

A Solid Electrolyte Interphase to Protect the Sulfurized Polyacrylonitrile (SPAN) Composite for Li-S Batteries: Computational Approach Addressing the Electrolyte/SPAN Interfacial Reactivity

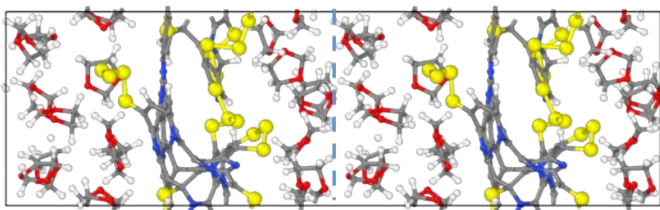
Saul Perez Beltran¹, and Perla B. Balbuena^{1,2,3,*}

¹Department of Chemical Engineering, ²Department of Materials Science and Engineering, and

³Department of Chemistry, Texas A&M University, College Station, Texas 77843, United States

*e-mail: balbuena@tamu.edu

(a) Li/S = 0.0 with DOL @ 22.6 ps



(b) Li/S = 3.0 with DOL @ 23.4 ps

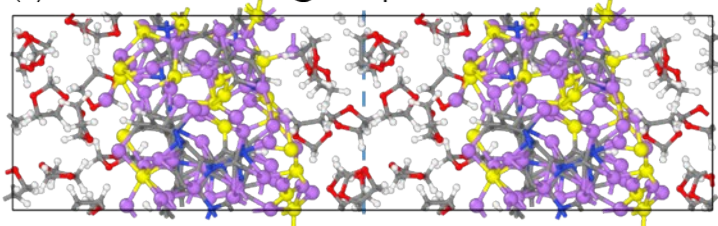


Figure S1: DOL/SPAN interfacial AIMD simulations: (a) Li/S = 0.0, (b) Li/S = 3.0. Color code for atoms as in Figure 2.

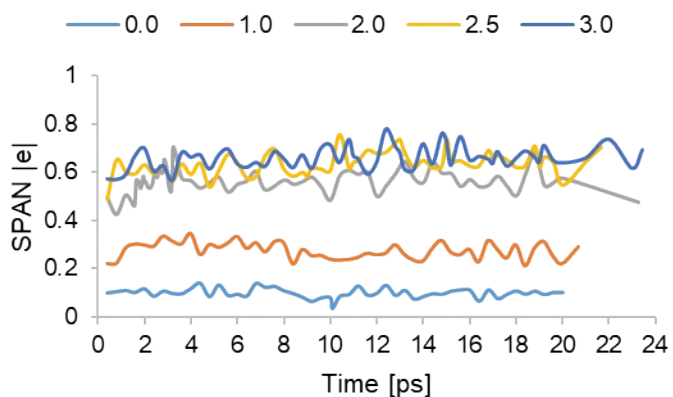


Figure S2: Time evolution of the SPAN surface electronic charge with increasing Li/S molar ratios. DOL/SPAN interfacial interaction.

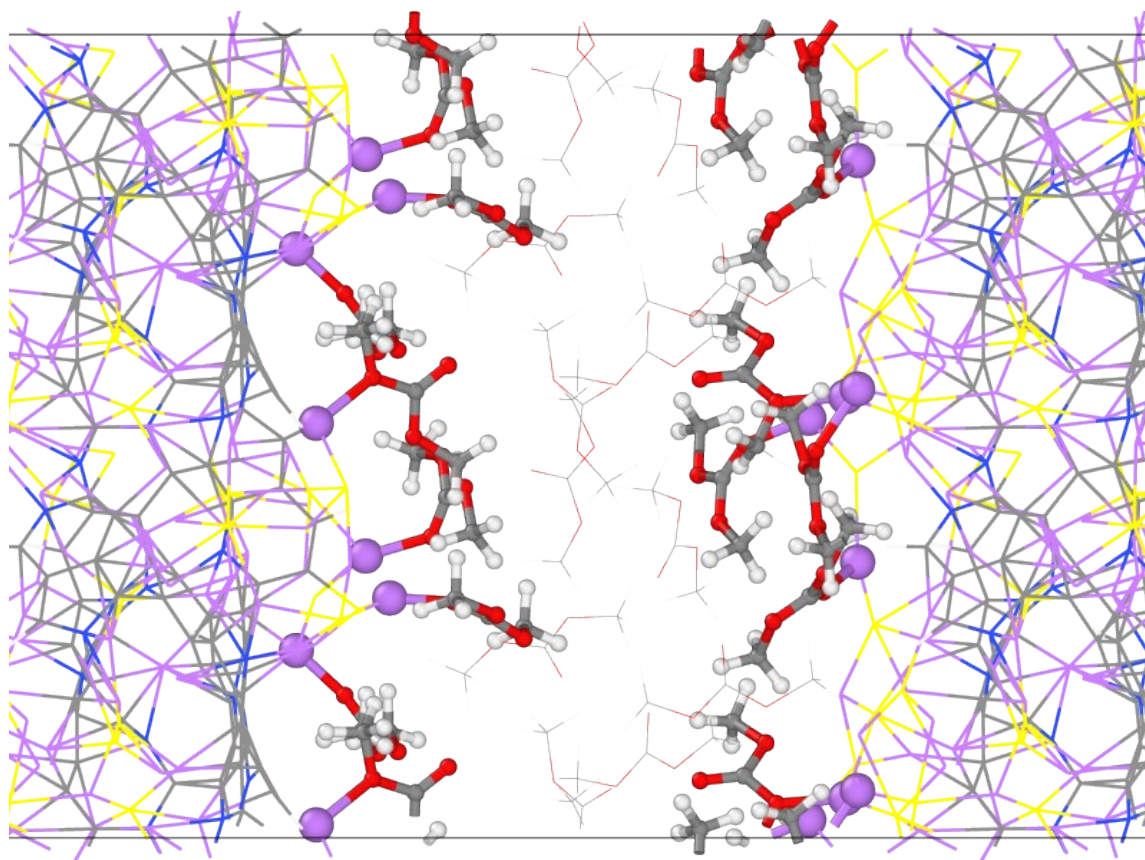


Figure S3: DMC interaction with the lithiated SPAN surface with a 3.0 Li/S molar ratio. Color code for atoms as in Figure 2.

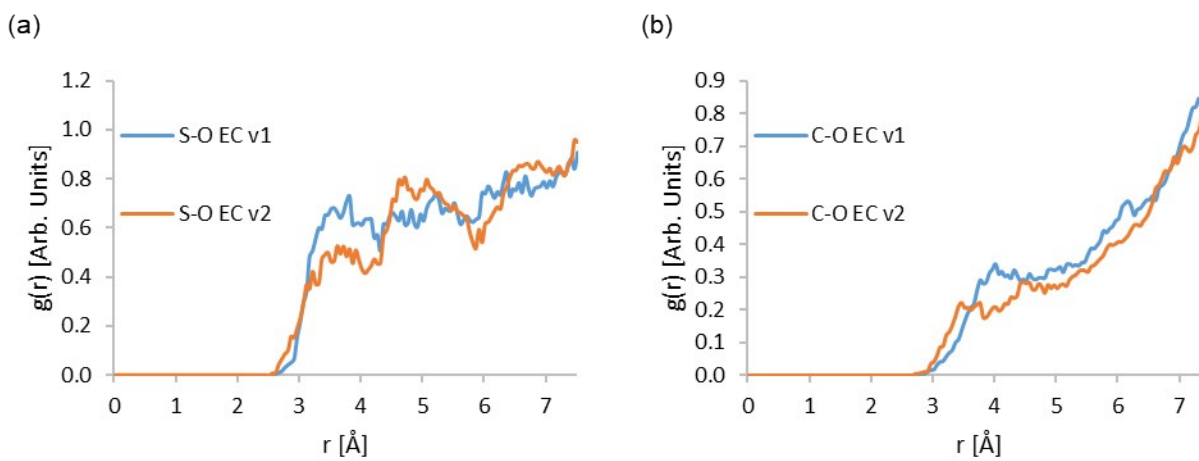


Figure S4: Partial radial distribution function (PRDF) for (a) S-O and (b) C-O and interactions for the EC interfacial interactions with the SPAN surface before lithiation.

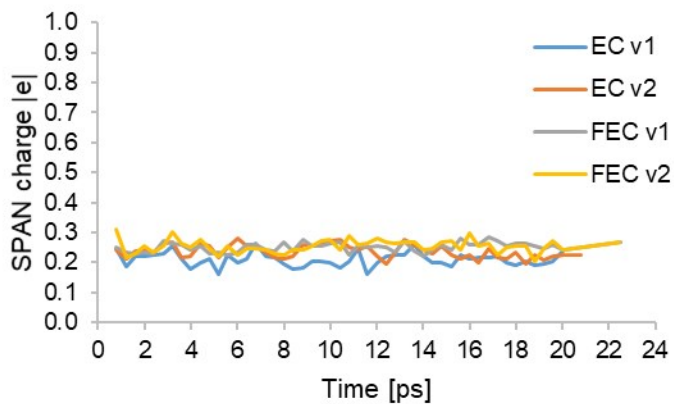


Figure S5: Time evolution for the SPAN electronic charge before lithiation: Simulations with EC and FEC.

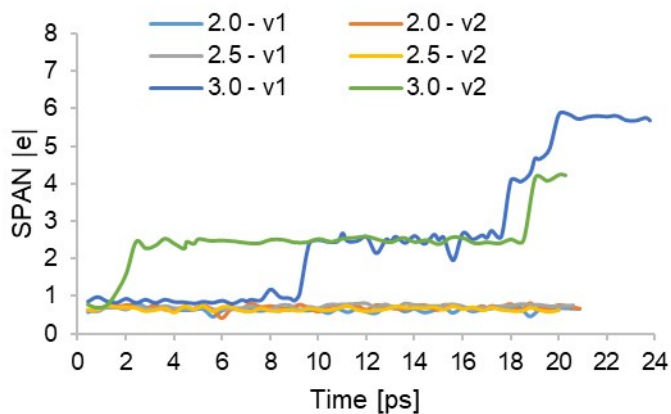


Figure S6: Surface charge time evolution for EC with increasing lithium loadings.

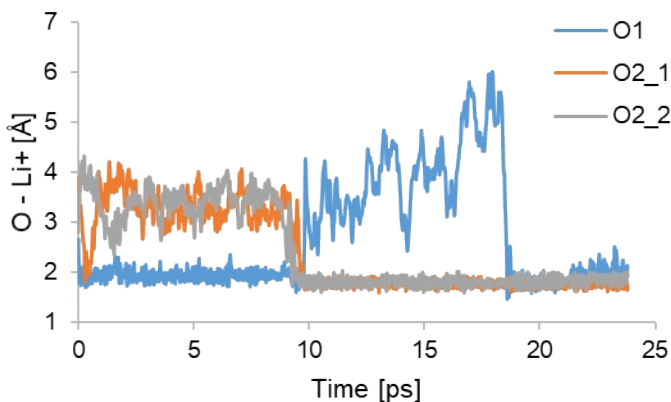


Figure S7: O-Li⁺ distances for 4.ec molecule.

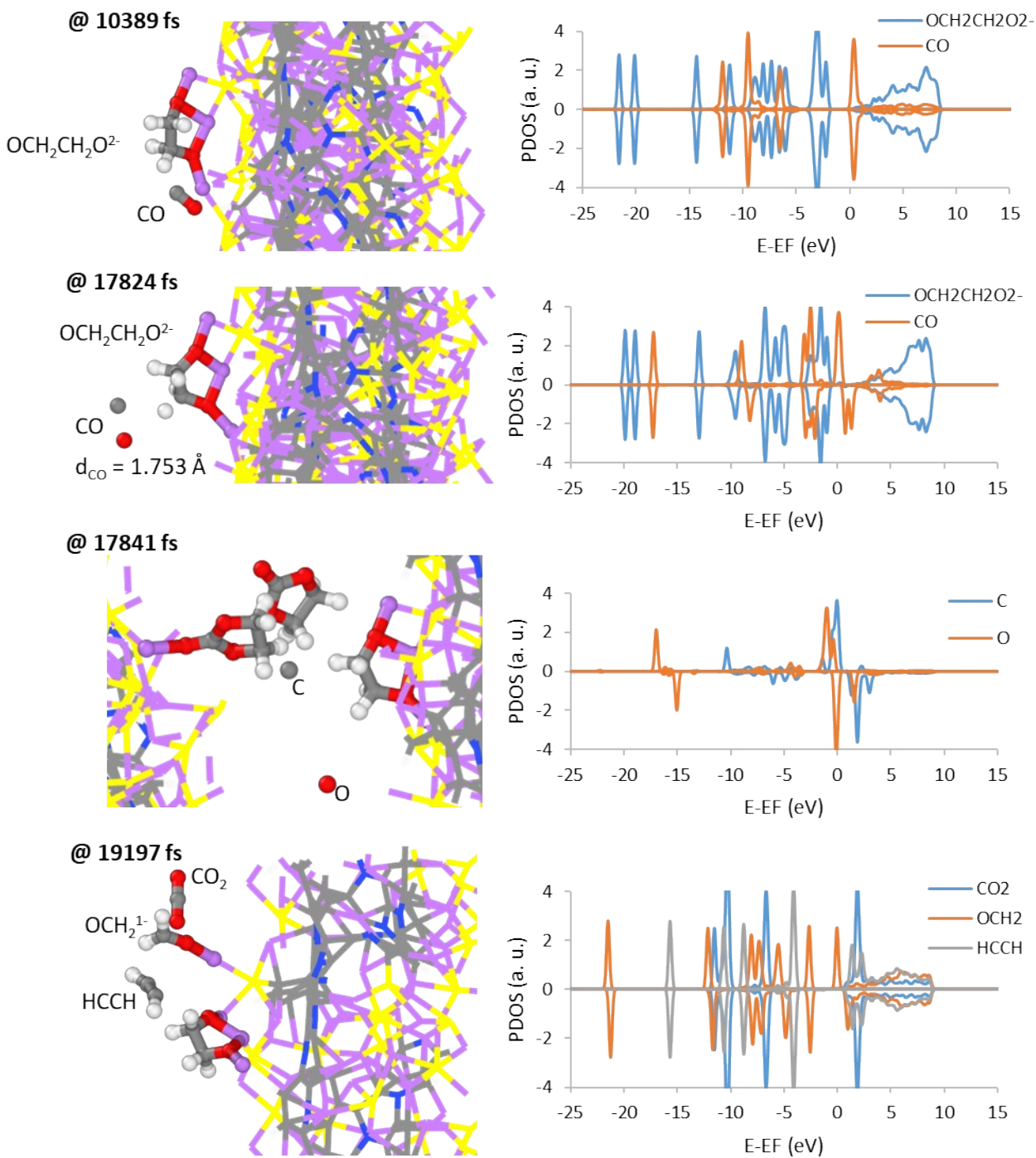


Figure S8: Projected density of states (pDOS) analysis performed on selected frames for the reaction depicted in Figure 8. Color code for atoms as in Figure 2.

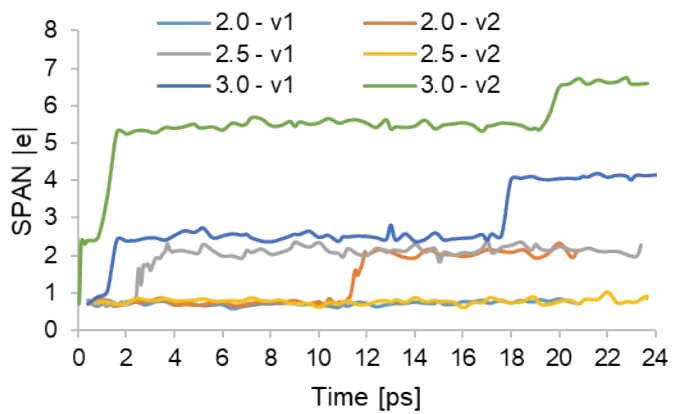


Figure S9: Surface charge time evolution for FEC with increasing lithium loadings.