

Supporting Information

In-situ growth of polyimide nanoarrays onto conductive carbon supports for high-rate charge storage and long-lived metal-free cathodes

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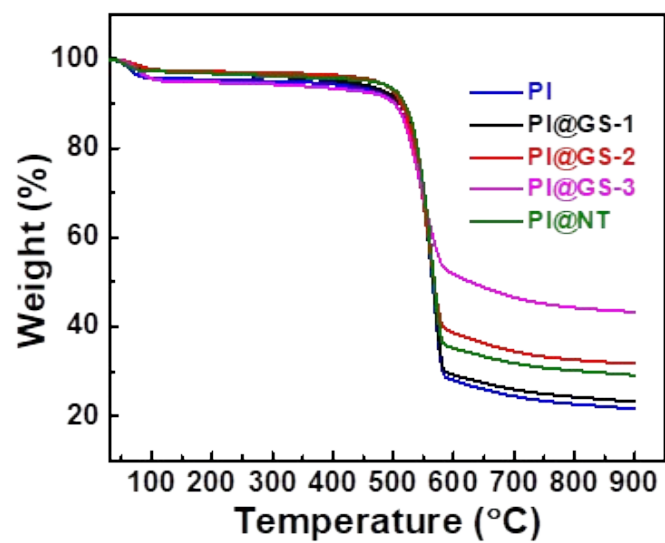


Fig. S1. TGA curves of pure PI, PI@GS, and PI@NT composites.

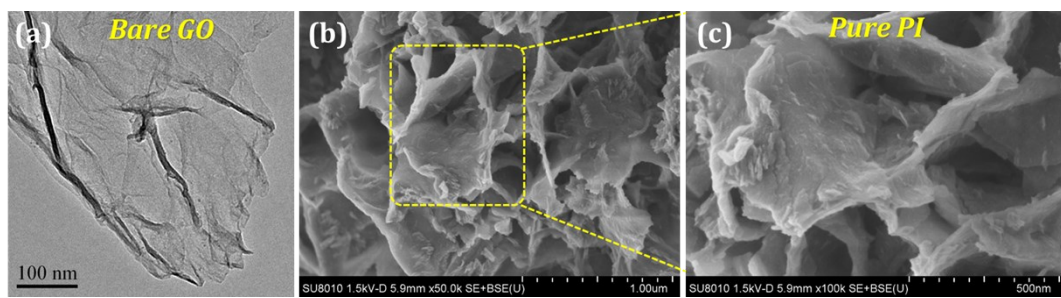


Fig. S2. (a) TEM of bare GO and (b, c) SEM images of pure PI.

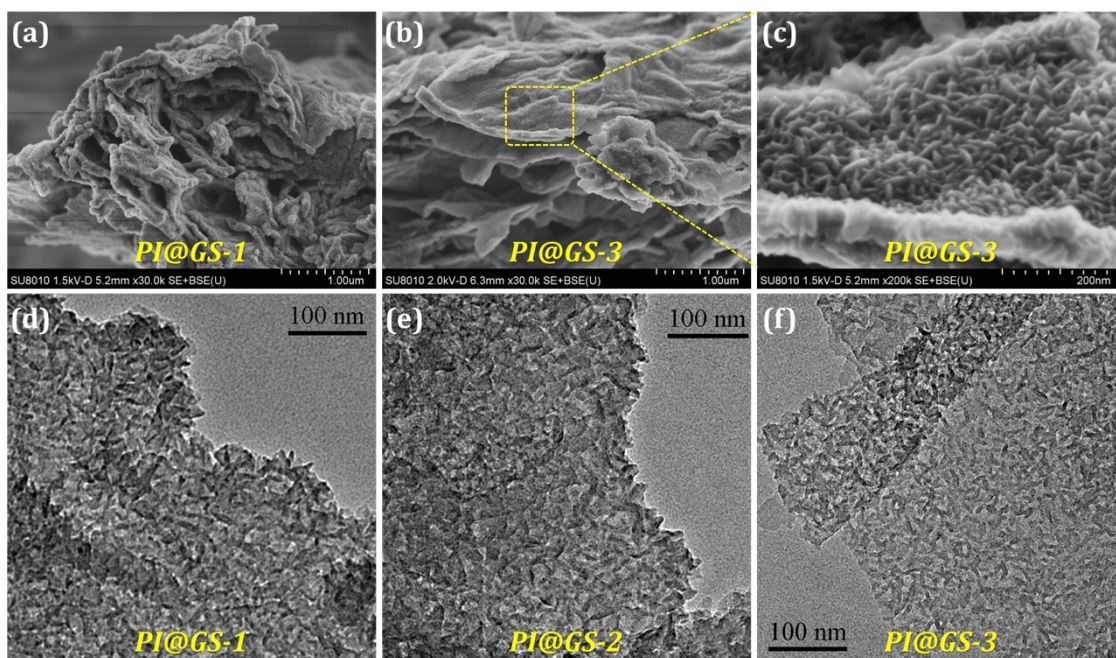


Fig. S3. SEM (a-c) and TEM (d-f) images of PI@GS composites.

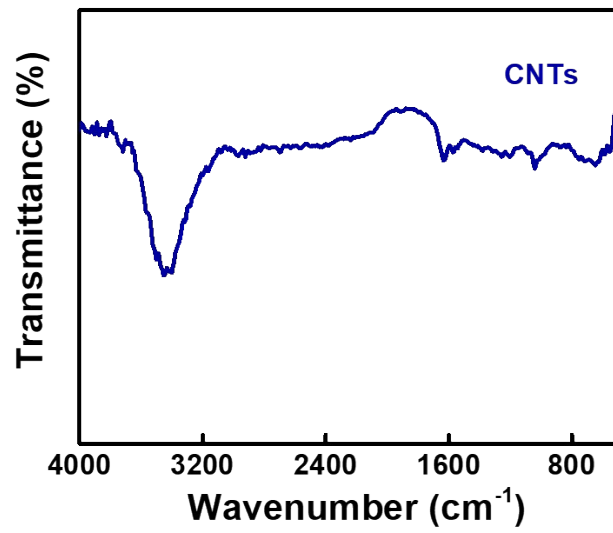


Fig. S4. FT-IR spectrum of acid-treated CNTs.

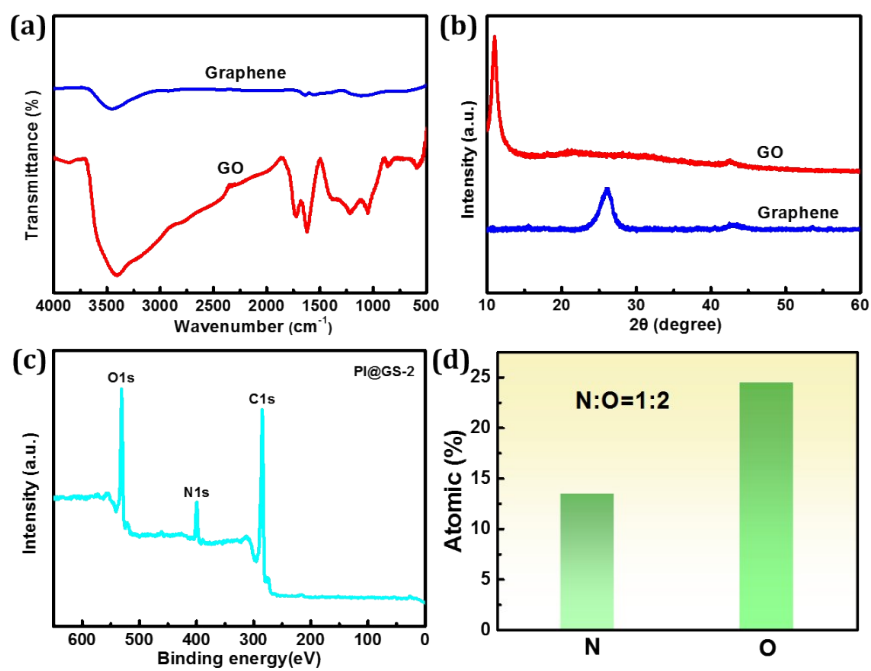


Fig. S5. (a) FT-IR spectra and (b) XRD patterns of GO and graphene, (c) typical XPS survey spectrum and (d) EDS of PI@GS-2.

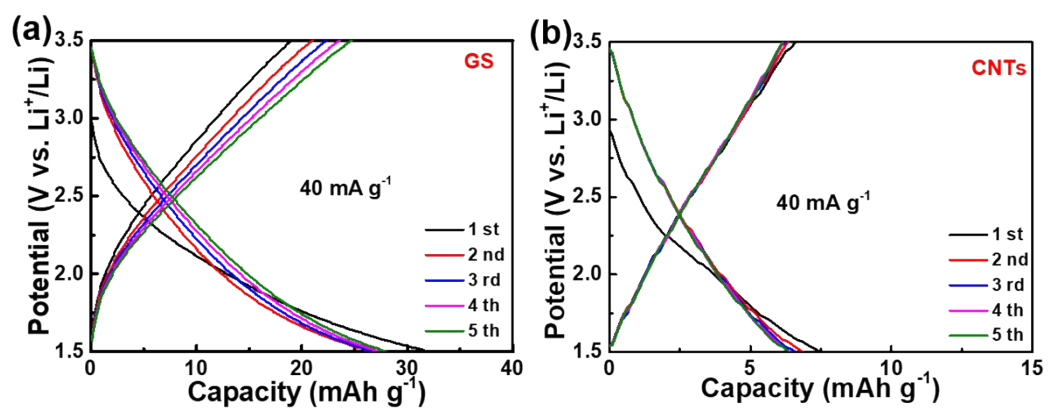


Fig. S6. The capacity contribution of GS and CNTs for LIBs.

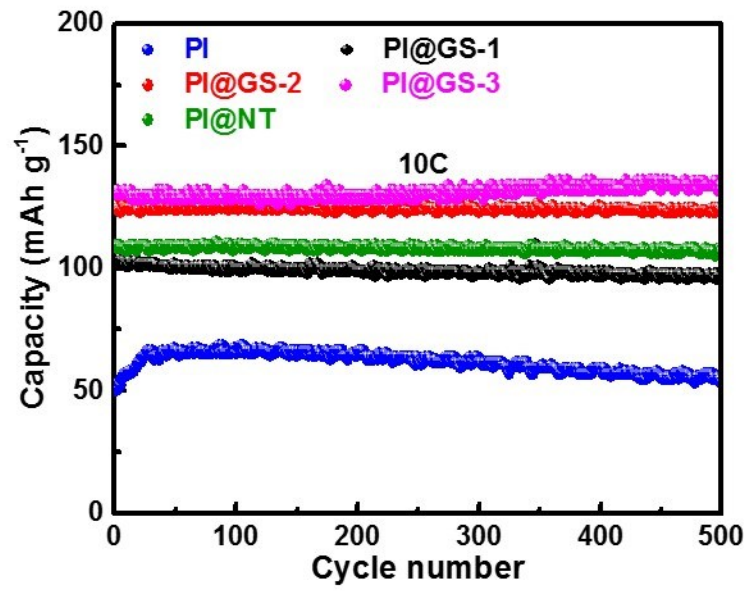


Fig. S7. Cycling performances of coin-type Li/LiTFSI/PI, Li/LiTFSI/PI@GS, and Li/LiTFSI/PI@NT cells at 10C over 500 cycles.

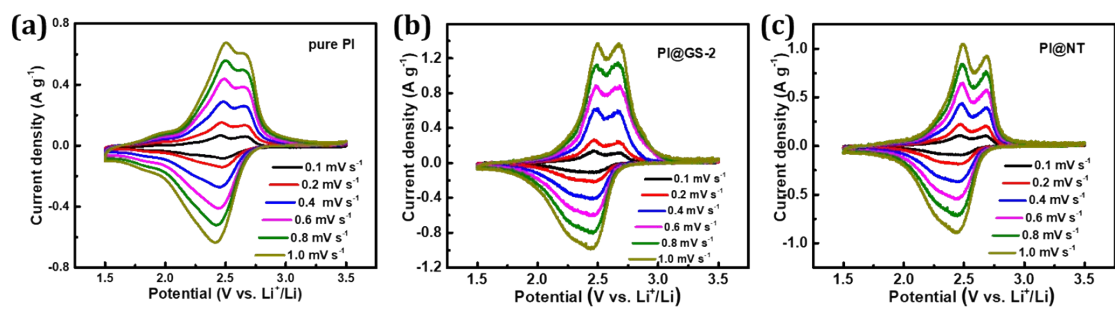


Fig. S8. CV curves of (a) pure PI, (b) PI@GS-2, and (c) PI@NT cathodes at sweep rates of 0.1 to 1.0 mV s⁻¹.

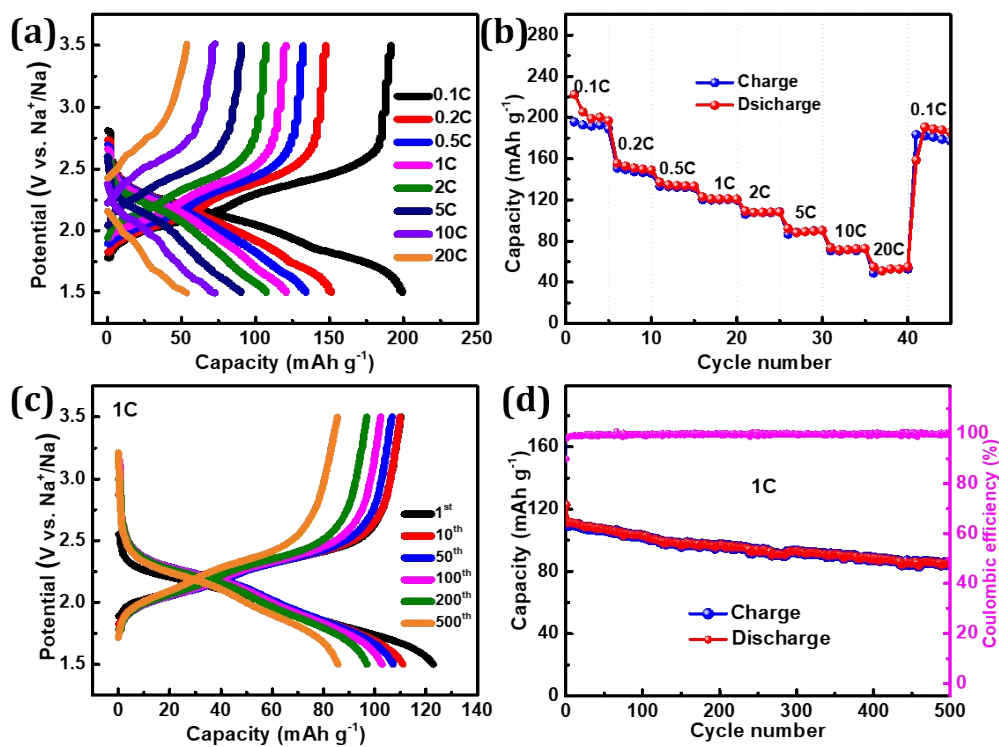
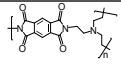
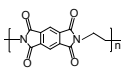
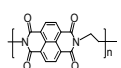
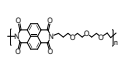
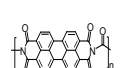
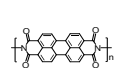
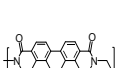
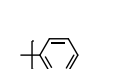
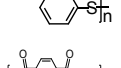


Fig. S9. Na-ion storage performance of Na/NaPF₆/PI@GS-2 cells: (a) charge/discharge curves and (b) rate performance measured in the rate range of 0.1C to 20 C, (c) charge/discharge curves at different cycles and (d) cycling stability measured at 1C for 500 cycles.

Table S1. The comparison of reported carbonyl polymers for LIB cathodes.

Materials	Theoretical capacity (mAh g ⁻¹)	Practical (Initial) capacity (mAh g ⁻¹), rate	Ratecapacity (mAh g ⁻¹), rate	Capacity (mAh g ⁻¹), cycle numbers, rate	Ref.
	343 (4e)	160 38.3 mA g ⁻¹	74 3.83 A g ⁻¹	147 200 191.5 mA g ⁻¹	1
	443 (4e)	175 0.1C	101 2C	101 150 0.5C	2
	367 (4e)	127.3 20 mA g ⁻¹	108 500 mA g ⁻¹	121 60 50 mA g ⁻¹	3
	245 (4e)	125 25 mA g ⁻¹	86 250 mA g ⁻¹	77.6 100 250 mA g ⁻¹	4
	322 (5e)	80 50 mA g ⁻¹	95 200 mA g ⁻¹	130 50 50 mA g ⁻¹	5
	276 (4e)	130 50 mA g ⁻¹	98 200 mA g ⁻¹	110 50 50 mA g ⁻¹	5
	257 (4e)	85 50 mA g ⁻¹	0 200 mA g ⁻¹	75 50 50 mA g ⁻¹	5
	225 (2e)	156 0.1C	102 20C	132 1000 0.5C	6
 Graphene	367 (4e)	165 0.1C	125 20C	112 (88%) 5000 5C	This work

References

1. H. Wu, Q. Meng, Q. Yang, M. Zhang, K. Lu and Z. Wei, *Adv. Mater.*, 2015, **27**, 6504-6510.
2. Y. Meng, H. Wu, Y. Zhang and Z. Wei, *J. Mater. Chem. A*, 2014, **2**, 10842-10846.
3. C. Chen, X. Zhao, H.-B. Li, F. Gan, J. Zhang, J. Dong and Q. Zhang, *Electrochim. Acta*, 2017, **229**, 387-395.
4. G. Hernández, M. Salsamendi, S. M. Morozova, E. I. Lozinskaya, S. Devaraj, Y. S. Vygodskii, A. S. Shaplov and D. Mecerreyes, *J. Polym. Sci. Part A: Polym. Chem.*, 2018, **56**, 714-723.
5. P. Sharma, D. Damien, K. Nagarajan, M. M. Shaijumon and M. Hariharan, *J. Phys. Chem. Lett.*, 2013, **4**, 3192-3197.
6. Y. Zhang, Y. Huang, G. Yang, F. Bu, K. Li, I. Shakir and Y. Xu, *ACS Appl. Mater. Interfaces*, 2017, **9**, 15549-15556.