

**One-step fabrication of laser induced forward transfer graphene/Cu_xO
nanocomposite based electrocatalyst to promote hydrogen
evolution reaction**

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Supporting Figures

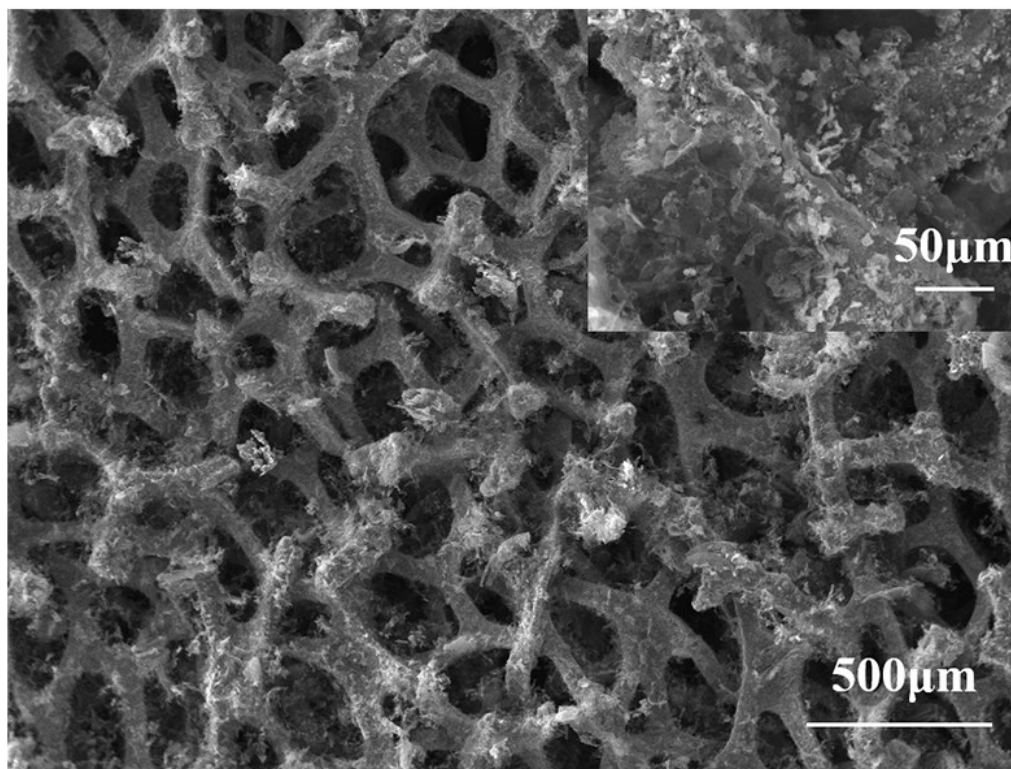


Figure S1. SEM images of the LIFT graphene/Cu_xO @Ni foam.

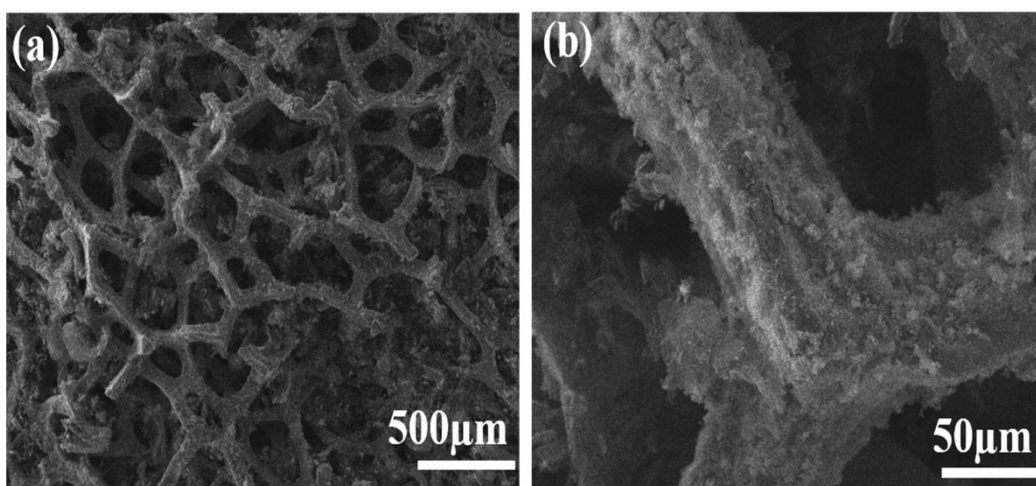


Figure S2. (a), (b) SEM images of the LIFT graphene/Cu_xO @Ni foam after 1000 cycles CV test.

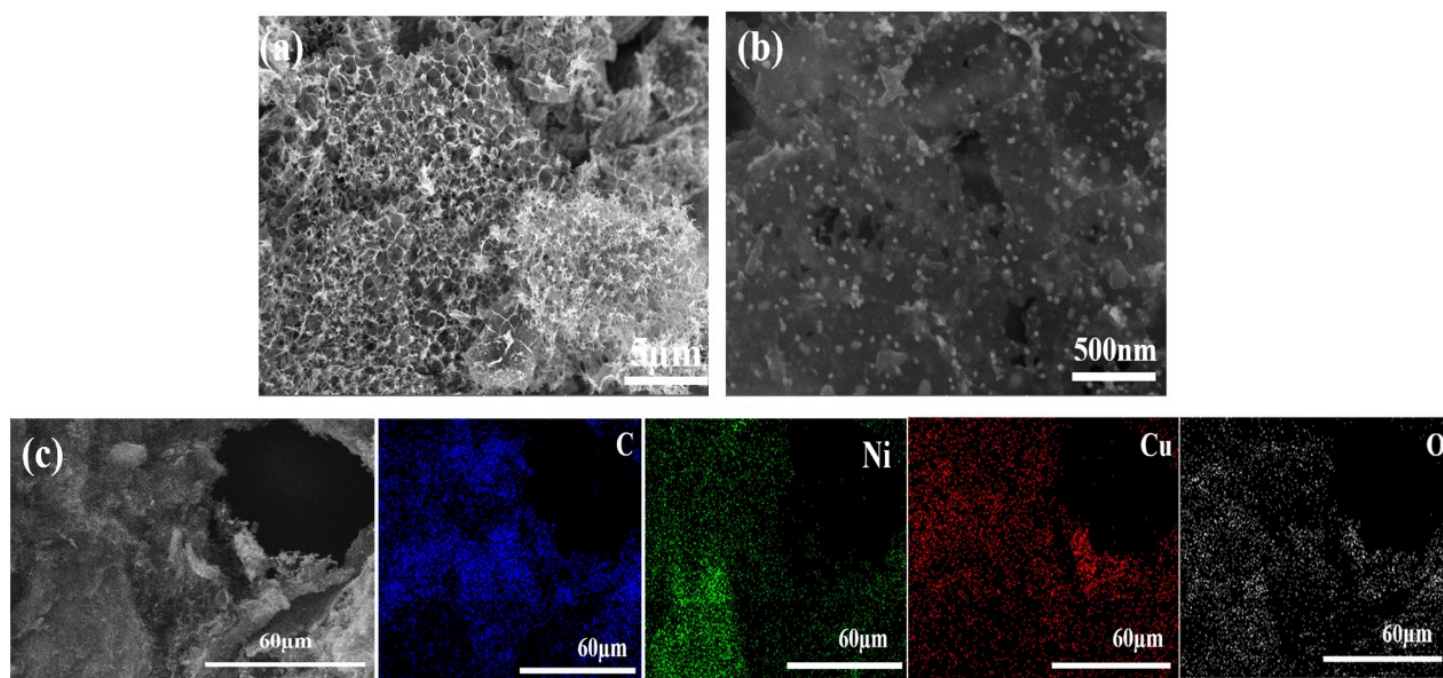


Figure S3. (a), (b) FESEM images of the porous LIFT graphene wrapped Cu_xO after 1000 cycles CV test.
 (c) EDX mapping of the LIFT graphene wrapped Cu_xO after 1000 cycles CV test.

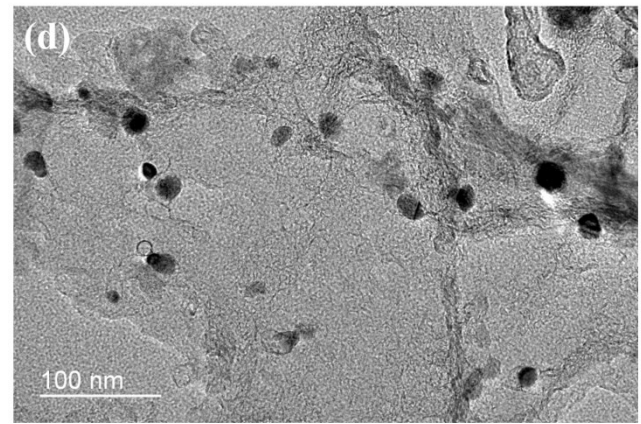
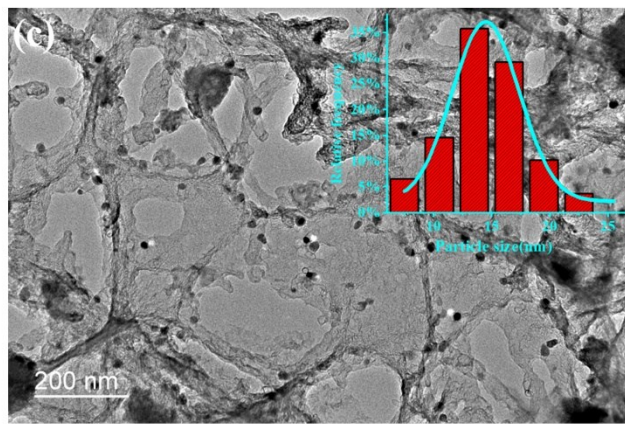
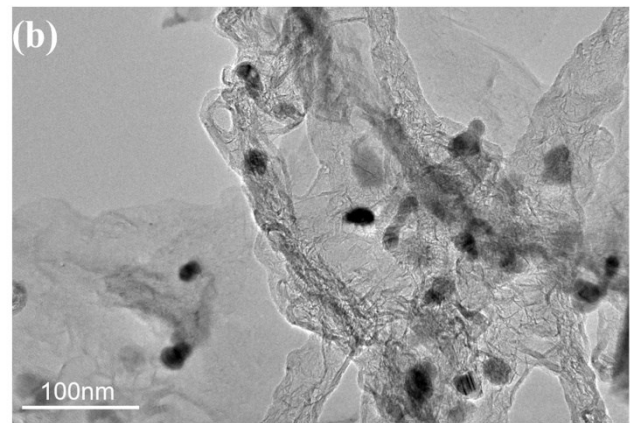
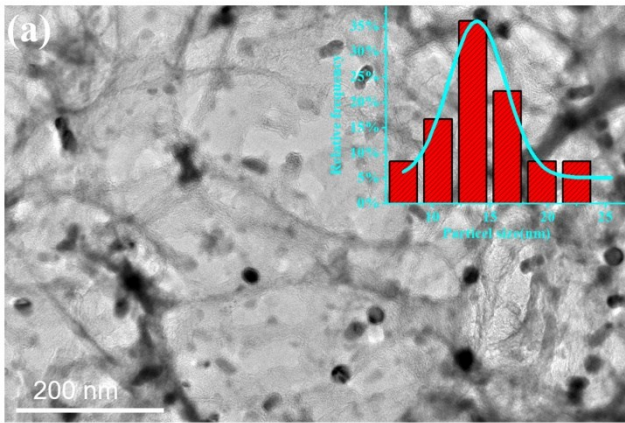


Figure S4. TEM images of (a), (b) initial LIFT graphene/ Cu_xO , (c), (d) LIFT graphene/ Cu_xO after 1000 cycles CV test.

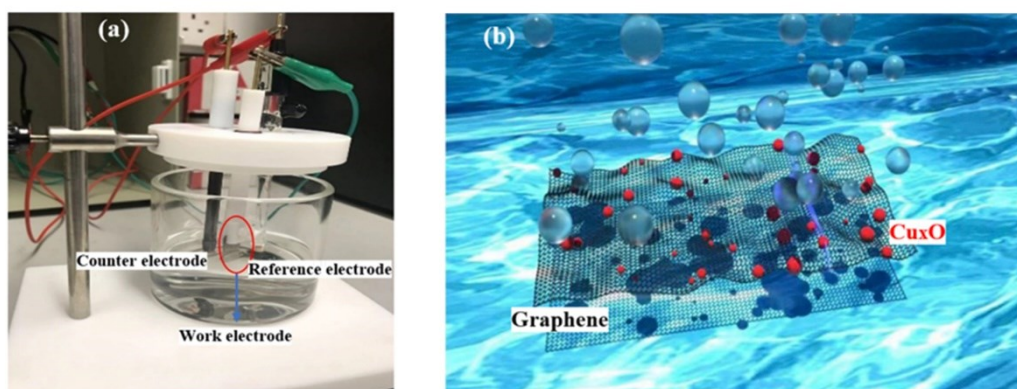


Figure S5. Pictures of (a) experiment set-up of the three-electrode: a graphite rod and a mercuric oxide electrode as the counter and reference electrode respectively, and (b) HER process in 1M KOH.

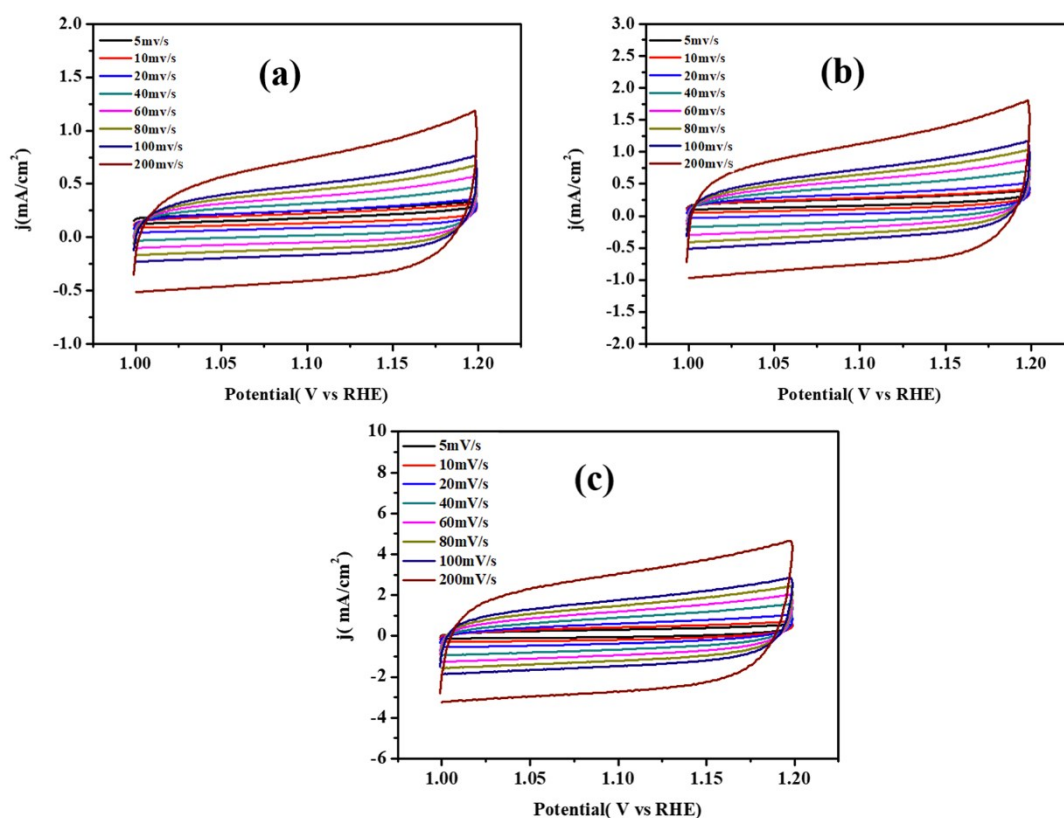


Figure S6. CV cures of bare Ni foam (a), LIFT graphene @Ni foam (b), and LIFT graphene/ Cu_xO @Ni foam(c) at different scan rates.

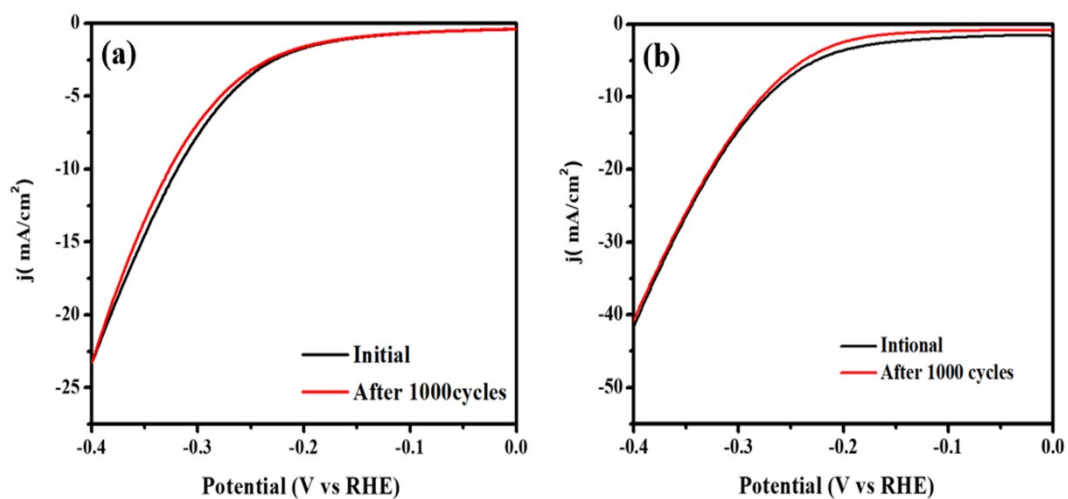


Figure S7. LSV curves for (a) initial bare Ni foam and after 1000 cycles CV tests, (b) initial LIFT graphene @Ni foam and after 1000 cycles CV tests

Table S1 Comparison of the HER performance of different Cu-based catalysts/electrodes obtained from different approaches.

| Catalyst/electrode | Electrolyte | $\eta(\text{mV})@j$ mA cm^{-2} | Tafel slope (mV dec^{-1}) | Preparation method | Reference |
|--|-------------------------------------|--|---|--|---|
| LIFT Graphene/Cu _x O/NiF | 1M KOH | 149.6@10 | 157 | Laser deposition | This work |
| Cu ₃ P/CF | 1M KOH | 210@20 | 127 | Anodization/calcination /Phosphorization | J. Phys. Chem. C 2017, 121, 25875-25881 |
| CuNPs/CNFs | 0.5M H ₂ SO ₄ | 200@10 | 152 | CVD/carbonization/ electrospinning / thermal reduction | International journal of hydrogen energy 41(2016) 18044-18049 |
| CoP@Cu ₂ P-Cu ₃ P/GC | 1M KOH | 118@10 | 72 | two-step nanocasting-solid phase phosphorization | Sustainable Energy Fuels DOI: 10.1039/d0se01745k |
| CuCo ₂ Se ₄ | 1M KOH | 125@10 | -- | hydrothermal method | ACS Appl. Energy Mater. 2020, 3, 3092- 3103 |
| CuO@Ni/NiFe hydroxides | 1M KOH | 125@10 | 86 | chemical oxidation-calcination /room-temperature interfacial Galvanic reaction | Electrochimica Acta 318 (2019) 695-702 |
| Cu ₃ P@NF | 1M KOH | 105@10 | -- | Hydrothermal method /phosphorization | ACS Appl. Mater. Interfaces 2017, 9, 2240- 2248 |
| Cu@NiFe LDH /CF | 1M KOH | 116@10 | 58.9 | chemical oxidation/air annealing /electroreduction/electrodeposition | Energy Environ. Sci., 2017, 10, 1820--1827 |
| Cu ₂ S@Cu | 1M KOH | 316@10 | 76 | sulfurization reaction | International journal of hydrogen energy 44(2019) 1620-1626 |
| CMS/Ni | 1M KOH | 213@50 | 80 | hydrothermal method | Mater. Adv., 2021, 2, 455-463 |
| Co-P/CuO CF | 1M KOH | 95@20 | -- | chemical oxidation/ Air annealing/ electrodeposition | Dalton Trans., 2019, 48, 891-897 |
| NiCoP@Cu ₃ P/CF | 1M KOH | 54@10 | 72 | Chemical oxidation /hydrothermal method | J. Mater. Chem. A, 2018, 6, 2100-2106 |
| Cu@CoFe LDH | 1M KOH | 171@10 | 36.4 | chemical oxidation | Nano Energy 41 (2017) |

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|---------------------------------------|--------------|-------|------|-----------------------------------|--|
| | | | | /electrodeposition | 327–336 |
| Copper-nickel-sulfide NWS | 1 M alkaline | 70@10 | 44 | multi-step wet chemistry strategy | Chem. Commun., 2019, 55, 8154--8157 |
| CuCo ₂ S ₄ TSAs | 1M KOH | 69@10 | 55.4 | hydrothermal method | J. Mater. Chem. A, 2020, 8, 1799–1807 |