

Supporting Information

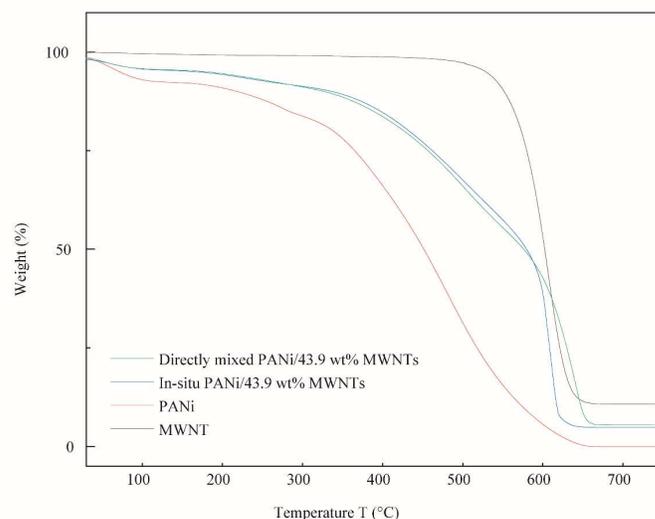
**A Gill-Mimicking Thermoelectric Generator (TEG) for Waste Heat Recovery and Self-Powering Wearable Devices**

Yuxiang Zhu,<sup>a</sup> Weiheng Xu,<sup>a</sup> Dharnedar Ravichandran,<sup>a</sup> Sayli Jambhulkar<sup>a</sup> and Kenan Song<sup>\*b</sup>

<sup>a</sup> Systems Engineering, The Polytechnic School (TPS), Ira A. Fulton Schools of Engineering, Arizona State University, Mesa, AZ, United States 85212

<sup>b</sup> The Polytechnic School (TPS) & School for Engineering of Matter, Transport and Energy (SEMTE), Ira A. Fulton Schools of Engineering, Arizona State University, Mesa, AZ, United States 85212

\*Correspondence author: [Kenan.Song@asu.edu](mailto:Kenan.Song@asu.edu)



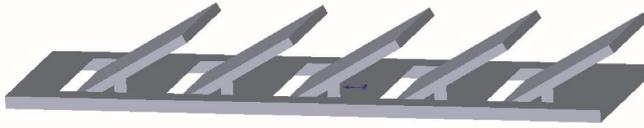
**Figure S1.** TGA comparison of in-situ polymerized and directly mixed PANi/43.9 wt% MWNTs confirmed the consistency of MWNT concentrations

The conjugated structure endows polyaniline with high thermal stability. The pure PANi experienced a three-step weight loss in air.

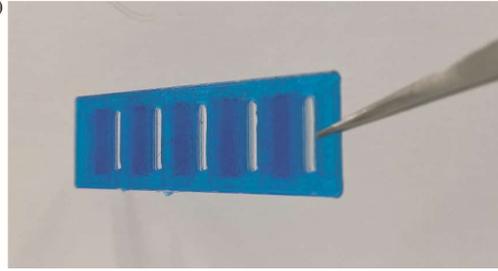
- (i) The first small step started right above room temperature and stopped at around 110 °C. This weight drop was due to the moisture captured in the powder.
- (ii) The second gentle slope from 200 °C to 340 °C corresponds to the loss of counter anion, in our case, the chloride ion.<sup>1</sup>
- (iii) At 340 °C, the slope became steep, and the significant weight loss happened within this range up to 640 °C. The enormous weight loss corresponds to the decomposition of the PANi backbone, leaving no mass residue.<sup>2</sup>

The in-situ polymerized and directly-mixed PANi/MWNTs with 43.9 wt% of MWNTs were also tested under the same condition. The three-step weight loss was similar to pure PANi below 590 °C. The MWNT addition improved the thermal stability of PANi, showing higher residue during the thermal sweeping.

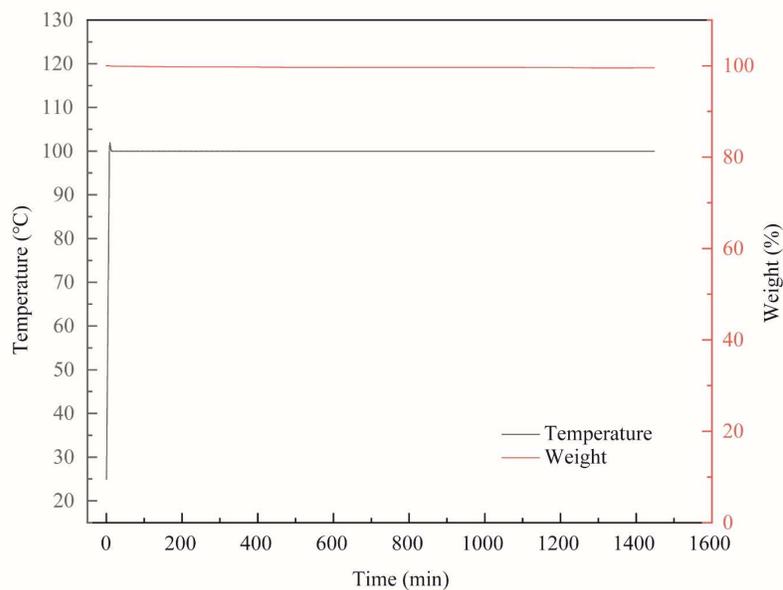
(a)



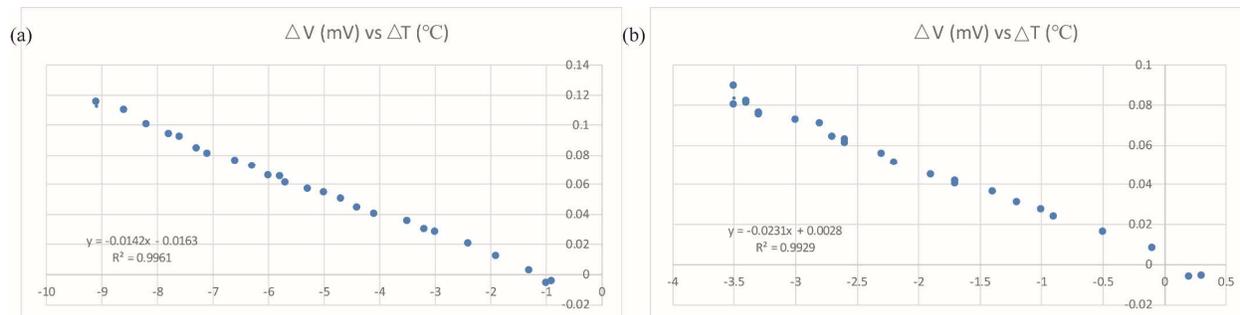
(b)



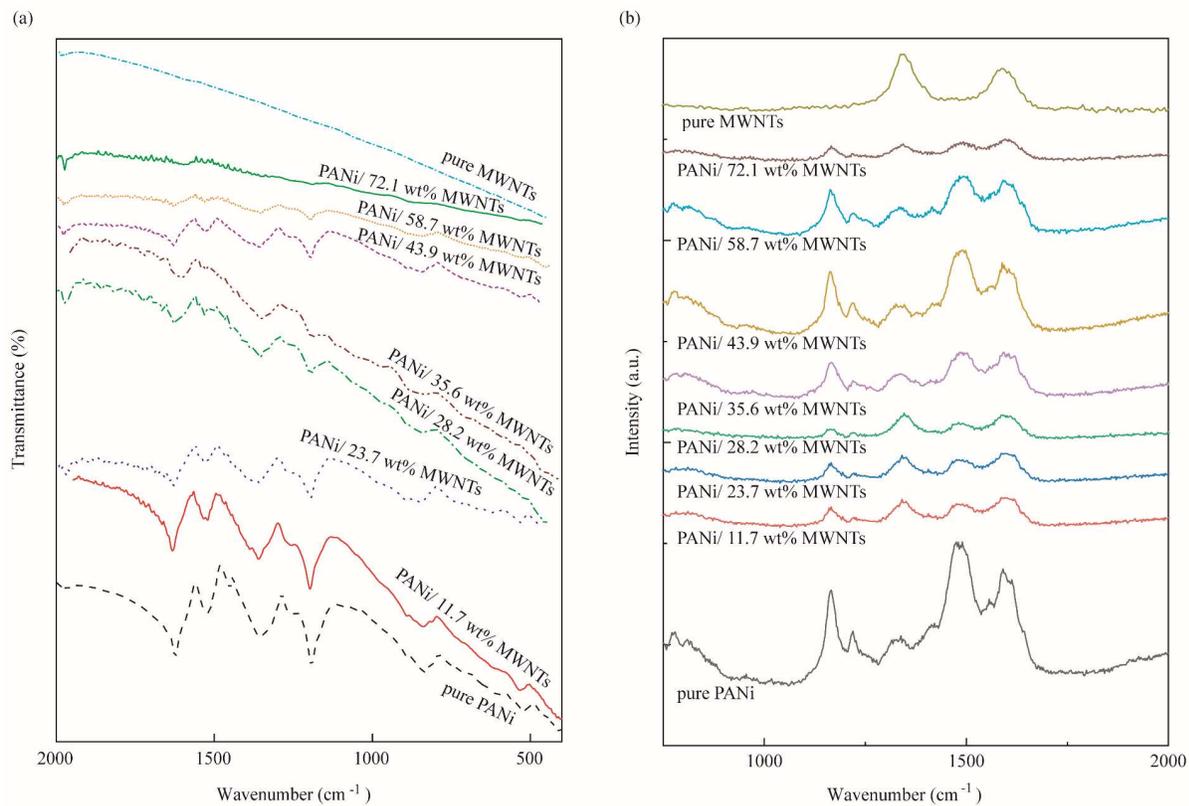
**Figure S2.** (a) The STL file of the TEG substrate for printing. (b) Photo of the printed substrate bottom. The openings on the bottom of the substrate were to assemble the thermoelectric bulks



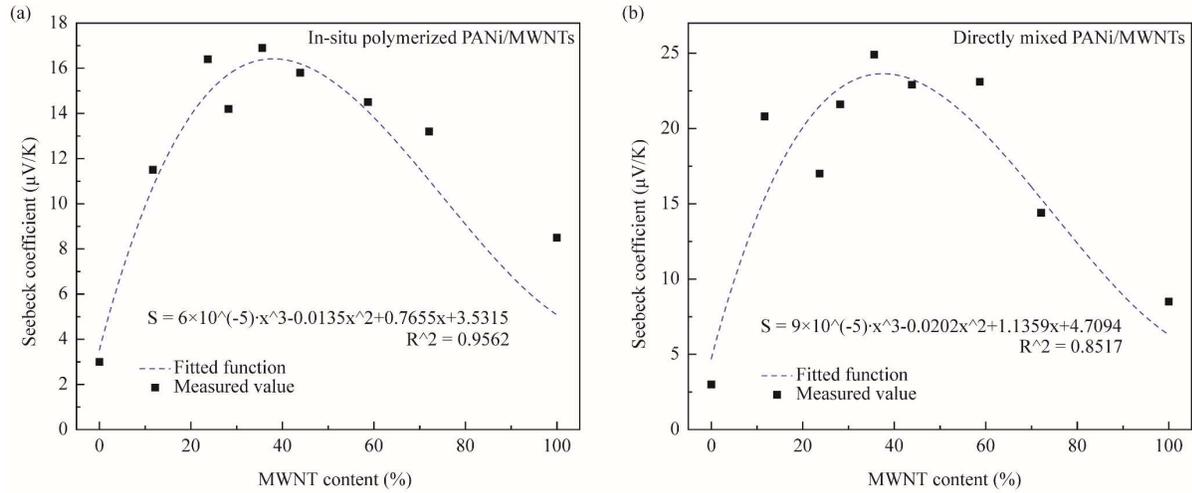
**Figure S3.** The thermogravimetric analysis (TGA) of the Kapton tape for aging characterizations. The tape's mass loss was negligible at a temperature of 100 °C for 24 h, indicating the thermal stability of the binding material at a working temperature  $\leq 100$  °C in the vicinity of the surfaces of skins, hot water bottles, and car roofs.



**Figure S4.** Examples of the linear fitting of the voltage difference ( $\Delta V$ ) vs. temperature difference ( $\Delta T$ ) to obtain Seebeck coefficient values. (a) In-situ polymerized PANi/28.2 wt% MWNTs, and (b) directly mixed PANi/58.7 wt% MWNTs



**Figure S5.** (a) FTIR and (b) Raman spectra for directly mixed pure-PANi, PANi/11.7 wt% MWNTs, PANi/23.7 wt% MWNTs, PANi/28.2 wt% MWNTs, PANi/35.6 wt% MWNTs, PANi/43.9 wt% MWNTs, PANi/58.7 wt% MWNTs, PANi/72.1 wt% MWNTs and pure MWNTs.

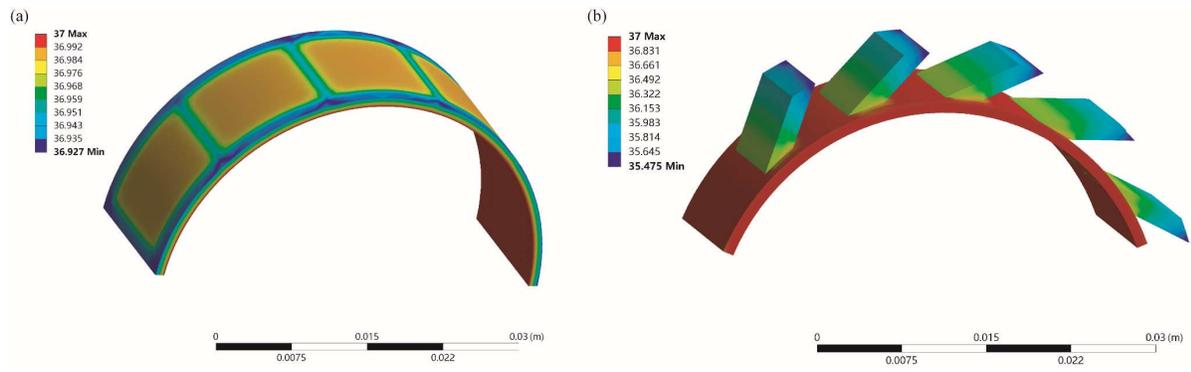


**Figure S6.** The Seebeck coefficients with different MWNT contents and the polynomial curve fitting of (a) in-situ polymerized PANi/MWNTs and (b) directly mixed PANi/MWNTs

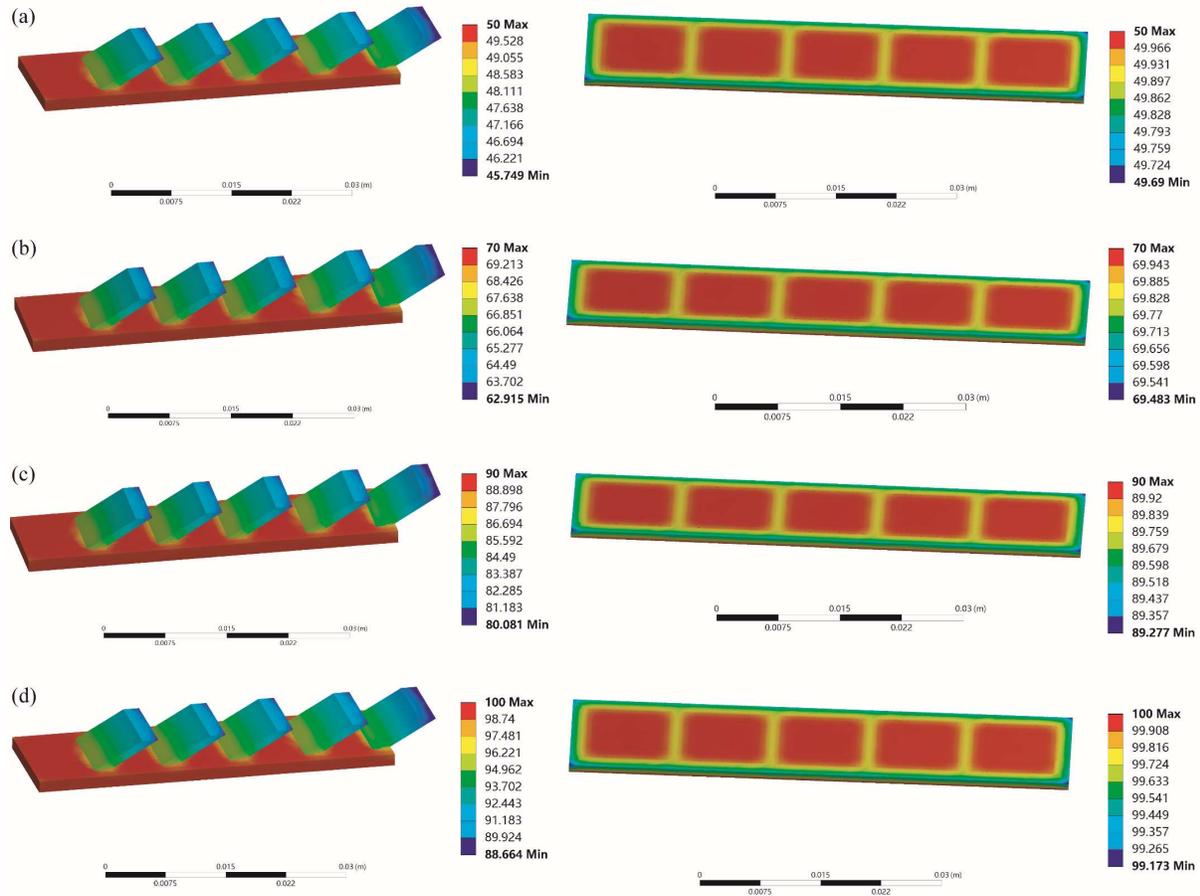
The Seebeck coefficients from the in-situ polymerized and directly-mixed samples were fitted with a cubic function ( $S = 6 \times 10^{-5} \cdot x^3 - 0.0135x^2 + 0.7655x + 3.5315$  for in-situ PANi/MWNTs and  $S = 9 \times 10^{-5} \cdot x^3 - 0.0202x^2 + 1.1359x + 4.7094$  for directly mixed PANi/MWNTs, respectively). The Seebeck coefficient's rapid increase at low MWNT content was due to the MWNT phase and their comparatively uniform dispersions. The interphase between the PANi/MWNT phases induced the energy filtering effect, enhancing the Seebeck coefficient.<sup>3,4</sup> With higher MWNTs content, the carrier concentration rises. According to  $S = \frac{8\pi^2\kappa_B^2}{3eh^2} m^* T \left(\frac{\pi}{3n}\right)^{2/3}$ ,<sup>5,6</sup> the deterioration of the coefficient dominated when MWNTs content is higher than 40%, in addition to the worse dispersion and widely distributed voids and defects, resulting in the overall drop of the Seebeck coefficient.

**Table S1.** Mechanical properties of cold-pressed pure PANi, pure MWNTs, in-situ polymerized and directly mixed PANi/MWNTs composite bulks

Sample	Tensile strength (MPa)	Young's modulus (GPa)	Tensile Strain (%)
Pure PANi	2.028 ± 0.3005	0.4651 ± 0.1607	0.8302 ± 0.3956
In-situ PANi/11.7 wt% MWNTs	2.122 ± 0.5496	0.9701 ± 0.2028	0.4860 ± 0.2176
In-situ PANi/23.7 wt% MWNTs	3.428 ± 0.7026	0.9896 ± 0.2787	0.5916 ± 0.1486
In-situ PANi/28.2 wt% MWNTs	4.022 ± 0.9793	1.134 ± 0.2087	0.5864 ± 0.1729
In-situ PANi/35.6 wt% MWNTs	4.432 ± 0.9275	1.434 ± 0.3190	0.4899 ± 0.0688
In-situ PANi/43.9 wt% MWNTs	6.540 ± 0.5350	1.127 ± 0.2487	0.8391 ± 0.1933
In-situ PANi/58.7 wt% MWNTs	11.470 ± 1.6770	2.017 ± 0.1843	1.0898 ± 0.2205
In-situ PANi/72.1 wt% MWNTs	5.586 ± 0.6222	0.9961 ± 0.2772	0.8055 ± 0.1675
Directly mixed PANi/11.7 wt% MWNTs	3.075 ± 0.2287	0.8399 ± 0.1697	0.5630 ± 0.1318
Directly mixed PANi/23.7 wt% MWNTs	2.774 ± 1.0420	0.8287 ± 0.1287	0.5177 ± 0.1654
Directly mixed PANi/28.2 wt% MWNTs	1.216 ± 0.2998	0.6003 ± 0.1667	0.4231 ± 0.1125
Directly mixed PANi/35.6 wt% MWNTs	1.446 ± 0.5200	0.4703 ± 0.2978	0.6260 ± 0.3319
Directly mixed PANi/43.9 wt% MWNTs	1.553 ± 0.3282	0.3606 ± 0.0587	0.7492 ± 0.0814
Directly mixed PANi/58.7 wt% MWNTs	1.423 ± 0.5633	0.4118 ± 0.0914	0.7040 ± 0.3519
Directly mixed PANi/72.1 wt% MWNTs	0.6568 ± 0.3049	0.1661 ± 0.0769	0.8661 ± 0.4651
Pure MWNTs	0.8910 ± 0.1853	0.1405 ± 0.0349	1.274 ± 0.3806



**Figure S7.** The simulated temperature distribution of (a) the in-plane design TEG and (b) the out-of-plane, gill-mimicking TEG at a curving state with a bending displacement of 2 cm. The maximum temperature difference of the flat design TEG is below 0.1 °C, while the gill-mimicking TEG can obtain a temperature difference of ~1.5 °C, showing the same consistent trend of the thermal mapping between the curving and flat substrates.



**Figure S8.** The Finite Element Method (FEM) simulation of a gill-inspired design of fTEG (left) and a flat design of TEG (right) with the substrate (hot side) temperature of (a) 50 °C, (b) 70 °C, (c) 90 °C, and 100 °C.

**Table S2.** TE property of polymer/carbon-based composites in literature as compared with our work

Material	Processing method	Power factor ( $\mu\text{W m}^{-1} \text{K}^{-2}$ )	Generator assembled	Organic solvent	Manufacturing time (h)	Flexibility for wearability	Out-of-plane design	Year
PANi/MWNTs	<i>in-situ</i> polymerization, cold pressing	1.254	Yes	No	~20	Yes	Yes	This work
PANi/PbTe	<i>in-situ</i> polymerization, cold pressing	0.8	No	Yes	>24			2011 <sup>7</sup>
PANi/Graphite	Mechanical mixing, cold pressing	4	No	No	~30			2011 <sup>8</sup>
PANi/CNT	<i>in-situ</i> polymerization, electro-spinning	0.17	No	Yes	~47			2012 <sup>9</sup>
PANi/GNPs	Cold pressing	14	No	No	~60			2013 <sup>10</sup>
PANi/GNPs	Solution dispersion, drop-casting	19	No	Yes	~60			2014 <sup>11</sup>
PEDOT:PSS/Ionic liquid/PU	Drop casting	~4.8	Yes	No	~21	Yes	No	2020 <sup>12</sup>
PEDOT:PSS & n-type CNT	Gelation process	4.77	Yes	Yes	>48	Yes	No	2018 <sup>13</sup>
Polypyrrole/Silver	One pot photo-polymerization	0.03	Yes	Yes	>1.5	Yes	No	2017 <sup>14</sup>
PEDOT:PSS/BiSbTe	Drop casting	8.3	Yes	Yes	>24	Yes	No	2019 <sup>15</sup>
PbTe-modified PEDOT	Interfacial polymerization	1.45	No	Yes	>24			2011 <sup>16</sup>
PEDOT/SWCNT	Direct mixing	25	No	Yes	>144			2010 <sup>17</sup>

Polythiophenes	Solution casting	~10	No	Yes	~12			2015 <sup>18</sup>
FBDPPV	Chemical synthesis	~7	No	Yes	Varies with different dopants			2019 <sup>19</sup>
Bacteria nanocellulose/CNT	Bacteria culture	~30	Yes	Yes	>295, including inoculum culture	Yes	No	2019 <sup>20</sup>
Carbonized polydopamine	Coating, heat treatment	0.2	No	No	~60			2017 <sup>21</sup>
Polypyrrole/rGO	Soft template polymerization	8.57	No	No	~36			2015 <sup>22</sup>
PVA/Bi <sub>2</sub> Te <sub>3</sub>	Solid-state reaction, screen printing	0.04	Yes	No	>38, excluding solid-state reaction	Yes	No	2019 <sup>23</sup>
P3HT-TFSI	Drop-casting	~24	No	Yes	>3, depends on doping time			2012 <sup>24</sup>
P3HT-F <sub>4</sub> TCNQ	Vapor doping, spin casting	27	No	Yes	~1, varies with dopant amount			2018 <sup>25</sup>
Polypyrrole	Chemical oxidative polymerization	~24	No	Yes	>32, depends on oxidation time			2017 <sup>26</sup>
Polypyrrole/PANi/SWCNT	<i>in-situ</i> polymerization	19.2	No	Yes	~48			2020 <sup>27</sup>

With easy availability and facile processing, current organic TE materials find it hard to compete with traditional inorganic alloys or their composites in the figure of merit or power factor (**Table S2**). However, the high power factor has usually been mostly achieved by sacrificing device density, flexibility, or assembly simplicity. For example, Yuan et al. were able to obtain a power of 190  $\mu\text{W}$  with the body temperature by using p-type and n-type semiconductors attached to metals/conductive polymers.<sup>28</sup> With the extremely high number of 576 pieces of TE grains, which is over 100 times more than ours, they still need a boost converter to raise the voltage to 2-3 V by sacrificing some power to be compatible with an LCD screen and other sensors. Put simply, our current TEG can work with a strain switch to reveal joint movements and respiration cycles. To further include multiple sensors, such as gyro sensors and humidity sensors, or improve user-friendliness by introducing an electronic display, we can increase the output power and voltage by simply increasing the number of gills in the TEG device. Our gill-mimicking structure can also enhance the thermoelectric performance via the following protocols. (i) Exploring the n-type TE bulks to match the current p-type PANi/MWNTs. Most conducting polymers are p-type, with positive charge holes as the primary carriers. A complete TE pair will improve energy efficiency. (ii) Using inorganic semiconductors with our MWNTs as hybrid materials but sacrificing a larger density and more expensive raw materials.

## References

- 1 N. Chandrakanthi and M. A. Careem, *Polym. Bull.*, 2000, **44**, 101–108.
- 2 L. Ding, X. Wang and R. V. Gregory, *Synth. Met.*, 1999, **104**, 73–78.
- 3 W. Park, H. Hwang, S. Kim, S. Park and K. Jang, *ACS Appl. Mater. Interfaces*, 2021, **13**, 7208–7215.
- 4 Y. Wang, C. Yu, M. Sheng, S. Song and Y. Deng, *Adv. Mater. Interfaces*, 2018, **5**, 7–9.
- 5 Y. Du, J. Xu, B. Paul and P. Eklund, *Appl. Mater. Today*, 2018, **12**, 366–388.
- 6 Y. Wang, L. Yang, X. Shi, X. Shi, L. Chen, M. S. Dargusch, J. Zou and Z. Chen, *Adv. Mater.*, 2019, **31**, 1807916.
- 7 Y. Y. Wang, K. F. Cai, J. L. Yin, B. J. An, Y. Du and X. Yao, *J. Nanoparticle Res.*, 2011, **13**, 533–539.
- 8 L. Wang, D. Wang, G. Zhu, J. Li and F. Pan, *Mater. Lett.*, 2011, **65**, 1086–1088.
- 9 Q. Wang, Q. Yao, J. Chang and L. Chen, *J. Mater. Chem.*, 2012, **22**, 17612–17618.
- 10 B. Abad, I. Alda, P. Díaz-Chao, H. Kawakami, A. Almarza, D. Amantia, D. Gutierrez, L. Aubouy and M. Martín-González, *J. Mater. Chem. A*, 2013, **1**, 10450–10457.
- 11 L. Wang, Q. Yao, H. Bi, F. Huang, Q. Wang and L. Chen, *J. Mater. Chem. A*, 2014, **2**, 11107–11113.
- 12 N. Kim, S. Lienemann, I. Petsagkourakis, D. Alemu Mengistie, S. Kee, T. Ederth, V. Gueskine, P. Leclère, R. Lazzaroni, X. Crispin and K. Tybrandt, *Nat. Commun.*, 2020, **11**, 1–10.
- 13 J. Liu, Y. Jia, Q. Jiang, F. Jiang, C. Li, X. Wang, P. Liu, P. Liu, F. Hu, Y. Du and J. Xu, *ACS Appl. Mater. Interfaces*, 2018, **10**, 44033–44040.
- 14 M. Bharti, A. Singh, S. Samanta, A. K. Debnath, D. K. Aswal, K. P. Muthe and S. C. Gadkari, *Energy Convers. Manag.*, 2017, **144**, 143–152.
- 15 M. Bharti, A. Singh, G. Saini, S. Saha, A. Bohra, Y. Kaneko, A. K. Debnath, K. P. Muthe, K. Marumoto, D. K. Aswal and S. C. Gadkari, *J. Power Sources*, 2019, **435**, 226758.
- 16 Y. Wang, K. Cai and X. Yao, *ACS Appl. Mater. Interfaces*, 2011, **3**, 1163–1166.
- 17 D. Kim, Y. Kim, K. Choi, J. C. Grunlan and C. Yu, *ACS Nano*, 2010, **4**, 513–523.
- 18 A. M. Glaudell, J. E. Cochran, S. N. Patel and M. L. Chabiny, *Adv. Energy Mater.*, 2015, **5**, 1–8.
- 19 H. Un, S. A. Gregory, S. K. Mohapatra, M. Xiong, E. Longhi, Y. Lu, S. Rigin, S. Jhulki, C. Yang and T. V Timofeeva, *Adv. Energy Mater.*, 2019, **9**, 1900817.
- 20 D. Abol-Fotouh, B. Dörling, O. Zapata-Arteaga, X. Rodríguez-Martínez, A. Gómez, J. S. Reparaz, A. Laromaine, A. Roig and M. Campoy-Quiles, *Energy Environ. Sci.*, 2019, **12**, 716–726.
- 21 H. Li, Y. V. Aulin, L. Frazer, E. Borguet, R. Kakodkar, J. Feser, Y. Chen, K. An, D. A. Dikin and F. Ren, *ACS Appl. Mater. Interfaces*, 2017, **9**, 6655–6660.
- 22 Z. Zhang, G. Chen, H. Wang and W. Zhai, *J. Mater. Chem. C*, 2015, **3**, 1649–1654.
- 23 A. L. Pires, I. F. Cruz, J. Silva, G. N. P. Oliveira, S. Ferreira-Teixeira, A. M. L. Lopes, J. P. Araújo, J. Fonseca, C. Pereira and A. M. Pereira, *ACS Appl. Mater. Interfaces*, 2019, **11**, 8969–8981.
- 24 Q. Zhang, Y. Sun, W. Xu and D. Zhu, *Energy Environ. Sci.*, 2012, **5**, 9639–9644.
- 25 E. Lim, K. A. Peterson, G. M. Su and M. L. Chabiny, *Chem. Mater.*, 2018, **30**, 998–1010.
- 26 L. Liang, G. Chen and C. Y. Guo, *Mater. Chem. Front.*, 2017, **1**, 380–386.
- 27 S. Wang, Y. Zhou, Y. Liu, L. Wang and C. Gao, *J. Mater. Chem. C*, 2020, **8**, 528–535.
- 28 J. Yuan, R. Zhu and G. Li, *Adv. Mater. Technol.*, 2020, **5**, 1–7.