

## Supporting Information

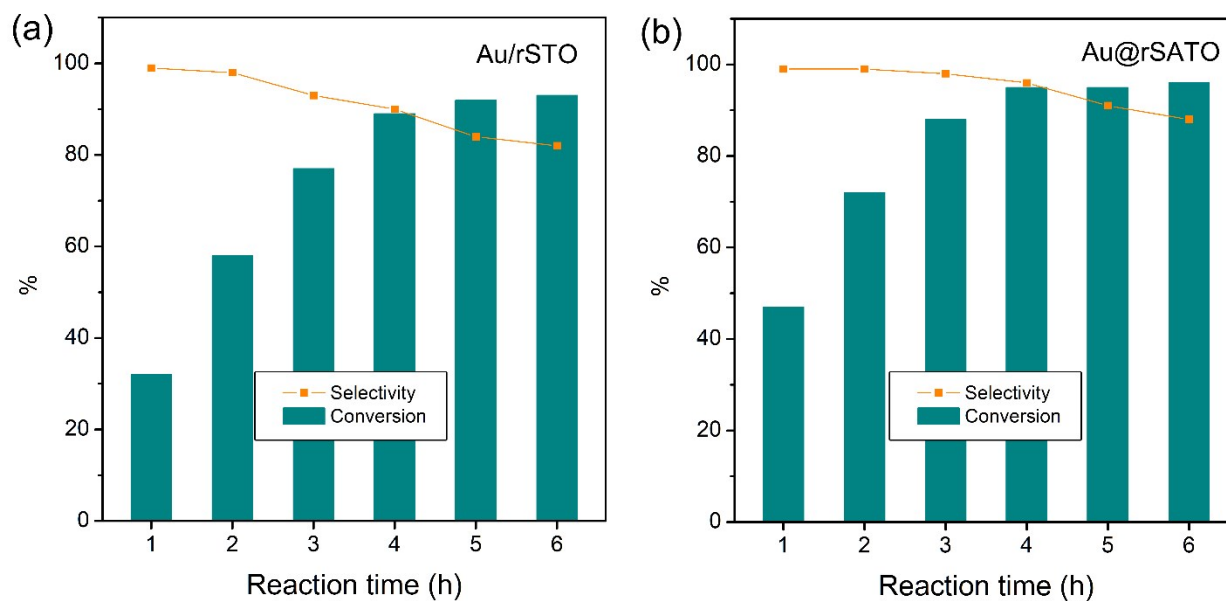
# In Situ Exsolved Au Nanoparticles from Perovskite Oxide for Efficient Epoxidation of Styrene

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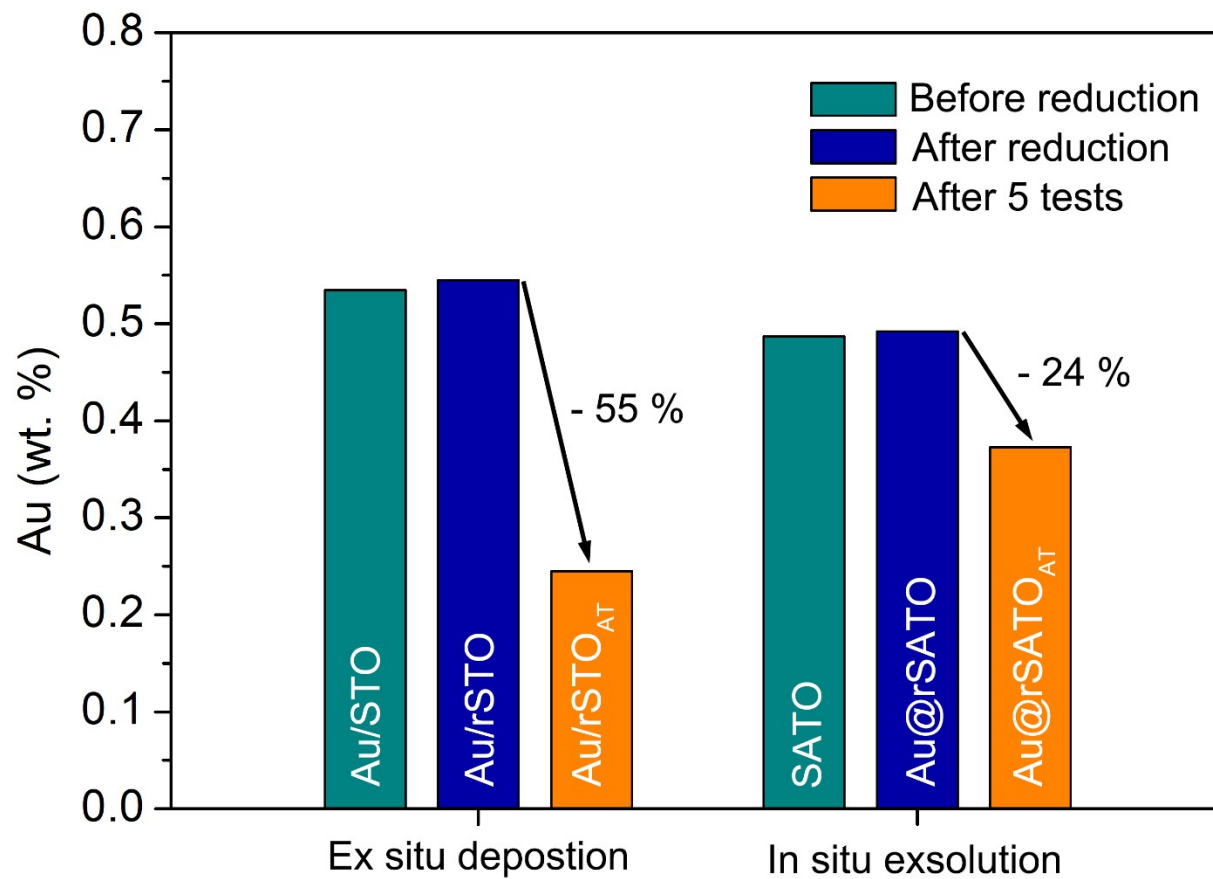
<sup>‡</sup> These authors contributed equally to this work.



**Figure S1.** Conversion and selectivity of the *ex situ* deposited Au/rSTO and *in situ* exsolved Au@rSATO versus reaction time in 6h.

**Table S1.** Catalytic activity of supported AuNPs for styrene epoxidation.

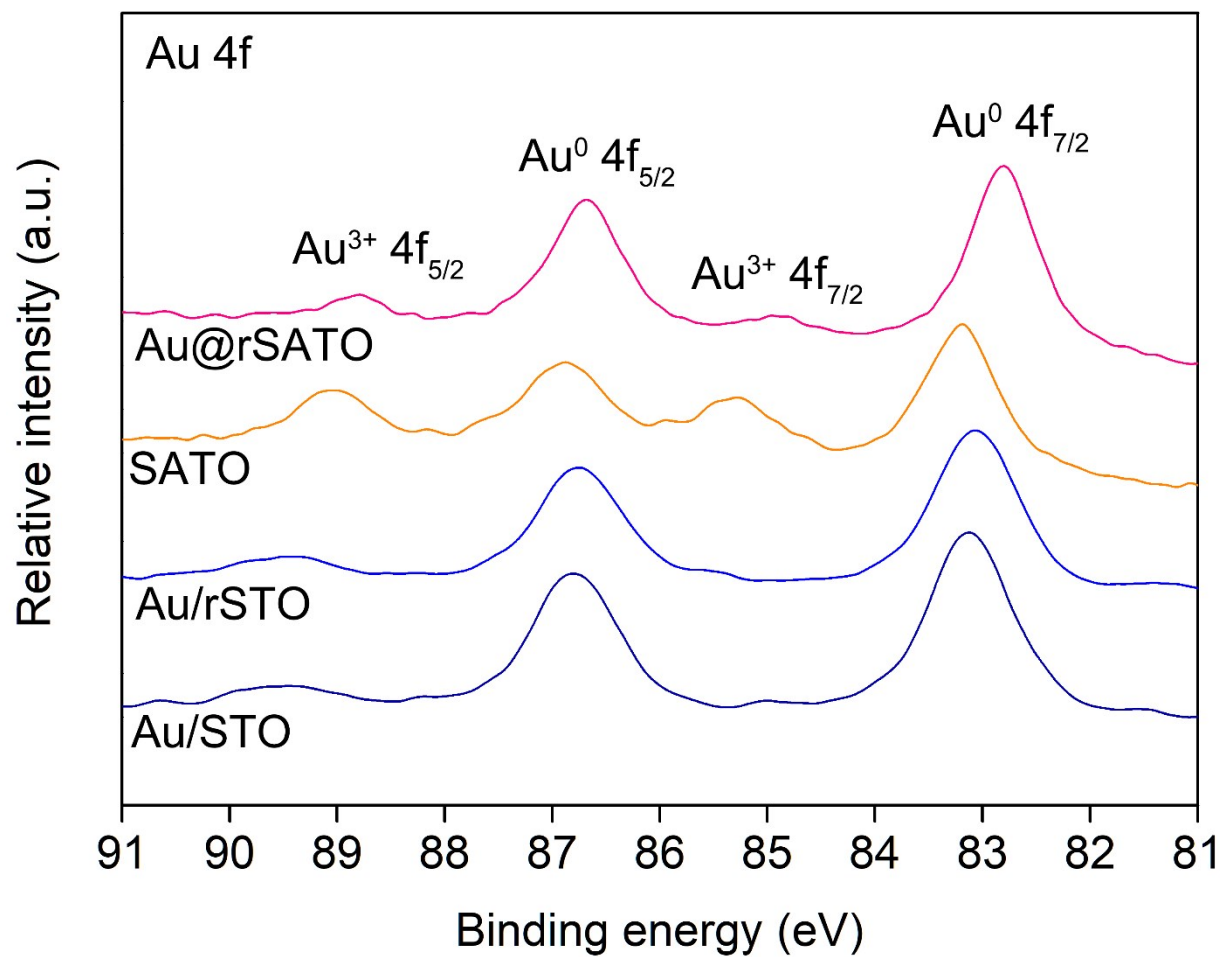
Catalyst	Au wt. %	Oxidant	Conversion %	Selectivity %	Ref.
Au/STO	0.53	H <sub>2</sub> O <sub>2</sub>	65.2	91.7	This work
Au/rSTO	0.53	H <sub>2</sub> O <sub>2</sub>	88.5	90.1	This work
SATO	0.53	H <sub>2</sub> O <sub>2</sub>	21.6	96.4	This work
Au@rSATO	0.53	H <sub>2</sub> O <sub>2</sub>	95.0	96.3	This work
Au/MgO	7.5	TBHP	62.6	54.3	[1]
Au/Al <sub>2</sub> O <sub>3</sub>	6.36	TBHP	44	28	[2]
Au/TiO <sub>2</sub>	6.0	TBHP	61	53	[3]
Au/Al <sub>2</sub> O <sub>3</sub>	2.0	TBHP	84.3	69.0	[4]
Au/LDH	0.66	TBHP	67.4	73.5	[5]
Au/BaTNT	1	TBHP	60.5	80.1	[6]
Au/Si	4	TBHP	98.5	82.8	[7]
Au/TS-1	1	H <sub>2</sub> O <sub>2</sub>	92.7	90.4	[8]
Au-2S-IL	4.5	H <sub>2</sub> O <sub>2</sub>	60.4	92.9	[9]



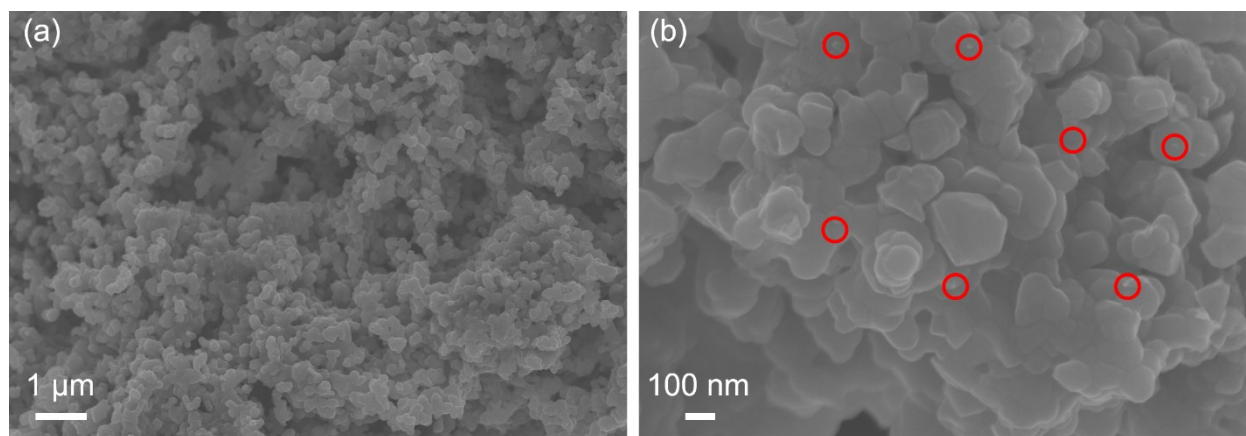
**Figure S2.** Weight ratio of Au in the as-prepared catalysts before and after catalysis obtained with ICP-OES.

**Table S2.** Elemental compositions of the as-prepared catalysts based on XPS analysis. The values show the relative atomic ratios of Sr and Au compared to Ti.

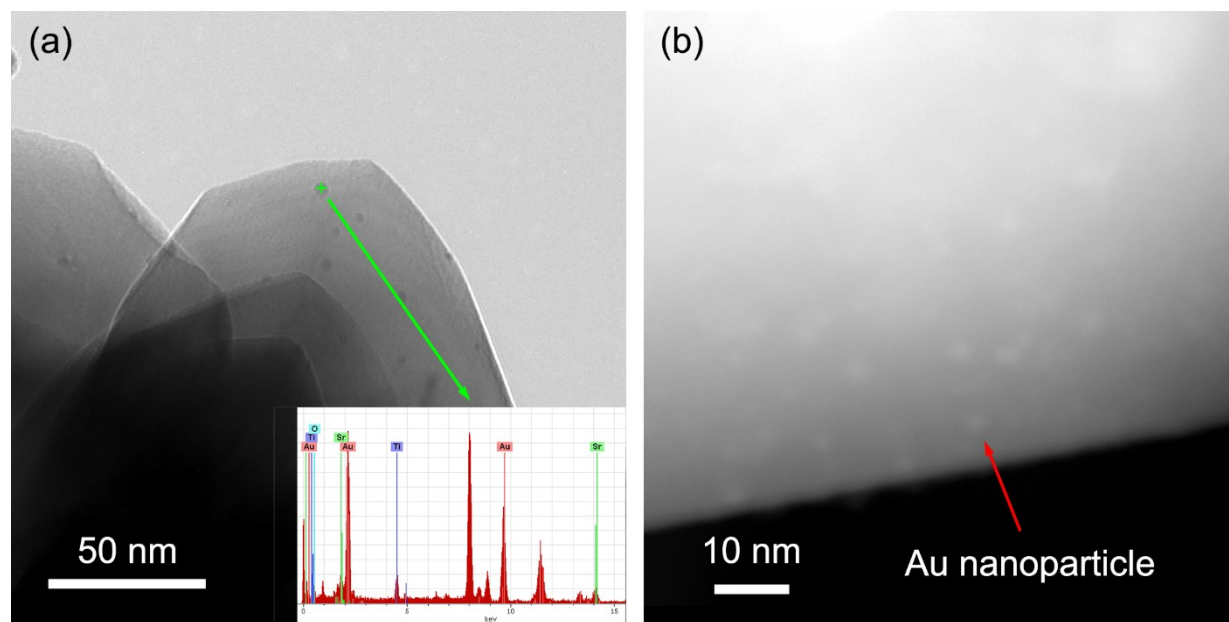
Elemental	Sr/Ti	Au/Ti
Au@rSATO	0.851	0.0055
SATO	0.806	0.0048
Au/rSTO	0.948	0.0050
Au/STO	0.915	0.0049
rSTO	0.953	/
STO	0.915	/



**Figure S3.** High-resolution Au spectra of the Au-containing samples.

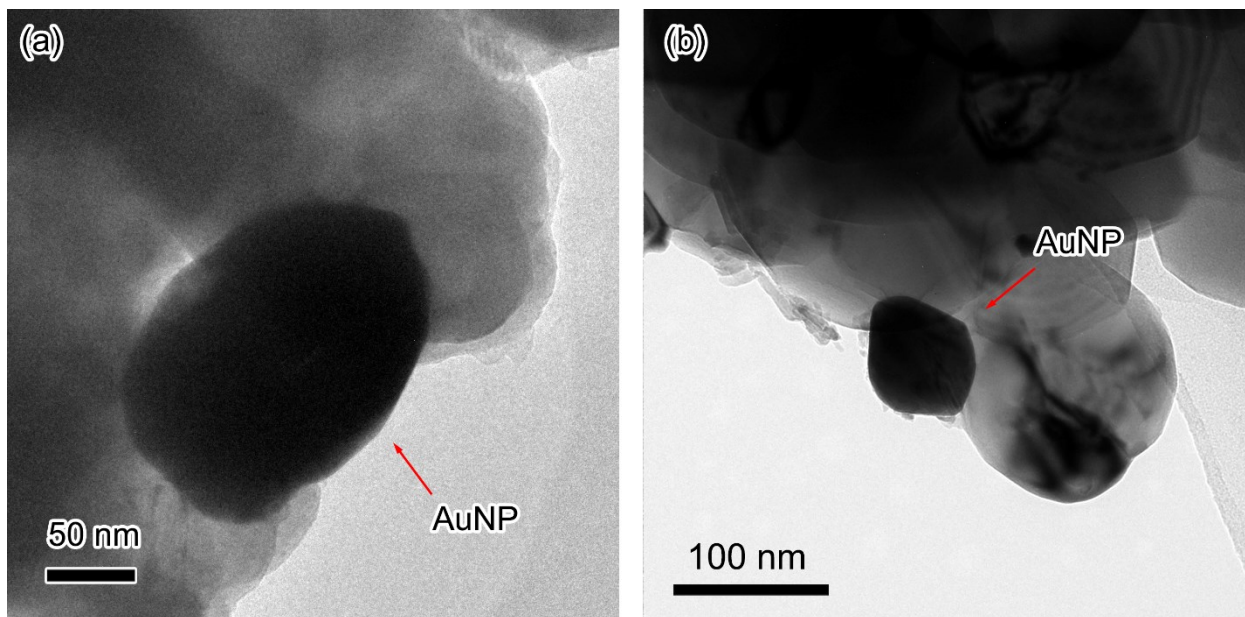


**Figure S4.** SEM images of Au@rSATO with different magnifications. The red circles highlight some of the exsolved nanoparticles.

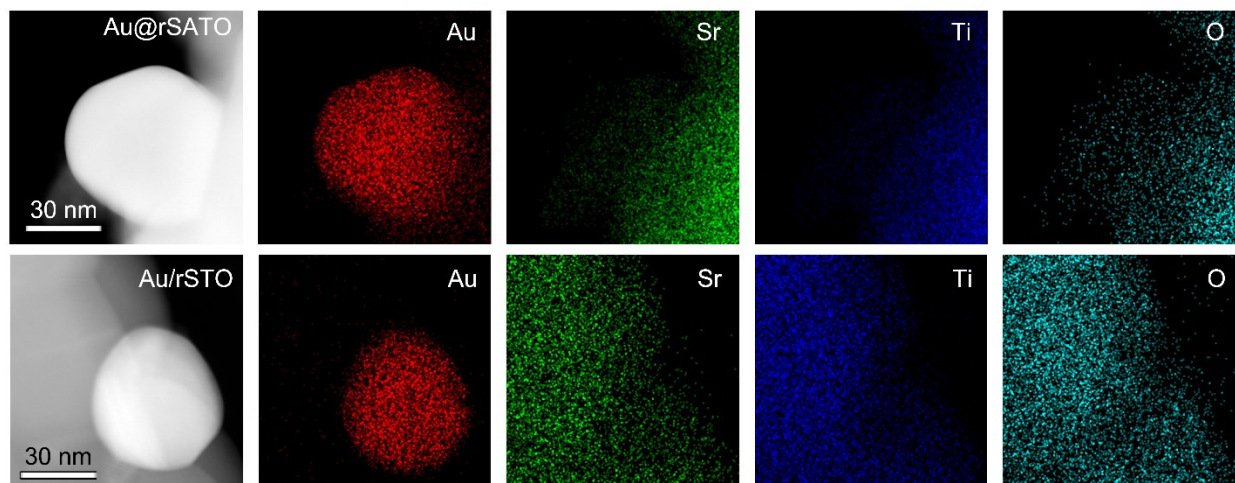


**Figure S5.** (a) EDX analysis of a selected Au nanoparticle for Au@rSATO. (b) HAADF-STEM image of Au@rSATO. Bright dots represent for the AuNPs.

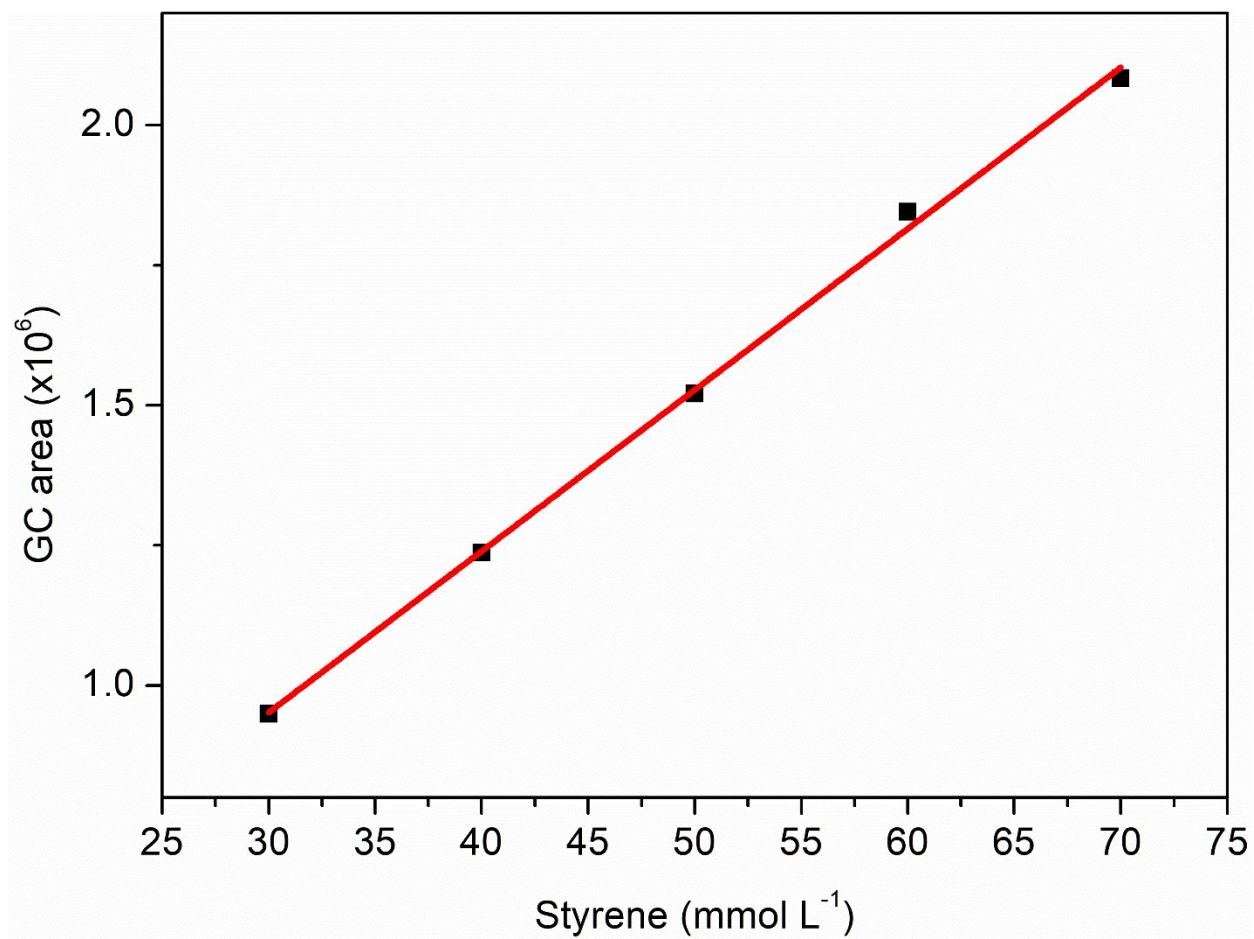




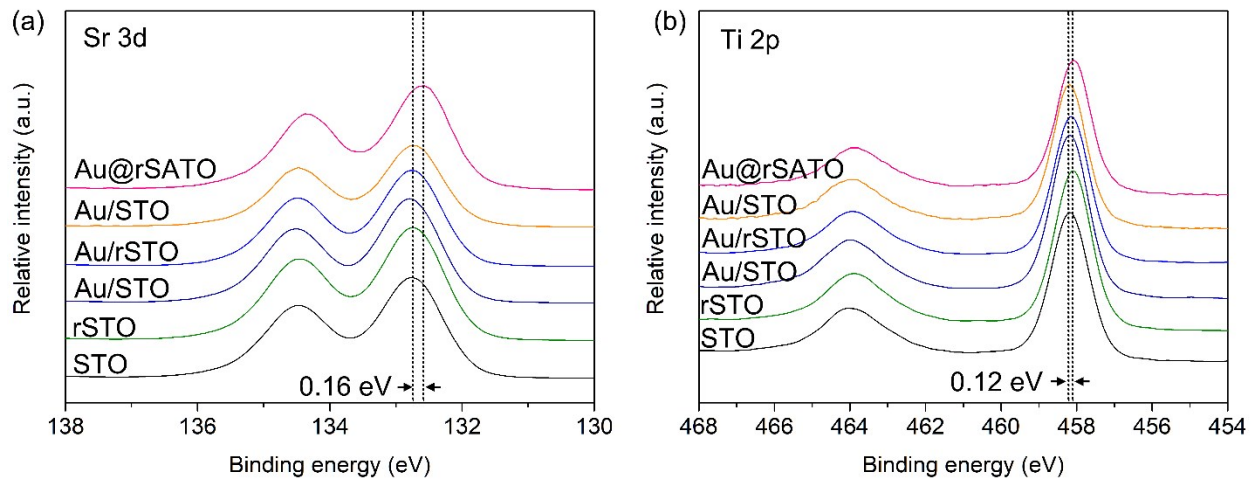
**Figure S6.** TEM images of the AuNPs in Au/rSTO.



**Figure S7.** HAADF image and the corresponding EDX mappings of selected AuNPs for Au@rSATO (top) and Au/rSTO (bottom).

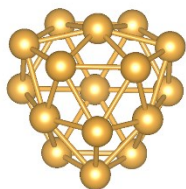


**Figure S8.** Calibration of the GC-area with standard styrene solutions of different concentrations (1  $\mu\text{L}$  styrene solution with a split ratio of 1:50).

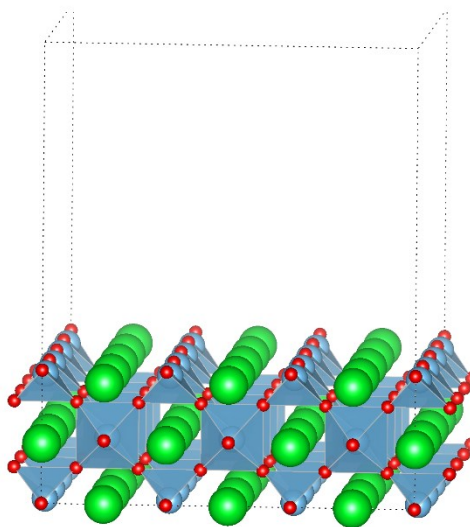


**Figure S9.** High-resolution Sr 3d and Ti 2p spectra of the as-prepared catalysts.

(a)



(b)



**Figure S10.** Relaxed slab models for (a)  $\text{Au}_{16}$  cluster, and (b)  $\text{SrTiO}_3$  (110) surface with the bottom two layers are fixed.

**Table S3.1.** Coordinates and Bader charge of the Au atoms for Figure 6(f) in the main text.

<b>Atom number</b>	<b>X</b>	<b>Y</b>	<b>Z</b>	<b>Charge (e<sup>-</sup>)</b>
<b>1</b>	13.707	6.905	7.912	0.498443
<b>2</b>	11.901	5.399	9.37	0.130842
<b>3</b>	7.672	5.399	9.352	0.129377
<b>4</b>	5.962	3.833	7.761	0.694315
<b>5</b>	7.704	8.282	9.51	0.070997
<b>6</b>	5.873	6.924	7.903	0.498174
<b>7</b>	13.607	3.831	7.761	0.70034
<b>8</b>	9.758	9.928	7.842	0.60733
<b>9</b>	9.783	6.636	10.566	0.028279
<b>10</b>	13.599	9.744	7.82	0.593205
<b>11</b>	9.779	6.997	7.76	0.484984
<b>12</b>	11.855	8.276	9.484	0.066647
<b>13</b>	9.788	9.575	10.645	0.182214
<b>14</b>	5.99	9.761	7.828	0.600224
<b>15</b>	9.793	4.148	7.955	0.439763
<b>16</b>	9.791	1.412	7.767	0.638329

**Table S3.2.** Coordinates and Bader charge of the Au atoms for Figure 6(g) in the main text.

<b>Atom number</b>	<b>X</b>	<b>Y</b>	<b>Z</b>	<b>Charge (e<sup>-</sup>)</b>
<b>1</b>	9.523	9.159	7.638	0.035284
<b>2</b>	6.467	9.099	12.194	-0.02098
<b>3</b>	8.757	10.498	12.092	0.019555
<b>4</b>	5.635	7.692	9.953	0.013918
<b>5</b>	7.684	5.828	9.726	-0.09355
<b>6</b>	8.821	5.02	12.176	0.021363
<b>7</b>	7.648	9.661	9.686	-0.09514
<b>8</b>	10.35	5.048	9.873	0.021877
<b>9</b>	7.142	7.757	7.729	0.01242
<b>10</b>	11.182	6.461	12.11	-0.02293
<b>11</b>	9.558	6.395	7.689	0.00954
<b>12</b>	10.323	10.529	9.812	0.02386
<b>13</b>	10.983	7.794	9.623	-0.0949
<b>14</b>	11.182	9.145	12.046	0.048927
<b>15</b>	6.473	6.337	12.216	0.054393
<b>16</b>	8.829	7.765	12.802	-0.08855

**Table S3.3.** Coordinates and Bader charge of the Au atoms for Figure 6(h) in the main text.

<b>Atom number</b>	<b>X</b>	<b>Y</b>	<b>Z</b>	<b>Charge (e<sup>-</sup>)</b>
<b>1</b>	5.513	6.809	12.307	0.037373
<b>2</b>	10.226	9.615	12.35	0.079694
<b>3</b>	7.848	8.225	12.999	-0.032404
<b>4</b>	8.754	6.942	7.892	0.33344
<b>5</b>	10.223	6.931	12.367	0.05145
<b>6</b>	10.111	8.306	9.899	-0.017959
<b>7</b>	9.446	11.038	10.112	0.132677
<b>8</b>	8.722	9.707	7.888	0.27777
<b>9</b>	7.801	10.97	12.335	0.08216
<b>10</b>	5.509	9.571	12.333	0.086698
<b>11</b>	6.81	6.342	9.852	0.055361
<b>12</b>	4.755	8.204	10.04	0.129941
<b>13</b>	6.776	10.175	9.877	0.058941
<b>14</b>	7.861	5.49	12.326	0.079822
<b>15</b>	6.338	8.306	7.872	0.570837
<b>16</b>	9.469	5.557	10.079	0.132963



## References

- [1] N. Patil, B. Uphade, P. Jana, S. Bhargava, V. Choudhary, Epoxidation of styrene by anhydrous t-butyl hydroperoxide over reusable gold supported on MgO and other alkaline earth oxides, *J. Catal.*, 223 (2004) 236-239.
- [2] N. Patil, R. Jha, B. Uphade, S. Bhargava, V. Choudhary, Epoxidation of styrene by anhydrous t-butyl hydroperoxide over gold supported on Al<sub>2</sub>O<sub>3</sub>, Ga<sub>2</sub>O<sub>3</sub>, In<sub>2</sub>O<sub>3</sub> and Ti<sub>2</sub>O<sub>3</sub>, *Appl. Catal., A*, 275 (2004) 87-93.
- [3] N. Patil, B. Uphade, D. McCulloh, S. Bhargava, V. Choudhary, Styrene epoxidation over gold supported on different transition metal oxides prepared by homogeneous deposition–precipitation, *Catal. Commun.*, 5 (2004) 681-685.
- [4] D. Yin, L. Qin, J. Liu, C. Li, Y. Jin, Gold nanoparticles deposited on mesoporous alumina for epoxidation of styrene: Effects of the surface basicity of the supports, *J. Mol. Catal. A: Chem.*, 240 (2005) 40-48.
- [5] F. Zhang, X. Zhao, C. Feng, B. Li, T. Chen, W. Lu, X. Lei, S. Xu, Crystal-face-selective supporting of gold nanoparticles on layered double hydroxide as efficient catalyst for epoxidation of styrene, *ACS Catal.*, 1 (2011) 232-237.
- [6] D. Nepak, D. Srinivas, Spectroscopy and catalytic activity study of gold supported on barium titanate nanotubes for styrene epoxidation, *Appl. Catal., A*, 523 (2016) 61-72.
- [7] J. Liu, F. Wang, S. Qi, Z. Gu, G. Wu, Highly selective epoxidation of styrene over gold–silica catalysts via one-pot synthesis: synthesis, characterization, and catalytic application, *New J. Chem.*, 37 (2013) 769-774.
- [8] J. Huang, C. Liu, D. Sun, Y. Hong, M. Du, T. Odoom-Wubah, W. Fang, Q. Li, Biosynthesized gold nanoparticles supported over TS-1 toward efficient catalyst for epoxidation of styrene, *Chem. Eng. J.*, 235 (2014) 215-223.
- [9] Y. Jin, D. Zhuang, N. Yu, H. Zhao, Y. Ding, L. Qin, J. Liu, D. Yin, H. Qiu, Z. Fu, Epoxidation of styrene over gold nanoparticles supported on organic–inorganic hybrid mesoporous silicas with aqueous hydrogen peroxide, *Microporous Mesoporous Mater.*, 126 (2009) 159-165.