

Supporting information

Nanosized High Entropy Spinel Oxide (FeCoNiCrMn)₃O₄ as a High-active and Ultra-stable Electrocatalyst for Oxygen Evolution Reaction

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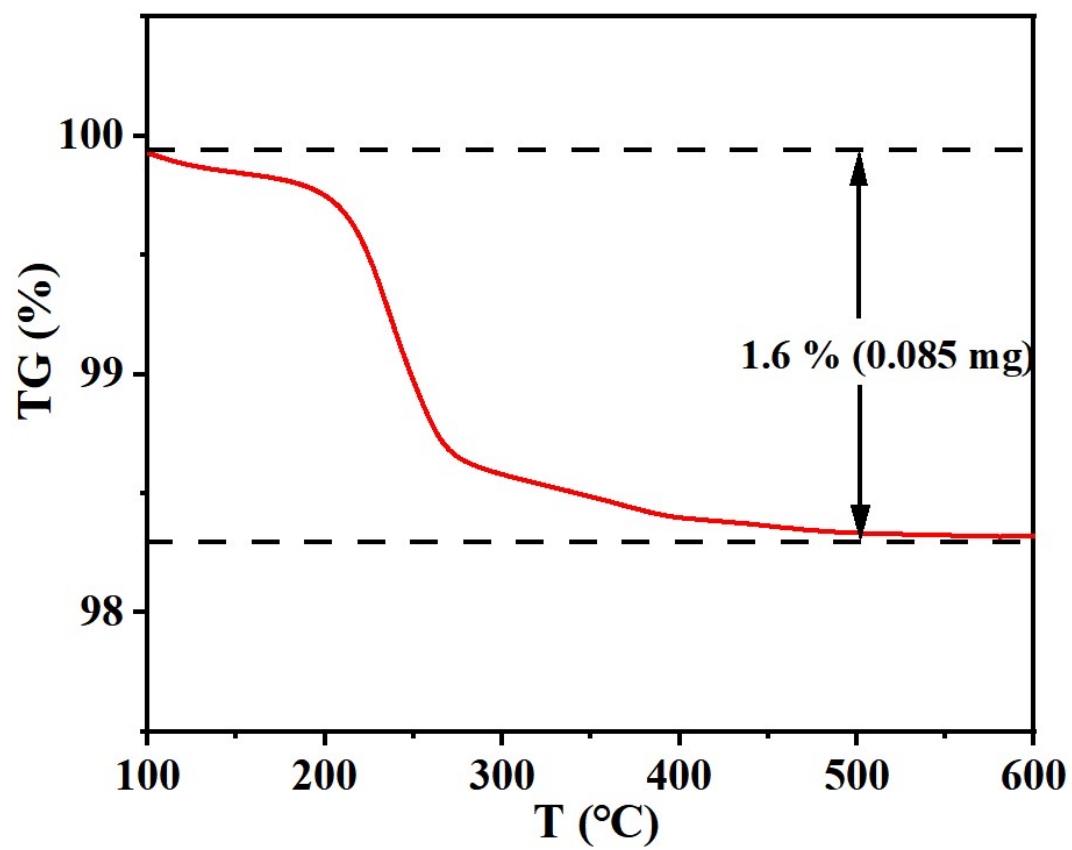


Fig.S1 The TG pattern of the HEO precursor

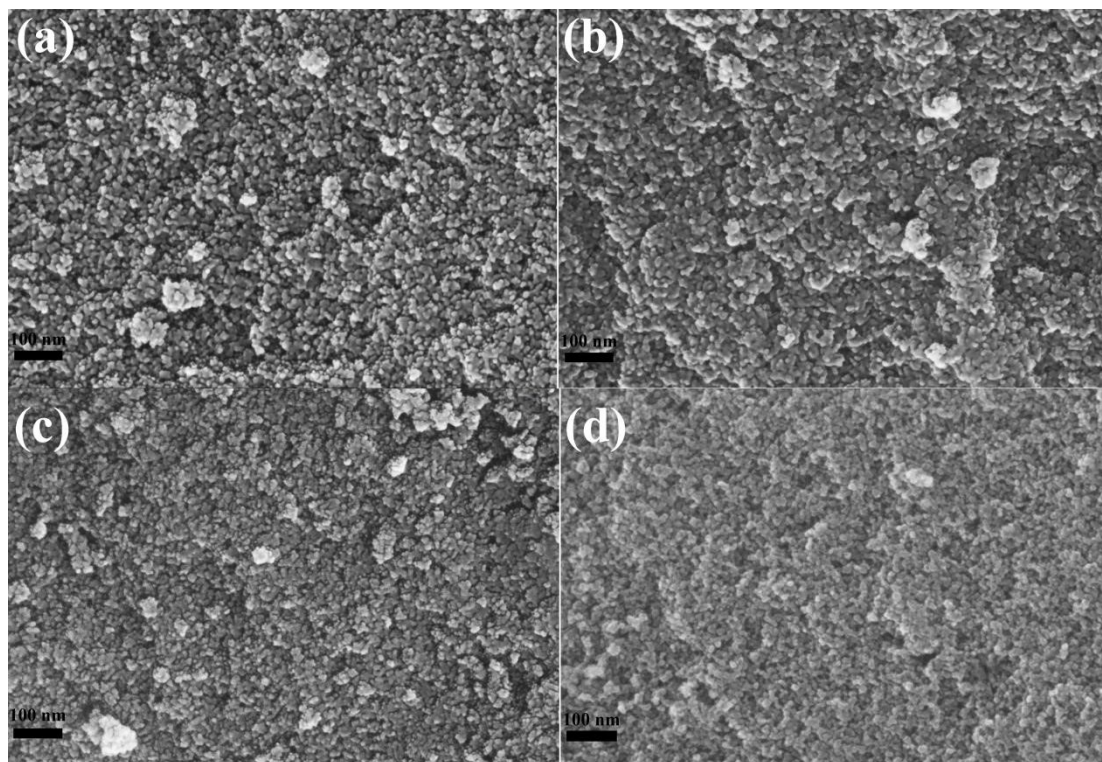


Fig.S2 SEM image of M_3O_4-T ($T=0$ 、 300 、 400 、 $500^{\circ}C$)

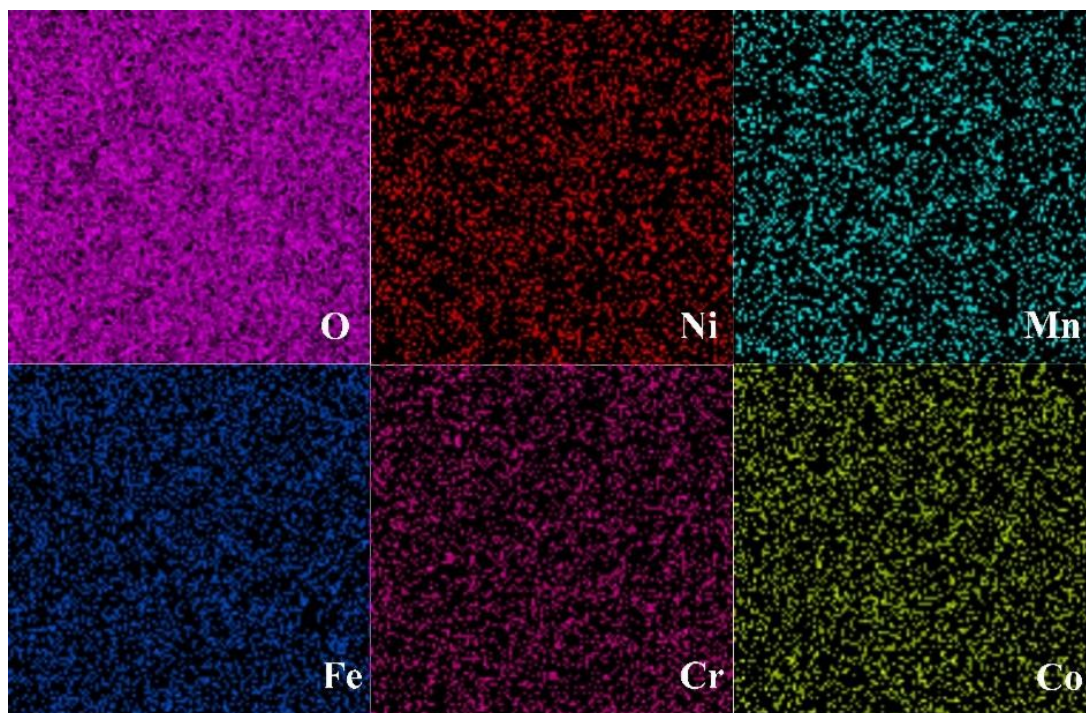


Fig.S3 SEM-EDS-Mapping of M₃O₄-400

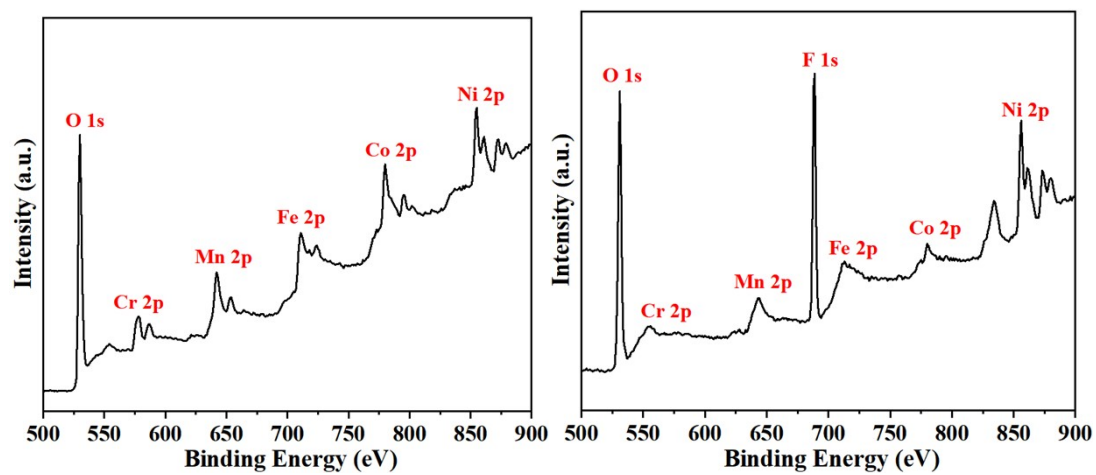


Fig.S4 The survey XPS spectra of M_3O_4-400 : (a) before OER test (b) after OER test

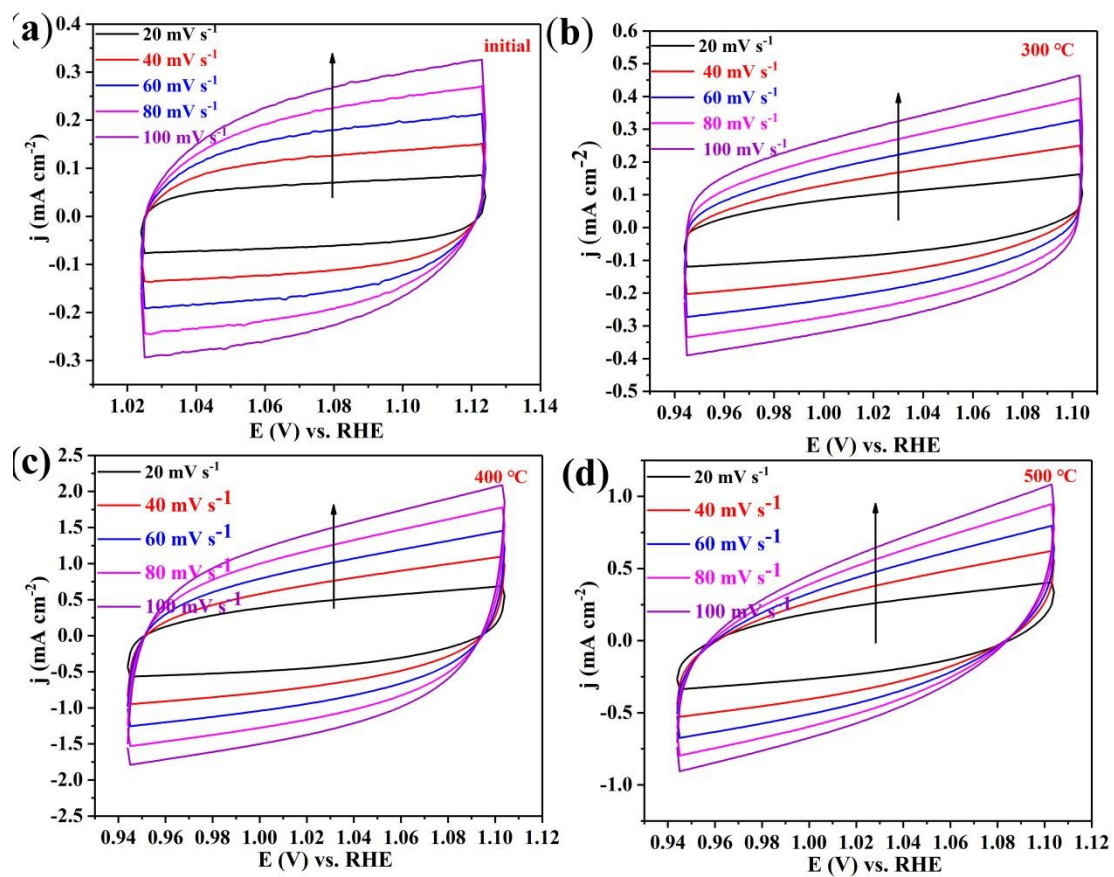


Fig.S5 CV curves at 20 -100 mV s⁻¹ scan rates of M₃O₄-T (T=0 、300 、400 、500 °C)

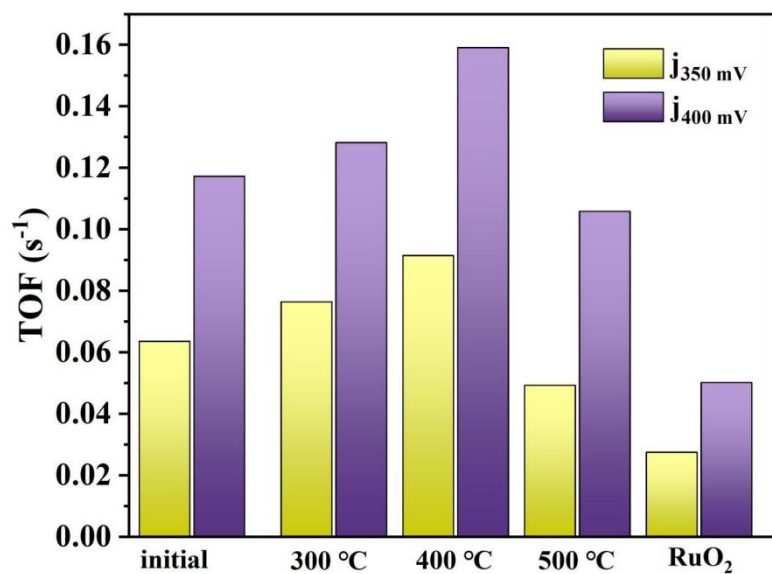


Fig.S6 TOF values of M_3O_4 -T (T=0、300、400、500 °C) and RuO_2 at $j_{350 \text{ mV}}$ and $j_{400 \text{ mV}}$
 The TOF was calculated at the overpotential of 400 mV and 350 mV as the follow equation:

$$TOF = \frac{jA}{4Fn}$$

where, j is current density ($A \text{ cm}^{-2}$), A is the geometric area of the electrode ($3 \times 3 \text{ cm}^2$), F is Faraday constant ($96485 \text{ s A mol}^{-1}$), and n is the number of moles of the metal in the electrode.

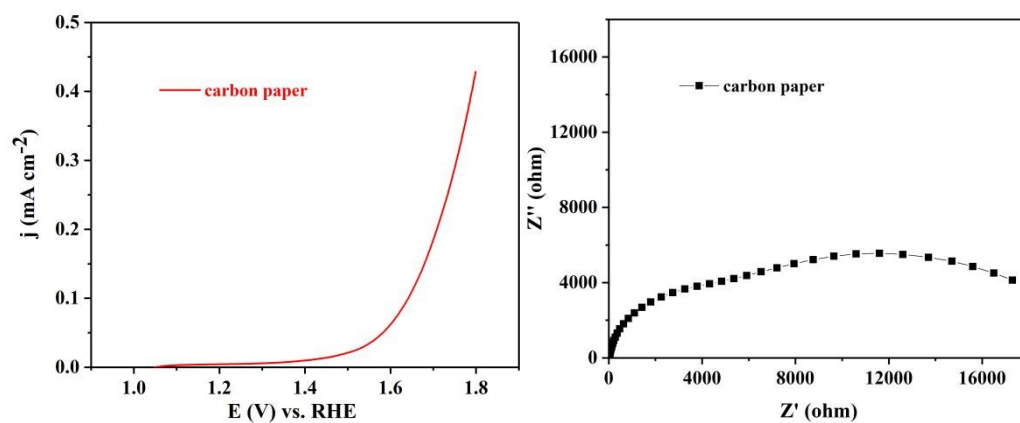


Fig. S7. Carbon paper performance test: (a) LSV curves, (b) Nyquist plots

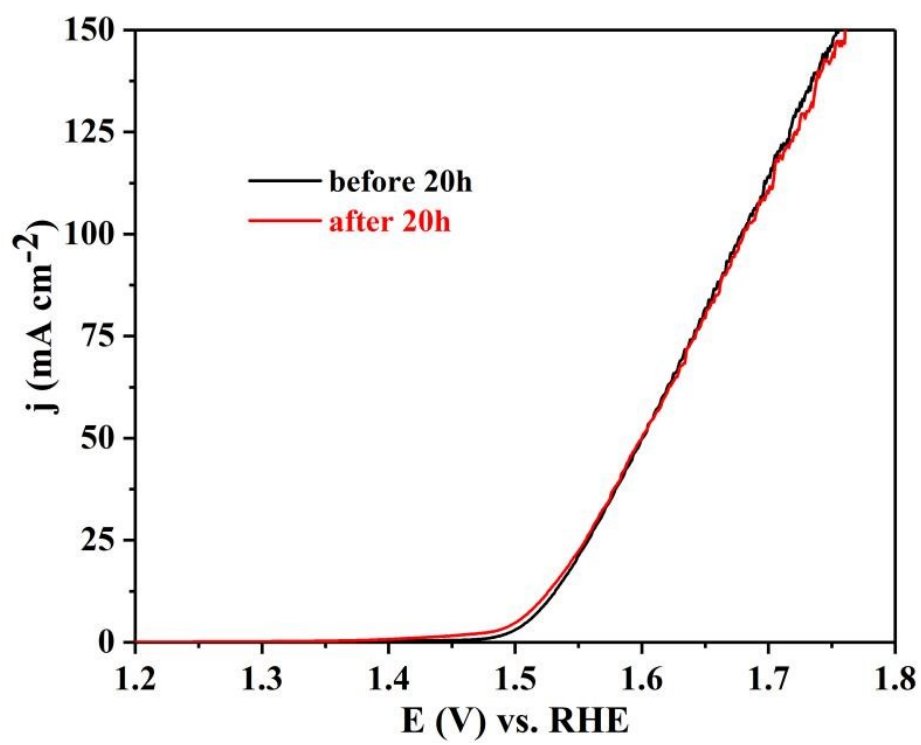


Fig.S8 before and after time-dependent current density for 20h

Table S1 Element composition of M₃O₄-400 (SEM-EDS)

element	atomic percentage (%)
O	65.54
Cr	5.01
Mn	3.53
Fe	8.54
Co	8.48
Ni	8.91

Table S2 Element composition of M₃O₄-400 (TEM-EDS)

element	atomic percentage (%)
O	67.19
Cr	4.79
Mn	3.70
Fe	8.82
Co	7.98
Ni	7.52

Table4 S3 summary of metal oxide for oxygen evolution reaction

Catalysts	η_{10} (V)	Tafel slope (mV dec ⁻¹)	journal	Ref.
(Fe,Co,Ni,Cr,Mn) ₃ O ₄	1.51	60		This work
(Co, Cu, Fe, Mn,Ni) ₃ O ₄	1.58	59.5	J. Mater. Chem. A	[1]
(CoNiMnZnFe) ₃ O _{3.2}	1.57	47.5	J. Alloy. Compd.	[2]
La(CrMnFeCo ₂ Ni)O ₃	1.56	51.2	Adv. Funct. Mater.	[3]
NiCo-(FeCrCoNiAl _{0.1})O _x	1.56	97.4	J. Alloy. Compd.	[4]
(Co _{0.2} Mn _{0.2} Ni _{0.2} Fe _{0.2} Zn _{0.2})Fe ₂ O ₄	1.57	53.6	ACS Appl. Mater. Interfaces	[5]
La _{0.6} Sr _{0.4} Co _{0.8} Fe _{0.1} Mn _{0.1} O _{3-δ}	1.57	63	Chem. Eng. J.	[6]
meso-NPC/Co ₂ NiO _x	1.56	54	ACS Appl. Mater. Interfaces	[7]
S-Co ₃ O ₄	1.54	67.4	Appl. Surf. Sci.	[8]
NiCo ₂ O ₄	1.52	53	Angew. Chem.-Int. Edit.	[9]
Fe-La _{0.6} Sr _{0.4} CoO ₃	1.57	63	Chemical Engineering Journal	[10]
Mn _{0.6} Zn _{0.4} Co ₂ O ₄	1.55	80.6	Int. J. Hydrog. Energy	[11]
CoFe ₂ O ₄ /CNTs/FA	1.62	82	Ceramics International	[12]
Vo-MnCo ₂ O ₄	1.63	87	Chem. Eng. J.	[13]
NiO	1.55	52.4	Electrochimica Acta	[14]
Mn ₃ O ₄	1.55	85.6	Int. J. Hydrog. Energy	[15]

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