-DMMP/HFE electrolytes (LHCE).